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Formwork Pressure of Self-Consolidating Concrete: Influence of Flocculation Mechanisms,
Structural Rebuilding, Thixotropy and Rheology

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ABSTRACT

Formwork Pressure of Self-Consolidating Concrete: Influence of Flocculation Mechanisms, Structural Rebuilding, Thixotropy and Rheology

Raissa Patricia Douglas Ferron

While self-consolidating concrete (SCC) may no longer be considered a “new concrete”, there are still significant challenges to overcome before there is broader acceptance of SCC. One of these challenges concerns the formwork pressure exerted by SCC. A major advantage of SCC is the accelerated casting process due to the elimination of external vibration. However, faster casting rates may induce higher formwork pressure; this is a major concern for cast-in place applications, especially when casting tall elements.

It has been reported that the formwork pressure of SCC can be less than hydrostatic pressure. This is due to the build-up of a three-dimensional structure when the concrete is left at rest. The development of this structure and the mechanisms behind it are of particular interest to users of SCC.

The research presented in this manuscript was carried out at the Center for Advanced Cement-Based Materials at Northwestern University and the Universidad Complutense de Madrid. This dissertation focuses on the structural rebuilding SCC and its implications for formwork pressure. Special emphasis was given to the influence of flocculation mechanisms and the impact of material constituents. A rheological protocol to characterize structural rebuilding was developed. This protocol can be used to assess the contributions from irreversible structural

build-up from hydration and reversible structural rebuilding from thixotropic effects. The impact of various mixture ingredients, including cement type, mineral admixtures, chemical admixtures and clays, on the structural rebuilding was examined. The results showed that the rheological properties of the paste matrix and its evolution over time can be used as an indication of the formwork pressure behavior. Formwork pressure is highly impacted by the structural rebuilding that occurs in the paste matrix, and the results showed that formwork pressure is related to the rate at which structural rebuilding occurs and the total amount of structural build-up. In addition, a novel experimental technique was used to examine the flocculation and breakage mechanisms of cement paste suspensions subjected to shear using. It is believed to be the first time that direct investigation of the microstructural response of concentrated cement paste suspensions subjected to shear-induced stresses has occurred.

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“No woman is an island”

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think I can finish this?" Because you know what, I did...we did it. Thank you. I love you.

Saltfish and Bakes. Let's get it started.

List of Key Abbreviations and Symbols

[X]	Concentration of element X
ACI	American Concrete Institute
ASTM	American Society of Testing Materials
C ₂ S	Dicalcium silicate, 2CaO*SiO ₂
C ₃ A	Tricalcium aluminate, 3CaO*Al ₂ O ₃
C ₃ S	Tri-calcium silicate, 3CaO*SiO ₂
C ₄ AF	Tetracalcium aluminoferrite, 4CaO*Al ₂ O ₃ *Fe ₂ O ₃
DI	Deionized water
FA	Fly ash
FBRM	Focused Beam Reflectance Measurement
G'	Storage (elastic) modulus
G''	Loss (viscous) shear modulus
G*	Complex shear modulus
I _c	Ionic strength
ICP	Inductively coupled plasma spectroscopy
K	Lateral pressure coefficient in Chapter 7
k ₁	Aggregation kinetic constant in Chapter 8
k ₂	Disaggregation kinetic constant in Chapter 8
LVER	Linear viscoelastic region

MTS	Mechanical Testing System
PC	Polycarboxylate
SCC	Self-consolidating concrete
SF	Silica fume
sfd	Slump flow diameter
SRE	Specific rebuilding energy
T_{50}	Time required to reach a slump flow diameter of 50 centimeters
T_{final}	Time required to reach final slump diameter
t_{rest}	Resting time at which sample is undisturbed
VC	Vibrated concrete
VMA	Viscosity modifying admixture
w/c	Water-to-cement ratio
w/p	Water-to-powder ratio
$\dot{\gamma}$	Shear strain rate
ΔSRE	Rate of structural rebuilding (calculated from the difference in SRE values)
η	Viscosity
μ	Bingham viscosity
ξ	Zeta potential
τ	Shear stress
τ_{eq}	Equilibrium shear stress
τ_y	Yield stress

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PART I: Overview

1 CHAPTER 1: Introduction

1.1 Problem Overview

Concrete, a composite material composed of cement, water, sand, and gravel, is the world's most widely used construction material. In its fresh state, concrete is a dense suspension consisting of aggregate particles suspended in a mortar matrix. Traditionally, concrete must be vibrated to remove entrapped air voids and to consolidate it; however, this can lead to durability problems due to improper consolidation (either from over-vibration or under-vibration). In the late 1980's, a new generation of concrete, self-consolidating concrete (SCC), was developed to combat the problems due to poorly vibrated concrete. SCC is a highly flowable concrete that does not require any external vibration during placing. One of the major advantages of SCC is that it can accelerate the construction process since casting can occur more quickly as there is no need to stop the pouring to vibrate the concrete. However, increasing the casting rate leads to higher lateral pressure against the formwork [1], and if not properly accounted for, the high pressure can lead to formwork failure. In North America, the pre-cast industry has spearheaded the usage of SCC. However, its usage in ready-mix (i.e. cast-in place) industry has progressed at a slower rate due to concerns about quality control and uncertainty about formwork pressure development.

Presently, formworks for SCC are designed for at least full hydrostatic pressure [2]. However, recent studies have shown that not all types of SCC achieve hydrostatic pressures [3-6] and that formwork pressure behavior can be influenced by factors such as casting rate, aggregate

volume, and mineral and chemical admixtures. Thus, the conservative assumption of hydrostatic pressure can lead to unnecessary costs due to the construction of over-dimensioned formworks. Considering the fact that in the United States the cost of formwork can be as much as 60% (and sometimes even more) of the total cost of the completed concrete structure in place [7], it is evident that there is a dire need to understand the factors, particularly as it pertains to the mix composition, that can influence the formwork pressure of SCC.

In addition to concerns related to formwork pressure, there is a need to increase the fundamental knowledge about the rheological behavior of cement-based materials. Cementitious fluids have complex rheology and typically exhibit shear-thinning viscosity, yield stress, elasticity, and thixotropy. Thixotropy is defined as “a decrease of the apparent viscosity [in time] under constant shear stress or shear rate, followed by a gradual recovery when the stress or shear rate is removed” [8]. In cementitious materials, the phenomenon of thixotropy results from the heterogeneity of the material and the finite time that it takes for the microstructure to rearrange when subjected to shear-induced stresses. The specific causation of thixotropy depends on interactions at the molecular level, which, unfortunately, are poorly understood. The rheological behavior of cement-based materials is directly linked to the aggregation, de-aggregation, re-aggregation, and dispersion of the solid particles. Research that can provide information about any of these phenomena would advance the state of knowledge on the flow behavior of cement-based materials.

1.2 Research Objectives

The over-arching goal of this research was to advance the understanding of the role of mixture ingredients on the structural rebuilding and the formwork pressure behavior of SCC. To accomplish this goal the following objectives were established:

1. Evaluate evolution of structural rebuilding during the matrix phase.
2. Identify the role of material constituents that have a major influence on structural rebuilding.
3. Relate structural rebuilding to formwork pressure.
4. Gain fundamental understanding of the aggregation behavior.

To accomplish the first objective, a rheological protocol based on hysteretic loops was developed. To achieve the second goal, a comprehensive research program was undertaken to evaluate the influence of various mixture ingredients on the paste matrix phase of SCC. The third objective was achieved by developing a formwork pressure simulation apparatus. Using the rheological protocol developed in the first objective and the information gained from the second objective, the influence of mixture proportioning was investigated. The last objective was accomplished by conducting studies that focused on evaluating the role of pore solution chemistry and *in situ* particle size measurements within the cement paste matrix.

1.3 Structure of the Dissertation

This chapter provided an overview of the research significance and objectives of the proposed research. In addition, an introduction to concrete and SCC were given. Chapter 2 reviews the

major factors governing formwork pressure of SCC, and Chapter 3 provides background information regarding rheology and flocculation. In Chapter 3, the mechanisms governing yield stress and thixotropy are highlighted, and the role of interparticle forces and their influence on rheology are discussed. Chapters 4 through 8 should be considered the major body of this dissertation. Chapter 4 presents a rheological protocol that was developed to evaluate structural rebuilding and Chapter 5 provides the results from preliminary studies concerning the influence of cement composition. Chapter 6 offers a systematic investigation of the role of mixture ingredients, and discusses the impact of mixture proportioning on structural rebuilding. Chapter 7 focuses on the implications of structural rebuilding on formwork pressure, and Chapter 8 explores the impact of aggregation and breakage mechanisms on structural rebuilding and formwork pressure. Chapter 9 provides key conclusions from this study and thoughts on future work. A list of all the works cited in this manuscript can be found in Chapter 10.

PART II: Background/Literature Review

2 CHAPTER 2: Self-Consolidating Concrete and Lateral Pressure

2.1 Introduction

This chapter provides an overview of SCC and the issues concerning the formwork pressure generated by SCC.

2.2 Self-Consolidating Concrete

As its name implies, SCC is a special class of concrete that consolidates only from self-weight and needs no extra external compaction methods. It is generally agreed that Professor Hajime Okamura first proposed the idea of SCC in 1986, and that the first prototype was developed in 1988. However, precursors to SCC have been identified before 1988. Mario Collepardi states that as early as 1975 self-compacting concretes were studied and that as early as 1981 commercial applications of non-vibrated self-leveling concretes were used in the United States, Hong Kong, and Italy [9].

SCC is characterized by its filling ability (flowability), passing ability, and stability (resistance to segregation and bleeding). Filling ability, or flowability, is the ability of the concrete to completely flow (horizontally and vertically upwards if necessary) and fill all spaces in the formwork without the addition of any external compaction. Passing ability is the ability of the concrete to flow through restricted spaces without blocking. Stability, or resistance to segregation, is the ability of the concrete to remain uniform and cohesive throughout the entire construction process (mixing, transporting, handling, placing, casting, etc.). There should be

minimum segregation of the aggregates (both fine and coarse) from the matrix and little bleeding. Hence, SCC is a highly flowable, non-segregating concrete that is able to spread and flow through restricted orifices, fill formwork, and completely cover any reinforcement without any mechanical vibration. In the field, the quality of SCC is assessed based on its slump flow diameter (sfd), which is measured using the slump flow test (ASTM C1611) [10], a modified version of the popular slump test (ASTM C143) [11] for vibrated concrete (VC). Depending on the application, the sfd of SCC can range from 18.5 to 31.5 inches (470 to 800 mm) [12]. At low sfd ranges, extra precaution should be taken to ensure that SCC is truly consolidated, whereas at high sfd ranges care should be taken to guarantee that the concrete is stable.

2.2.1 Advantages and Disadvantages

Over the last two decades, SCC has revolutionized the practices of concrete construction in the United States and throughout most of the world. Typically, the term high performance concrete is bestowed upon concretes that exhibit extraordinary properties in the hardened state (ex. ultra-high compressive strength, crack-resistant concrete, long-term durability, etc.); however, SCC can also be considered a high performance concrete due to its high performance in the fresh state. In fact, SCC was developed in order to alleviate durability problems due to poorly vibrated concrete [13]. The degree of compaction significantly affects the quality of the concrete. Insufficient compaction can result in entrapped air voids or poor bonding between the reinforcement and concrete, whereas excessive compaction can lead to segregation and cracking. As a result, the skill level and assiduousness of the construction workers became crucially important, and it was thought that the development of SCC could eliminate problems caused by

negligence and lack of skill during the compaction stage. In addition to the above-mentioned

advantage, other benefits to using SCC are:

- increased construction productivity
- improved jobsite safety
- improved hardened properties
- reduction in noise due to elimination of vibration
- reduction in labor and equipment costs
- better surface finish and little to no remedial surface retouching for bugholes
- ability to flow into tight spaces and openings (This allows for the opportunity to cast intricate structural and architectural shapes and is especially beneficially in high seismic zones where dense reinforcement is used.)
- higher bond strength and less significant top bar effect
- faster casting rates and quicker unloading of ready mixed concrete trucks

However, the material costs for SCC are generally higher than for conventionally placed concrete due to the usage of chemical admixtures, which are necessary in order to achieve the fresh state properties. Other challenges for users of SCC are:

- greater technical expertise and quality control measures needed to develop and control mixtures
- lack of standardized mix design and test procedures
- increased sensitivity of fresh state properties due to small changes during processing (i.e. guaranteeing robustness)
- increased shrinkage due to high paste volume

- increased formwork costs due to concerns about formwork pressure

2.2.2 Mix Proportioning

SCC was made possible through the development of advanced chemical admixtures coupled with proper understanding of concrete materials. SCC constituent materials and mixture proportions must be properly selected to achieve the required flow properties. Mixtures are typically designed with high powder contents and contain chemical admixtures, such as superplasticizers (also known as high range water reducers) and viscosity modifying admixtures (VMA).

2.2.2.1 Superplasticizers and VMAs

Plasticizers are chemical admixtures that are added to concrete to improve workability or reduce the water content. For a given slump/slump flow, incorporation of plasticizers can yield a 5 - 10% reduction in the water content [14]. This enables one to achieve higher-strength concrete (due to a lower water-to-cement (w/c) ratio) without sacrificing workability. As their name implies, superplasticizers are more powerful than plasticizers and yield water reductions of 12% or more [15], and as much as 20 – 30% water reduction in some cases [15]. Superplasticizers are commonly derived from sulphonated melamine, sulphonated naphthalene, modified lignosulphonates, or carboxylated polymers [15-17]. They increase workability by dispersing the cement particles; however, the dispersion efficiency strongly depends on cement-superplasticizer interactions. Hence, for a given cement drastically different dispersion effects (and hence workability) can occur when different superplasticizers are employed. Other factors such as the dosage and time at which the superplasticizer is added to the mixture will also influence workability. Typically, polycarboxylate (PC) superplasticizers (as the carboxylated

polymer superplasticizers are commonly called) are used in SCC mixtures. PC superplasticizers are synthetic organic polymers and were introduced in the early 1980s [15]. They are generally more effective dispersants than superplasticizers derived from sulphonated melamine, sulphonated naphthalene, or modified lignosulphonates.

VMAs are water-soluble polysaccharides that are added to increase cohesiveness of concrete [18, 19]. The first application of VMAs was in Germany during the mid-1970s, and it was not until the late 1980s that VMAs were used in North America [18]. Initially, VMAs were used in specialty grouts and underwater concrete applications to enhance water-retention and reduce the washout; however, they are frequently used in SCC to improve homogeneity, deformability, robustness, and stability. According to Bury and Christensen [20], VMAs work by two different mechanisms: binding or thickening. The binding type VMA works by adhering to water molecules within the mixture, whereas the thickening type VMA functions by thickening the concrete to make it more cohesive without significantly influencing the fluidity. In SCC, the most commonly used VMAs are semi-synthetic polymers derived from cellulose ethers (such as hydroxypropyl methyl cellulose) and biopolymers polymers (such as welan gum). In the case of welan gum and cellulose derivatives, it is believed that these VMAs function by adsorbing onto water molecules and expanding [18]. Furthermore, the polymer chains intertwine and develop attractive forces in adjacent polymer chains, which further restrict the movement of water and cause a gel formation. A good review on VMAs can be found in a paper published by Kamal Khayat [21].

2.2.2.2 Mix Design

When designing SCC one must consider both the matrix phase and the aggregate phase. With regard to the matrix phase, there are two commonly used approaches for developing SCC: the powder approach and the VMA approach (c.f. Figure 2-1). In the powder SCC approach, the amount of fine materials (ex. cement, fly ash, limestone filler, blast furnace slag, silica fume, and/or fine aggregate particles smaller than 0.125mm) is significantly increased when compared to conventionally vibrated concrete (VC)*. Superplasticizers are used to increase the flowability and the water-to-powder (w/p) ratio (by volume) ranges from 0.9 to 1.0, compared to VC whose w/p ratio (by volume) range from 1.5 to 2.0 [22]. This method, also referred to as the “Japanese approach”, was the original approach used to develop SCC and does not employ the use of VMA. In the method suggested by Okamura [13], the coarse aggregate content is fixed at 50% of the total solid volume of concrete, and the fine aggregate is fixed at 40% of the mortar volume. Self-compactability is obtained by adjusting the superplasticizer dosage and w/p ratio (by mass). In the VMA-SCC approach, VMA is incorporated to enhance stability while the powder content remains less than that used in the powder-type SCC approach but still higher than that of a VC. A third approach combines the powder approach and the VMA approach to form an SCC with a moderate powder content and VMA.

* In this manuscript, conventional concrete that is subjected to external vibration will be referred to as vibrated concrete (VC).

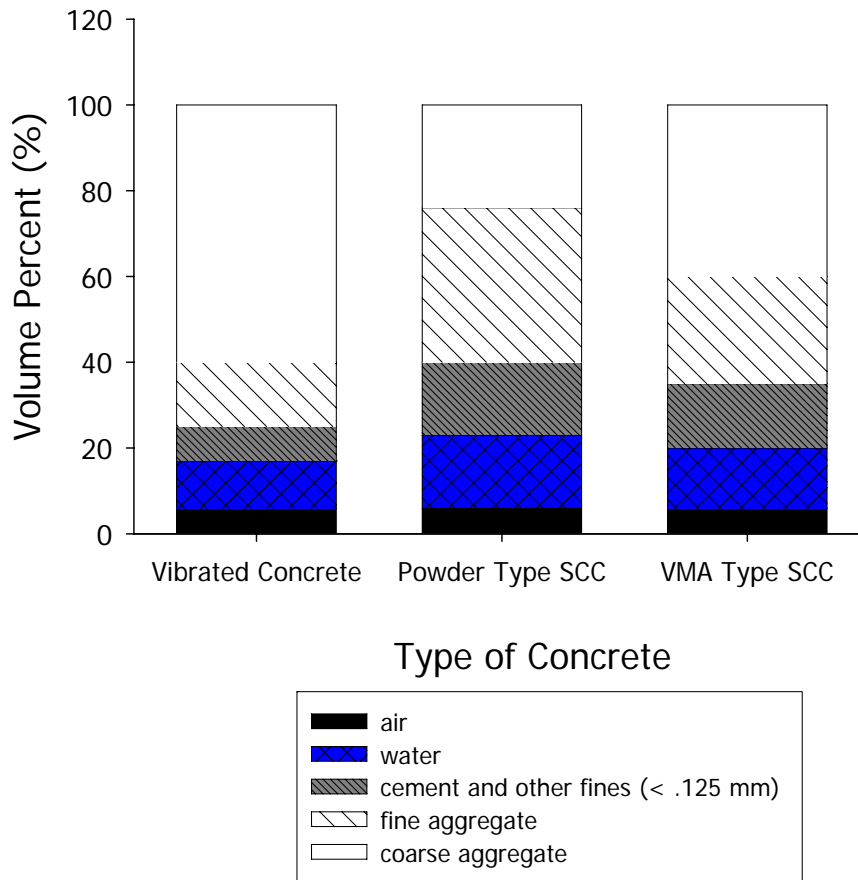


Figure 2-1. Comparison of vibrated concrete and SCC proportioned with 6% nominal air (adapted from [23])

Regardless of the method used to design the matrix phase, there should be enough paste to ensure good deformability of the concrete. However, volume instability increases with an increase in paste content. Thus, it is desirable to use larger size coarse aggregates since the reduction in surface area of the aggregates will lower the paste demand requirement. Because passing ability is also an important consideration, the grading of the aggregates should be

optimized to minimize blockage and paste requirements. Segregation resistance and passing ability is also improved by reducing the overall content of coarse aggregate [9].

2.3 Formwork Pressure of Self-Consolidating Concrete

The lateral pressure exerted by freshly placed SCC against vertical form surfaces has been the subject of many research papers during the last couple of years [2-5, 23, 24]. Underestimating the pressure may cause deformed structural elements or even failure of the formwork, while overestimating the pressure leads to unnecessary costs due to over-dimensioned formworks. With respect to conventional VC, numerous studies were necessary to obtain sufficient understanding of the main variables controlling lateral pressure. In VC, the major factors affecting pressure development are the depth of fluidized concrete and the development of shear strength and wall shear by the concrete [1]. However, these factors are significantly influenced by the height of cast, casting rate, casting method, consolidation method, ambient temperature, temperature of the concrete, maximum aggregate size, consistency of fresh concrete, setting time, and the shape of formwork. As shown in Table 2-1 these variables can be grouped into three categories: concrete material properties, formwork characteristics, and processing influence.

Table 2-1. Variables influencing formwork pressure

Concrete Properties	Formwork Characteristics	Processing
Consistency	Dimensions	consolidation method and depth
mix composition	material	rate of cast
set time	permeability	method of placement
Temperature	surface roughness	ambient temperature (weather)
Density	shoring and bracing	depth of placement (lift)

VC is poured into forms in lifts, and the concrete is consolidated through mechanical vibration. Typically, poker-type vibrators, which are immersed into the top 3 feet of concrete, are used for consolidation [1]. Figure 2-2 to Figure 2-5 show the lateral pressure development for concrete placed into a wall or column. When vibrated, the fresh concrete fluidizes, and the fluidized layer develops hydrostatic pressure (c.f. Figure 2-2 and Figure 2-3). However, once the head of the concrete reaches a certain height (c.f. Figure 2-4) the effect of the vibrator is reduced and the lowest layer of the concrete can not be fluidized by the poker vibrator. This layer will begin to stiffen, the shear resistance will increase, and the pressure at the bottom will be less than hydrostatic.

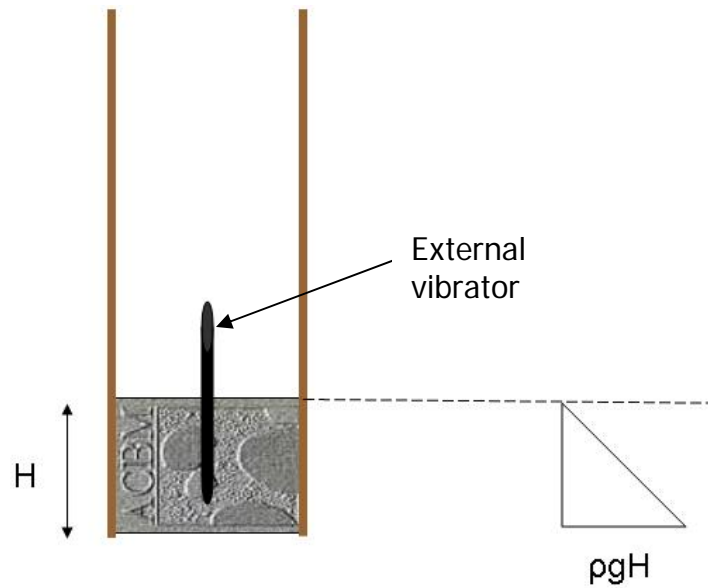


Figure 2-2. Lateral pressure envelope development when concrete is initially poured and subjected to vibration. Figure on right shows depth of concrete pour. Figure on left shows lateral pressure exerted by concrete. Lateral pressure is equal to hydrostatic pressure. Hydrostatic pressure equals $\rho g H$, where ρ is density of concrete.

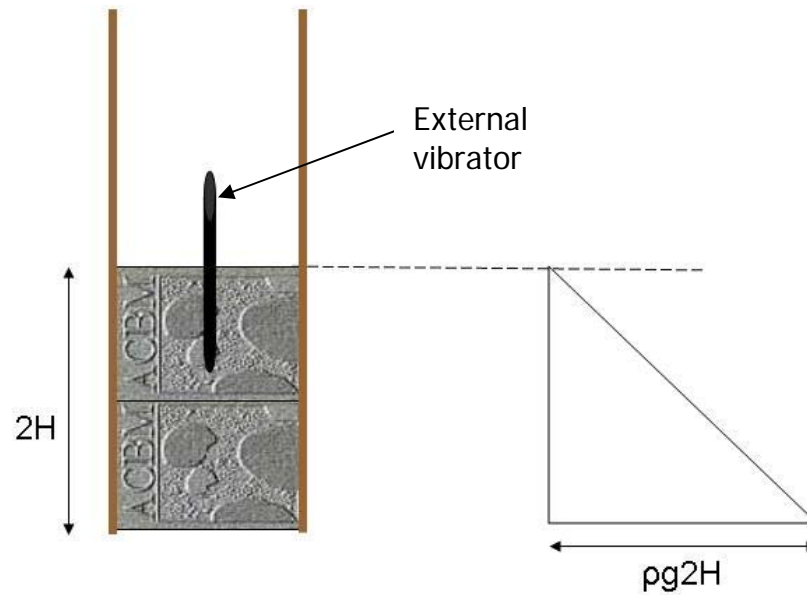


Figure 2-3. Lateral pressure envelope development after second layer of concrete is cast and is subjected to vibration. Figure on left shows depth of concrete pour. Figure on right shows lateral pressure exerted by concrete. Lateral pressure is equal to hydrostatic pressure. Hydrostatic pressure equals $pg2H$, where ρ is density of concrete.

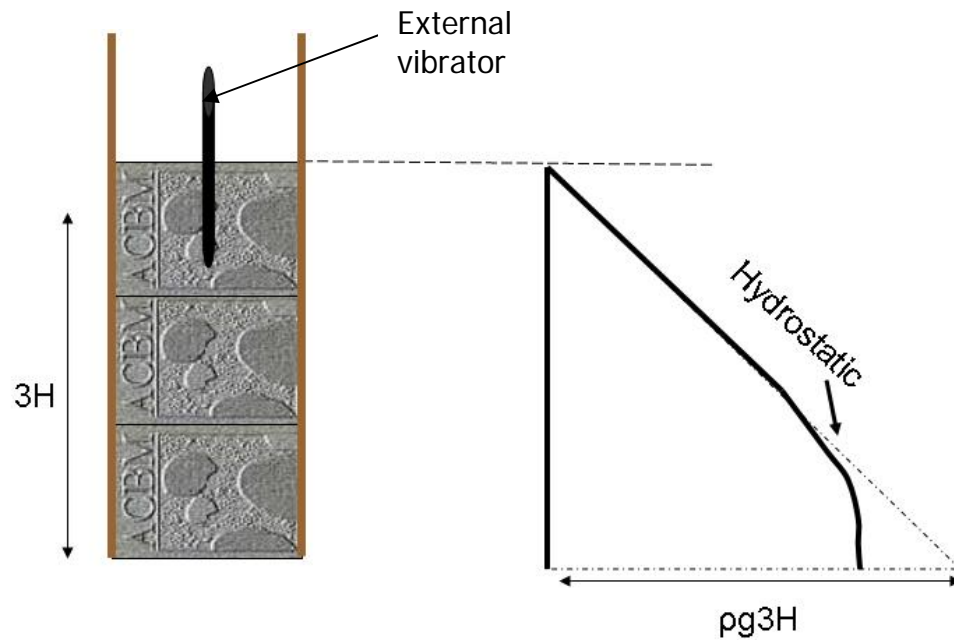


Figure 2-4. Lateral pressure envelope development after third layer of concrete is cast and is subjected to vibration. Figure on left shows depth of concrete pour. Thick solid line in figure on right shows lateral pressure exerted by concrete, and dashed line shows the hydrostatic pressure. Effects from vibrator are not felt throughout the entire depth of concrete and the lateral pressure is less than hydrostatic pressure. Hydrostatic pressure equals $\rho g 3H$, where ρ is density of concrete.

If the concrete head is increased further (c.f. Figure 2-5), the shear strength of the lower layer will become even more significant, and the lateral pressure will reach a maximum at some elevation above the base of the formwork. Figure 2-5 shows that the lateral pressure along the depth of the form will be hydrostatic from the free surface, reach a maximum, and then decrease. Hence, VC formworks are designed taking into account the stiffening of the lower layer of

concrete, and the forms are designed for a maximum pressure that is lower than the hydrostatic pressure.

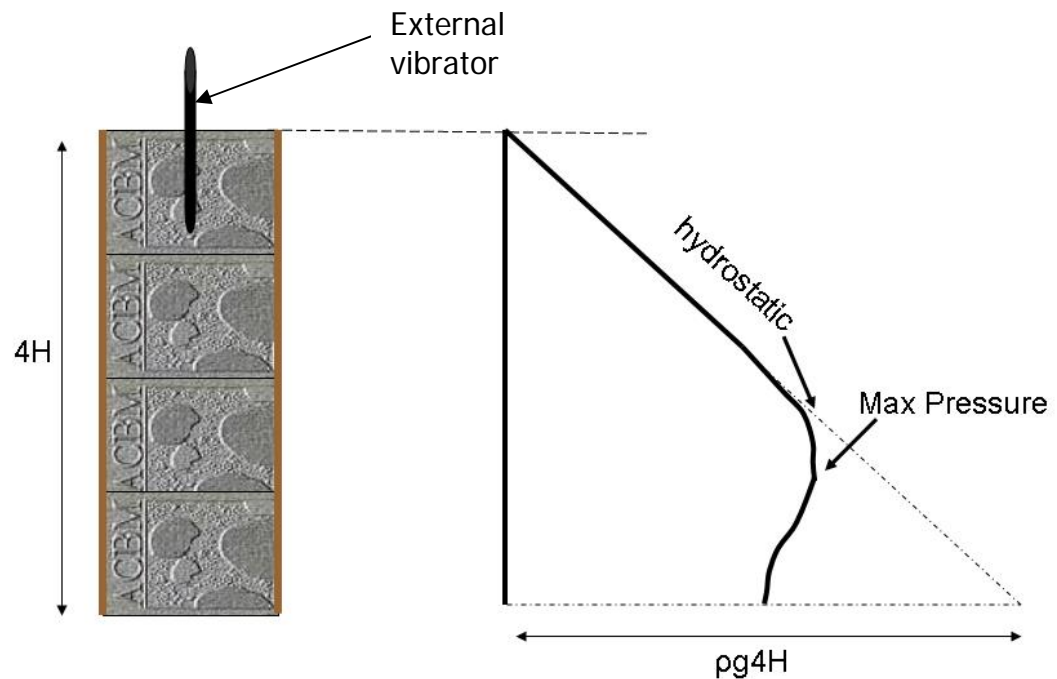


Figure 2-5. Lateral pressure envelope development after final layer of concrete is cast (i.e. end of pour) and concrete is subjected to vibration. Figure on left shows depth of concrete pour. Thick solid line in figure on right shows lateral pressure exerted by concrete, and dashed line shows the hydrostatic pressure. Effects from vibrator are not felt throughout the entire depth of concrete and the lateral pressure is significantly less than hydrostatic pressure due to shear strength resistance in lower layers. Hydrostatic pressure equals $\rho g 4H$, where ρ is density of concrete.

SCC quickly flows into the formwork and full consolidation can occur without the need for vibration. SCC is particularly convenient in cases of heavily reinforced structures and for narrow cross sections since it can flow readily through restricted spaces. Due to the high fluidity of SCC, it is cast at faster rates than VC. This induces higher formwork pressure and raises questions about the adequacy of using current formwork design practices for SCC. The American Concrete Institute Committee 347, Formwork for Concrete, recommends that formworks be designed with the assumption that full liquid pressure head will be achieved unless trial study measurements can prove otherwise [2]. Similarly, the European Federation of Producers and Contractors of Specialist Products for Structures (EFNARC) recommends that forms higher than 3 meters be designed for full hydrostatic head [25]. Currently, forms for SCC are constructed with more structural ties or heavier steel-framed gang forms than those constructed for vibrated concrete [2, 6]. This is of particular concern because formwork construction (i.e. materials and labor) can account for a significant amount of the construction costs and oftentimes is greater than the cost of the concrete itself [26].

Studies have shown that not all SCCs achieve hydrostatic pressures (c.f. Table 2-2) [3-6]. Research on instrumented formwork up to 12 meter high has shown that the pressure exerted by the SCC can consistently deviate from the linear hydrostatic distribution (up to 30-35% at the casting bottom) [27]. In that case, the difference between the expected (hydrostatic) theoretical values and the measured pressure values was attributed to the friction between the concrete and the formwork's internal surface [27]. An extensive laboratory study conducted by Assaad, Khayat and Mesbah showed that mix composition was a key factor that influenced formwork pressure [28]. This was attributed to the belief that mix composition affects the kinetics of a structural build-up phase and thixotropy. In 2001, Peter Billberg conducted full scale tests using

wall forms to evaluate the effect of different SCC compositions on formwork pressure [6].

The results indicated that casting rate strongly influenced pressure development, and that mix composition was not a significant factor affecting form pressures. Another study [3] showed that the pressure generated by SCC was very close to the hydrostatic profile. This was attributed to the aggravating effect of a faster formwork filling rate (3 m/h). The reason for the diversity of results reported by the literature is still not clear given that pressure values equal to the 60% of hydrostatic values have been reported also in cases of SCC castings with filling rates higher than 5 m/h [3, 6].

Table 2-2. SCC formwork pressure as reported by different authors (adapted from [29])

Author	Max. pressure (% hydrostatic)	Casting Rate (m/hr)	Height (m)	Time to fill (hr)
Leemann-Hoffman	87 – 90/102*	7.3/18.3	2.7/4.7	0.37/0.26
Vanhove	75/80*	25/19.5	11.9	0.48/0.61
Assaad-Khayat	87 – 100	10	1.0	0.09
Billing	18/38/65	1/1.3/2	3.0	3/2.3/1.4
Fedroff-Frosch	35	1	3.7	4
Tejeda-Lange	24/32/80	1.2/2.7/18.3	8.5/6.6/3.2	7/2.4/0.18

* Concrete was pumped from bottom

The study conducted by Assaad, Khayat and Mesbah showed that in addition to casting rate, mix composition was a key factor that influenced formwork pressure [28]. This was attributed to the belief that mix composition affects the kinetics of structural build-up and thixotropy.

3 CHAPTER 3: Rheology and Flocculation

3.1 Introduction

Cement paste is a dense flocculated suspension. As such, its flow behavior will be influenced by the degree of flocculation within the paste. This depends on the attractive and repulsive forces between the molecules. This chapter gives an overview of rheology and the influence of flocculation.

3.2 Rheology

Rheology is the study of the deformation and flow of matter under stress [30]. Its origin as an independent branch of natural sciences began in 1929, and, typically, rheologists are primarily concerned with trying to explain the behavior of industrially relevant materials with properties intermediate between those of ideal liquids (ex. Newtonian fluid) and ideal solids (Hookean solid) [31]. The deformation (strain) of a Hookean solid only depends on the applied load; once the load is removed, the deformation is immediately recovered (i.e. the solid returns back to its original shape). For these materials, the shear deformation relates linearly to the force causing the deformation:

$$\tau = G\gamma \qquad \text{Equation 3-1}$$

where τ is the shear stress, G is the constant of proportionality (shear modulus), and γ is the shear deformation (shear strain).

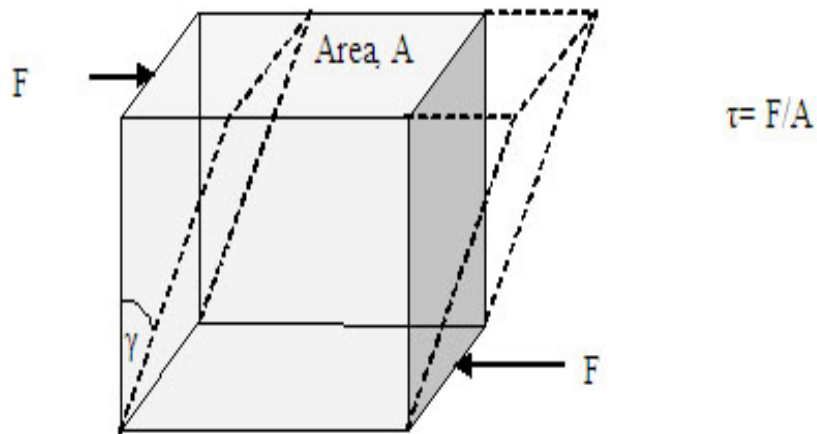


Figure 3-1. Shear deformation for a Hookean solid

For a Newtonian liquid, the application of a load does not produce a definite shear strain; rather, the shear stress is linearly proportional to the velocity gradient (in the direction perpendicular to the plane of shear):

$$\tau = \eta \dot{\gamma} \quad \text{Equation 3-2}$$

where η is the viscosity, and $\dot{\gamma}$ is the velocity gradient (also known as the shear rate). The viscosity represents the ability of the fluid to resist shear-induced flow, and it is an indication of fluidity. Immediately, it follows that the Newtonian viscosity is the constant of proportionality that represents the characteristic behavior of the liquid, and that the viscosity is constant for any $\dot{\gamma}$.

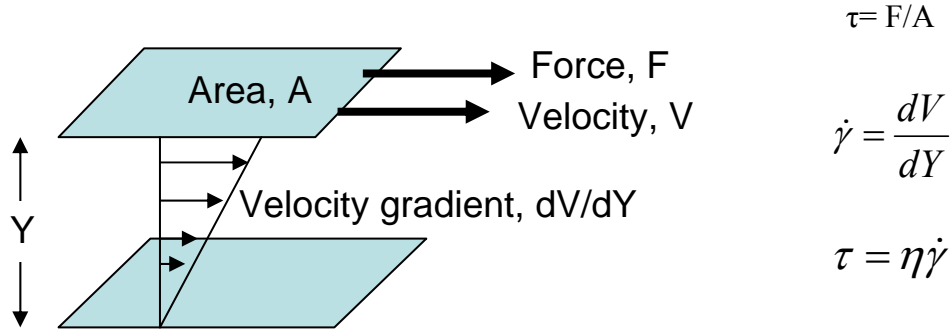


Figure 3-2. Viscous flow in a Newtonian fluid (bottom plate is fixed)

Consider the case of a Newtonian fluid confined between two plates where the bottom plate is fixed. The application of an applied load to the top plate causes the plate to move with a velocity, V . A layer of fluid next to the top plate moves with the plate, and a layer of fluid, which is essentially stationary, exists next to the stationary plate. Thus, a velocity gradient develops in the fluid from the top plate to the stationary plate.

Viscoelastic materials simultaneously exhibit the properties of a viscous fluid and an elastic solid. Thus, the shear stress will be proportional to the shear rate and the shear strain. For a solid, the deformation and recovery may be virtually instantaneous, whereas for fluid virtually none of the deformation is recovered. In a viscoelastic material, the response of the material may be slow and the recovery is not immediate.

3.2.1 Rheological Categories

The field of rheology characterizes and quantifies the different viscous behaviors existing in fluids and suspensions. There are two general categories of rheological properties: time-independent and time dependent.

Within the time-independent category, there are six rheologies. These rheologies are considered time-independent because the duration of shear has no effect on the rheological properties. Newtonian, dilatant, and pseudoplastic behaviors are time-independent rheologies that do not exhibit a yield stress (τ_y), whereas the Bingham, yield-dilatant and, yield-pseudoplastic rheologies have a yield stress. The yield stress is the shear strength of the concrete, and, ideally, it is the strength that must be exceeded in order for flow to begin. In a plot of shear stress versus shear rate, the yield stress is the point at which the slope of the line crosses the ordinate axis.

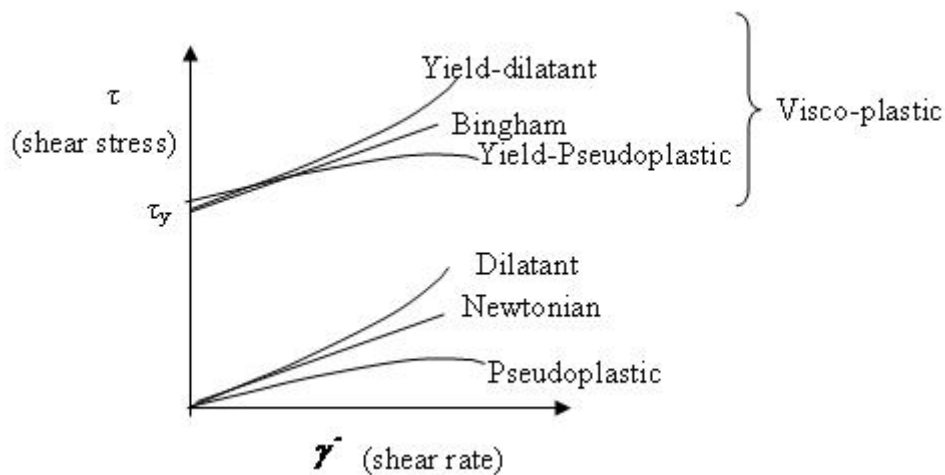


Figure 3-3. Time independent rheologies

For many liquids, the ratio between shear stress and shear rate is not constant; rather, the viscosity depends on the shear rate and, therefore, is called apparent viscosity. Liquids that exhibit an increase in apparent viscosity with increase in shear rate are characterized as dilatant (shear-thickening), whereas liquids that exhibit a reduction in apparent viscosity with increase in shear rate are characterized as pseudo-plastic (shear-thinning). The same type of behavior occurs

in time-independent liquids with a yield stress. For these liquids, the flow behavior cannot be described by a single parameter. The simplest model for this type of fluid is expressed by the Bingham model [32]:

$$\tau = \tau_y + \mu\dot{\gamma} \quad \text{for } \tau \geq \tau_y \quad \text{Equation 3-3}$$

and

$$\dot{\gamma} = 0 \quad \text{for } \tau \leq \tau_y \quad \text{Equation 3-4}$$

where μ is the plastic or “Bingham viscosity”. In an ideal Bingham fluid, once the yield stress is exceeded the material flows, and the viscosity is the material’s resistance to flow. For a Bingham fluid, the relationship between (Newtonian) viscosity and plastic viscosity is:

$$\eta = \mu + \frac{\tau_y}{\dot{\gamma}} \quad \text{Equation 3-5}$$

Hence, the Newtonian viscosity is the viscosity limit at high shear rates. It must be stressed that the Bingham model is not the only model that can be used to describe visco-plastic materials. Models such as the Casson and Herschel-Bulkley models are slightly more complicated but are also used to represent yield-pseudoplastic fluids. Examples of some fluids that appear to have a yield stress are clays, toothpastes, ketchups, and drilling mud.

Cementitious fluids have complex rheology and exhibit yield-pseudoplastic and thixotropic behavior. Further complicating matters is the fact that the rheological properties of

cementitious fluid are constantly changing with time. G.H. Tattersall and P.F.G. Banfill wrote the first comprehensive book on the rheology of cement paste and concrete in 1983 [30]. In this classic text, it was suggested that concrete and cement paste should be described using a minimum of two parameters. The flow behavior of cement paste and concrete is most often modeled as Bingham fluids (two parameters are τ_y and μ). However, a pseudoplastic model is probably a more accurate representation of the flow behavior since oftentimes the viscosity above the yield stress tends to show shear-dependent behavior.

Within the time-dependent category, there are two rheologies: thixotropy and rheopexy. Thixotropic fluids display a reduction in magnitude of rheological properties (ex. elastic modulus, yield stress, and viscosity) with a distinct time dependence on the application of the shear strain [33]. The changes in the system occur reversibly and isothermally. A rheopectic, or antithixotropic, fluid exhibits behavior that is the opposite of a thixotropic fluid. Namely, the viscosity increases with time under a constant rate of shear, and once the shear is removed the viscosity decreases. Oftentimes, thixotropy is confused with the shear-thinning behavior of non-Newtonian fluids. It is important to point out that thixotropy is related to non-Newtonian time-dependent changes, whereas shear-thinning refers to Non-Newtonian time-independent changes. A pseudo-plastic fluid should “solidify” almost immediately after being sheared, whereas a finite amount of time is required before a thixotropic fluid “solidifies”. As shown in Figure 3-4, in a plot of viscosity versus time (for a given shear rate), a shear thinning fluid will have a constant viscosity for any given time, whereas a thixotropic fluid will display a decrease in viscosity.

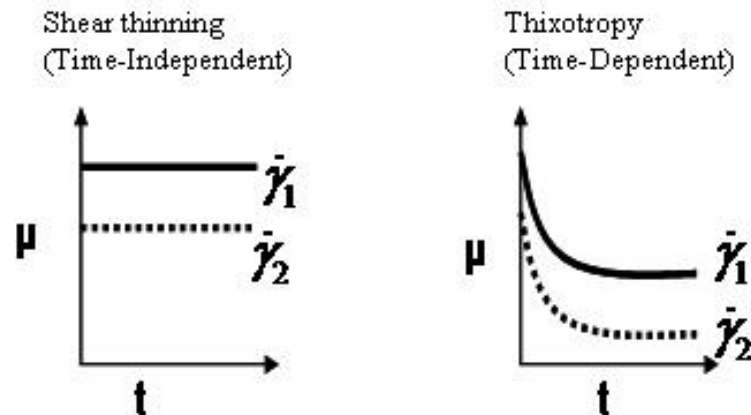


Figure 3-4. Shear thinning and thixotropic fluid

As mentioned in Section 3.2, viscoelasticity is also due to time effects, and this can be another source of stress relaxation at constant strain rate. When a shear stress is applied to a viscoelastic material part of the mechanical energy supplied is stored as elastic energy. When the maximum stored energy for a particular stress is reached, the rate of supply of the energy falls to that necessary to maintain flow, and the strain rate decreases [34]. When the stress is removed, the elastic recovery of the system can be observed. In these fluids, there will be partial recovery.

3.2.2 Measurement

Commercially available devices in the form of coaxial cylinder, parallel plate, and vane geometry are commonly used to measure the rheological properties of cement paste. High fluidity cement pastes are measured most often using a coaxial cylinder or vane configuration in order to prevent the material from flowing out of the test apparatus. Originally developed to

measure polymer rheology, the sample holders for commercially available rheometers are small, and, thus, are limited to cement paste. Slippage at the walls of the viscometer, sedimentation of the particles, and plug flow [35] are some experimental challenges that can be encountered during rheological testing. The yield stress and viscosity of the sample can be underestimated due to the occurrence of a water rich layer that is formed at the surface of the rheometer. For concentrated suspensions, a low viscosity layer can develop near cylinder surfaces which leads to an apparent wall slip [32]. Slip is more prevalent in concentric cylinder rheometers due to the constant strain rate in the material. The slip velocity can be determined by making measurements with two different inner bobs with different radii, and this velocity can be used to correct shear rate readings to give the true viscosity. Other ways to minimize slip are to use a vane bob or by serrating the rheometer surface. Sedimentation is of particular concern to high fluidity cement pastes. At high water contents, the particles in cement pastes may separate gravitationally and centrifugally, which can cause errors. Plug flow occurs when the shear stress does not exceed the yield stress everywhere in the sample. Thus, there will be a region within the material that remains stationary and acts as a solid plug. Plug flow becomes more significant at low shear rates. In the light of these experimental challenges, care should be taken to minimize these problems when possible.

A number of custom-designed rheometers (BML, BTRHEOM, IBB, CEMAGREG-IMG and Two-point workability test rheometer) have been developed to measure the rheological properties of mortar and concrete. The reader is referred to reference [36] for more information about the configuration and measurement principles of concrete rheometers. A comparison of concrete rheometers was conducted at Laboratoire Central des Ponts et Chaussées (LCPC) in Nantes, France [37]. The yield stress and plastic viscosity of twelve concrete mixes were

measured in five different rheometers in order to compare the data measured from the different rheometers and to establish correlation functions. It was found that even in controlled rheology experiments (in which the same materials and mixing protocol were used) significantly different values of yield stress and viscosity were obtained. However, good empirical correlation of yield stress and viscosity measurements between any two rheometers existed. A second comparison of concrete rheometers was conducted in 2003 in Cleveland, Ohio to determine the reliability of the rheometers [38]. A high degree of variation was found among the measurements, which may be an indication that the concrete rheometers were more sensitive to small variations in concrete mixtures than originally thought.

3.3 Origin of yield Stress in cement paste

An amorphous C-S-H gel structure forms around the cement grain immediately after the cement reacts with water. This gel structure arises from a combination of colloidal forces and the chemical reaction that occurs during hydration. While cement may not be technically defined as a colloid, a large proportion of cement consists of colloidal (particles ranging from 1 nm – 10 μ m [39]) particles, and for a typical cement powder the percentage of particles smaller than 2 μ m is 20% by number [40]. It is likely that the interparticle forces resulting from these colloidal particles largely contribute to the yield stress of the paste. Furthermore, as the cement paste ages, its structure changes and the yield stress and viscosity increases. The microscopic mechanism of yield stress in cement paste is likely due to the gel structure's resistance to deformation and reorganization of individual cement particles under applied stresses. If the applied stress exceeds the threshold of the particles and the gel structure to resist deformation and reorganization, flow will occur. The yield stress can be used as a measure of the strength

and number of interparticle bonds that are ruptured due to the applied shear or stress [30]. It has been hypothesized that interparticle links exist within cement paste that are broken irreversibly by shear [30]. Tattersall and Banfill reported initial yield values for a normal consistency cement paste ranging from 50 – 200 Pa [30]. Similarly, the yield stress values for VC range from 500 Pa to a several thousands Pa, and for SCC these values tend to range from 0 – 60 Pa [41].

There are several ways to measure yield stress, and one of the most common techniques is to extrapolate the shear rate – shear stress curve to the point in which the line intercepts the stress axis. The point of intersection is considered the yield stress. In actuality, an exact yield point is often difficult to measure, and the existence of the yield stress as a material property is a topic of much debate [42-44]. It has been suggested that the yield stress does not exist at all [45]. For most structured liquids the transition from the point of no flow to the point of flow (i.e. yield point) is not a single defined point. Barnes and Walters showed that if the shear rate is sufficiently low, there exists a lower Newtonian viscosity zone, and at sufficiently high shear rates, an upper Newtonian viscosity zone. Between the upper and lower Newtonian viscosity zones the viscosity decreases. Thus, there exists a range of stresses over which the flow behavior changes dramatically. However, it has been argued that if the transition occurs over a small range of stresses, then, for practical engineering purposes, this region can be considered the yield stress, and, therefore, it is defined as an apparent yield stress [46].

Further complicating the matter, is the fact that different results can be achieved based on the materials, testing protocol, and measuring device [30, 47-49]. Rahman and Nehdi [50] investigated the effect of different rheological test geometries (smooth parallel plates serrated parallel plate, coaxial cylinder, vane rotor) on the properties of cement paste. Rheological

properties were found to be strongly affected by the surface friction and gap of the test geometry. Test geometries with lower friction capability (i.e., smooth parallel plate and coaxial cylinder) exhibited lower yield stress and higher viscosity compared to test geometries with higher friction capability. Hodne [48] compared the influence of propeller blade shape and size on the energy applied to oil-well cement slurries. It was found that the mixing energy was dependent on the type of blade used and that the type of propeller blade used during mixing influenced the rheological performance of the cement slurries. In 2002, Geiker et al. [49] investigated the effect of measuring protocols on the rheological properties of SCC. They found that shorter measurement sampling time resulted in higher values for yield stress and viscosity and that the amount of time needed to achieve steady state flow is an important parameter that should be taken into account when determining rheological protocols.

3.4 Origin of thixotropy in cement paste

In cementitious materials, the phenomenon of thixotropy results from the heterogeneity of the material and changes in the resulting microstructure due to hydration. The specific causation of thixotropy depends on interactions at the molecular level. Unfortunately, these mechanisms are poorly understood. Thixotropy develops due to the finite time that it takes for the microstructure to change from one state to another, and Helmuth (as reported by Struble [40]) suggested that mixing breaks down the flocculent structure responsible for thixotropic behavior. Thus, from a microstructural perspective, thixotropy is a result of structural degradation due to the rupturing of flocs or linked particles [51]. As shown in Figure 3-5, one can visualize thixotropy as a coagulation of particles when the system is at rest, and then as a separation of the particles during shearing or agitation due to the rupturing of the interparticle bonds. In cement paste, it is likely

that thixotropy is governed by a combination of reversible coagulation, dispersion, and then re-coagulation of the cement particles [52].

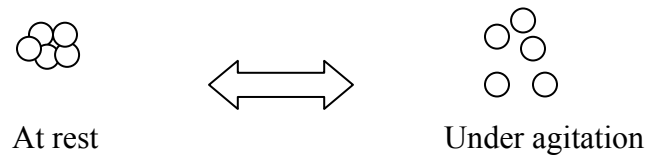


Figure 3-5. Schematic of thixotropy phenomenon

The decrease in viscosity that is accompanied with thixotropy is due to the resulting flow altering the microstructure. When a specific microstructure is agitated, the viscosity will decrease with the shearing time until an equilibrium state (the lowest energetically possible state) is achieved. Thus, the time scale in which the microstructural changes take place is an important parameter in the consideration of thixotropy. During macroscopic flow, alignment of particles along flow lines and/or separation of flocculated particles can occur, which results in a decrease in viscosity. Reflocculation of the particles may offset the degree to which the viscosity reduces.

From the above discussion, it may be apparent that yield stress is coupled together with thixotropy. The yield stress of cement paste is influenced by thixotropy because it takes time for the microstructure to rebuild. For example, suppose a sample of cement paste is sheared for 5 minutes and then is left at rest. The resulting yield stress will be a function of the time that the sample is allowed to rest. For a thixotropic fluid, longer resting times result in higher yield stress because the fluid becomes more structured with time. The rate at which the yield stress changes with time of rest can be used to quantify thixotropy [53].

There are several papers in the literature that confirm the existence of the thixotropic phenomena in cementitious materials [30, 54, 55]. In 1941 Pierce conducted one of the earliest experimental studies on thixotropy in which he attempted to measure the gel strength of cement slurries under alternating cycles of rest and agitation [56]. More recently, in 2002, researchers at the University of Sherbrook [57] investigated the influence of thixotropy on grout and concrete by examining the difference between the peak stress and equilibrium stress. Another study [28] was conducted in 2003 to assess thixotropy in SCC. In both of these studies, the mixtures were not brought to an equilibrium state prior to performing the experiments. This is an important consideration because the initial stress value depends on the shear history of the mixture. The researchers at the University of Sherbrook conducted experiments in order to determine the variation of thixotropy over time after 1, 2, and 3 hours, which provides a useful way to describe the rate in which the material is rebuilding.

3.4.1 Rheological experimental characterization of thixotropy

Measuring the rheology of cement paste gives an indication of the colloidal state and interactions that are occurring. There are no standard methods to measure thixotropy but typical thixotropic experiments often consist of either rheological tests conducted at a constant shear rate (equilibrium flow curves) or using varied sheared rates (hysteresis loops) [8, 58, 59]. In the varied sheared rate protocol, the shear rate is increased to a pre-determined maximum and then decreased, and the rheological behavior is evaluated by plotting the resulting shear stress as a function of the applied versus shear rate. If the plot shows a hysteresis loop (the “up increasing shear-rate” curve is above the “down decreasing shear-rate” curve) then the material is thixotropic (Figure 3-6a), whereas if there is no hysteresis loop (the “up” curve and “down”

curve coincide) the material is shear-thinning (see Figure 3-6b). In actuality, nearly all shear-thinning materials are thixotropic because it takes time for the microstructure to realign itself.

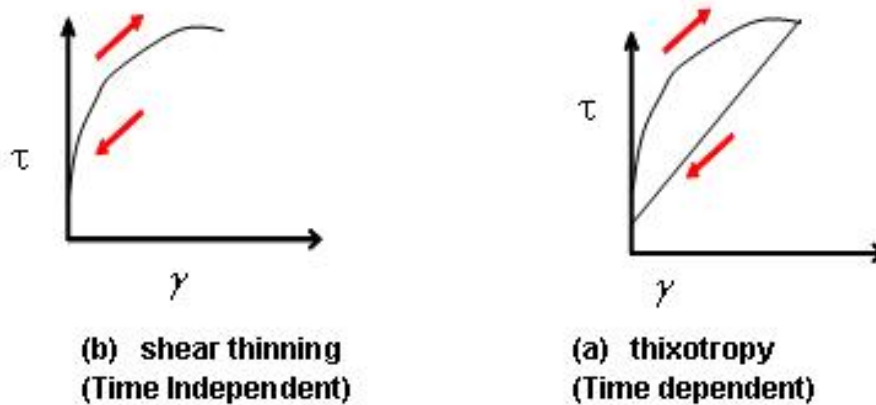


Figure 3-6. Hysteresis approach to examining thixotropy

For thixotropic fluids, the area between the “up” and “down” curve can be used as a measure of thixotropy [58]. If a sample is left at rest for a period of time before testing, then the formation of the up curve is a result of the initial resistance of the sample to flow. Once the initial resistance of the sample is exceeded, flow will occur and the sample will undergo partial breakdown. When the shear rate is decreased, there is a natural tendency for the structure to try to rebuild. However, it takes time for the bonds to rebuild, and, thus, the up-curve and down-curve do not coincide. An ideal shear thinning fluid is able to rebuild rapidly, and the up and down curve will coincide.

Tattersall [60] was the first to report the use of varied shear rates to study the structural breakdown and thixotropic behavior of cement paste. Pastes with a water-to-cement (w/c) ratio ranging from 0.28 – 0.32 were tested using a coaxial cylinder viscometer. The speed of rotation

of the outer cylinder was increased to a pre-defined maximum, reduced to zero, and then increased again. This protocol (increasing and decreasing the speed of rotation of the outer cylinder) was repeated about four times. Hysteresis loops were obtained for all paste mixtures (see Figure 3-7), which provided evidence of the occurrence of structural breakdown due to shear. Unfortunately, it is unclear if the specimen was allowed to rest for a period of time before the outer cylinder's speed of rotation was increased. Nevertheless, under the experimental conditions used in Tattersall's research, the structural buildup did not appear to occur at a fast rate.

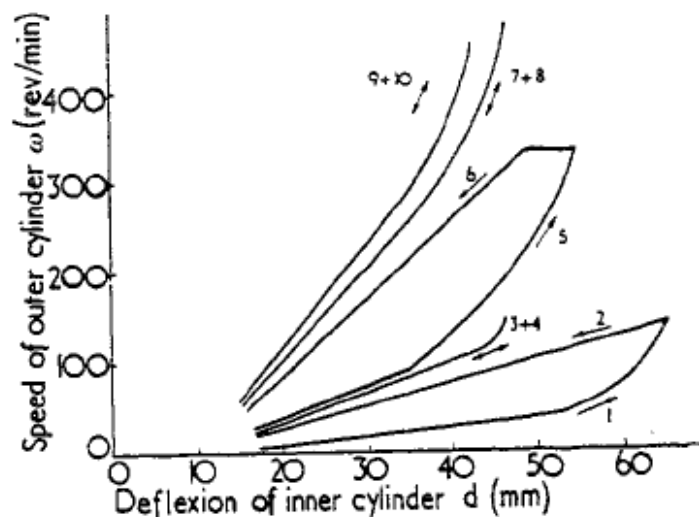


Figure 3-7. Hysteresis curves for cement paste [60]

In 1999, Ur'ev et al. [55] conducted experiments using the hysteresis testing method to evaluate the variation of yield stress in different cement paste systems. Various superplasticizers and different cement compositions were used in this study. In all mixtures, positive hysteresis loop behavior was seen (up curve is at higher stress levels than the down curve), and it was

supposed that this behavior was due to irreversible destruction of the crystalline component of the coagulation-crystallization structure. Regardless of the type of superplasticizer added, the area of the hysteresis loop significantly decreased..

If the structure experiences complete breakdown then the intersection of the down curve of the hysteresis loop with the ordinate axis (τ_{od}) may also be used as an indication of yield stress. Atzeni et al. [61] conducted a study in which τ_{od} was compared with the equilibrium yield stress extracted from the equilibrium flow curve. Good agreement was found between the two values. In general, a disadvantage of using a hysteresis loop testing method to evaluate thixotropy is that the entire structural breakdown of the material may not be captured since the loop captures a range between the peak and equilibrium stress for a given shear rate.

The shape of the hysteresis loop is highly influenced by the ramp rate. Studies conducted by Saak [51] show that as the ramp rate decreases, the hysteresis loops shift to lower stress values since more energy is inputted into the system to break the interparticle bonds of the cement paste. However, if the ramp rate is too slow then it is possible to get a “figure 8” hysteresis due to structural recovery. The varied shear rate protocol is a useful measurement technique to measure the flow characteristics of cement systems, and information about the transient flow properties can be obtained from testing one sample. However, the shape of the hysteresis loop is very sensitive to the testing protocol (ramp rate, shear history), and, thus, it is important that a detailed account of the experimental procedure is given.

Equilibrium flow curves are generated through monitoring the stress as a function of time for a constant shear rate (see Figure 3-8). The initial stress, τ_i , is generally considered to correspond to the initial structural condition [5], and the equilibrium stress, τ_{eq} , represents a structural condition in which rate of structural breakdown and buildup are in equilibrium. τ_{eq} is

independent of shear history and should only depend on the shear rate. It should be noted that the material's initial response in such tests may be elastic [8], and in controlled strain experiments the thixotropic response may be seen as an overshoot in the stress. Typically, the equilibrium flow curve is preferred over hysteresis loops since the effects of time and shear rate are not coupled.

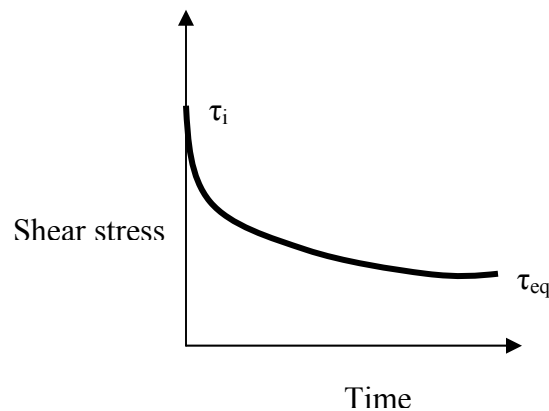


Figure 3-8. Equilibrium flow curve

Figure 3-9 shows the results that were obtained from testing an oil-well cement slurry made from an ordinary Portland cement as reported by Hodne [48]. In this study, the rheological properties of the cement slurries were tested after specific time intervals in a consistometer (an apparatus that measures the torque exerted on a stationary paddle immersed in a rotating container). In the figure, each curve represents the shear stress measured at a particular shear rate plotted as a function of time spent in the consistometer. Although it was not pointed out in the paper, it can be seen that for almost every shear rate, no significant increase in shear stress occurred until after 120 minutes. Thus, two possible hypotheses can be made: 1) for the shear

rates used in the study, the rate of structural build-up and structural break-down are in equilibrium; 2) breakdown dominates and rate of structural buildup is too slow to be significant.

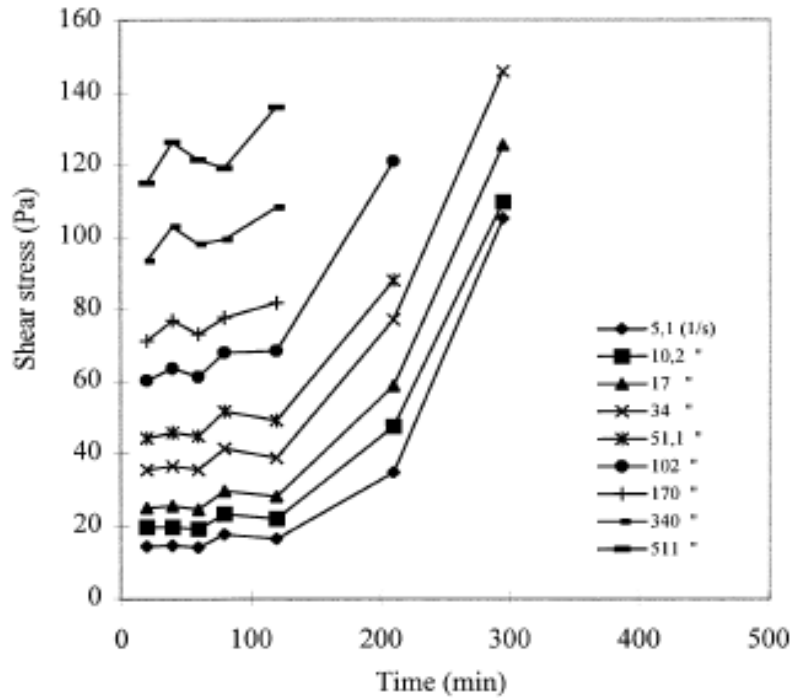


Figure 3-9. Shear stress measured at different shear rates as a function of preconditioning time [48]

3.4.2 Mathematical theories for thixotropy

An ideal model to describe thixotropic behavior should start from some fundamental, rheology-determining physical entity (ex. floc size, orientation angle of particles) that takes time to change when the flow field around it is changing or has changed [8]. With information about the rate at which the changes occur and how the microstructure relates to the stress, a predication about the overall behavior may be possible.

3.4.2.1 Viscous theories:

A large number of thixotropic materials hardly show any elastic effects [58]. There are numerous models in the literature [58] to describe viscous thixotropic behavior. In general they can be divided into two classes: generalized continuum theory and structural kinetics.

Generalized continuum approach: These models specify the properties of the fluid through a constitutive equation (equation that relates stresses (forces) to strains (deformations)). The Reiner-Rivlin constitutive equation describes the nonlinear viscosity and some normal stress effect behavior of (visco-inelastic) fluids [33]. Existing theories for thixotropic materials generalize the Reiner-Rivlin constitutive equation by making the relationship between stress and strain rate dependent on time [58]. The model can also be modified to include plastic behavior (yield stresses). Slibar and Paslay (as reported in Mewis [58] and Mujumdar [62]) were the first to apply a phenomenological approach to thixotropy. They used a Bingham equation with a non-constant yield stress that was a function of shear rate and time to describe the flow of drilling mud. In the model, flow will occur if the critical yield stress required for initiating flow (τ_{crit}) exceeds the stress necessary to sustain it (sliding yield stress, $\tau_{sliding}$). The important modeling features can be shown as [62]:

$$\tau = \tau_{sliding} + \mu\dot{\gamma} \quad \text{for } \tau \geq \tau_{crit} \quad \text{Equation 3-6}$$

where $\tau_{sliding} = \tau_{sliding}(\dot{\gamma}, t)$ and τ_{crit} depends on shear history through an exponential memory function. Similar models can be developed by modifying the assumptions used by Slibar and Paslay (ex. allow for pseudoplastic behavior or introduce Hookean elastic behavior before yielding). It should be noted that Equation 3.6 is a simplified expression of the equations

developed by Slibar and Paslay in order to highlight the key terms. In reality, models based on the continuum mechanics approach may consist of terms in which the physical meaning is not clear [58] and may not be directly connected to the basic processes responsible for the structural changes [62].

Structural kinetics approach: The flow behavior of thixotropic fluids involves the competition of between the interparticle bonds broken from shear-induced stresses and the interparticle bonds formed from shear-induced collisions and attractive forces. Thus, the behavior of both can be modeled as a rate process in which the two major parameters are structural breakdown and structural buildup. A state of dynamic equilibrium is reached when the rates between the two are equal. The methodology of this approach consists of a rheological constitutive equation and a kinetic equation.

3.4.2.2 Indirect Microstructural Theories

This approach is based on a scalar structural parameter, λ , which is used to represent the extent of breakdown. λ can vary from 0 to 1 where a completely broken down structure is represented by $\lambda = 1$ and a fully built up structure is represented by $\lambda = 0$. The time derivative of λ is used to represent thixotropy, and, in the simplest theories, it is only controlled by the shear rate and the current level of structure λ [8]. Balance equations can be written for the structural buildup and structural breakdown

$$\frac{d\lambda}{dt} = k_2(1 - \lambda)\dot{\gamma}^b - k_1\lambda\dot{\gamma}^a \quad \text{Equation 3-7}$$

where a is the exponent for the rate of structural breakdown, b is the exponent for rate of structural buildup, and k_1 and k_2 are rate constants for breakdown and build-up, respectively. λ is then related to the shear stress or viscosity through rheological models (ex. Bingham equation, etc.). It is assumed that the rate of break-down depends on the shear rate and on the fraction of links existing at any instant and that the rate of buildup is proportional to the fraction of links remaining to be formed [63]. In Equation 3-7, the mechanisms of breakdown and buildup are described using a single rate constant; however, considering the likelihood that there are different types of bonding in the system (van der Waals, ionic, etc.) and that these bonds have different strength, using a distribution of rate constants may provide a more accurate characterization of the mechanisms.

In 1964, Worrall and Tuliani [64] used the concept of structural parameter to account for yield stress during the aging in clay-water suspensions. For deflocculated clay-water suspensions, the rate of structure breakdown was given as

$$\frac{d\lambda}{dt} = k_2(\lambda_0 - \lambda) - k_1\lambda\dot{\gamma} \quad \text{Equation 3-8}$$

where λ_0 is the maximum amount of structure in the suspension and λ varied from 1 (for complete build-up) to 0 (for complete breakdown). The relationship between λ and stress was found to be

$$\tau = \eta_0\dot{\gamma} + \tau_{yield} + \frac{g \cdot b \cdot \dot{\gamma}}{a\dot{\gamma} + b} \quad \text{Equation 3-9}$$

where τ_{yield} is the yield stress.

In the classic paper by Cheng and Evans [65], general models for thixotropic and antithixotropic fluids were developed. They proposed that the change in shear stress with respect to the change in shear rate (for a constant structure) is always greater than zero and that the structure will always tend towards the equilibrium condition. Allesandrini et al. [66] used the Cheng and Evans approach to derive a constitutive equation characterizing the rheological behavior of gypsum plaster pastes. The pastes were tested under a constant shear rate condition, and the model was used to characterize the shear stress decay prior to the occurrence of set. The kinetic equation for the degree of structure was considered in respect to the Herschel-Bulkley equation and the Bingham equation. While a better fit to the data was obtained with the Herschel-Bulkley equation, the Bingham equation better described the shear and time-dependent behavior due to variations in the solid volume content. In 1988, Papo [67] also employed the Cheng and Evans approach to characterize the initial shear stress decay with time for white cement pastes. The constitutive equation used in this study was based on the Casson model. Roussel et al. [68] also employed the Cheng Evans approach to fresh cement pastes. Their constitutive relationship was based on an approach by Coussot et al [69] in which the apparent viscosity was an increasing function of the structural parameter and was given as

$$\tau = \mu_0 (1 + \lambda^a) \dot{\gamma} \quad \text{Equation 3-10}$$

and

$$\frac{d\lambda}{dt} = \frac{1}{T} - k_1 \lambda \dot{\gamma} \quad \text{Equation 3-11}$$

where μ_0 is the viscosity at infinite shear rate when λ tends towards zero (completely broken down structure) and $1/T$ is the flocculation term, and T is the characteristic time of aging. In 2003, Cheng [70] revisited the models proposed in his earlier work in order to take into account negative slope behavior of the equilibrium flow curve for thixotropic materials.

3.4.2.2.1 Direct Structural Theories

In these models, the rate of change in time of rheological parameters is caused by changes in the internal structure [58]. Typically, it has been assumed that the number of bonds within the material is an adequate way to characterize the changes within the internal structure. However, it should be noted that there are studies in which the apparent viscosity has been used as a measure of structure [58]. Similar to the methodology used in Section 3.4.2.2, balance equations can be used to represent the change in the interparticle bonds:

$$\frac{dn(t)}{dt} = k_2 [n_0 - n(t)]^d \dot{\gamma}(t)^\beta - k_1 n^c \dot{\gamma}(t)^\alpha \quad \text{Equation 3-12}$$

where n is the number of unbroken bonds, n_0 is the total number of bonds in the existing virgin structure, α is the exponent for structure breakdown, β is the exponent for structure build-up, and k_1 , k_2 , d , and c are constants [58]. A change in dn/dt results in a proportional change in the material behavior (ex. viscosity), and the rheological constitutive equation characterizes instantaneous stress as a function of the degree of structure, $\tau = [an + b]\dot{\gamma}^\beta$. As reported by Barnes [8], Denny and Brodkey used a structural kinetics approach that looked at the distribution of broken and unbroken bonds and then later connected the number of these bonds to viscosity.

3.4.2.3 Viscoelastic Theories

The mechanism of thixotropy can be introduced into viscoelastic theories if the parameters for viscous and elastic responses are modified to take into account variable structure. For direct structural theories, the instantaneous response to stress is now viscoelastic [58], and the generalized continuum approach should be modified to include elasticity.

3.5 Flocculation Mechanisms

3.5.1 Attractive Forces

The way in which the atoms are arranged is primarily determined by nature and the directionality of the interatomic bonds holding the fluid together [71]. Ionic, covalent, and metallic bonds can be considered strong primary bonds. These bonds are formed when outer orbital electrons are transferred or shared between atoms [72]. Van der Waals force is the general term for all the weaker secondary intermolecular attractive forces that exist between neutral molecules. These forces are significant only when molecules are very close together. These secondary bonds can be thought of as physical bonds, and they are the result of more subtle attraction between positive and negative charges without any actual transfer or sharing of electrons [72]. Rather, the attraction depends on the dipole (asymmetrical distribution of positive and negative charges) within each atom or molecular unit being bonded. In regard to thixotropy, the interatomic bonds are easily broken by shear; thus, it is likely that the weaker secondary bonds are at issue and that these are the bonds that will control reversible coagulation and deflocculation. There are four types of secondary forces: ion-dipole, dipole-dipole, hydrogen bonds, and London dispersion force. The following paragraph will discuss these forces in more detail.

The shape of a molecule and the polarity of its bonds determine the charge distribution in the molecule. A molecule in which the centers of negative and positive charge do not coincide is called a polar molecule. Dipole-induced dipole (also known as Debye) occurs when a polar molecule having a permanent electric dipole moment induces a dipole moment in a nonpolar molecule [73]. Dipole-dipole (also known as Keesom) forces are the attractive forces that occur among molecules with a permanent electric dipole moment. These forces are generally weaker than ion-dipole forces [74]. For molecules of similar mass and size, an increase in dipole polarity increases the strength of intermolecular attraction. Hydrogen bonds are a special type of intermolecular attraction occurring between the hydrogen atom in a polar bond and an unshared electron pair on a nearby ion or atom [74]. These bonds can be considered a particular type of dipole-dipole bond, but, they tend to be stronger than the typical dipole-dipole bond. London dispersion forces arise from the temporary (instantaneous) dipole that is created from the oscillation of electron clouds in an atom or molecule. The temporary dipole in an atom can induce a similar dipole on an adjacent atom causing the atoms to be attracted to one another. Since all molecules have electron clouds that can oscillate, London forces always contribute to intermolecular attractions. Forces like dipole-dipole will act in addition to the London dispersion forces.

Van der Waals forces are always attractive between particles of the same nature. As noted by Morrison [75], in 1937, Hamaker proposed a molecular model to calculate the van der Waals force of attraction between particles. This model is based on the summation of the van der Waals energies between all pairs of molecules in the separate particles and is proportional to the inverse sixth power of distance [76]. The general form of the Hamaker equation for the energy of attraction between two particles is given by:

$$V_{A,12} = -A \oint_{V_1} dV_1 \oint_{V_2} \frac{dV_2}{\pi^2 r_{12}^6} \quad \text{Equation 3-13}$$

where dV_1 and dV_2 are differential volume elements of the particles, and A is Hamaker constant [75]. Hamaker constant is a material-dependent term dependent on the properties of the particles and the dispersion fluid. The particular case of the attraction between two equal spherical particles of radius, a , is given by:

$$V_A = -\frac{A}{6} \left[\frac{2a^2}{R^2 - 4a^2} + \frac{2a^2}{R^2} + \ln \frac{R^2 - 4a^2}{R^2} \right] \quad \text{Equation 3-14}$$

where R is the distance between the centers of the two particles. The attractive force is directly proportional to the size of the particle and inversely proportional to the square of the distance separating the particles (Figure 3-10).

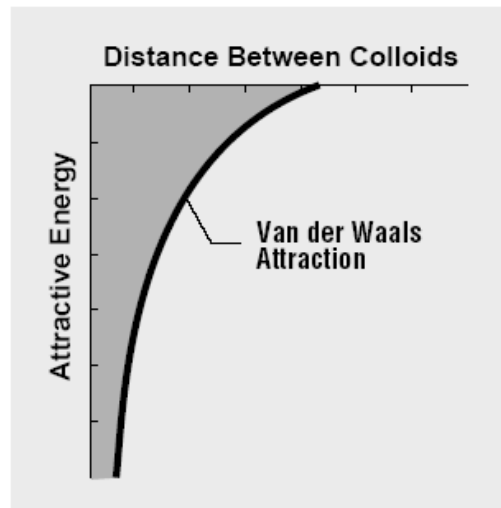


Figure 3-10. Attractive energy as a function of distance between particles [77]

It should be noted that the energy of attraction is retarded if the distance between the particles is too large. This is due to the large amount of time that is needed for the electric field to propagate from one particle to another. However, in most practical examples of colloid stability, only the nonretarded van der Waals force is important [76].

As stated earlier, the van der Waals force is always present between atoms, and, in the absence of any other repulsive or attractive forces, the London dispersion force will be responsible for the creation of flocs in a suspension [78]. However, in cement paste it is unclear whether flocculation of cement particles is due to the London dispersion force or electrostatic attraction among particles of dissimilar surface potential [40]. The strength of the force of attraction from the London dispersion forces is related to the size of the particle, and, for colloidal particles, the forces are significant. As stated in Section 3.3, although cement is not a colloid, a large percentage of the particles are colloidal. Thus, it is likely that a large number of particles may flocculate due to the London dispersion forces. In addition, since cement is a heterogeneous material and it is composed of many different phases, compositional differences are also likely to cause electrostatic attraction between particles [40]. A recent study by Plassard et al. in 2005 [79] involved the use of atomic force microscopy to investigate if non van der Waals forces contribute to the flocculation of cement pastes. The pore solution of the hydrated cemented paste was simulated using a calcium hydroxide solution. It was found that the intensity of the attractive force is related to the calcium hydroxide concentration. For high calcium hydroxide concentrations, the interaction force was purely attractive, and it was suggested that the attraction was due to ion-ion interaction. This indicates that there are forces other than the weak van der Waals force that are holding the structure together, and it is possible that the different bonds will break and reform at different rates.

3.5.2 Repulsive Forces

Repulsive forces can occur either through electrostatic repulsion or by steric repulsion. The next section will discuss these two mechanisms in more detail.

3.5.2.1 Electrostatic Repulsion

In an aqueous media, particles can be electrically charged. As a result, the ions in the water that have a sign opposite of the particle charge will be drawn toward the particle [32]. Thus, near the surface of the particle, a layer of these charges will develop. This layer is called the Stern layer. The Stern layer is strongly attached to the surface of the particle. Outside of the Stern layer is a diffuse layer of counter ions. The concentration of counter ions is high near the surface of the Stern layer, and the concentration of counter ions decreases linearly with distance within the Stern layer. Outside of the Stern layer, the potential decreases exponentially with distance until equilibrium is reached with the counter-ion concentration in the bulk solution (Figure 3-11).

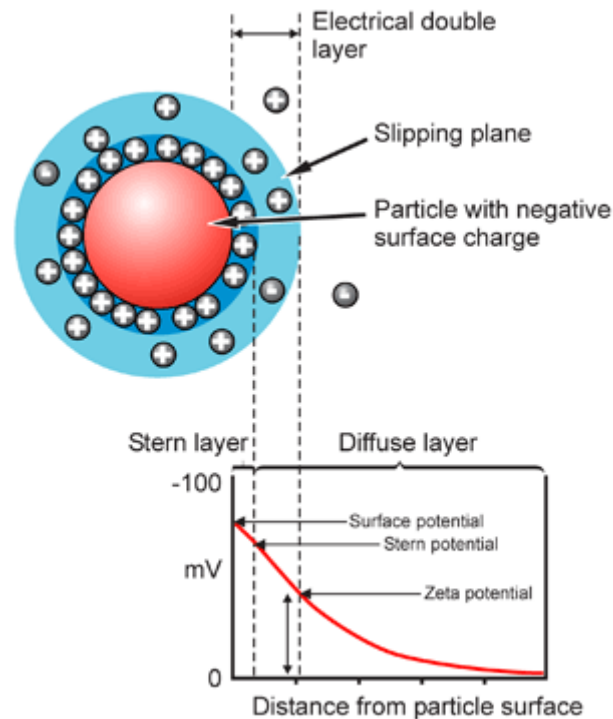


Figure 3-11. Schematic representation of zeta potential

Because cement particles in water are positively charged (Powers as reported in Saak [51]), let us suppose that the Stern layer is composed of negative ions. Additional negative ions are attracted to the positive cement particle, but they are repelled by the negative Stern layer and the other negative ions that are also trying to approach the positively charged cement particle. Thus, the diffuse layer is the result of the dynamic equilibrium of competing counter-ions. If the particles are of the same chemical nature and have surface charges and surface potentials of the same sign and magnitude, then the electrostatic forces will always lead to repulsion between the particles [76]. In cement paste, the electrostatic repulsion force will depend on the pore solution.

The attached counter-ions in the Stern layer and the charged atmosphere in the diffuse layer together are called the double layer. The Debye-Hückel length ($1/\kappa$) represents the thickness of the double layer, which depends upon the type and concentration of ions in solution. The range of the electrostatic repulsion force is of the order of the Debye-Hückel length:

$$\frac{1}{\kappa} = \sqrt{\frac{\varepsilon\varepsilon_0 RT}{F^2 \sum c_i z_i^2}} \quad \text{Equation 3-15 (from [76])}$$

where ε_0 is the permittivity of the vacuum, ε is the dielectric constant (relative permittivity) of the dispersion medium, R is the gas constant, T is the absolute temperature, F is the Faraday constant, c_i is the ionic concentration of the ions of type i in the dispersion medium, and z_i is the charge number of the ions of type i in the dispersion medium. In colloidal dispersions, the Debye-Hückel length ranges from 1 – 100 nm [80].

For small surface potentials, Φ_0 and small κa (thus the double layer is extended beyond the radius of the particle), the energy of repulsion, V_r , is

$$V_r = 2\pi\varepsilon\varepsilon_0 a\phi_0^2 \frac{\exp(-\kappa H)}{1 + H/2a} \quad \text{Equation 3-16}$$

where H is the distance between the surfaces of the two particles. Thus, the distance between the two centers of two particles is given by $R = 2a + H$. For small values of Φ_0 and small values of $\exp(-\kappa H)$, the repulsion energy is

$$V_r = 2\pi\varepsilon\varepsilon_0 a\phi_0^2 \exp(-\kappa H) \quad \text{Equation 3-17}$$

V_r becomes significant when the double layers between two particles begin to interact.

An electrostatic repulsion curve is used to indicate the energy that must be overcome for the particles to be forced together. A maximum value for the repulsive energy is obtained when the particles are in close contact, and it decreases to zero outside of the double layer (see Figure 3-12).

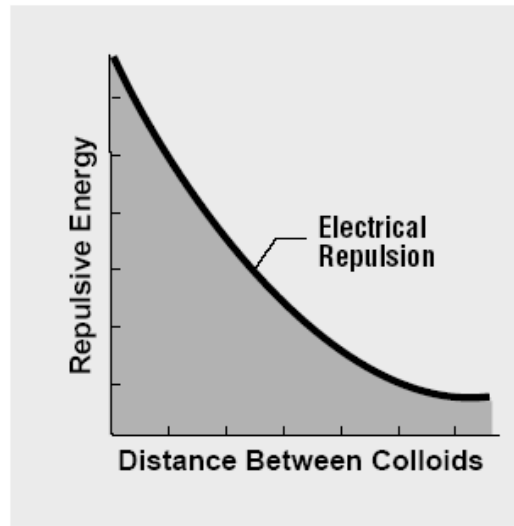


Figure 3-12. Repulsive energy as a function of distance between particles [77]

It is assumed that the surface potential is constant during the course of action of the particles; however, for cementitious materials this may not be a valid assumption due to changes that result from the hydration reaction. Furthermore, the value of the actual surface potential is often not known, and, thus, the zeta potential, ξ , is often used as an approximation of Φ_0 . The zeta potential is defined as the electric potential at the plane of shear between the moving particle and the liquid constituting the bulk liquid phase [81]. This plane of shear (also called the slip plane) is usually defined as the point where the Stern layer and the diffuse layer meet, but it can vary anywhere within the double layer. The electric potential within the plane of shear can vary

right up to the surface of the particle due to the adsorption of ions within the Stern layer [75].

The zeta potential and the Debye-Hückel length are the two parameters of the double layer that can be determined experimentally. However, neither of these properties can be measured directly. The zeta potential is computed from measurements that use electrokinetic motion of particles, while the Debye-Hückel length is computed from the ionic composition of the dispersion medium and where the ionic strength of the bulk solution is defined as

$$I_c = \frac{1}{2} \sum c_i z_i^2 \quad \text{Equation 3-18 [80]}$$

If the zeta potential is too small then the maximum electrical repulsion will not be strong enough to overcome the London-van der Waals attraction between the particles, and, as a result, the particles will flocculate. In pure cement paste, the zeta potential typically ranges from -10 to 20 mV [82]. Chemical admixtures, such as water-reducing agents and viscosity-modifying agents (VMA), are commonly used to alter the flocculation mechanisms. Water-reducing admixtures are commonly used in concrete to lower water demand requirements while still maintaining good workability. Regular water-reducing admixtures are capable of reducing water demand requirements by 5 - 10% while mid-range water reducers reduce water demand requirements by 10 - 15%. Further reductions in water requirements can be obtained by using higher dosages, though it is not recommended due to problems that can occur with setting, segregation, bleeding, and air content of the concrete. High range water-reducing admixtures (also known as superplasticizers) are able to impart much higher amounts of water-reduction (ranging from 15 – 30%) and are commonly used in SCC and high-performance concrete. Until

recently, all superplasticizers were highly ionic polymers that adsorbed at the interface between the cement particle and water. As mentioned in Section 3.5.2.1, cement particles carry a surface charge, and the molecules of the water-reducing admixture interact to neutralize these surface charges so that the surfaces will carry uniform charges of the like sign [83]. As a result, electrostatic repulsion acts between the cement particles. The cement particles are dispersed and the fluidity of the cement paste increases. The addition of water-reducing agents typically induces negative zeta potentials, ranging from -30mV for strong electrolytes to -5mV for weaker electrolytes [82, 84]. In ordinary cement pastes, zeta potential studies have been used successfully to investigate particle coagulation [85], flocculation mechanisms [86], and the effect of superplasticizers on rheology [87]. If the zeta potential is too small then the maximum electrical repulsion will not be strong enough to overcome the London-van der Waals attraction between the particles, which will result in flocculation/coagulation.

3.5.2.2 Steric Repulsion

Surface-active agents are organic molecules that when dissolved in a solvent at low concentration have the ability to adsorb at the interfaces, thereby altering the physical properties at those interfaces [39]. The addition of large molecules, such as proteins or gums, inhibits flocculation in colloidal suspensions due to interaction of long-chain molecules adsorbed on the particles. The source of the steric repulsion between dispersed particles is due to an increase in free energy resulting from the overlap of adsorbed layers [75]. This will occur if the chain length of the adsorbed particle is longer than the distance between the particles. Hence, for a given system, surfactants must be carefully selected. If the surfactant chain molecules are shorter than the size of the particles in the system, steric repulsion will not occur.

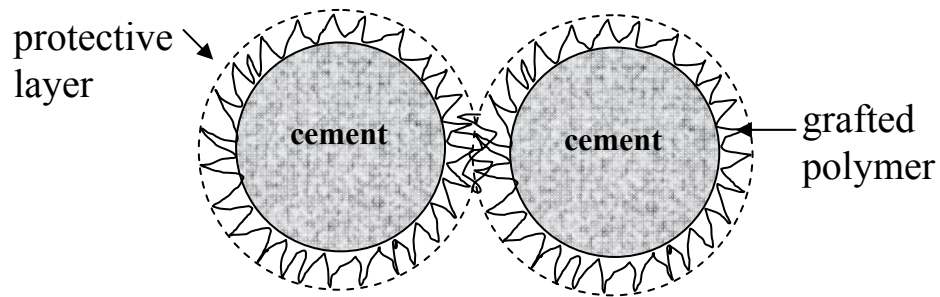


Figure 3-13. Schematic of steric repulsion mechanism

Advances in chemical engineering and concrete/cement rheology have led to the development of new admixtures with better and longer dispersing ability, and the use of polycarboxylate-based superplasticizers have been important in the development of SCC. These superplasticizers use negatively charged polymers to adsorb at the interface. It is generally accepted that steric repulsion from the bulky protective layer of the adsorbed polymers on the cement particles prevents the cement particles from agglomerating (see Figure 3-13). In 1997, Uchikawa et al. [88] experimentally proved for the first time the existence of a steric repulsive force in the dispersion of cement pastes with organic admixtures.

3.5.3 DLVO Theory and Stability

Developed in the 1940's by Derjaguin, Landau, Verwey, and Overbeek, the DLVO theory is the general theory that describes the stability of colloidal suspensions. This theory is based on the linear addition of the electrostatic potential and van der Waals forces potential to determine the total force (net interaction energy) acting on the system.

$$V_T = V_A + V_R$$

Equation 3-19

Substituting in 3-20 and 3-21 yields

$$V_T = -\frac{A}{6} \left[\frac{2a^2}{R^2 - 4a^2} + \frac{2a^2}{R^2} + \ln \frac{R^2 - 4a^2}{R^2} \right] + 2\pi\epsilon\epsilon_0 a \phi_0^2 \exp(-\kappa H) \quad \text{Equation 3-22}$$

Figure 3-14 shows a schematic of the V_T as a function of the separation between the particles. The point of maximum repulsive energy is called the energy barrier (shown as ψ_{\max} in Figure 3-15), and the height of this barrier indicates the stability of the dispersion. A higher height indicates a more stable dispersion. In order for two particles to flocculate, there must be sufficient kinetic energy (due to the mass and velocity of the particle) to get over the energy barrier. If the kinetic energy is sufficient to overcome the energy barrier, then the net interaction is attractive and the particles will flocculate. This inner region is referred to as an energy trap (shown as ψ_{\min} in Figure 3-15) since the particles can be considered to be trapped together by van der Waals forces [77].

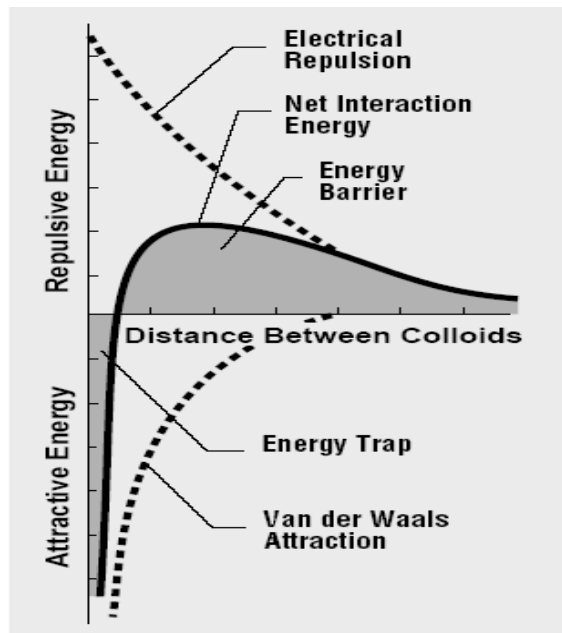


Figure 3-14. Total interaction potential curve for a stable dispersion [77]

Flocculation will occur when the particles move into an attractive minimum, and more than one minimum is possible (shown as ψ_{\min} and ψ_{\sec} in Figure 3-15). The depth of the attractive minimum determines the nature of the flocculation. The larger the depth of the attractive minimum, the more strongly the particles will stick together. A dispersion is considered stable when the maximum value of the total interaction curve is large compared to the kinetic energy [75]. Thus, the magnitude of the repulsive potential must be greater than the magnitude of the attractive potential to prevent flocculation.

It was shown in Section 3.5.2.1, that $V_r = \text{fn}(\text{ionic concentration})$, and, if the zeta potential and particle size is held constant while varying the ionic strength, then the energy barrier (and hence the flocculation mechanisms) within the dispersion can be altered. A stable dispersion (represented by Curve A in Figure 3-15) is obtained at low ionic strength and high surface potential. Similar to the curve shown in Figure 3-14, the repulsive potential is greater than the attractive potential at all separation distances. A flocculated suspension (Curve B in Figure 3-15) is obtained at moderate ionic concentrations when the two particles are located in the secondary minimum. Particles in this region may be strongly flocculated or weakly flocculated depending on the magnitude of ψ_{\sec} . When $-\psi_{\sec}/k_B T \approx 1$, the state of flocculation is reversible [80]. When the ionic concentration increases to a critical concentration, the energy barrier disappears and the system becomes a coagulated suspension (Curve C in Figure 3-15). In this state, the particles are irreversibly flocculated at the primary minimum.

In a dilute dispersion, it is sufficient to consider interactions between only two particles at a time; however, for a concentrated suspension like cement paste, multi-particle (more than two) interactions should be considered. Furthermore, the addition of superplasticizers may also result in electrosteric repulsion forces between the particles. The zeta potential is a measure of the

strength of the electrical force that exists between particles in a fluid. However, it does not capture the contribution due to steric repulsion mechanisms. For highly plasticized cement pastes (especially those used in SCC) an additional term should be added to the DLVO theory in order to describe the contribution of the steric repulsive mechanism to the repulsive force. With the inclusion of the steric forces, the total interaction force can be expressed as [89]

$$V_T = -\frac{A}{6} \left[\frac{2a^2}{R^2 - 4a^2} + \frac{2a^2}{R^2} + \ln \frac{R^2 - 4a^2}{R^2} \right] + \frac{(2\pi)^{5/2}}{27} \langle r^2 \rangle^{3/2} v^2 (1 - \alpha) a M k_B T \quad \text{Equation 3-23}$$

where v is the number of adsorbing chains per unit area, α is the surface potential, $\langle r^2 \rangle^{1/2}$ is the mean distance between both ends of the chain in the solvent, and M is the coefficient concerning the distance H .

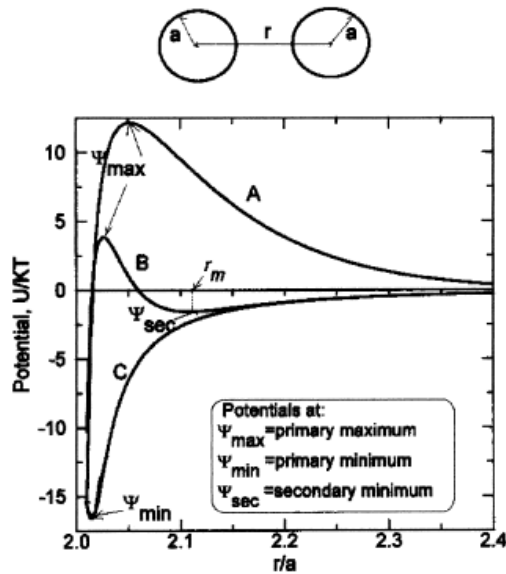


Figure 3-15. Effect of ionic concentration on total interaction potential curves (from [80])

PART III: Evaluation of Structural Rebuilding

4 CHAPTER 4: Development of a Rheological Protocol to Measure Structural Rebuilding

4.1 Introduction

The main hypothesis of this dissertation is that the structural changes that occur within and during the induction period (i.e. prior to initial setting) are primarily governed by the paste matrix since the aggregates can be thought of as inert materials. Rheological tests can provide a useful way to characterize the structure of a cement paste prior to initial set. This chapter presents a newly developed protocol for measuring structural rebuilding.

4.2 Experimental Program

4.2.1 Materials and Mix Composition

All of the paste mixtures were prepared using Type I cement and tap water. In select mixtures, silica fume and fly ash were used to replace portions of the cement. The chemical compositions of the cement, fly ash, and silica fume are listed in Table 4-1. A polycarboxylate superplasticizer (specific gravity = 1.04) was employed in all of the mixtures.

Table 4-1. Chemical composition of cement, fly ash, and silica fume

	Cement	Fly ash	Silica fume
SiO ₂ (%)	20.1	47.5	95.5
Al ₂ O ₃ (%)	4.8	26.4	0.41
Fe ₂ O ₂ (%)	2.7	12.2	0.28
CaO (%)	63.8	5.4	0.49
MgO	2.5	0.9	0.24
SO ₃	2.5	1.1	-
LOI	1.42		2.04
Fineness (+325 mesh) (%)	95	16	-
Oversize 45 μ m (0.0017 in.) (%)	-	-	0.42
Specific gravity	3.1	2.3	2.07

4.2.2 Sample Preparation

A Hobart mixer was used to mix all samples. Prior to mixing, a portion of the water was used to dilute the superplasticizer. In addition, another portion of the water was set aside to rinse the cylinder containing the diluted superplasticizer. The remaining water (total water minus rinse water and water used for dilution) was added to the powder materials and mixed for 3 minutes at a low speed (136 rpm). Then the mixer was turned off and within one minute the sides of the bowl were scraped with a spatula and the superplasticizer was added. Then, the paste was mixed for 15 seconds at low speed. Finally, the paste was mixed for an additional 2 minutes and 45 seconds at high speed (281 rpm). Hence, the total mixing time for the cement paste was 7 minutes. After mixing, a portion of the paste was placed into a HAAKE-RS150 rheometer for rheological testing and the remainder of the paste was used for the paste flow tests. Paste flow tests were conducted 1 minute after the sample preparation was completed (hence 8 minutes after the induction of water) using a glass plate and a mini-slump cone (c.f. Figure 4-1). The

dimensions of the cone were 70 mm (2.76 in.), 100 mm (3.94 in.), and 50 mm (1.97 in.), for the upper inner diameter, lower inner diameter, and height. After pouring the sample into the cone, the cone was vertically lifted to allow the paste to flow freely. The diameters in two right angle directions were measured one minute after lifting the cone. The slump flow diameter (sfd) was determined by taking the average of the two measured diameters.

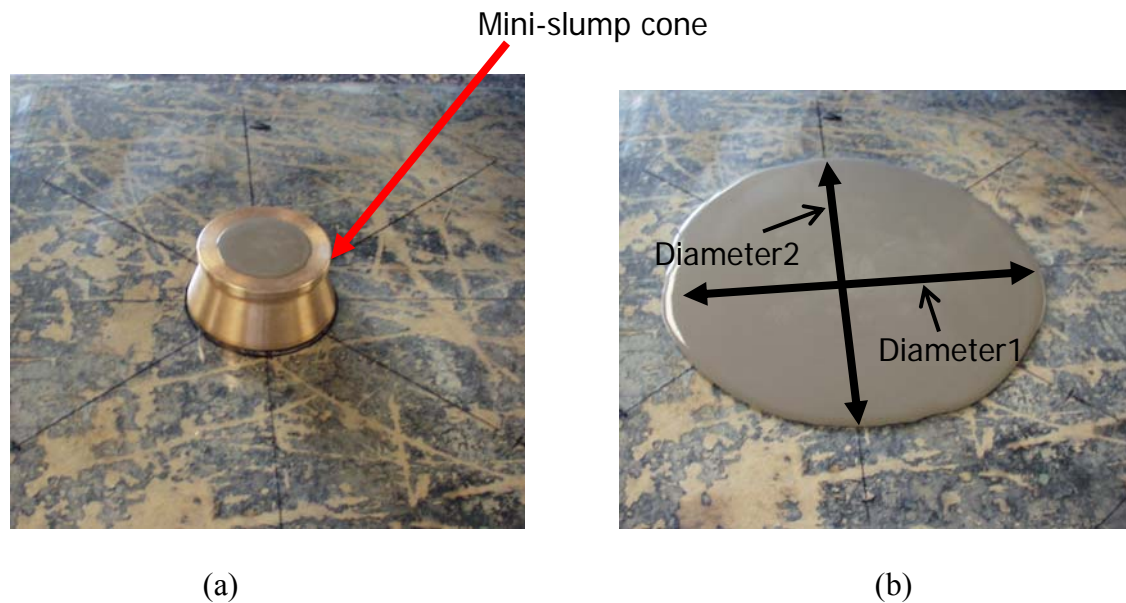


Figure 4-1. Paste slump flow diameter test (a) prior to lifting mini-slump cone; (b) slump flow diameter: $sfd = (\text{diameter 1} + \text{diameter 2})/2$

4.2.3 Rheometer

A HAAKE Rheostress150 rheometer with a concentric cylinder geometry conforming to the German DIN 53019 sensor system was used. The radius of the outer cylinder (R_a), radius of the inner cylinder (R_i), and the length of the inner cylinder (L) are 10.85 mm, 10.00 mm, and 30.00 mm, respectively (c.f. Figure 4-2). For time-dependent fluids, it is desirable for the annular gap

to be very small in order to minimize the strain variation and difference in shear history for the materials near the inner and outer cylinder. The shear rate ($\dot{\gamma}$) is estimated to be proportional to

the angular velocity of the inner rotating cylinder (Ω) and a geometry factor (M), $\dot{\gamma} = \Omega \cdot M$

where $M = \frac{R_i^2 + R_o^2}{R_o^2 - R_i^2}$ for the concentric cylinder geometry. The angular velocity is measured

in units of reciprocal seconds and is determined from the $\Omega = 2 \cdot \pi \cdot n / 60$ where n represents the speed of the rotor (measured in reciprocal minutes) [90]. All tests were conducted using a temperature control unit (maintained at 23°C) to limit strain variations due to sample heating.

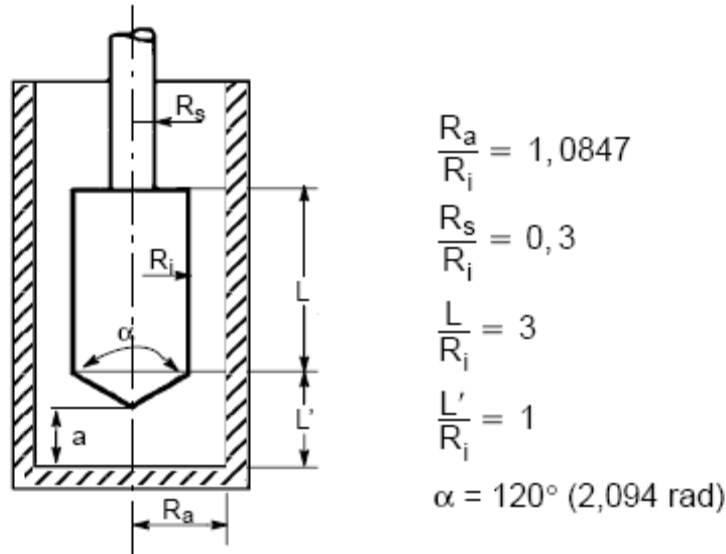


Figure 4-2. Concentric cylinder rheometer system conforming to German DIN 53019 sensor system with standardized dimensions, where R_a = radius of outer cylinder, R_i = radius of inner cylinder, R_s = radius of the rotor shaft, a = distance from bottom of outer cylinder to apex of cone, α = angle of the cone, L = length of inner cylinder, L' = distance of the rotor [90]

4.3 Preliminary attempts

Different rheological techniques that could be used to measure the structural rebuilding were considered. One of the key considerations was that the method provided a relatively fast way to measure the rebuilding and that it was not too labor-intensive.

4.3.1 Dynamic rheology

The application of dynamic tests using small-amplitude oscillatory shearing to measure the viscoelastic properties was investigated. When the strain amplitude in an oscillatory shear flow is very small, the stress curve as a function of time can be described by a single trigonometric function

$$\sigma(t) = \gamma_0 [G'(\omega) \sin(\omega t) + G''(\omega) \cos(\omega t)]$$

where γ_0 is the strain amplitude imposed on the fluid, ω is the frequency of oscillation, $G'(\omega)$ is the storage (elastic) shear modulus, and $G''(\omega)$ is the loss (viscous) shear modulus. The storage shear modulus represents the elastic behavior or the energy storage of the material, and the loss shear modulus represents the viscous behavior or energy dissipation of the material. The complex shear modulus, G^* , is a measure of the material's resistance to deformation and defined by $G^* = G' + iG''$ where i is the imaginary unit. By limiting the strain to small amplitudes, the particles stay in close contact with one another and are able to recover elastically so that the microstructure is not disturbed and the paste behaves as a solid [91]. Thus, an important aspect to conducting this type of test is determining the linear viscoelastic region (LVER) in which the moduli are independent of the strain amplitude. An additional parameter for the LVER is that

the frequency of the applied oscillation provides sufficient time for the sample to relax so that there is no build up of residual energy. A practical way to determine the critical shear strain for the LVER is to conduct a strain sweep for a constant frequency. Ideally, there should exist a low shear plateau zone in which the moduli are constant. This plateau is seen when the moduli are plotted against the applied strain in a log-log scale. The critical shear strain is determined at the point in which the low shear plateau ends. In order to determine whether oscillatory shear rheology can be used to evaluate the structural rate of rebuilding in SCC pastes, the LVER should be first established.

A strain sweep was performed on 2 different plasticized cement pastes. Each paste had a w/b ratio of 0.35, but the superplasticizer dosage was varied in order to achieve a slump flow diameter which would be more representative of paste for SCC (Table 4-2). The rheometer was specified to sweep the shear strain from 3.87×10^{-5} (which is the lowest capacity of the machine) to 1.00×10^2 , and the frequency was set at a level of 1 Hz. Prior to the start of the strain sweep each sample was sheared at 600s^{-1} for eight minutes and then allowed to rest for 15 minutes. The rest time was given so that the internal structure would have time to rebuild.

Table 4-2. Slump flow diameter for pastes used in oscillatory rheology studies

Paste	Slump flow diameter
A	10.5 inch (267 mm)
B	12 inch (305 mm)

Figure 4-3 and Figure 4-4 show the variation of G'' and G' with strain amplitude. It can be seen that regardless of the slump flow diameter, the low shear strain plateau is not well defined and that the critical shear strain is very small. Small critical strains are associated with strong interparticle forces [92]. A critical shear strain of approximately 1.2×10^{-4} was seen in mix A. This result is in agreement with data from Sun for a non-plasticized cement paste with a w/c ratio of 0.6 [93]. When the slump flow diameter was increased to 12 inches the critical shear strain decreased to the point that the observance of the low shear strain plateau (and thus LVER) ceased to exist. This indicates that Paste B has a much shorter LVER than Paste A and, therefore, would be expected to break down much more easily with small movements or agitations. This is expected due to the higher dosage of superplasticizer, which decreases the elasticity and stress resistance of the sample.

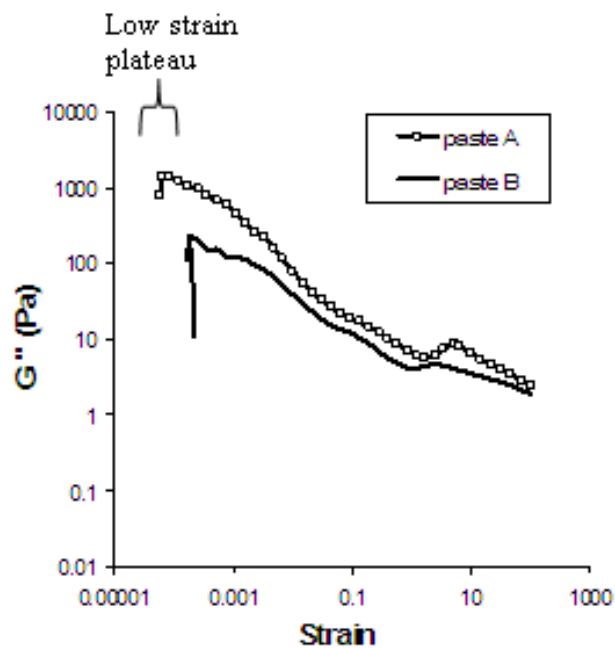


Figure 4-3. Dependence on loss modulus on strain ($f = 1$ Hz)

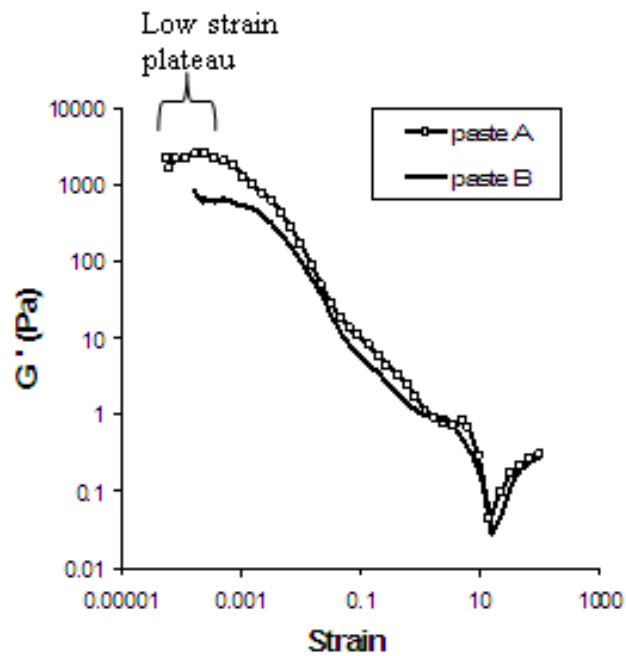


Figure 4-4. Dependence on elastic modulus on strain ($f = 1$ Hz)

As seen in Figure 4-5, the addition of superplasticizer caused a reduction in the rheological parameters (compare G' and G'' of Paste A and Paste B). This reduction is indicative of microstructural changes occurring due to the interaction of the polymer chains of the superplasticizer with the cement particles. Difficulty in determining the extremely small critical strain values makes it hard to ensure that the sample is tested within the LVER. For this reason, alternative ways of investigating the structural build-up were investigated.

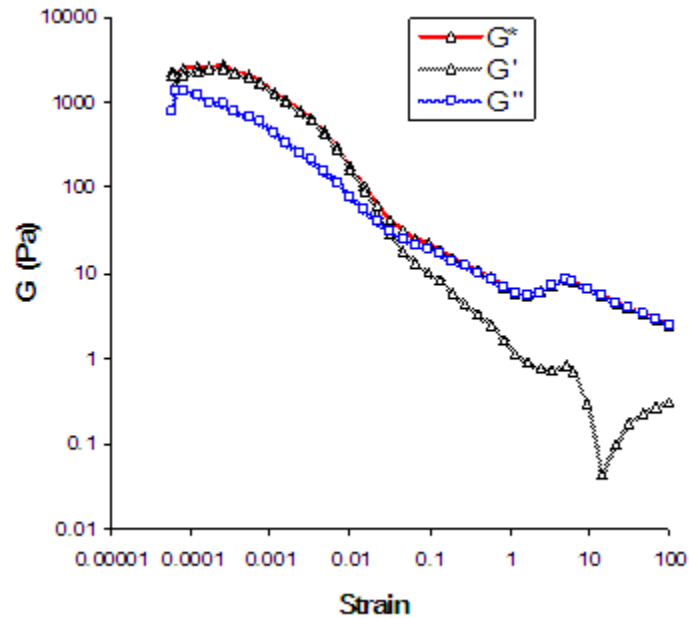


Figure 4-5. Strain sweep and viscoelastic behavior of cement paste

4.3.2 Shear Rheology

To accurately evaluate and compare the rate of rebuilding of different materials, it is important that the specimen starts from a well-defined initial condition. After placing the cement paste in the rheometer, the sample was immediately sheared at 600 s^{-1} (c.f. Figure 4-6). The time needed to establish the equilibrium shear stress (τ_{eq}) depends on the mix composition, but it was found that shearing the mixes at 600 s^{-1} for 8 minutes was enough to establish an initial equilibrium condition in which a steady-state shear stress was achieved.

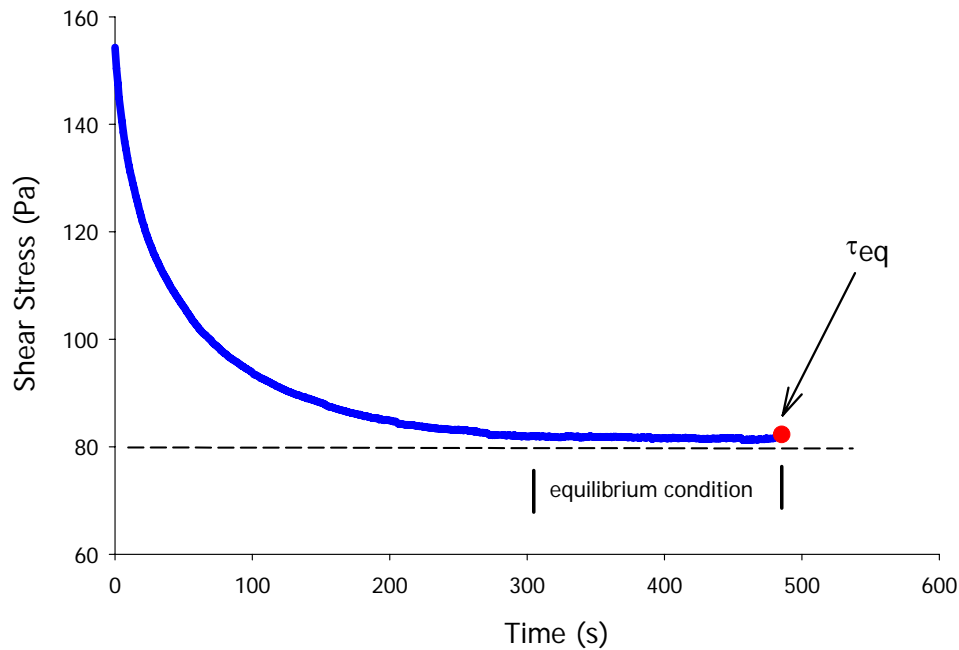


Figure 4-6. Establishment of equilibrium condition (shear rate = 600 s^{-1})

The shear rate was then ramped down from 600 s^{-1} to 0 s^{-1} over a 30 second interval. This process establishes the *equilibrium loop*—that is, the hysteresis loop used to establish the equilibrium downline. This is so named because if immediately after ramping the shear rate from 600 s^{-1} to 0 s^{-1} the shear rate is then ramped from 0 s^{-1} to 600 s^{-1} , the resulting line will be perfectly superimposed on the descending branch of the equilibrium loop. Thus, the descending portion of the equilibrium loop is called the equilibrium line (c.f. Figure 4-7). It was shown that for a given shear rate the stresses in the equilibrium line correspond to τ_{eq} for a given shear rate [51].

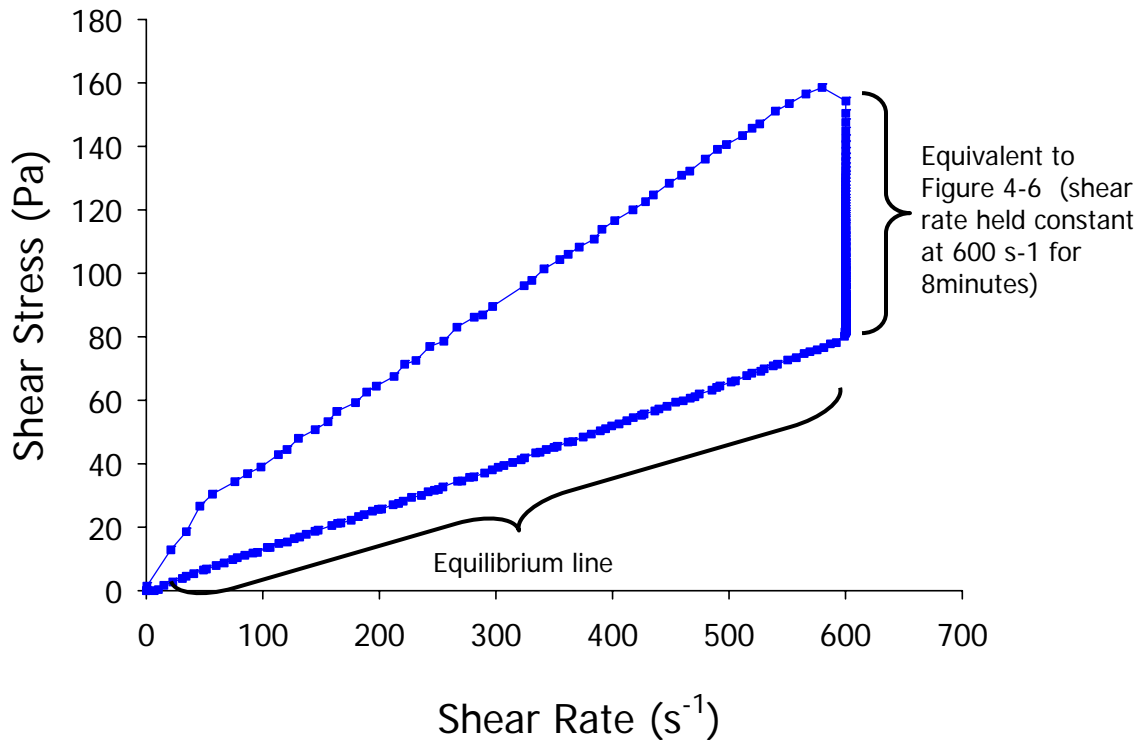


Figure 4-7. Equilibrium loop

4.3.3 Evaluation of Rebuilding

Once the equilibrium state is established, the rate of rebuilding can be investigated by testing the specimen after different resting times (t_{rest}). Initially, the rebuilding of the specimens was evaluated by performing an equilibrium loop and allowing the sample to rest for 1 minute before applying a hysteresis loop from 0 s^{-1} to 300 s^{-1} at a ramp rate of 20 s^{-2} . Immediately after the hysteresis loop, a new equilibrium condition was achieved by applying the equilibrium loop. Then the sample was allowed to rest for 2 minutes before another hysteresis loop was applied. This process (equilibrium loop, rest, hysteresis loop) was repeated with the new rest period being

twice the amount of the previous rest period. Thus, the resting periods were 1, 2, 4, 8, and 16 minutes.

For a given resting period, the area on the rheogram plane between the up curve of each hysteresis loop and the corresponding equilibrium line was used to evaluate the rebuilding that occurred in the specimen. This area has the physical dimension of energy per unit time per unit of volume (or power per unit volume) and is called the *specific rebuilding energy (SRE)*. A plot of the SRE versus its corresponding t_{rest} is used to determine the rate of rebuilding by looking at the slope of the line in this plane. For a given specimen, the slope of this line indicates the degree of thixotropy. For various mixes, the slope of the line in the SRE- t_{rest} plane can be used to compare the thixotropic potential of different mixes.

Conceptually, the protocol described above can be easily applied to thixotropic materials that do not experience changes due to additional chemical or physical processes (ex. ketchup, crude oil, grease). But, in the case of cement and concrete, there are other mechanisms influencing material behavior. In particular, cementitious materials also undergo a hydration reaction, which makes the amount of structural rebuilding also a function of the age of the specimen. Therefore, an amendment in the original protocol was made in light of the following observations:

- 1) The achievement of a τ_{eq} in a specimen became gradually more difficult as more energy (and hence time) was required to rupture bonds. As a result, the usage of the equilibrium loop described in Section 4.3.2 was not effective. A screening would have to be conducted for every t_{rest} of each specimen in order to determine the time needed to achieve τ_{eq} . This makes it difficult to use the same protocol for all the specimens because the time needed to establish each new τ_{eq} may vary significantly among specimens.

2) It is practically impossible to separate the effect of aging and t_{rest} . In other words, can the rate of rebuilding be different if the same resting time is used at different ages of the specimen?

With the above protocol, the effects of thixotropy and structural rebuilding associated with hydration are coupled. Furthermore, the maximum resting time investigated was only 16 minutes, but the age of the specimen was already 71 minutes (and that is only if the time needed to reach τ_{eq} for each t_{rest} was 8 minutes). Since it is important to conduct the test within the induction period, it is obvious that the amount of data that can be obtained using this approach is limited.

4.4 Final protocol

Based on the results of the preliminary tests, the final procedure used to evaluate thixotropy is described below:

- 1) Immediately after mixing and the paste flow test, the sample was poured in the rheometer. Rheological testing was initiated when the age of the specimen was 12 minutes (counted from the introduction of water) by using the equilibrium loop protocol as specified in Section 4.3.3.

- 2) The specimen was allowed to rest in the rheometer for 10 minutes, and then a hysteresis loop was performed. The protocol for the hysteresis loop was as follows: during a 30 second time interval, the shear rate was ramped up from 0 s^{-1} to 300 s^{-1} ; then, the shear rate was ramped down from 300 s^{-1} to 0 s^{-1} over a 30 second interval. This procedure was repeated 8 more times (rest for 10 minutes, hysteresis loop, rest for 10 minutes, hysteresis loop, etc.). Thus, the hysteresis loop applied after the second rest period was conducted 20 minutes after the equilibrium loop;

similarly, the hysteresis loop applied after the ninth rest period was conducted 90 minutes

after the equilibrium loop. As expected, an increase in shear stress can be seen with t_{rest} (c.f.

Figure 4-8).

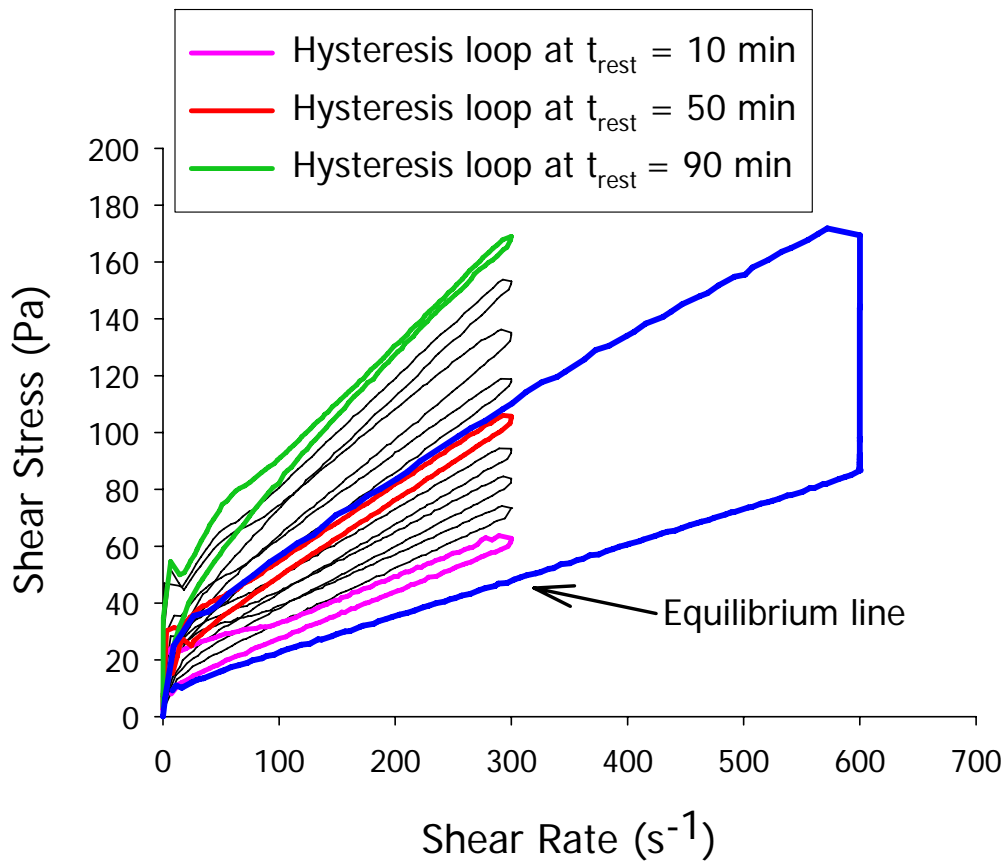


Figure 4-8. Hysteresis loops for each t_{rest} (with select t_{rest} emphasized at 10 minutes, 50 minutes, and 90 minutes).

3) The area within 100 s^{-1} and 200 s^{-1} of the up curve of each hysteretic loop and the initial equilibrium line (in the plane of shear stress versus shear rate) was calculated and used as the SRE. Figure 4-9 shows the SRE for a specimen with a corresponding t_{rest} of 90 minutes. The

shear rate interval of $100 - 200 \text{ s}^{-1}$ was chosen in order to reduce errors caused by the initial elastic response of the sample.

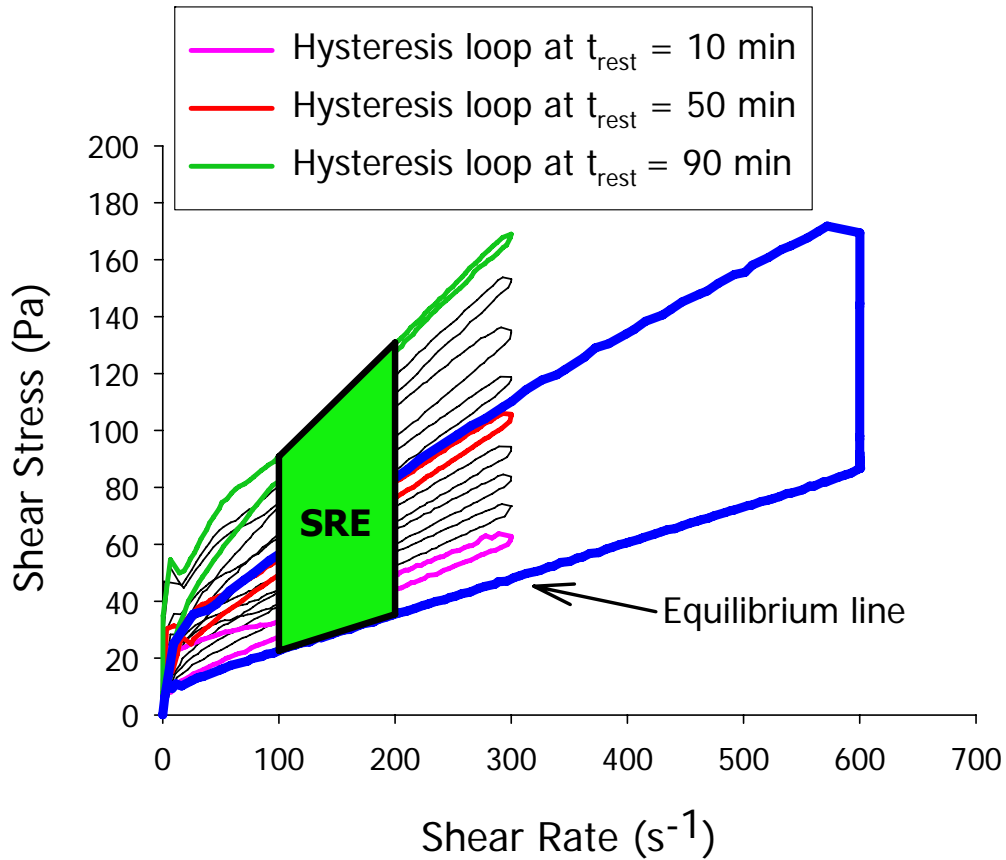


Figure 4-9. Area between up curve of hysteresis loop and equilibrium line is used as a measure of the rate of rebuilding and denoted as SRE (Specific Rebuilding Energy). The figure shows SRE for $t_{\text{rest}} = 90$ minutes.

4) SRE versus t_{rest} is plotted and the slope of the line in this plane describes the rate of rebuilding. This slope is used as a measure of the rate of rebuilding and provides a way to compare different mixtures. In Figure 4-10, material 1 has a faster rate of rebuilding than materials 2, 3, and 4 because its slope is the steepest.

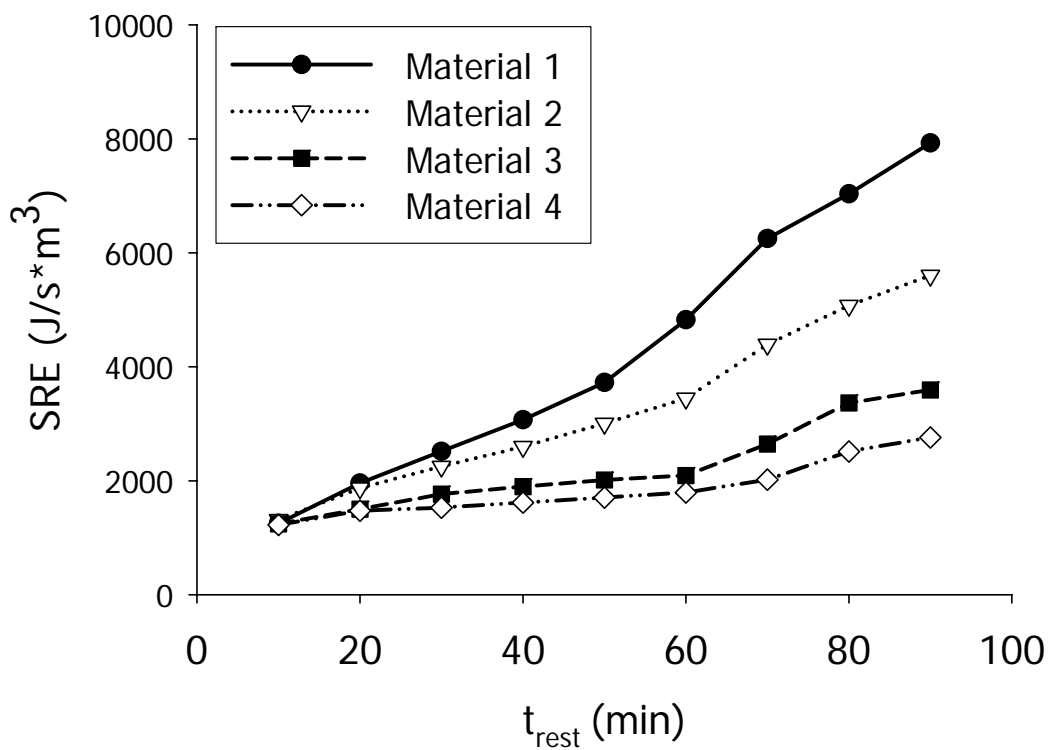


Figure 4-10. Application of protocol to compare rate of structural rebuilding of different mixtures. Slope of line indicates degree of rebuilding.

The approach described above gives an indication of the initial degree of structural rebuilding and the way it varied with time. In order to assess the total amount of rebuilding that

occurred in a specimen hysteresis loops would have to be repeated again and again until the up curve and down curve completely overlap. The addition of area between each hysteresis loop would represent the total amount of energy needed to bring the structure to its minimum degree of organization. However, this would require a new specimen for each t_{rest} and an extensive amount of time.

4.4.1 Effect of Fluidity

Initial studies showed that the rate of structural rebuilding was strongly related to the initial fluidity of the specimen [94]. When the initial slump flow of the sample was increased, the rate of rebuilding decreased. In order to better understand how structural rebuilding is influenced by different mixture ingredients, comparisons among specimens were only made for mixes that had the same initial fluidity level (as determined by the sfd with the slump flow test). The target fluidity level can be obtained by two methods: varying the superplasticizer dosage or varying the water-to-cement ratio. It was determined that the adoption of a lower w/c ratio with a greater dosage of superplasticizer was more effective in increasing structural rebuilding thixotropy than using a higher w/c ratio and a lower superplasticizer dosage [94]. Increasing the water content increases the thickness of the water layer around the particles. Therefore, the inter-particle distance increases and particle flocculation is reduced because the inter-particle coagulation forces decrease.

4.4.2 Effect of Shear History

An important aspect that should be considered is the effect of the previous hysteresis loops and the SRE. In other words, it is important to know if the hysteresis loop conducted after the second

resting period will significantly affect the hysteresis loop that is conducted after the third resting period and so on. The effect of shear history will depend on the amount of time required by the cement paste to recover its structure (i.e. rebuild its bonds) after the cessation of shear. If the cement paste is subjected to a hysteresis loop before it fully recovers from the previous hysteresis loops, then the area between the up curve of the hysteresis loop and the equilibrium line will be smaller. In addition, the shear stress values will be lower.

Figure 4-11 shows the results from experiments that were conducted in order to investigate the effect of shear history. In order to do this, experiments were performed using a new sample for each t_{rest} . In these tests, the equilibrium loop was performed as usual, and then the sample was allowed to rest for the required amount of time (ex. 10 minutes, 20 minutes, etc.) without any agitation until the hysteresis loop was performed. As shown in the figure, during the early stages of the testing, there was little variation between the results of the tests in which a new sample was used for each t_{rest} versus the one in which the same sample was used for each t_{rest} . However, at later ages, the results from the test using the same sample at each may deviate from the results in which a new sample was used for each test. This indicates that the rate of rebuilding may be underestimated due to the effects of shear history. In effect, the protocol is also giving some indication of the sensitivity of the structural network to shear and accounting for the influence of disturbances on the rate of rebuilding (which may actually be more representative of the process that concrete would experience in the field).

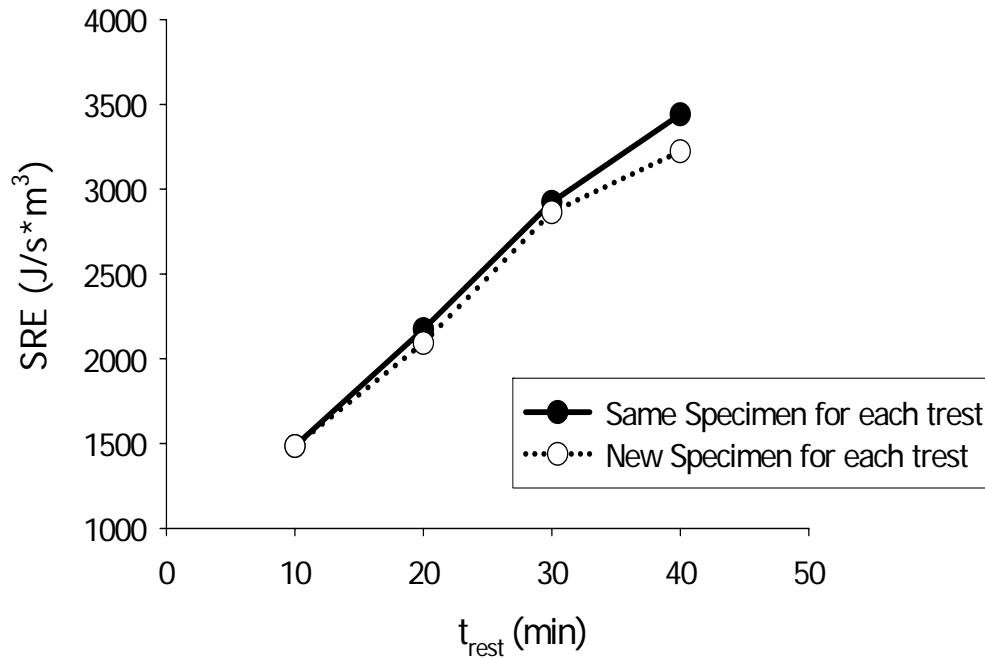


Figure 4-11. Effect of shear history on structural rebuilding measurements ($w/c = 0.35$, SP dosage = 0.2%, initial slump flow = 280 mm)

4.4.3 Sensitivity and Repeatability

Tests were conducted using various mix compositions and initial slump flow conditions in order to evaluate the reliability of the protocol. As shown in Figure 4-12, good repeatability exists among the experiments. The developed protocol was also sensitive enough to detect changes in structural rebuilding due to the changes in the fluidity level and mix proportions. The fluidity level was evaluated using the slump flow diameter (sfd) obtained from the paste flow tests.

Three target fluidity levels were evaluated:

- low fluidity: sfd = 165 mm (6.50 in.)
- mid fluidity: sfd = 240 mm (9.45 in.)

- high fluidity: sfd = 330 mm (13.0 in.)

All mixtures were within ± 7 mm (0.28 in.) of the target fluidity level.

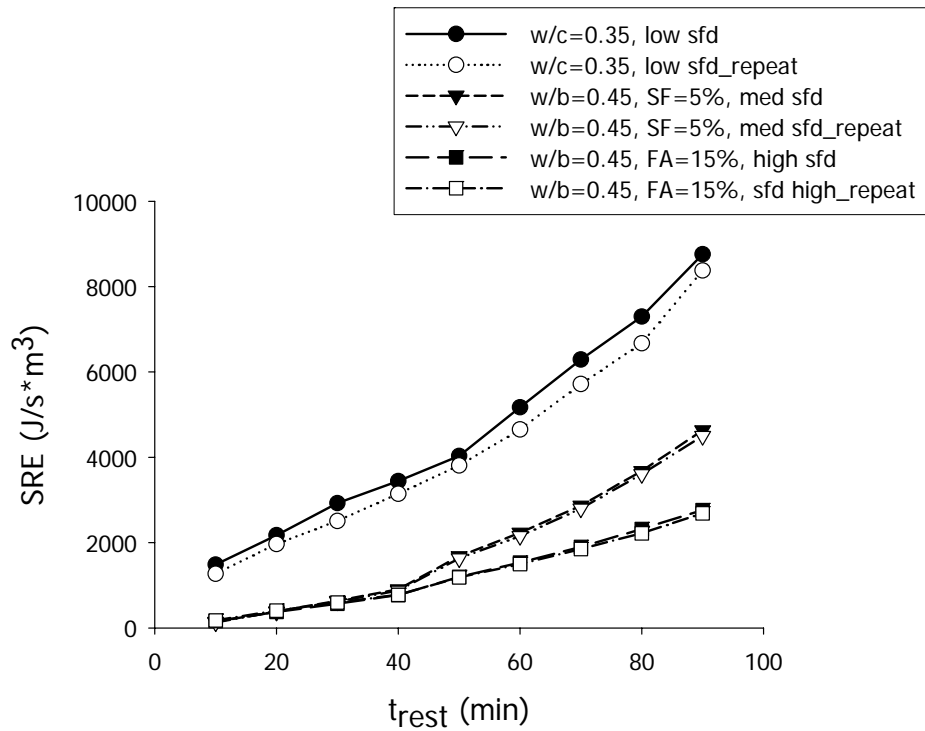


Figure 4-12. Sensitivity and repeatability for final protocol (SF=silica fume, FA=15)

4.5 Chapter Summary

The feasibility of using small amplitude oscillatory shear rheology as an evaluation technique to monitor the rate of structural rebuilding in high fluidity cement pastes was examined. A critical shear strain of approximately 1×10^{-4} was measured in a cement paste with an initial slump flow diameter of 240 mm, but when the initial slump flow diameter was increased to 330 mm the

Due to the highly fluidized nature of SCC it was difficult to establish the linear viscoelastic range, which is needed in order to accurately determine the moduli. Shear rheology can provide an alternative to measuring structural rebuilding, and a shear rheological protocol based on hysteresis loops and rebuilding energy was proposed to evaluate the rebuilding of cementitious materials. The protocol was shown to be effective in evaluating the rate of structural buildup. The rheological protocol developed in this chapter will be used throughout this research program to evaluate and compare the rate of structural rebuilding of various SCC cement paste mixtures.

PART IV: Impact of Mixture

Proportioning on Structural Rebuilding

5 Chapter 5: Screening Studies - Influence of cement composition

5.1 Introduction

The key objective in this phase of the research was to determine if structural rebuilding is influenced by the type of cement used. The rheological protocol presented in Chapter 4 was used to assess the rate of structural rebuilding.

5.2 Materials

Seven different cements were selected for this study. The chemical composition of these cements is given in Table 5-1. The chemical composition provided in Table 5-1 was based on mill analyses received from the cement manufacturer, and items denoted with * indicate that the value was calculated using Bogue Equation. The cements have a wide compositional range, varying from C_3A contents of 2 to 11% and equivalent alkali content ranging from 0.45 to 1 (%). Cements 2, 3, 5, 6, and 7 are manufactured as an ASTM Type I cement, and cements 1 and 4 are manufactured as an ASTM Type V. Each cement was tested with two different polycarboxylate superplasticizers (c.f. Table 5-2). The w/c ratio was held constant at 0.35, and the superplasticizer dosage was adjusted so that the initial slump flow diameter of each mix was 330 mm (13 in.) \pm 13 mm (0.5 in.).

Table 5-1. Chemical analysis, fineness, and initial set time for cements in screening

studies

	Cement						
Oxide Analysis	1	2	3	4	5	6	7
CaO (%)	63.2	63.0	61.9	63.9	62.0	63.7	64.2
SiO ₂ (%)	22.0	19.6	18.9	21.6	18.7	20.4	19.8
Al ₂ O ₃ (%)	3.4	4.9	5.7	3.8	5.7	4.1	4.7
Fe ₂ O ₃ (%)	4.3	2.5	2.6	3.5	2.5	2.4	2.7
SO ₃ (%)	2.1	3.0	4.4	2.3	4.3	2.3	2.6
MgO (%)	3.4	3.4	2.3	2.7	2.4	3.9	2.6
Equivalent Alkali (%)	0.45	0.7	1	0.52	0.85	0.26	0.52
Free lime (f-CaO) (%)	0.6	1.4	----	----	----	1.3	0.6
LOI (%)	0.7	2.2	1.8	1.1	1.9	1.5	2.3
Blaine specific surface (m²/kg)	373	394	368	----	388	365	382
Compound							
C ₃ S (%)	55	55	54	59	56	67	59
C ₂ S (%)	21	15	16*	18	11	14*	8*
C ₃ A (%)	2	9	11	4	11	7	8
C ₄ AF (%)	13	8	8*	11	8	7*	8*
Initial Set	274	125	120	----	120	101	274

Table 5-2. Polycarboxylate-based superplasticizers used during screening studies

Superplasticizer	Active Solids Content (%)	Specific Gravity	Application
SP1	25 - 50	1.04	Ready-mix
SP2	---	1.05	Precast

5.3 Results

Figure 5-1 shows the rate of rebuilding curves for paste mixtures made with SP1. It can be seen that the structural rebuilding rate is significantly influenced by the cement. Faster rates of structural rebuilding were seen in paste mixtures made with cement 4, 6, and 7. Slower rates of rebuilding were seen in the paste mixtures made with cement 4, 6, and 7. Slower rates of rebuilding were seen in the paste mixtures made with cement 2, 1, 3, and 5. With the exception of cement 1, the C_3A contents varied from 9 to 11% in the mixtures displaying the slower rates of rebuilding. Figure 5-2 shows the rate of rebuilding curves for paste mixtures made with SP2. Similar to SP1 results, faster rates of structural rebuilding were exhibited in the pastes prepared with cements that had lower C_3A and alkali contents.

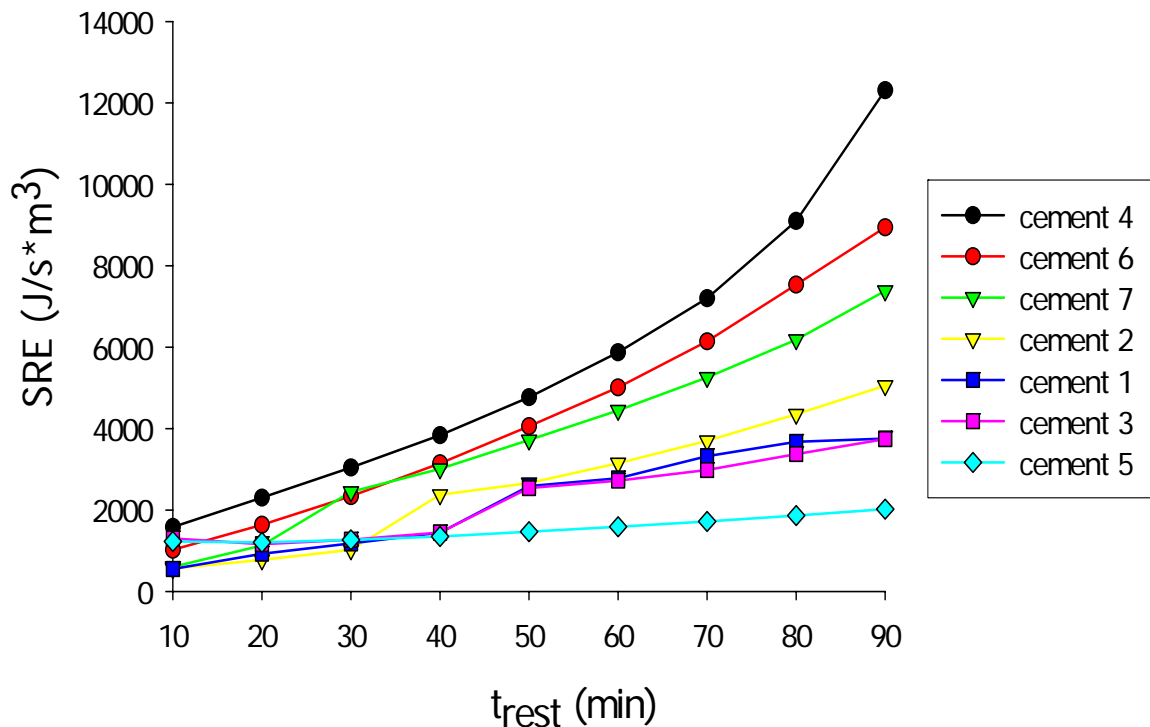


Figure 5-1. Rate of rebuilding for paste mixtures made with superplasticizer 1

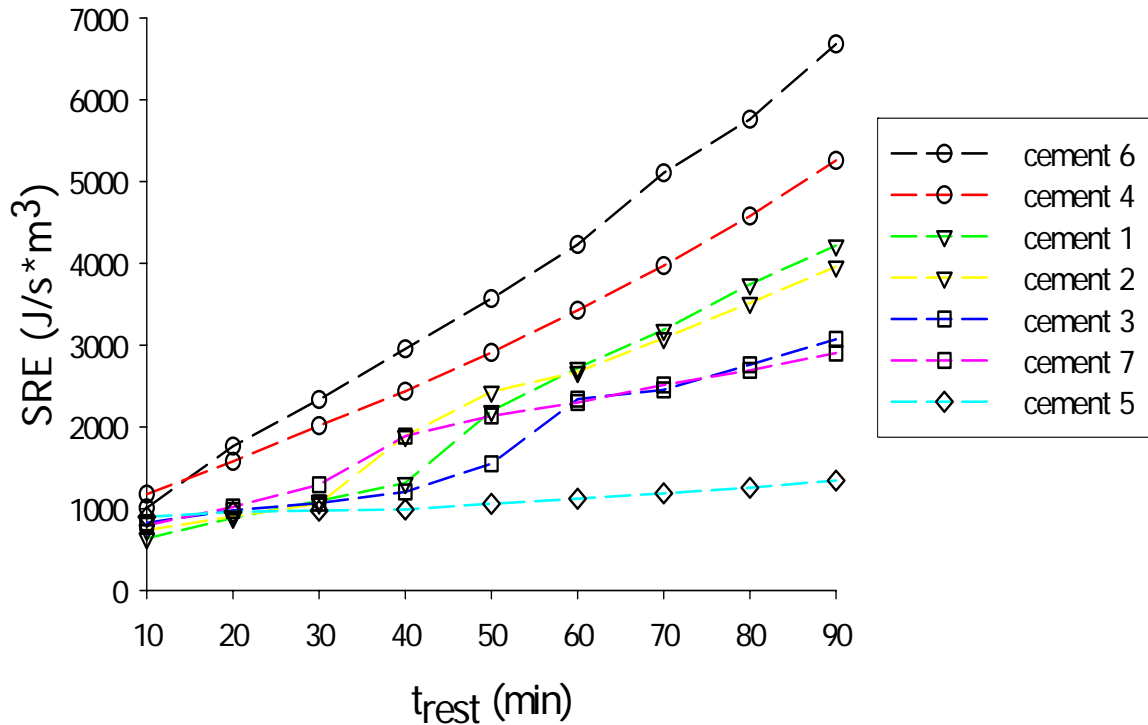


Figure 5-2. Rate of rebuilding for mixtures made with superplasticizer 2

5.4 Discussion

Based on the literature, it was believed that structural build up would occur faster in the cements that contained higher contents of C_3A and alkalis. This is due to the fact that early stiffening of Portland cement paste occurs predominantly from the hydration of C_3A [95, 96]. In addition, the alkalis from potassium and sodium are quickly dissolved into the pore solution, which tends to increase the attractive forces among the cement particles. However, as shown in Figure 5-1 and Figure 5-2, the opposite trend was seen. A correlation between the C_3A content and the rate of rebuilding appears to exist. With the exception of cement 1, the rate of structural rebuilding increased as the C_3A content decreased (c.f. Figure 5-3). In Figure 5-4, the equivalent alkali

content is plotted with respect to the rate of rebuilding, and the rate of rebuilding decreases from left to right (i.e. mix 4 had the fastest rate of rebuilding, and mix 5 had the slowest rate of rebuilding). Even if cement 1 is ignored, a positive correlation between the total equivalent alkali and structural rate of rebuilding could not be established. In terms of structural build-up, it is often cited that it is the soluble alkali content, and not necessarily the total alkali content, that is of importance [16, 97-99].

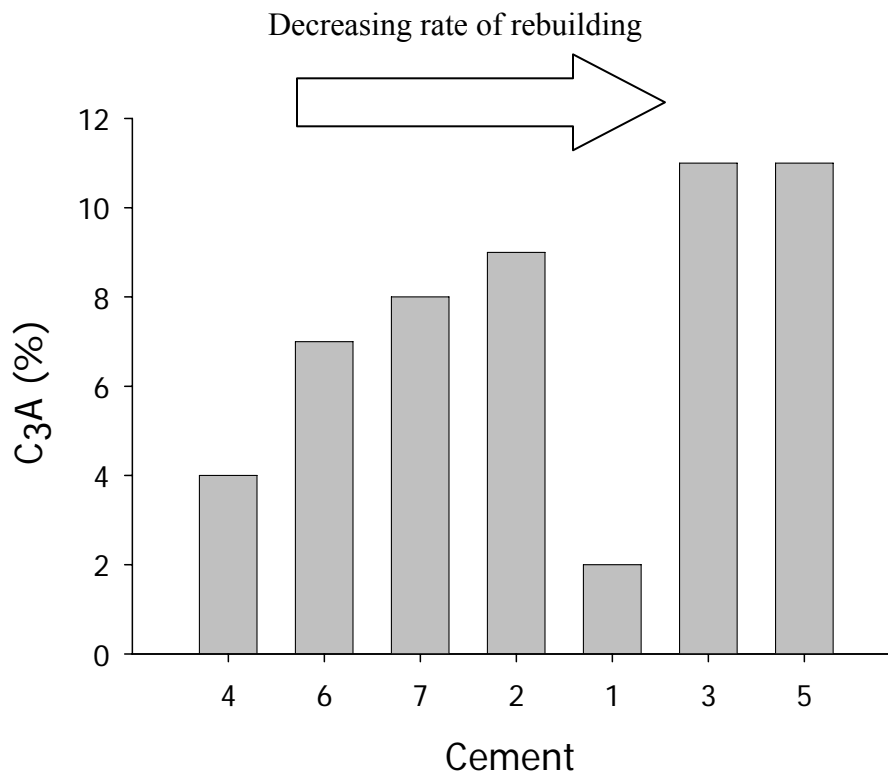


Figure 5-3. Rate of rebuilding versus C₃A content (for pastes made with SP1)

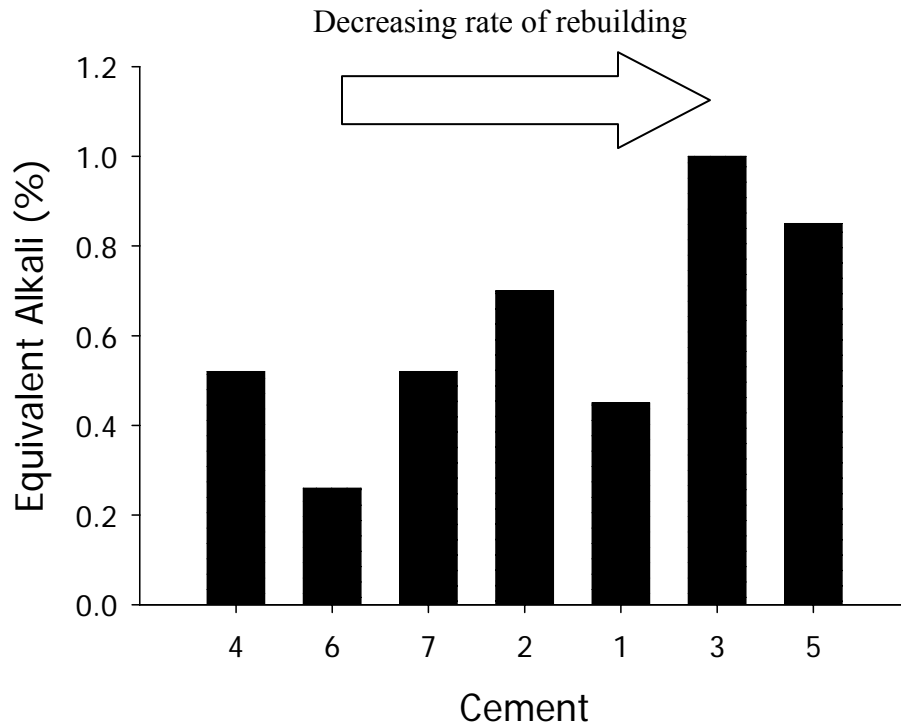


Figure 5-4. Rate of rebuilding versus equivalent alkali content (for pastes made with SP1)

Table 5-3 provides information about the superplasticizer demand required to achieve the target slump flow diameter. It is likely that there is a greater propensity for particle aggregation in cements pastes with the high C_3A and alkali contents, and, as seen in Figure 5-5 and Figure 5-6, higher superplasticizer dosages were required as the C_3A and alkali content increased. It is known that setting time of cements can be delayed when superplasticizer dosage is increased. This is due to the higher amount of superplasticizer molecules that are present to disperse the cement particles; this prevents the particles from aggregating and retards the development of the 3-dimensional network required set. The results indicate that the same effect occurred with regards to structural rebuilding.

Table 5-3. Superplasticizer dosage required to achieve initial slump flow of 330 mm

Cement	C3A (%)	Equivalent alkali (%)	SP1 dosage (% by mass of cementitious materials)	SP2 dosage (% by mass of cementitious materials)
1	2	0.45	0.20	0.32
2	9	0.7	0.30	0.43
3	11	1	0.75	0.65
4	4	0.52	0.23	0.45
5	11	0.85	0.75	0.83
6	7	0.26	0.14	0.28
7	8	0.52	0.25	0.47

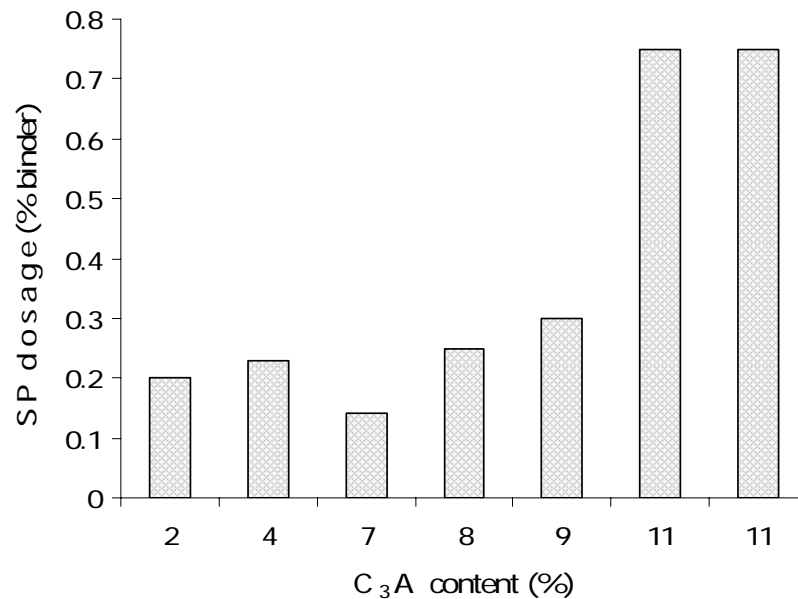


Figure 5-5. Relationship between superplasticizer dosage and C₃A contents for cement pastes evaluated during screening study. Superplasticizer dosage was varied such that all mixtures had same initial slump flow diameter.

With regard to cement 1, the slower rate of rebuilding may be due to many reasons, including, but not limited to, the superplasticizer adsorption, dispersing effect of the superplasticizer, soluble alkali content, hydration kinetics and hydrates formed, and particle morphology. Taking into account that the setting time of cement 1 is approximately twice that of the other cements, it is likely that the mechanisms influencing setting and structural rebuilding are interconnected, and the results indicate that mixtures which have longer setting times will have slower rates of rebuilding.

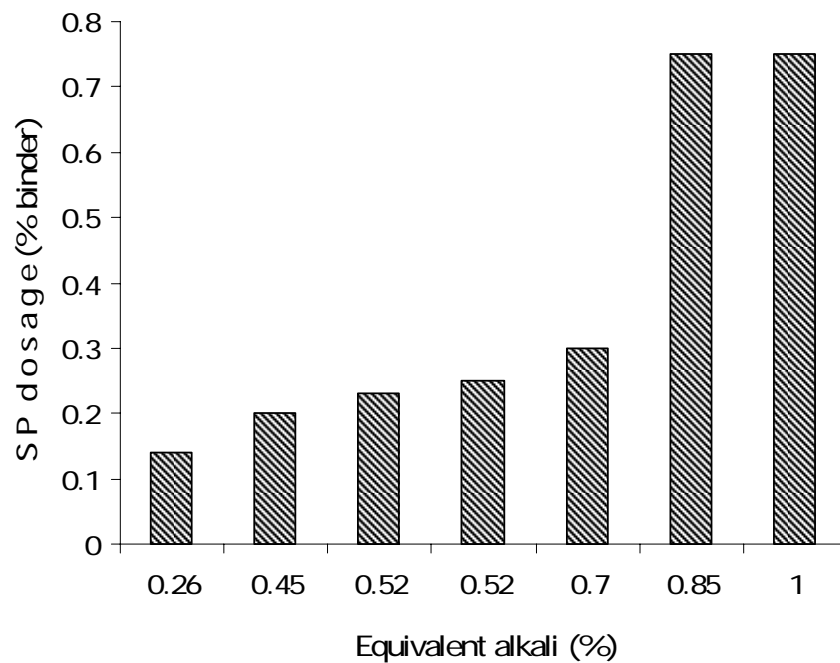


Figure 5-6. Relationship between superplasticizer dosage and equivalent alkali for cement pastes evaluated during screening study. Superplasticizer dosage was varied such that all mixtures had same initial slump flow diameter.

A comparison of the structural build-up between the two superplasticizers is shown in Figure 5-7 through Figure 5-13. In each case, the rate of rebuilding was either the same or faster when SP1 was used. Mixtures made with SP2 did not exhibit a faster rate of rebuilding in any of the cases. These results can also be used to determine the sensitivity of the cements to different superplasticizers. Cement 1 appears to be the most robust cement exhibiting the same rate of stiffening regardless of the superplasticizer used.

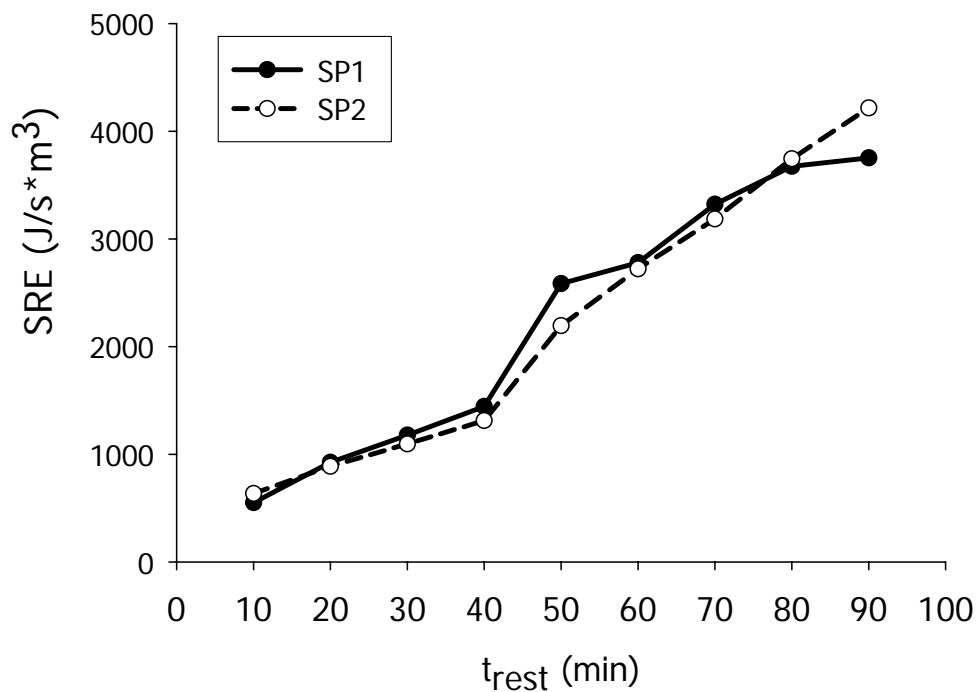


Figure 5-7. Structural rate of rebuilding for cement 1 with SP1 and SP2

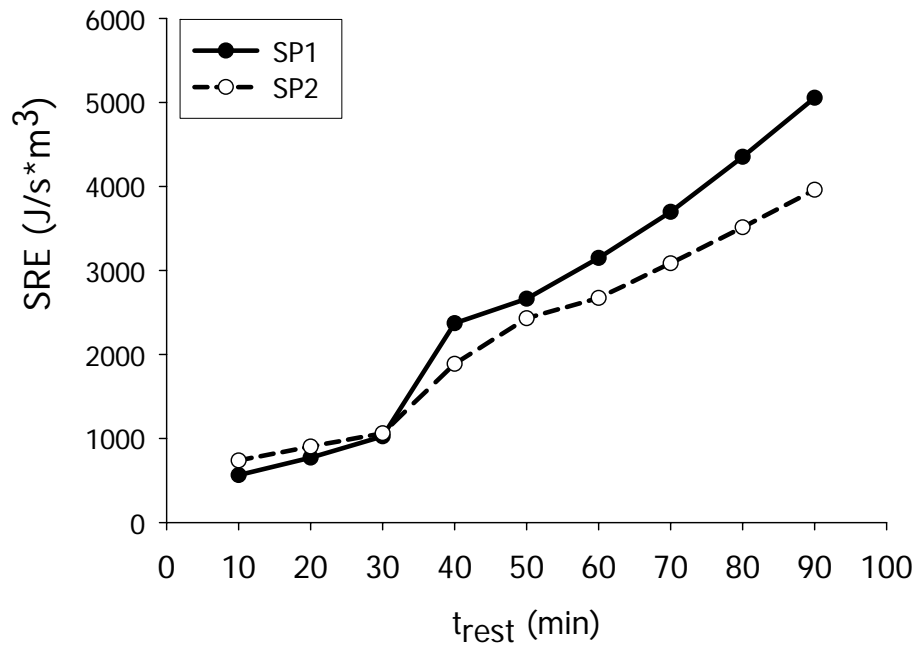


Figure 5-8. Structural rate of rebuilding for cement 2 with SP1 and SP2

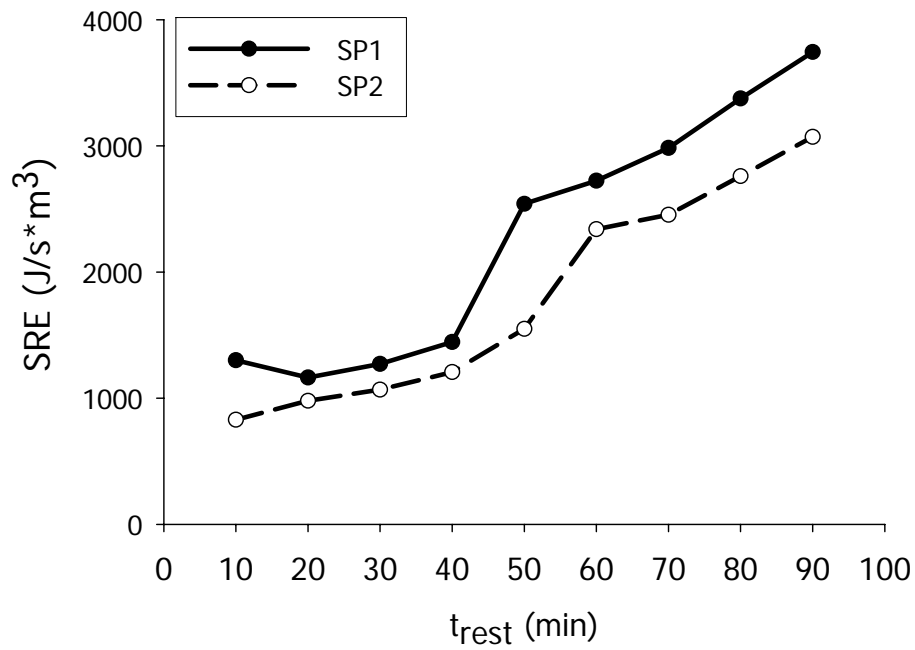


Figure 5-9. Structural rate of rebuilding for cement 3 with SP1 and SP2

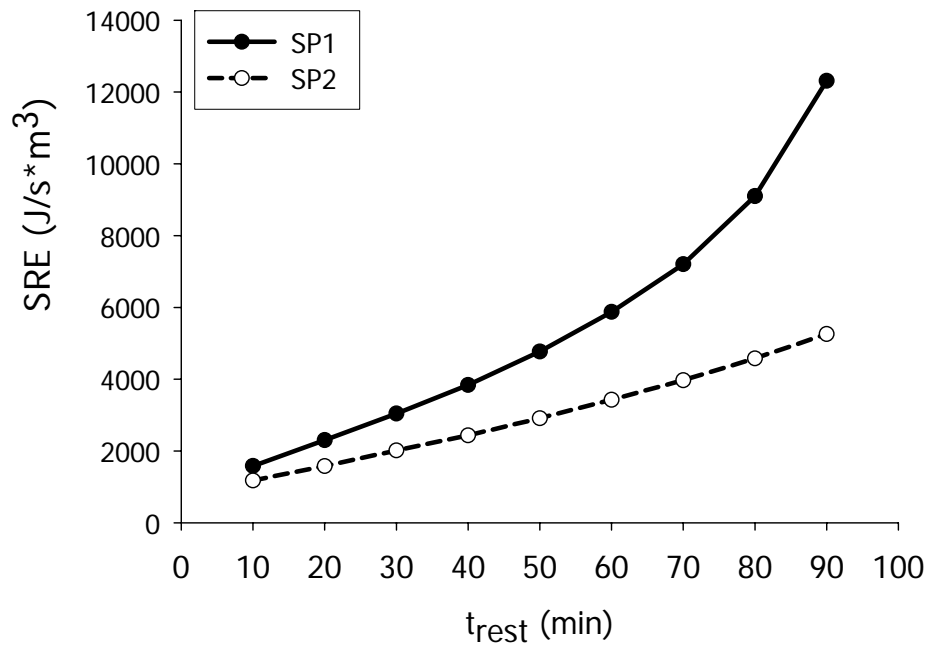


Figure 5-10. Structural rate of rebuilding for cement 4 with SP1 and SP2

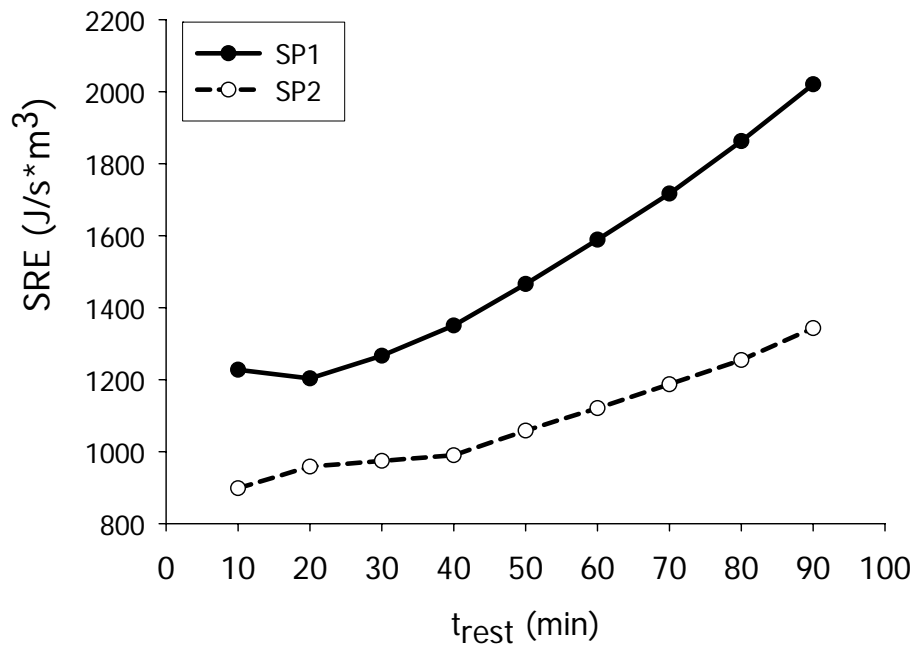


Figure 5-11. Structural rate of rebuilding for cement 5 with SP1 and SP2

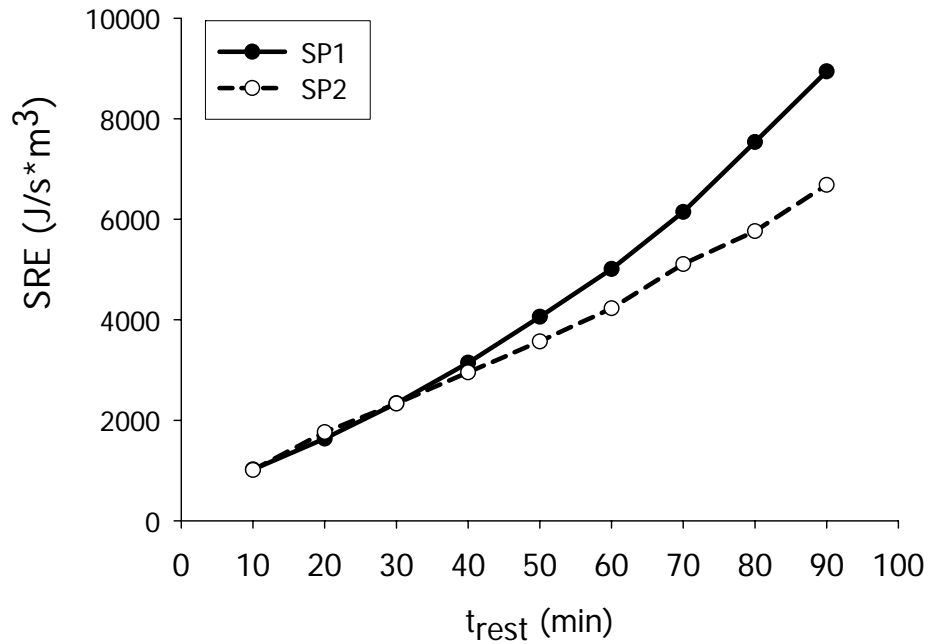


Figure 5-12. Structural rate of rebuilding for cement 6 with SP1 and SP2

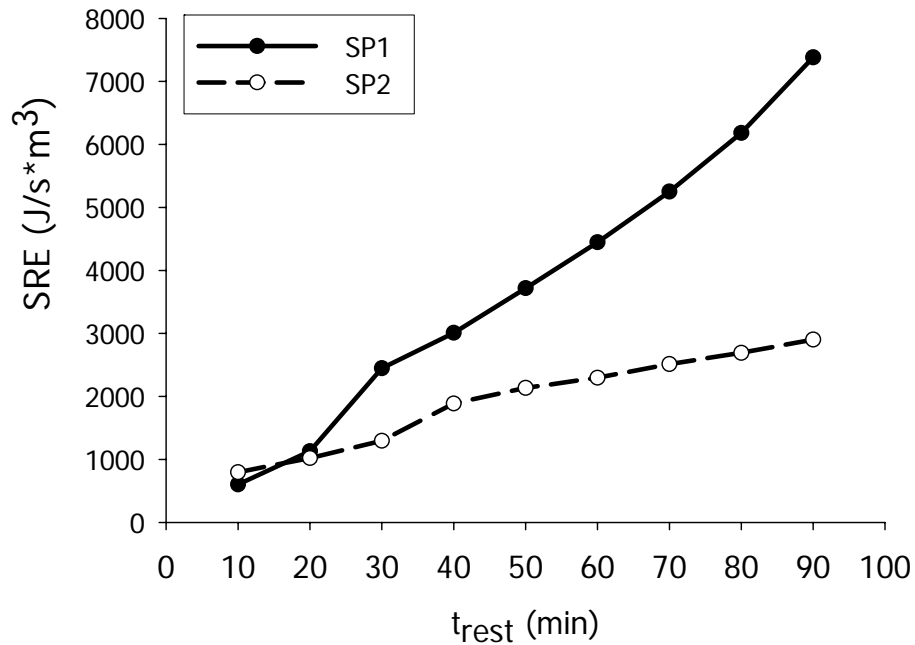


Figure 5-13. Structural rate of rebuilding for cement 7 with SP1 and SP2

5.5 Chapter Summary

The results showed that the rate of structural rebuilding can be influenced by the cement composition. Since SP1 was found to perform better than SP2 vis-à-vis the rate of stiffening, SP1 was selected for further studies. Cement 6 and Cement 5 were also selected for further studies because they exhibited significantly different rates of structural build-up. The general trend from the results indicates that slower rates of rebuilding occurred in cements with the higher C_3A and alkali contents. Cement 6 will be used to represent the cements with slow rates of structural build-up while Cement 5 will be used to represent the cements with fast rates of structural build-up.

6 Chapter 6: Detailed Studies - Influence of mix composition on structural rebuilding

6.1 Introduction

Due to the complex relationships among the various mixture ingredients, information about the interaction effects among the ingredients may also be significant in understanding the role of the paste matrix in controlling structural rebuilding. To evaluate the role of these potential interactions, a 2^{5-1} factorial design experiment was developed. This chapter presents the results from a comprehensive experimental program that included shear rheological studies, setting time measurements, and elemental analysis studies to examine the role of mix composition on structural rebuilding. Previous research showed that initial fluidity is a crucial factor affecting the rate of rebuilding [94]. Thus, the initial fluidity level was held constant so that the influence of the constituent ingredients could be highlighted.

6.2 Experimental Program

6.2.1 Materials

Sixteen cement paste mixtures were proportioned using Cement 5 or Cement 6 (c.f. Chapter 5). The C_3A and equivalent alkali content of Cement 5 was 11 and 0.8, respectively. The C_3A and equivalent alkali content of Cement 6 was 7 and 0.26, respectively. The maximum limit on equivalent alkalis when cement is to be used in concrete with reactive aggregates is 0.6% [100]. The usual equivalent alkali content of Portland cements, however, ranges from 0.3 to 1.3% [14,

101]. The C_3A contents of normal Portland cements generally range from 5 – 10 % [96].

Cement 6 was selected to represent the cements in Chapter 5 that showed relatively fast rates of structural rebuilding behavior. These cements generally contained lower C_3A and alkali contents. Cement 5 was selected to represent the cements in Chapter 5 that showed relatively slow rates of structural rebuilding behavior. These cements generally contained higher C_3A and alkali contents. For the purpose of this study, Cement 6 will be referred to as the “low alkali/ C_3A ”, and Cement 5 will be referred to as the “high alkali/ C_3A ” cement.

A portion of the cement was systematically replaced with Class F fly ash and/or silica fume. The fly ash and silica fume dosages were replaced on a mass basis. A polycarboxylate superplasticizer with a specific gravity of 1.04 (represented as SP1 in Table 5-2) was used in all mixtures. A liquid, water-soluble, cellulose-based VMA with a specific gravity of 1.002 and solids content of 70% was employed in select mixtures. The properties of the cements, fly ash, and silica fume are listed in Table 6-1 and Table 6-2. The particle size distributions of the cements and fly ash are shown in Figure 6-1.

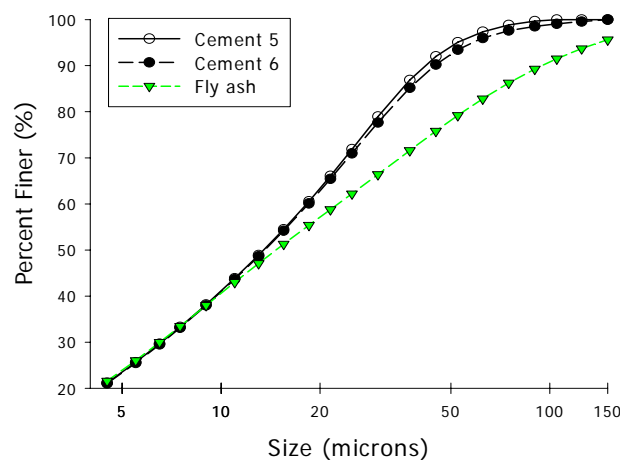


Figure 6-1. Volume size particle size distribution of cements and fly ash (determined by laser diffraction)

Table 6-1. Detailed properties for Cement 5 and Cement 6

Oxide Analysis	Cement	
	5	6
CaO (%)	62.0	63.7
SiO ₂ (%)	18.7	20.4
Al ₂ O ₃ (%)	5.7	4.1
Fe ₂ O ₃ (%)	2.5	2.4
SO ₃ (%)	4.3	2.3
MgO (%)	2.4	3.9
Equivalent Alkali (%)	0.85	0.26
free lime (f-CaO) (%)		1.3
LOI (1000°C) (%)	1.94	1.5
Specific surface (m²/kg)	388	365
Compound		
C ₃ S (%)	56	67
C ₂ S (%)	11	14*
C ₃ A (%)	11	7
C ₄ AF (%)	8	7*
Initial Set	120	101
Fineness, 45 um (% pass)	91.6	93
Blaine specific surface (m²/kg)	388	365
3 day f'c (psi/MPa)	3440/ 23.7	3272/ 22.6
28 day f'c (psi/MPa)	5920/ 40.8	6729/ 46.4
Air content (mortar) (%)	8	6

Table 6-2. Properties for fly ash and silica fume

Oxide	Fly ash	Oxide	Silica Fume
Al ₂ O ₃ (%)	17.86	SiO ₂ (%)	95.5
CaO (%)	12.50	Al ₂ O ₃ (%)	0.41
Fe ₂ O ₃ (%)	7.49	Fe ₂ O ₂ (%)	0.28
K ₂ O (%)	1.24	CaO (%)	0.49
MgO (%)	3.72	MgO (%)	0.24
Na ₂ O (%)	0.17	LOI (%)	2.04
P ₂ O ₅ (%)	0.10	Oversize 45 μ m (%)	0.42
SiO ₂ (%)	54.35	Specific gravity	2.07
SO ₃ (%)	0.53		
TiO ₂ (%)	1.15		
LOI (750°C) (%)	0.06		

6.2.2 Mix Proportions

A 2⁵⁻¹ factorial design was used to develop the mix proportions. Five factors—cement type, w/b ratio, fly ash (FA) dosage, silica fume (SF) dosage, and VMA dosage—were tested at two levels each. The high level and low level for each factor in un-coded units is given in Table 6-3. The superplasticizer dosage was adjusted such that all mixtures had an initial mini-slump flow of 330 mm (13 in.) \pm 7 mm (0.28 in.). The superplasticizer and VMA dosages were calculated based on the mass of the aqueous solution with respect to the mass of the cementitious materials (cement + fly ash + silica fume). When used, the dosage of VMA was set to 400 ml/100 kg of cementitious materials.

6.2.3 Methods

Brief descriptions of the different experimental methods used in this phase of the project are given below:

Vicat Needle Test: This test is used for normal consistency cement pastes and defines initial set as the time between the initial contact of cement and water and the time in which the Vicat needle penetrates into a depth of 25 mm or less into a cement paste sample [102]. The pressure applied to the paste can be calculated from the diameter and weight of the Vicat apparatus. For the standard Vicat apparatus this corresponds to a pressure of 3.7 MPa and force of 2.94 N. Research conducted by Struble et al. [103] showed that the Vicat needle tests could be used to determine the initial setting time of fluid pastes.

Inductively Coupled Plasma Spectroscopy (ICP): A Varian model ICP Spectrometer, which can cover a spectral range from 175 – 785 nm was used in this study. ICP is an instrumental elementary analysis technique for determining chemical composition. A sample is subjected to very high temperatures generated by a plasma of argon. The elements within the sample are excited and emit light. The elemental analysis of the sample is conducted by measuring the intensity of the characteristic emitted light at predetermined wavelengths. Each element has its own characteristic wavelength; thus, the intensity at a specified wavelength determines the concentration of the element in the specimen.

Table 6-3. Mixture proportions for evaluated paste mixtures (un-coded factorial design**units)**

Mix	Alkali/C ₃ A Cement Type	W/B	FA (% binder)	SF (% binder)	VMA
1	Low (Cement 6)	0.3	20	0	included
2	High (Cement 5)	0.3	20	0	0
3	Low (Cement 6)	0.4	20	0	0
4	High (Cement 5)	0.4	20	0	included
5	Low (Cement 6)	0.3	30	0	0
6	High (Cement 5)	0.3	30	0	included
7	Low (Cement 6)	0.4	30	0	included
8	High (Cement 5)	0.4	30	0	0
9	Low (Cement 6)	0.3	30	8	0
10	High (Cement 5)	0.3	20	8	included
11	Low (Cement 6)	0.4	20	8	included
12	High (Cement 5)	0.4	20	8	0
13	Low (Cement 6)	0.3	30	8	included
14	High (Cement 5)	0.3	30	8	0
15	Low (Cement 6)	0.4	30	8	0
16	High (Cement 5)	0.4	30	8	included

Rheology: A HAAKE Rheostress150 rheometer with a concentric cylinder geometry

conforming to the German DIN 53019 sensor system was used. The radius of the outer cylinder (R_o), radius of the inner cylinder (R_i), and the length of the inner cylinder (L) are 10.85 mm, 10.00 mm, and 30.00 mm, respectively.

6.2.4 Sample Preparation

A Hobart N-50 mixer was used to mix all samples. Prior to mixing, a portion of the water was used to dilute the superplasticizer. In addition, another portion of the water was set aside to rinse the cylinder containing the diluted superplasticizer. The remaining water (total water minus rinse water and water used for dilution) was added to the powder materials and mixed for 3 minutes at a low speed (136 rpm). Then the mixer was turned off and within one minute the sides of the bowl was scraped with a spatula and the superplasticizer was added. Then, the paste was mixed for 15 seconds at low speed. Finally, the paste was mixed for an additional 2 minutes and 45 seconds at a higher speed (281 rpm). The total mixing time for the cement paste was 7 minutes. Paste flow tests were conducted 1 minute after the sample preparation was completed (hence, 8 minutes after the introduction of water) using a glass plate and a mini-slump cone. The dimensions of the cone were 70 mm (2.76 in.), 100 mm (3.94 in.), and 50 mm (1.97 in.), for the upper inner diameter, lower inner diameter, and height. After preparing the paste, approximately 10 ml of paste was placed into the rheometer for rheological testing. Rheological tests commenced twelve minutes after the introduction of water. The rheological protocol was described in detail in Chapter 4.

The samples for the setting time experiments and ICP experiments were each batched separately. For the setting time tests, after preparing the paste, a portion of it was immediately

poured into the Vicat needle apparatus mold, covered with plastic, and left to rest. The first needle penetration test was conducted 30 minutes after the resting time followed by needle penetration tests every 10 minutes until a penetration of 25 mm or less was obtained. For the ICP tests, the cement paste samples were prepared using deionized (DI) water. After mixing, the paste was transferred into a nitrogen filled glove box. The pore solutions were extracted by vacuum filtering through a Buckner funnel (with 0.3 μm filter paper). Samples with VMA were centrifuged first, and then the supernatant was filtered through the Buckner funnel. All solutions were diluted with a nitric acid aqueous solution by a factor of 1:150 and then filtered through a 0.2 μm syringe filter. The pH was measured prior to any dilution in order to determine the concentration of the hydroxide ions.

6.3 Results

6.3.1 How to analyze factorial analysis

The results were analyzed using a statistical software program (Minitab 14, distributed by Minitab Inc.). A normal probability plot was used to assess the effect of each factor. All effects are assumed to be normally distributed with zero mean and the same variance. Effects that follow this pattern will fall along a straight line and are considered negligible, whereas significant effects will not lie along the straight line [104]. A Pareto bar chart was used in conjunction with the normal probability plot in order to insure that all significant effects would be considered. A Pareto bar chart plots the effects in descending order, and any effect above a critical horizontal line is considered significant.

6.3.2 Rheology

Figure 6-2 shows the rate of rebuilding for the 16 paste mixtures. As stated in Section 4.4, the slope of the curves is normally used as a measure of rebuilding, but this analysis method can not be used for mixtures that exhibit nonlinear behavior.

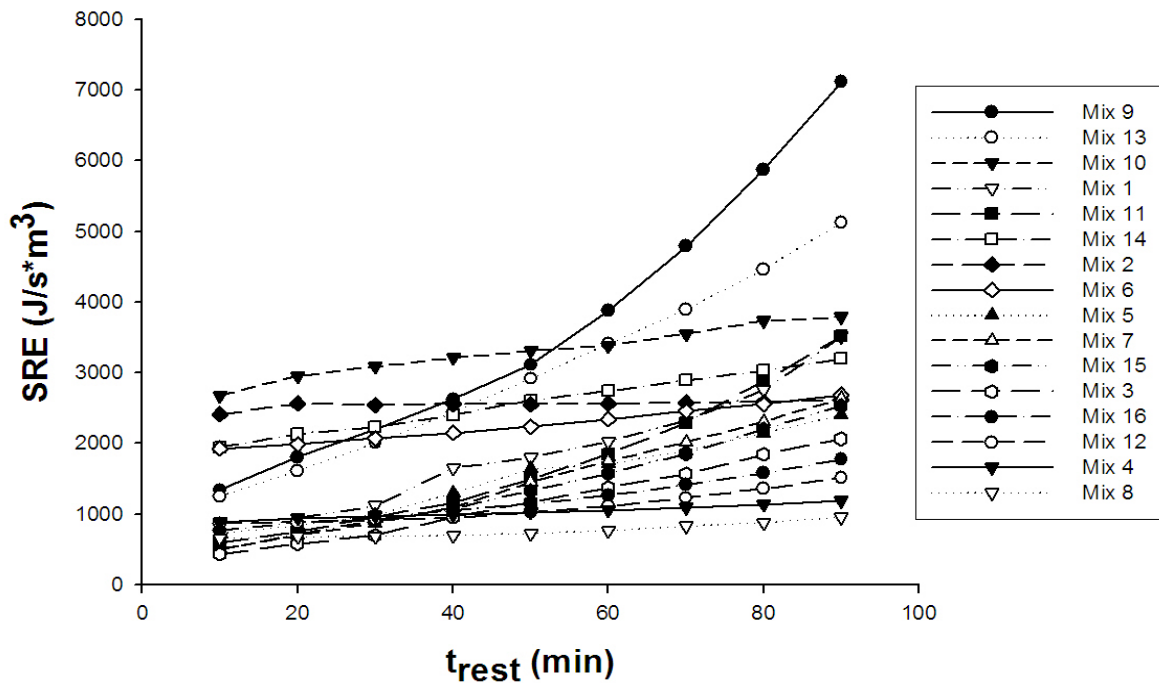


Figure 6-2. SRE vs. t_{rest} for mixes in detailed studies

The data was transformed according to the formula shown below so that all mixtures could be compared:

$$\Delta SRE = SRE_{final} - SRE_{initial}$$

where $SRE_{initial}$ corresponds to the SRE value at t_{rest} equals 10 min and SRE_{final} corresponds to

the SRE value at t_{rest} equals 90 minutes. Higher ΔSRE values correspond to mixtures with a faster rate of structural rebuilding. The equation above can be normalized so that the rate of structural rebuilding at different times can be evaluated. In this case the equation would be

$$\Delta SRE_{norm} = (SRE_2 - SRE_1) / (t_{rest2} - t_{rest1})$$

where the values 2,1 corresponds to different t_{rest} values and $t_{rest2} > t_{rest1}$. ΔSRE values are given in Table 6-4. The maximum ΔSRE was obtained by mix 9, whereas the minimum ΔSRE was obtained by mix 2. The average ΔSRE for all the mixtures was $1767 \text{ J/m}^3 \cdot \text{s}$.

Table 6-4. Rate of rebuilding determined using rheological protocol

Mix	Alkali/C ₃ A Cement Type	ΔSRE (J/s* m ³)
1	low	2633
2	high	210
3	low	1630
4	high	304
5	low	1680
6	high	765
7	low	2128
8	high	297
9	low	5777
10	high	1112
11	low	2929
12	high	645
13	low	3882
14	high	1251
15	low	2030
16	high	1005

The main effects of each factor on Δ SRE are plotted in Figure 6-3. The slope of the line indicates the strength of the response (i.e. steeper slope indicates that the factor is more significant). All plots are shown in coded units. Thus -1 corresponds to the low level and 1 corresponds to the high level of each factor listed in Table 6-3. For example, -1 for the w/b corresponds to mixtures having a w/b ratio of 0.3, whereas 1 corresponds to a w/b ratio of 0.4. Similarly, -1 for VMA corresponds to mixtures that did not include VMA, and +1 for VMA corresponds to mixtures that included VMA.

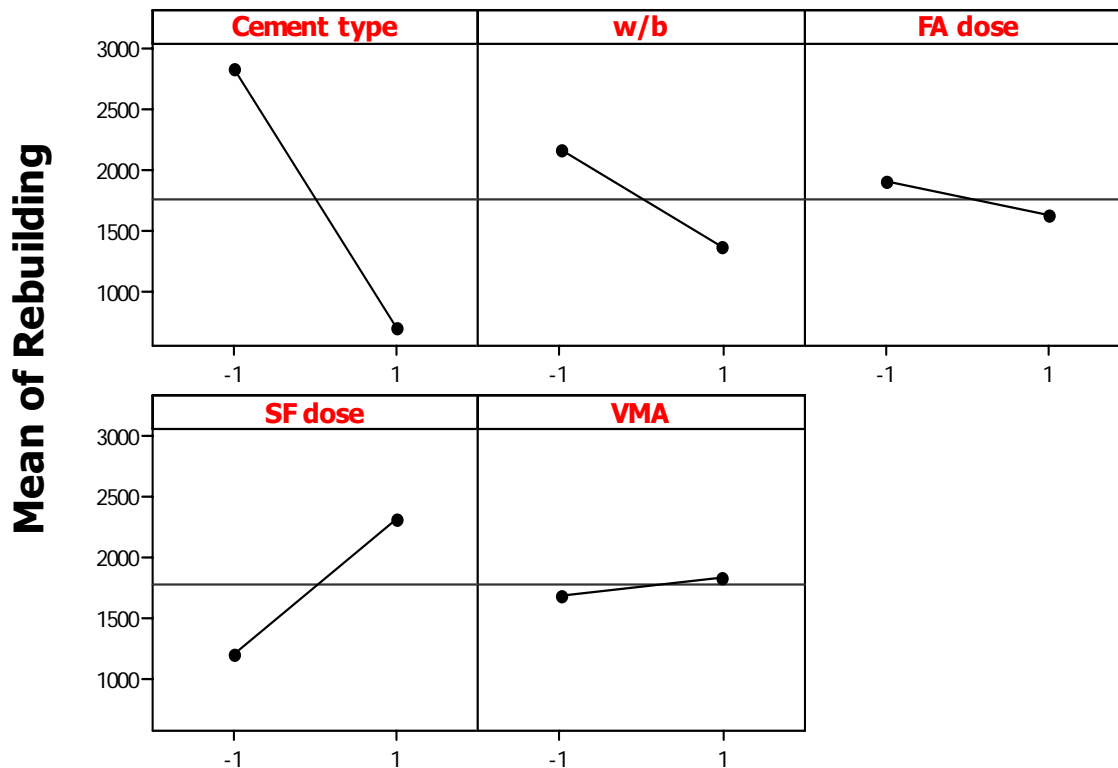


Figure 6-3. Main effect plots for structural rebuilding (-1 and 1 indicate low and high level for each factor; see Table 6-3 for the value for each factor)

The average effect of cement type and w/b ratio was negative, which means that the rate of rebuilding decreased when either one of these factors was increased. The main effect of increasing the silica fume dosage was positive, and on average the rate of rebuilding increased when silica fume was used as a cement replacement.

Nominal changes in the rate of structural rebuilding were seen when the fly ash dosage was increased or VMA was added to the mixtures. Thus, when only these main effects are considered, the rate of rebuilding is maximized when silica fume is set at the high level, w/b ratio is set at the low level, and cement type is set at the low level. The trends for the main effects generally hold true (i.e. decrease in rate of rebuilding when the cement with the higher alkali/C₃A content was used regardless of the w/b ratio). However, in certain cases interactions among the factors will influence the response. These interactions provide further insight into the structural rebuilding behavior.

Figure 6-4 shows the interaction plots for all the factors plotted in Figure 6-3. All plots are shown in coded units. Thus -1 corresponds to the low level and 1 corresponds to the high level of each factor listed in Table 6-3. For example, -1 for the w/b corresponds to mixtures having a w/b ratio of 0.3, whereas 1 corresponds to a w/b ratio of 0.4. Similarly, -1 for VMA corresponds to mixtures that did not include VMA, and +1 for VMA corresponds to mixtures that included VMA. The slope of the line indicates the strength of the response (i.e. steeper slope indicates that the factor is more significant). In addition, information about the degree of interaction among the factors can be obtained by comparing the slopes of the two lines in a given plot. Higher degrees of interactions are seen in plots in which the slopes of the two lines are significantly different.

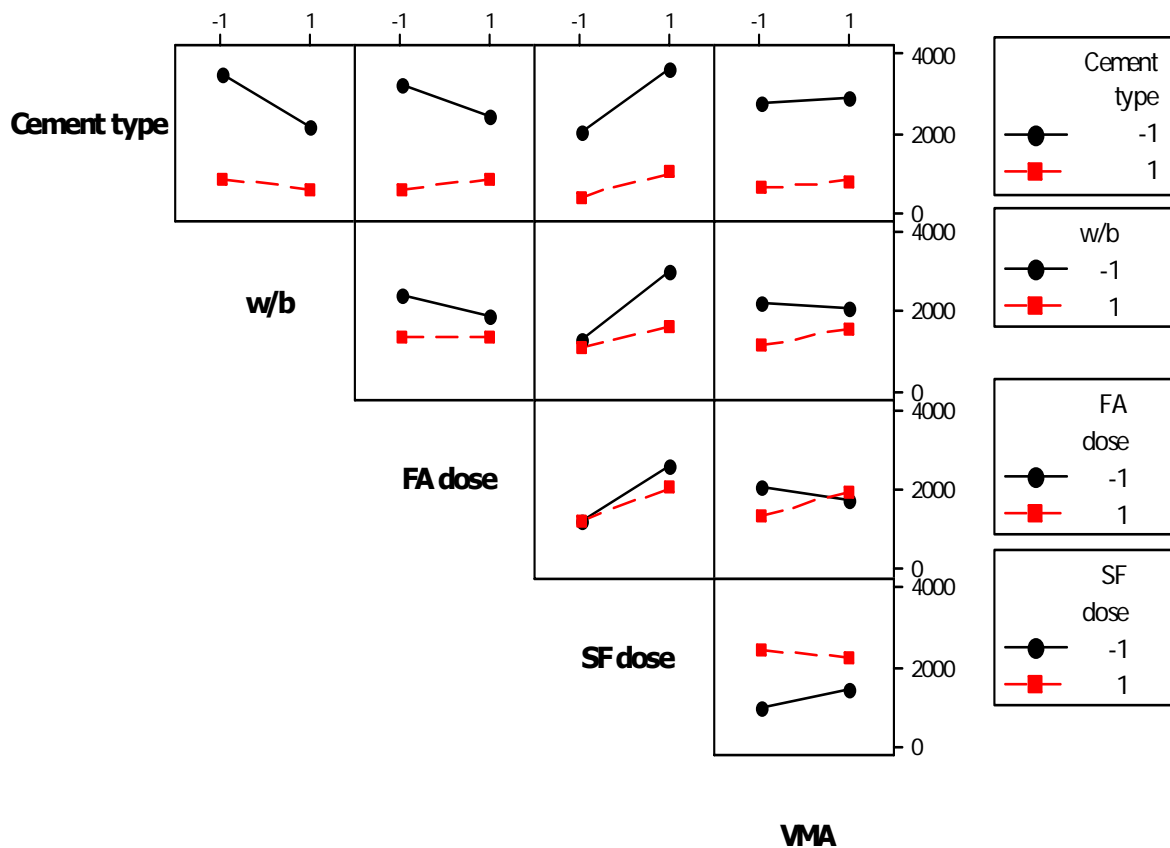


Figure 6-4. Interaction plots for structural rebuilding (-1 and 1 indicate low and high level for each factor; see Table 6-3 for the value for each factor)

It can be seen that the interaction between w/b ratio and silica fume resulted in a larger increase in the rate of rebuilding when silica fume was added to the lower w/b mixtures than when silica fume was incorporated in the higher w/b mixtures. Similarly, a larger increase in the rate of rebuilding was seen when silica fume was incorporated in the lower alkali/C₃A mixtures compared to when the higher alkali/C₃A cement was used. Different behaviors were seen in the rate of rebuilding when the fly ash dosage was increased in the high alkali/C₃A mixtures

compared to the low alkali/C₃A mixtures. Specifically, a small increase in the rate of rebuilding occurred when a higher dosage of fly ash was used with the higher alkali/C₃A cement, whereas the opposite trend was seen when the lower alkali/C₃A cement was used.

Analysis of the normal probability plot of effects (c.f. Figure 6-5) provides insight into which factors were most significant in influencing the rate of rebuilding. According to the normal probability plot, main effect A had a substantial influence on Δ SRE. The next effect that is furthest away from the straight line is the interaction effect AC. Analysis of the Pareto plot shows that if interaction effect AC is considered significant, then main effects D and B and interaction effect BD must also be considered. VMA did not significantly alter the rate of rebuilding.

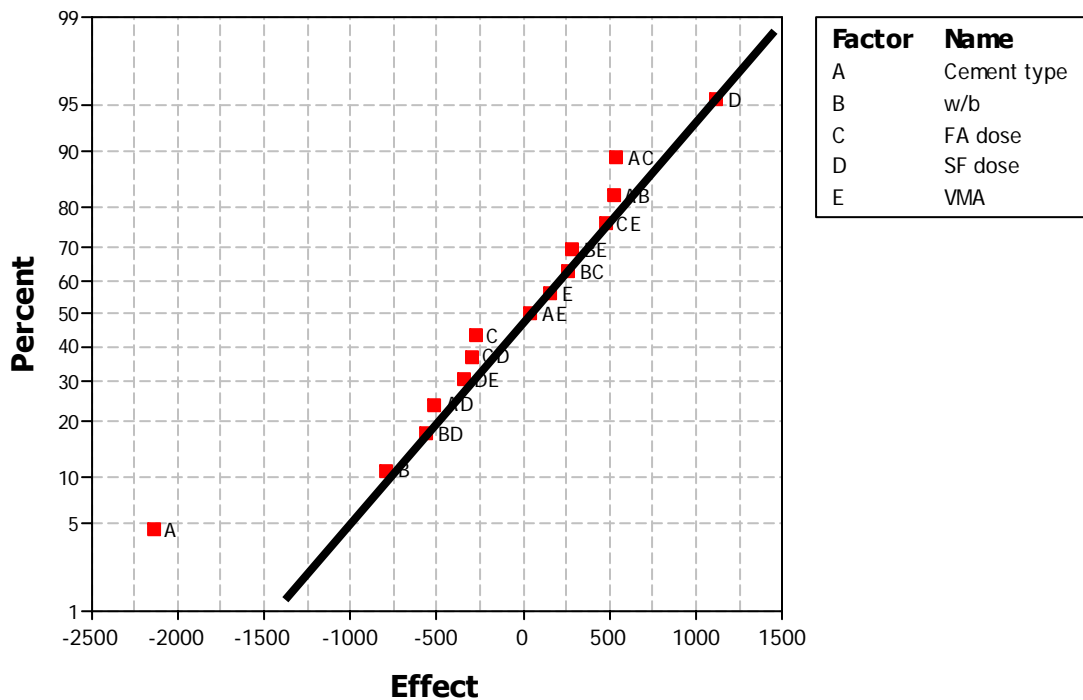


Figure 6-5. Normal probability plot of effects for structural rebuilding

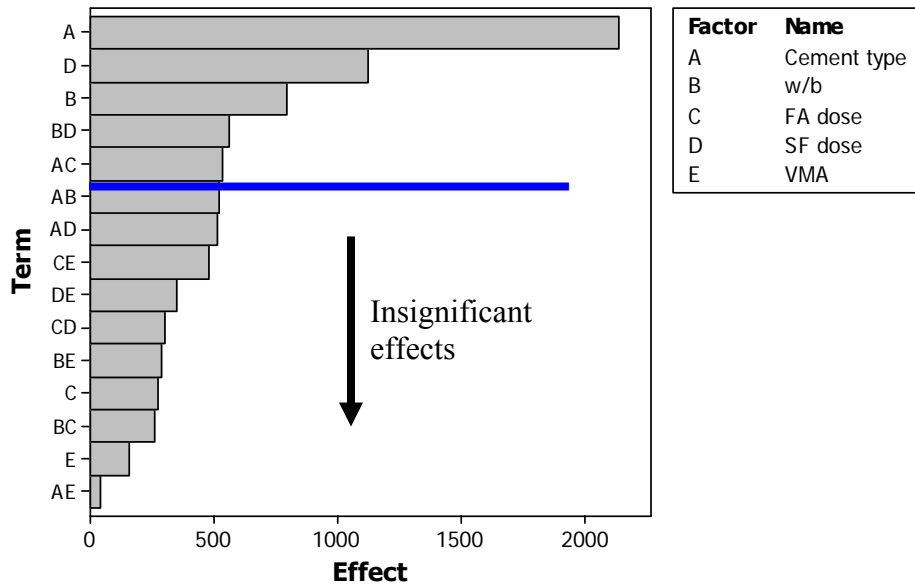


Figure 6-6. Pareto plot for structural rebuilding

Considering only the main effects, the regression model is

$$\Delta SRE = 1767 - 1069 \text{ cement type} - 396 \text{ w/b} - 138 \text{ FA dose} + 561 \text{ SF dose} + 77 \text{ VMA}$$

$$R^2 = 0.79$$

Equation 6-1

where 1767 is the average response and the coded variables (cement type, w/b, FA dose, SF dose, and VMA) take on values -1 or +1. The model can be improved by taken into consideration the BD and AC interaction effects. This yield

$$\Delta SRE = 1767 - 1069 \text{ cement type} - 396 \text{ w/b} - 138 \text{ FA dose} + 561 \text{ SF dose} - 280 \text{ w/c*SF} + 269 \text{ cement*FA dose}$$

$$R^2 = 0.86$$

Equation 6-2

Due to the principle of hierarchy [105], the insignificant term (FA dose) remains in the model as a main effect because it is included in an interaction effect. Using the regression model given in Equation 6-2, and taking into account the information gained from interaction plots, the fastest rate of rebuilding appears to be obtained when the factors are set at the following levels:

Cement: -1

w/b: -1

Silica fume: +1

Fly ash: -1

VMA: arbitrary, but set at +1 due to the interaction effects that shows ΔSRE is maximized at this level when silica fume dosage is set to +1 and fly ash dosage is set to -1

This corresponds to using the low alkali/C₃A cement at a w/b ratio of 0.30 with 8% silica fume and 20% fly ash. With the higher alkali/C₃A cement (Cement 5), the optimal levels to achieve the fastest rate of rebuilding are w/b of 0.30 with a silica fume dosage of 8% and a fly ash dosage of 30%.

6.3.3 Vicat Needle Tests

The Vicat needle test was used to determine the initial setting time, which is typically considered an indirect reflection of the physical state of the hydrating material. High fluidity pastes were used so that the results are more representative of the pastes used in SCC. Figure 6-7 shows results obtained from monitoring the time evolution of the Vicat needle penetration. The penetration resistance of the sample was relatively constant, and then after a critical point (which

in the figure is approximately 9 hours), an abrupt transition occurred in which the depth of penetration rapidly decreased towards zero. This constant behavior prior to the transition was seen in all mixtures, although the critical time in which the abrupt transition occurred was different for the various mixtures.

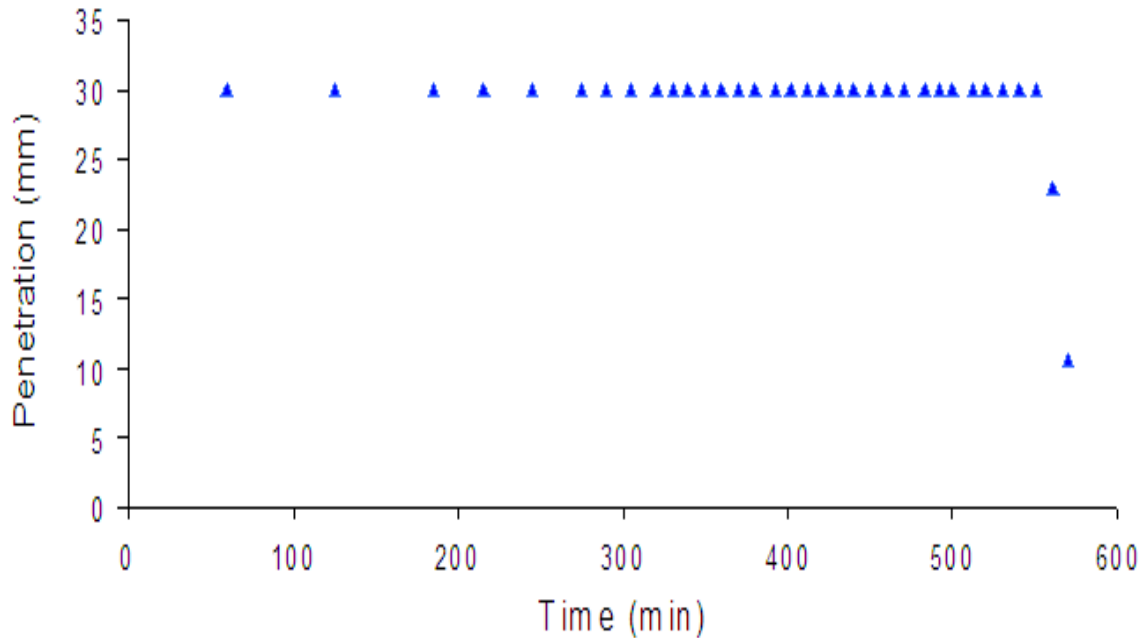


Figure 6-7. Evolution of Vicat needle penetration (representative mixture)

Figure 6-8 shows a chart of the initial setting time in order of increasing setting time. In general the mixtures made with the low alkali/ C_3A cement displayed the shortest initial setting time. Further understanding can be elicited from the analysis of the factorial design.

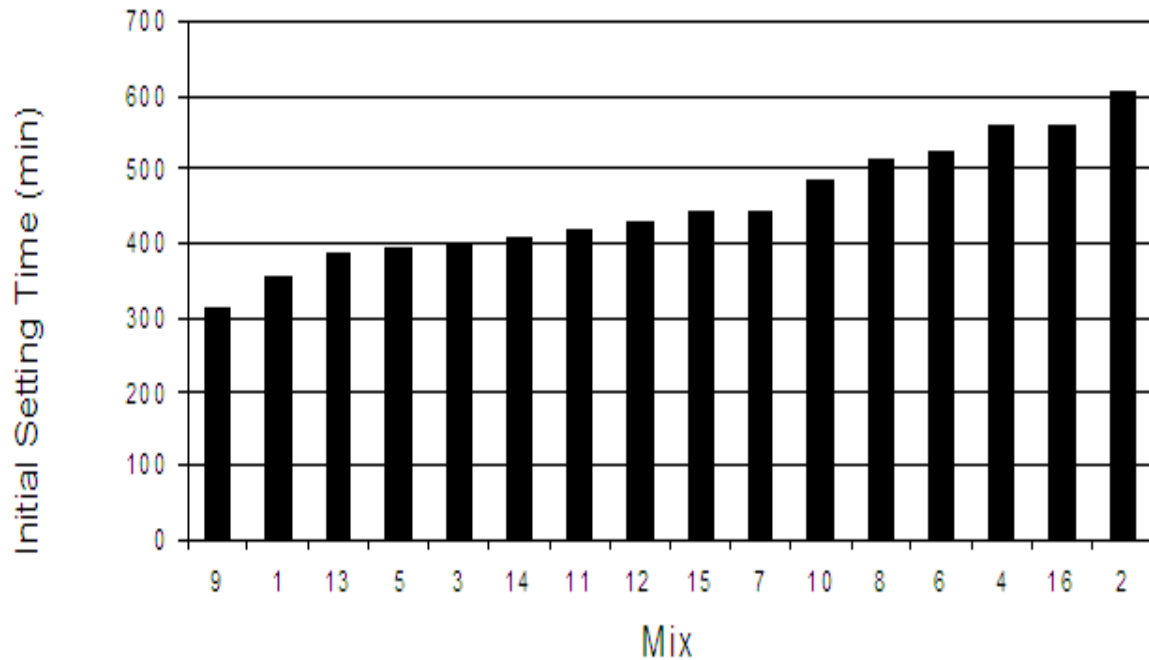


Figure 6-8. Initial Setting Times

The main effect plots are shown in Figure 6-9. In general, increasing the alkali/ C_3A content, w/b ratio, fly ash dosage, and VMA dosage resulted in an increase in the initial setting time. Incorporating silica fume into the mixtures resulted in a decrease in the initial setting time.

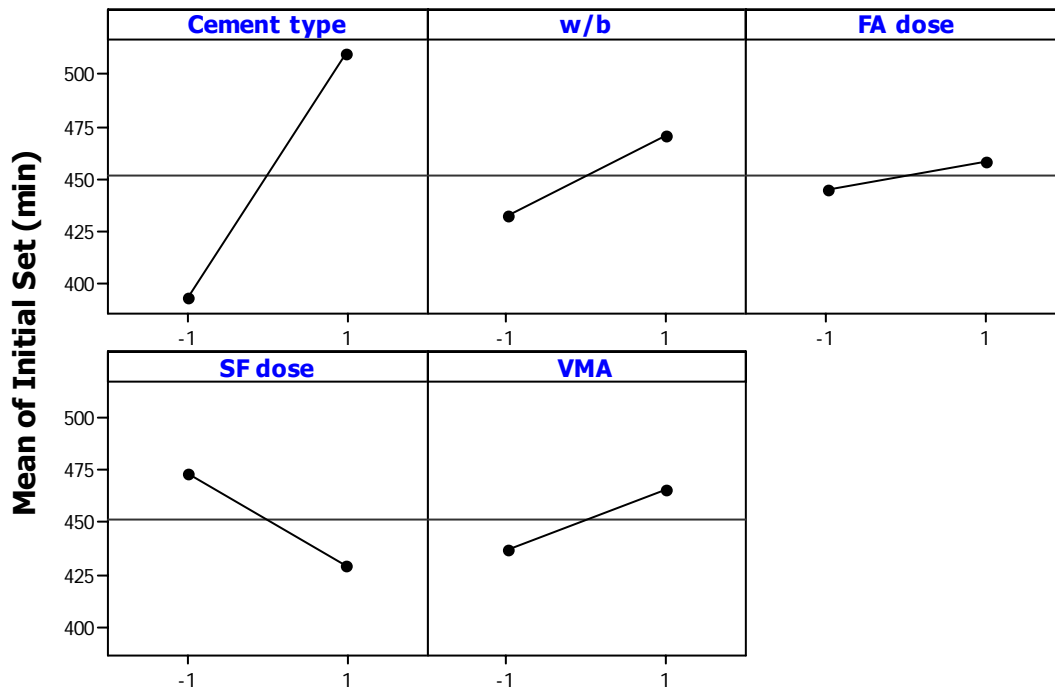


Figure 6-9. Main effect plots for initial set

The normal probability plot of the effects is shown in Figure 6-10. The significant effects that emerge from this analysis were the main effect A (cement type), main effect D (silica fume), and the AB (cement type*w/b), AC (cement type*fly ash), and AD (cement type*silica fume) interactions. The effect of cement was positive; whereas the effect of silica fume was negative (i.e. increasing alkali and C_3A content increases setting time while increasing silica fume dosage decreases setting time). Thus, if we considered only these main effects and ignored any interactions among the factors, in order to reduce the setting time a low alkali/ C_3A cement and high silica fume dosage should be used. However, it is necessary to examine any interactions that are important since main effects by themselves do not have much meaning when they are

involved in significant interactions. Figure 6-11 shows the Pareto plot for the main effects and interaction effects. A horizontal line was drawn to separate the significant effects of A, D, AB, AC, and AD from the non-significant effects. However, it is seen that main effects B (w/b) and E (VMA), as well as interaction effects DE (silica fume*VMA) and BD (w/b*silica fume), must also be included since they fall above the horizontal line. Thus, all main factors except fly ash were significant.

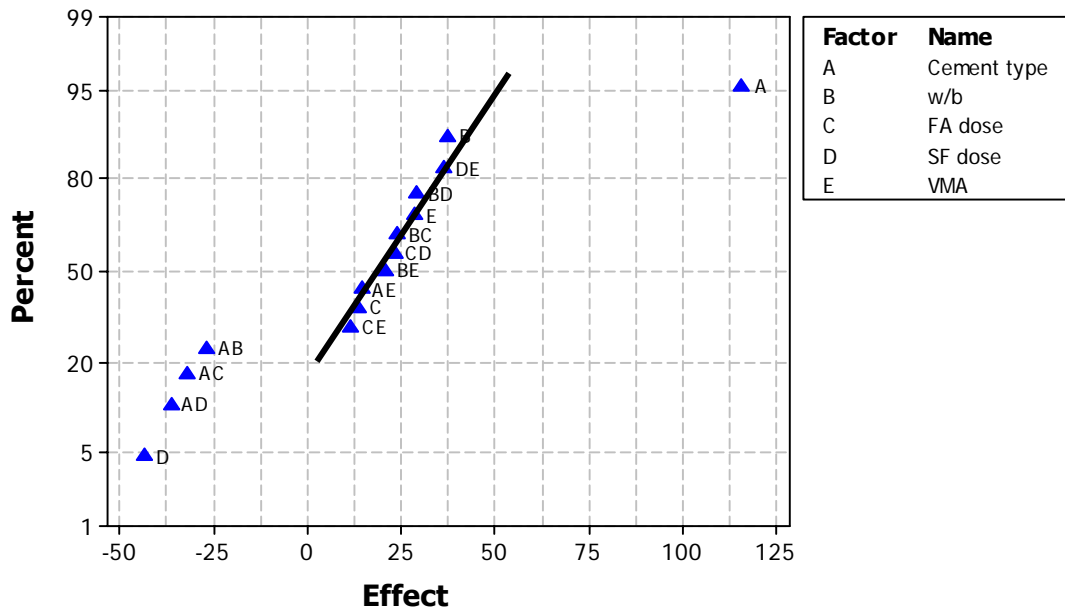


Figure 6-10. Normal probability plot of effects for initial setting time

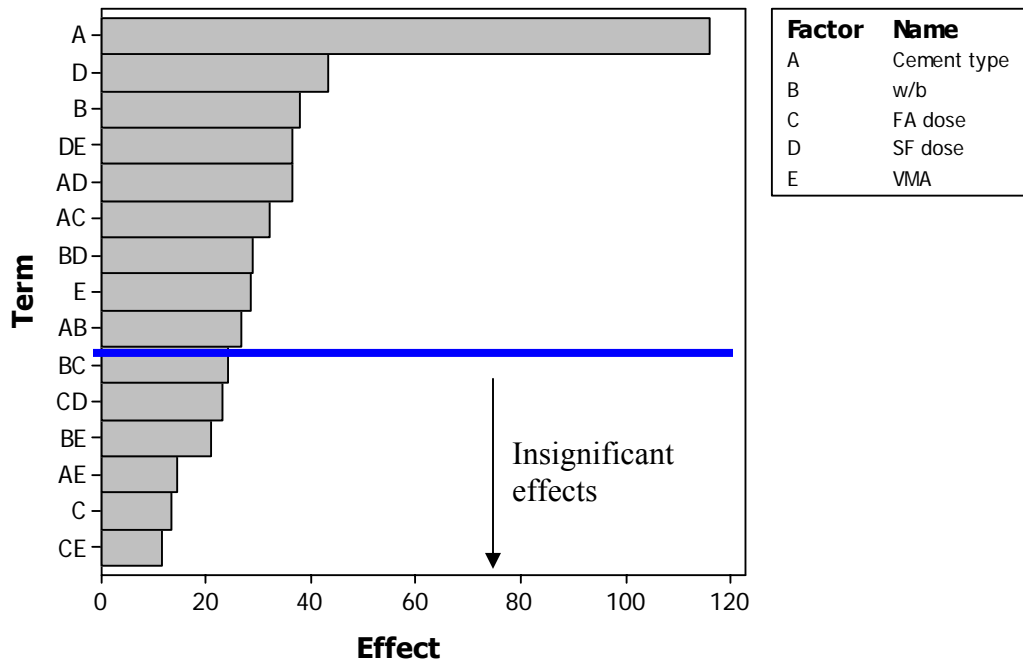


Figure 6-11. Pareto plot of effects for initial setting time

The AB, AC, and AD interactions are plotted in Figure 6-12, Figure 6-13, and Figure 6-14, respectively. Note from the AB interaction, that the influence of w/b was very small when the high alkali/C₃A cement was used and very large when the low alkali/C₃A cement was used. The AC interaction indicates that increasing the fly ash dosage resulted in a large increase in the initial setting time when the low alkali/C₃A cement was used. The reverse trend is seen when the fly ash dosage was increased when using the high alkali/C₃A cement. The AD interaction shows that a large negative effect (i.e. decrease in setting time) occurred when silica fume was used with the high alkali/C₃A cement.

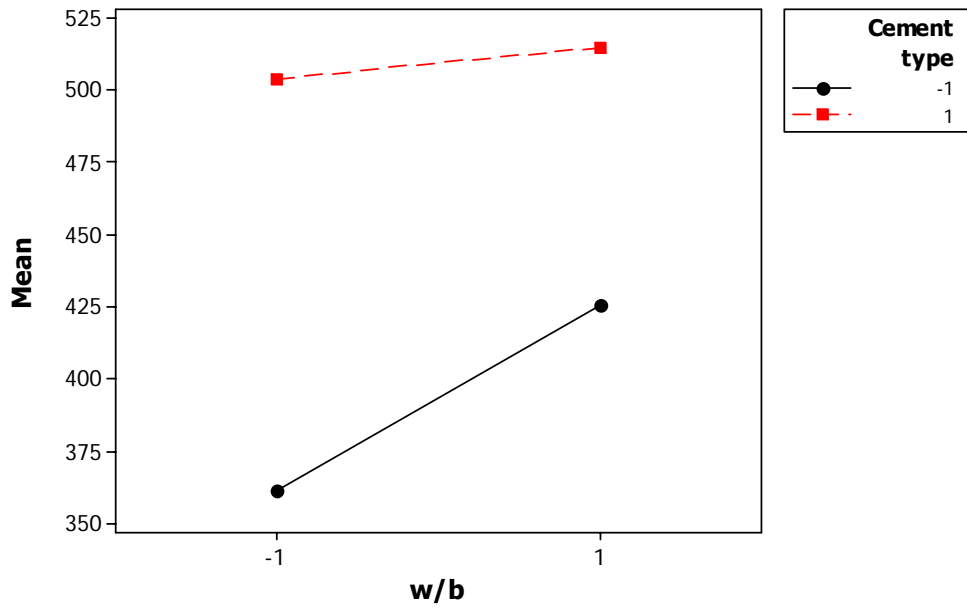


Figure 6-12. Interaction plot of cement type and w/b ratio for initial set (AB interaction)

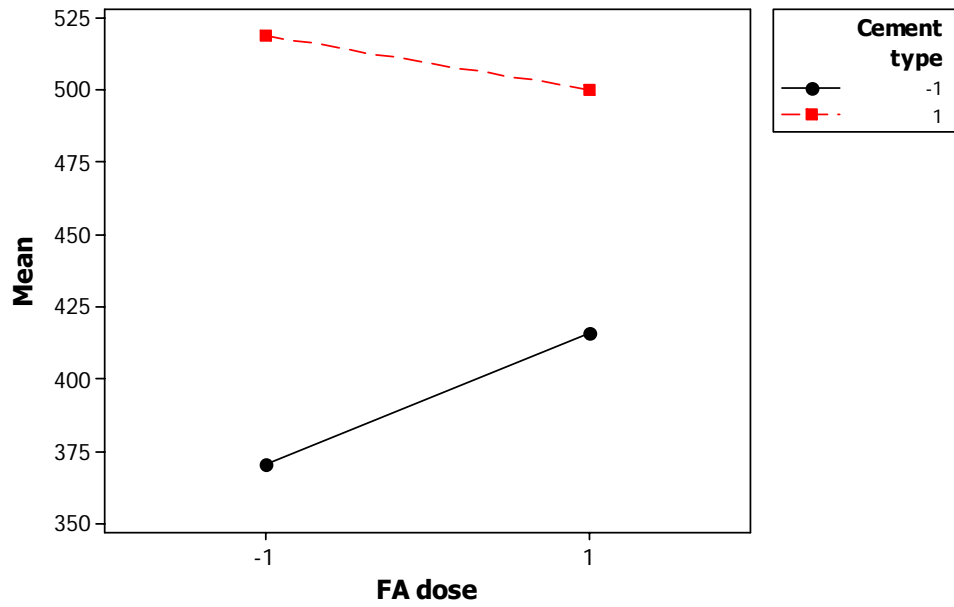


Figure 6-13. Interaction plot of cement type and fly ash dosage for initial set (AC interaction)

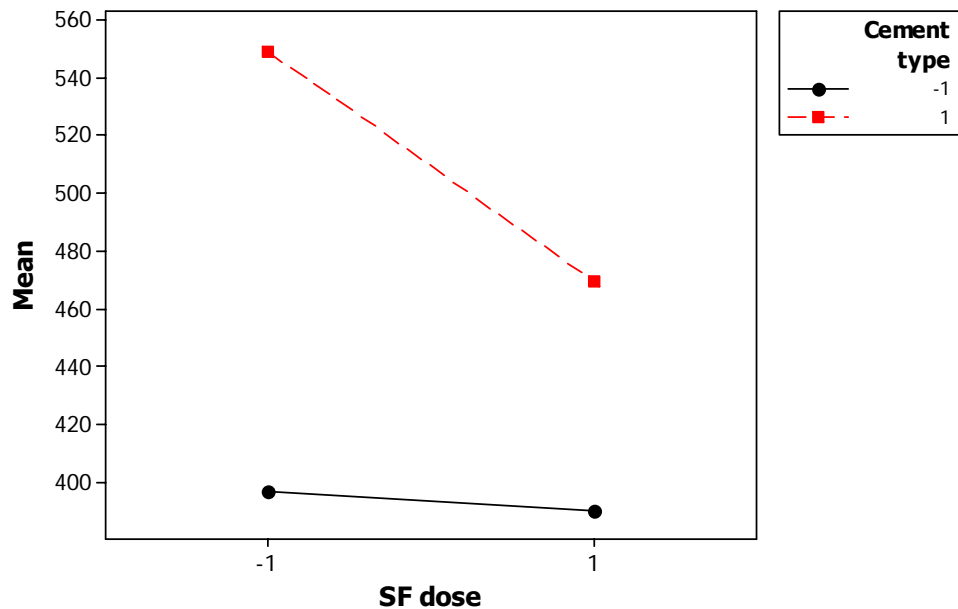


Figure 6-14. Interaction plot of cement type and silica fume dosage for initial set (AD interaction)

6.3.4 ICP Tests

The pore solutions were extracted at two distinct time intervals: immediately after mixing ($t=0$ min) and 120 minutes after the conclusion of mixing ($t=120$ min). The elemental concentration of aluminum (Al), calcium (Ca), iron (Fe), potassium (K), magnesium (Mg), sodium (Na), sulfur (S), and silicon (Si) were measured by ICP and the hydroxide (OH^-) ion concentration was determined by measuring the pH. The total concentration of sulfur was determined independent of its redox state. In Portland cement pastes, S is present in the oxidized form as sulfate [106]. Square brackets (with letter(s) inside) will be used to represent the concentration of an element or ion.

A rapid dissolution of ionic species into the aqueous solution occurred immediately after the cements were placed in contact with water. The alkali concentrations included alkalis released into the pore fluids from the hydrating cement and those released by fly ash and/or silica fume. No distinction was made between the amount of alkali contributed from the cement, fly ash, or silica fume.

Table 6-5 provides information about the pore solution chemistry of the samples 120 minutes after mixing. As expected, the concentrations of Al, Fe, Mg, and Si were lower than the Ca, K, Na, and S contents. This is due to the quick dissolution of the soluble alkali compounds, which deposit potassium, sodium, and sulfate ions into the pore solution. Calcium and additional sulfate ions are deposited into the pore solution from C_3S , C_2S , and calcium sulfate [107, 108]. All elemental concentrations measured in this study, except for silicon, were in accord with previously published results [97, 106, 108-112]. The silicon concentration was about 1 order of magnitude greater than those reported in the literature. (This may be due to errors in the equipment as the author was informed that the spectrometer was serviced a couple of months after these experiments were conducted). The pH ranged from 12.44 to 13.01, and mixtures made with Cement 5 (high alkali/ C_3A) generally exhibited a higher pH than paste mixtures made with Cement 6 (low alkali/ C_3A).

Table 6-5. Pore solution analysis measured 120 minutes after mixing (where 1 was determined using the pH meter, 2 was determined using the charge balance)

Mix	[Al] (mM)	[Ca] (mM)	[Fe] (mM)	[K] (mM)	[Mg] (mM)	[Na] (mM)	[S] (mM)	[Si] (mM)	[OH] ⁽¹⁾ (mM)	PH ⁽¹⁾	CB (%)	[OH] ⁽²⁾ (mM)	PH ⁽²⁾
1	0.62	31.55	0.08	40.67	0.61	17.41	20.87	2.28	36.31	12.56	36	79.44	12.90
2	3.93	26.23	0.95	262.38	6.13	91.95	150.82	4.44	117.49	13.07	-3	105.14	13.02
3	0.84	31.42	0.11	20.52	0.75	7.71	12.13	2.96	31.62	12.5	39	66.81	12.82
4	2.86	21.42	0.50	271.87	2.57	74.10	139.68	3.41	107.15	13.03	1	109.46	13.04
5	0.73	25.88	0.12	29.28	0.57	13.54	12.68	1.74	34.67	12.54	37	69.21	12.84
6	1.56	18.16	0.42	279.23	3.22	88.18	152.59	2.86	114.82	13.06	-4	98.56	12.99
7	1.15	35.34	0.50	23.22	0.58	10.83	14.17	3.14	30.90	12.49	43	76.39	12.88
8	5.57	55.98	1.42	238.02	2.98	66.82	128.71	9.88	75.86	12.88	3	102.73	13.01
9	0.13	27.07	0.00	44.32	0.23	20.04	22.29	1.54	29.51	12.47	37	73.92	12.87
10	7.19	41.64	0.00	296.95	6.99	104.68	182.60	11.56	107.15	13.03	3	119.73	13.08
11	0.20	28.10	0.00	29.69	0.23	12.34	14.97	0.76	43.65	12.64	25	68.29	12.83
12	0.01	0.14	0.00	2.00	0.01	0.58	1.08	1.68	100.00	13	1	102.69	13.01
13	0.01	0.30	0.00	0.34	0.01	0.17	0.18	3.82	37.15	12.57	45	111.41	13.05
14	0.01	0.14	0.00	2.22	0.01	0.72	1.18	4.31	100.00	13	6	129.05	13.11
15	0.41	31.25	0.00	30.47	0.28	14.62	16.32	1.08	26.30	12.42	45	74.93	12.87
16	0.69	14.36	0.03	221.34	0.73	63.74	125.68	1.99	81.28	12.91	-6	62.44	12.80

The charge balance error provides an indication of the quality of the results. Despite the fact that the extractions and analyses of the pore solutions were conducted in a random order, higher errors were obtained for the mixtures made with Cement 5 (c.f. Table 6-5). Charge balance errors are typical due to the inaccuracies in ICP measurements from spectral interferences and contamination, and, although much effort was taken to prevent contamination, it is impossible to eliminate all contaminants. Charge balance errors may also occur from carbonation (which would reduce the pH) and inaccuracies in determination of the hydroxyl ion concentration. The pH measurement is an activity measurement, and the hydroxyl concentration was calculated from the pH measurement with the assumption that the activity and concentration of the hydrogen ions was the same. In actuality, it is impossible to calculate the concentration of the hydrogen ion directly with an ion-selective electrode unless activity is known [113]. Furthermore, since other ions exist in the pore solution, it is highly likely that the activity coefficient is not equal to unity, and, thus, the actual hydrogen ion concentration in the solution and the hydrogen activity derived from the pH meter [97, 113] are not equal. If this is the case, hydroxyl ion concentration, as estimated from the pH measurement, would also be inaccurate. The aqueous phase of cement pastes is characterized with high ionic strengths (greater than 0.1M). Table 6-6 provides an estimate of the ionic strength (I_c) of the pore solutions using the measured concentrations reported in Table 6-5. (In order to actually determine the ionic strength, speciation should be conducted to determine the concentration of each ion. However, since this requires sophisticated thermodynamic modeling and knowledge of stability constants and activity coefficients, which in themselves are not agreed upon within the cement community, the elemental concentrations (as measured by ICP) were used to approximate the ionic strength.) In general, when the ionic strength increases, the activity coefficient decreases [113]. This

would lead to an under-prediction in the hydrogen ion and hydroxyl ion concentration (see equation below) and higher charge balance errors:

$$a_H = \gamma[H]$$

where a_H is the hydrogen ion activity and γ = activity coefficient. Theoretical hydroxyl ion concentrations and pH values using the elemental concentrations from the ICP test were shown in Table 6-5. For mixtures made with Cement 6 (odd number mixtures), the hydroxyl ion concentration calculated from the pH meter was substantially lower than those calculated from the ICP elemental concentrations.

Table 6-6. Ionic strength

Mix	I _c (M)
1	0.17
2	0.66
3	0.15
4	0.60
5	0.13
6	0.62
7	0.17
8	0.68
9	0.16
10	0.84
11	0.14
12	0.63
13	0.24
14	0.72
15	0.14
16	0.48

As mentioned earlier, the experiments were randomized, and if the charge balance errors were only attributed to under prediction in the hydroxyl ion concentration then one would expect that these errors would be evenly distributed across all the test results. However, this was not the

case. A surplus of cations were measured in the mixtures made with Cement 6 (lower alkali/C₃A), which suggests the presence of unaccounted anions in the pore solution. Although a significant amount of research has been devoted to understanding the chemistry of the aqueous phase of Portland cement pastes (and concretes), charge balance checks are often not reported due to researchers selecting to calculate the hydroxyl ion concentration from the charge balance of elemental concentrations instead of measuring the pH or lack of sufficient data (i.e. not measuring all the major ions). Charge balance errors at early hydration times have ranged from 1 – 26% [106, 111, 112, 114].

In general, little change was seen between the measurements conducted immediately after mixing (not shown) and those conducted after 120 minutes. This suggests that some type of equilibrium condition was established, or the rates of dissolution of the cement phases and the precipitation of cement hydration products were equal [96].

Based on the experimental program used in this study, changes in the mix proportioning did not significantly change the [Si], [Fe], and [Ca]. Figure 6-15 to Figure 6-23 show the normal probability plots for all the measured species. Significant increases in [Al], [K], [Mg], [Na], [S], and [OH⁻] occurred when Cement 5 (higher C₃A/alkali) was used instead of Cement 6 (lower C₃A/alkali). Low total alkali contents (as determined by the chemical composition) are usually associated with low soluble alkali concentrations and low ionic strengths; however, high total alkali content does not always imply high soluble alkali content and high ionic strengths. Alkalis in cements are present as alkali sulfates or trapped in the calcium aluminates and calcium silicate phases (incorporation into the aluminate and silicate phases occurs when there is not a sufficient amount of S₀₃) [99, 101]. The alkalis inside the aluminate and silicate phases are often considered as the insoluble portion of the total alkali content, whereas the solubility of alkali

sulfates are very high and are released rapidly into the pore solution upon the contact of cement with water. Studies [97, 99] have shown that the rate of dissolution of the alkalis are influence by the type and quantity of the sulfate, and that clinkers containing relatively large amounts of SO_3 have larger fractions of alkali sulfates [99, 115]. The high concentration of [Na] and [K] in the pore solutions of pastes made with Cement 5 indicates that the majority of the total alkalis were present as alkali sulfate. In order to maintain the mass balance of the solution, higher $[\text{OH}^-]$ and [S] were released in mixtures made with Cement 5 to compensate for the cations released from the dissolution of the alkali sulfates. Furthermore, changes in the w/b ratio and silica fume dosage also influenced the pore solution chemistry. In particular, the main effect of increasing the w/b ratio reduced [Mg], [Na], and [S]. Incorporation of silica fume yielded an increase in the [Na]. Cement replacement with silica fume increased [Al] when the w/b ratio was 0.30, but the opposite trend was seen when silica fume was added in mixes with a w/b ratio of 0.40 (c.f. Figure 6-24).

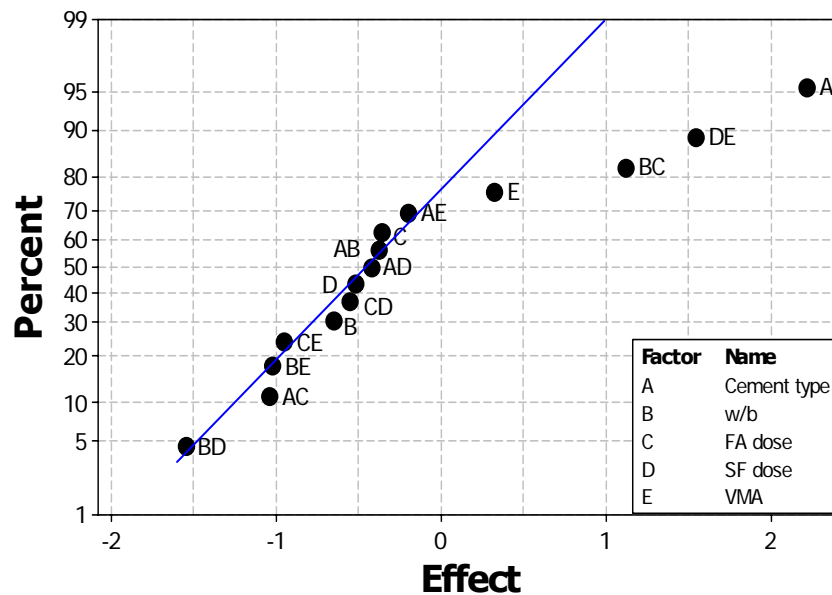


Figure 6-15. Normal probability plot for [Al]

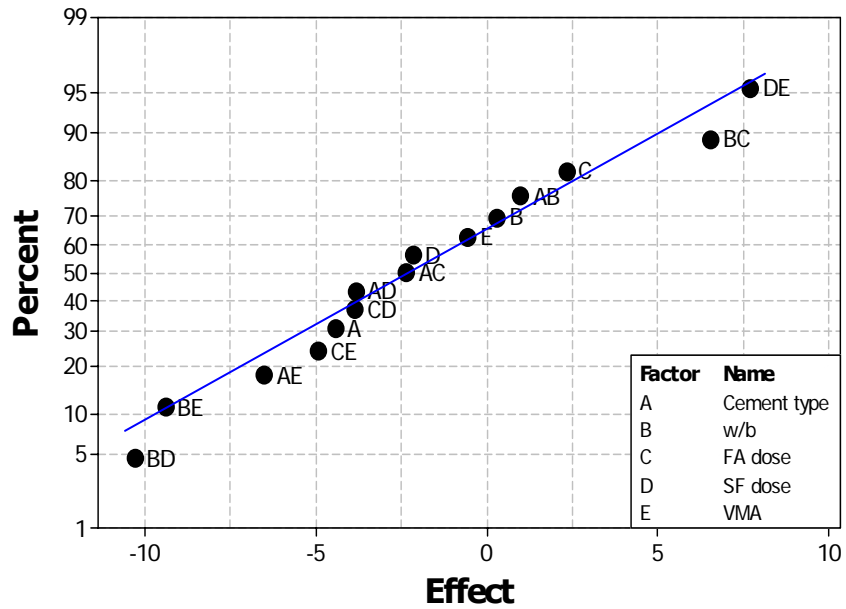


Figure 6-16. Normal probability plot for [Ca]

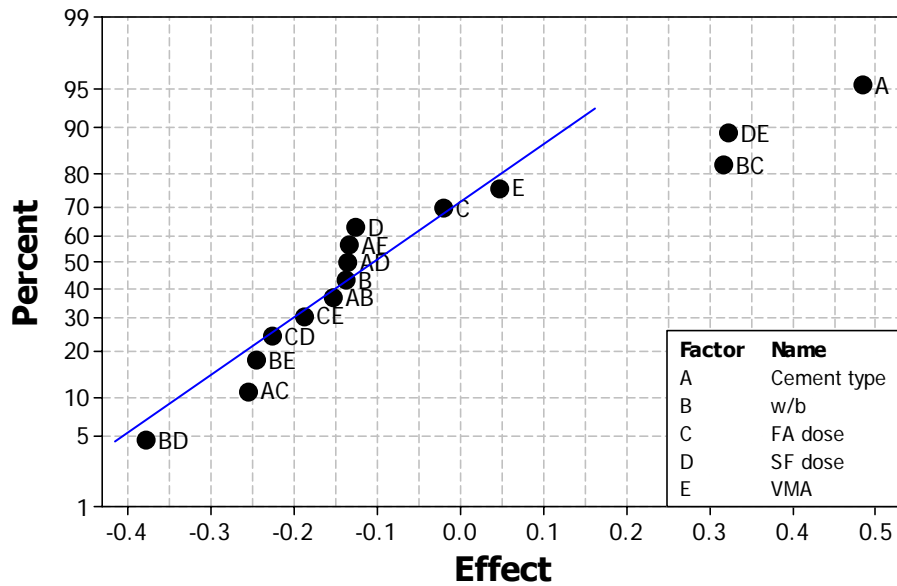


Figure 6-17. Normal probability plot for [Fe]

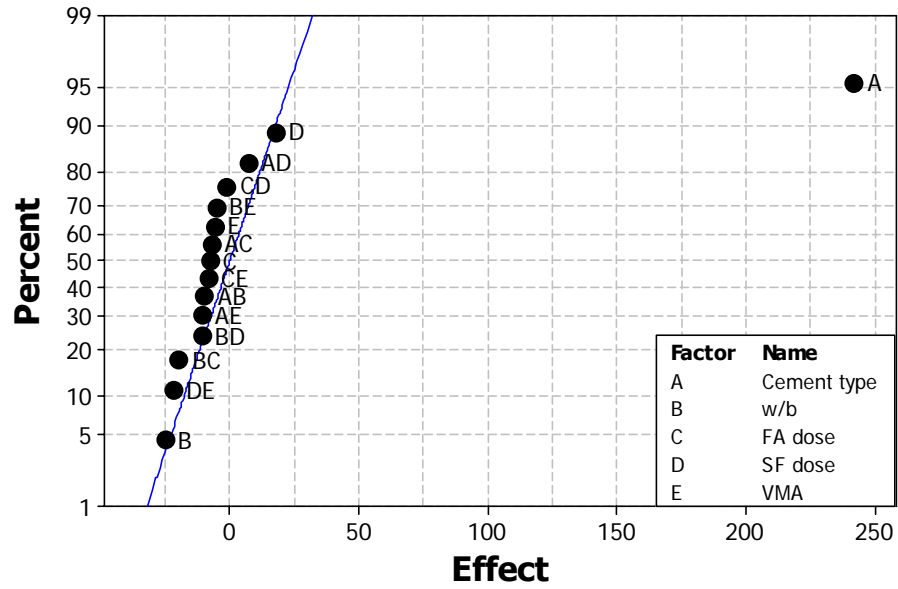


Figure 6-18. Normal probability plot for [K]

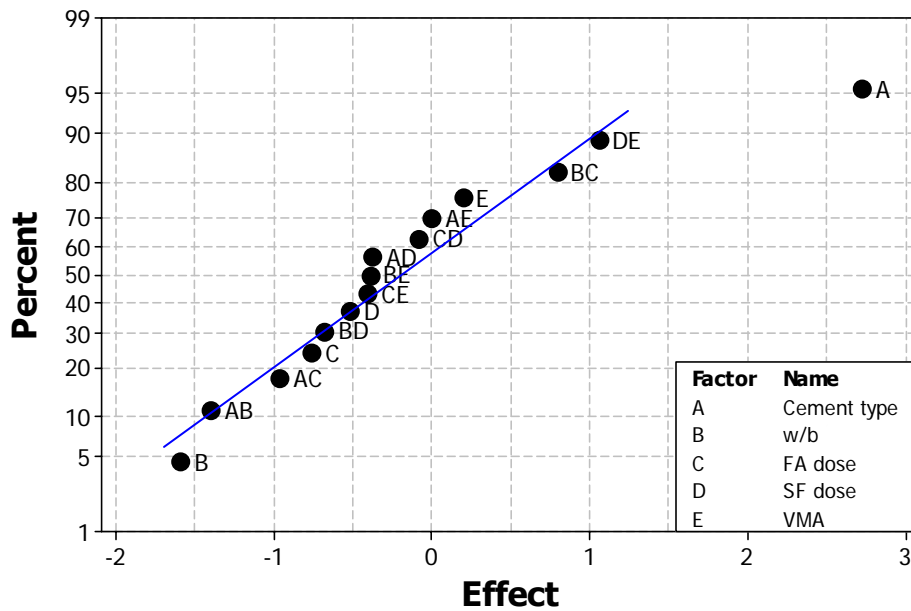


Figure 6-19. Normal probability plot for [Mg]

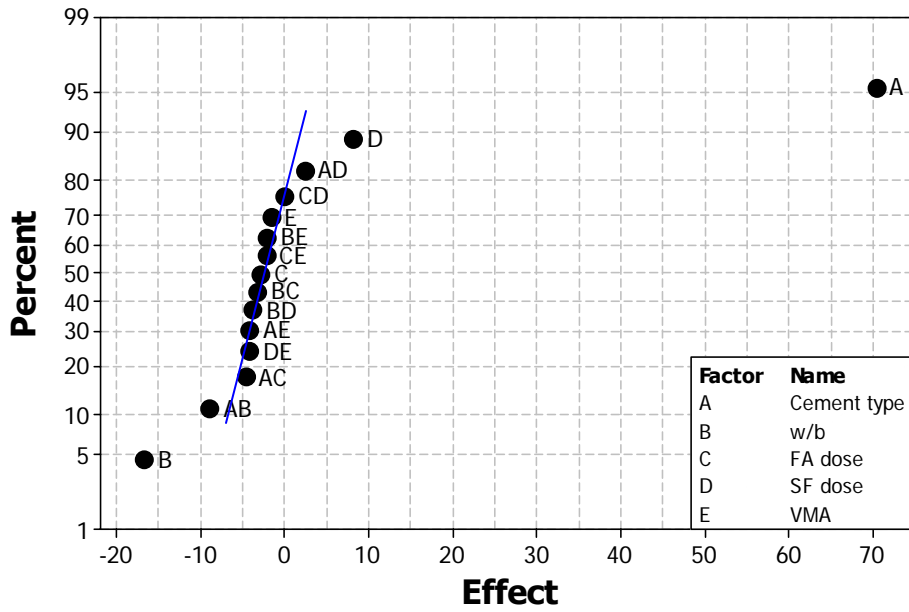


Figure 6-20. Normal probability plot for [Na]

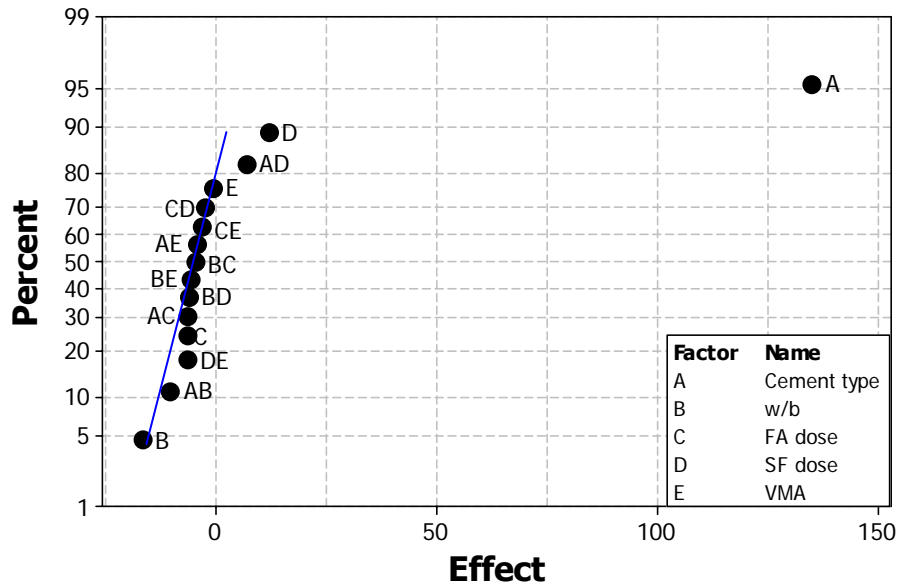


Figure 6-21. Normal probability plot for [S]

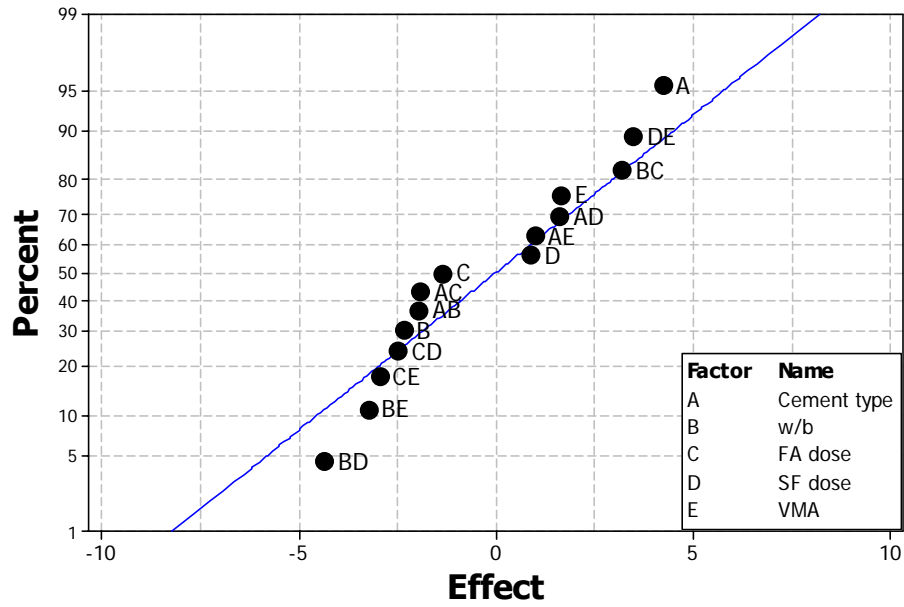


Figure 6-22. Normal probability plot for [Si]

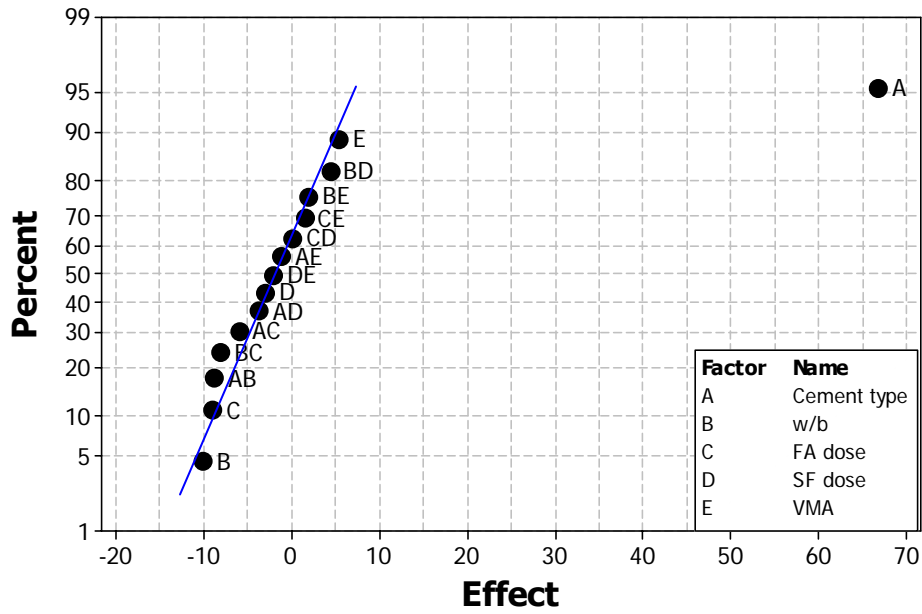


Figure 6-23. Normal probability plot for [OH]

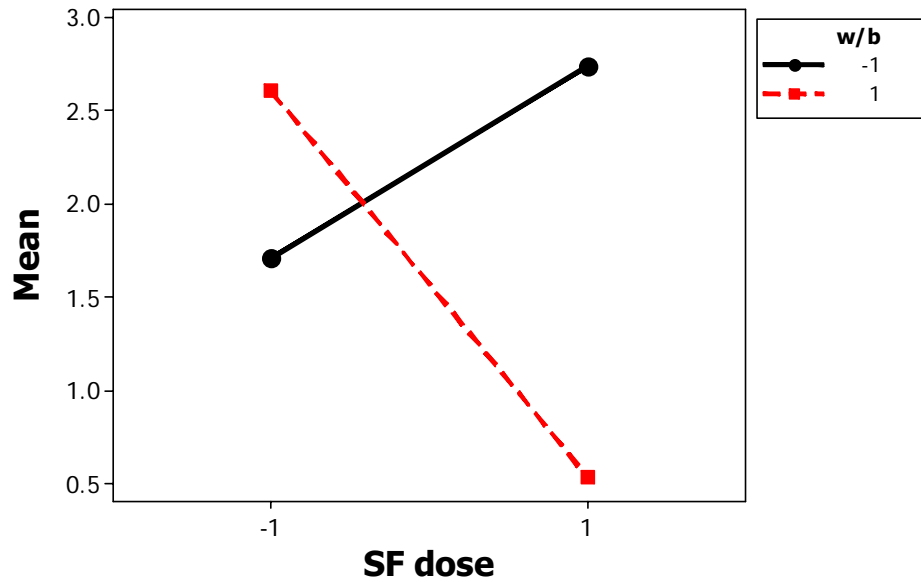


Figure 6-24. Effect of silica fume dosage and w/b ratio on Aluminum concentration (in coded units)

6.4 Discussion

In this section, the structural rebuilding of SCC cement paste suspensions will be linked to changes in mixture proportioning.

6.4.1 Understanding the correlation between paste matrix and structural rebuilding

6.4.1.1 Cement composition

The experimental results clearly show that the most significant factor influencing structural rebuilding within fresh cement pastes is the cement composition when the initial fluidity level is kept constant. All mixtures made with Cement 5 (higher C_3A /alkali) had significantly lower rates of rebuilding (an order of magnitude difference) than mixtures made with Cement 6 (lower C_3A /alkali). This is attributed in part to the different cement-superplasticizer interactions. As illustrated in Figure 6-25, mixtures made with Cement 5 (represented by even numbers in Figure 6-25) are shown to increase the superplasticizer demand compared to those made with Cement 6 (represented by odd numbers).

Figure 6-26, Figure 6-27, and Figure 6-28 show a comparison between the SP dosage and the sodium, sulfur, and potassium concentrations, respectively. It is postulated that this increase in superplasticizer dosage is necessary to counteract the increased cohesiveness of the pastes due to the reduction in repulsive forces. The double-layer thickness of ions surrounding a particle depends markedly on the ionic strength of the pore fluid, and, as the ionic strength increases, the thickness of the double-layer thickness rapidly decreases, which allows particles to come in closer contact.

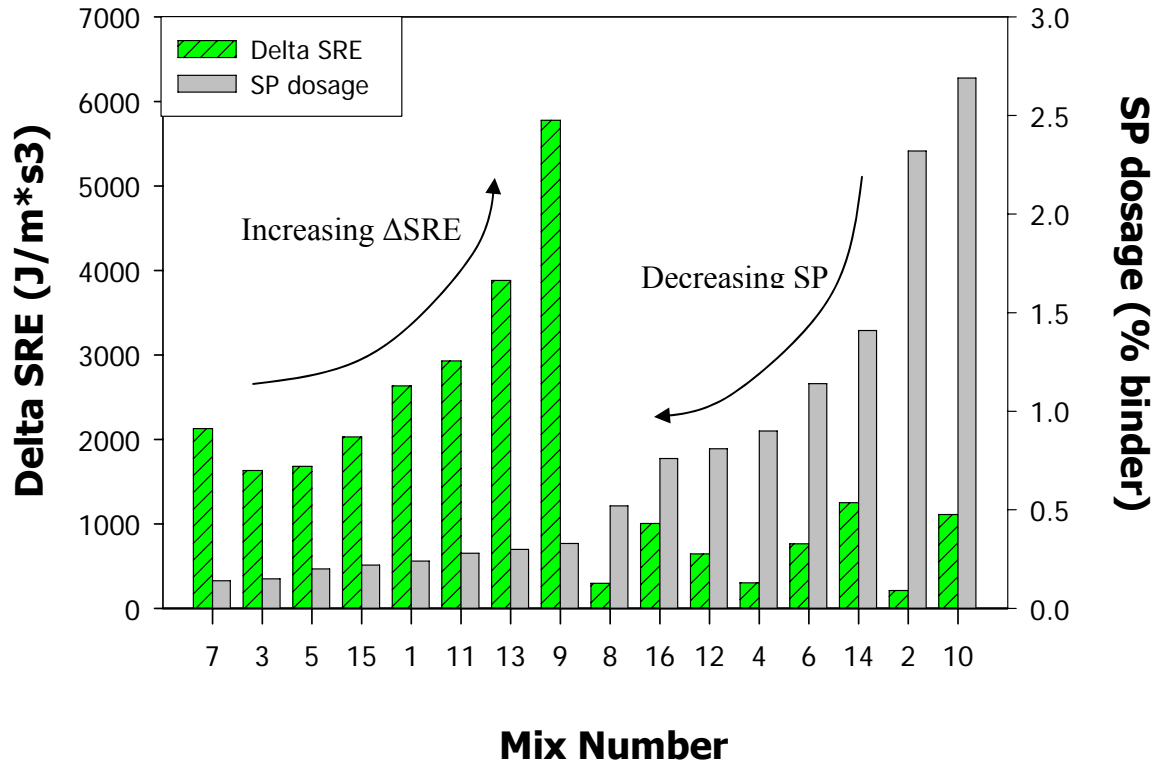


Figure 6-25. SP dosage and SRE (odd numbers indicate mixtures made with Cement 6)

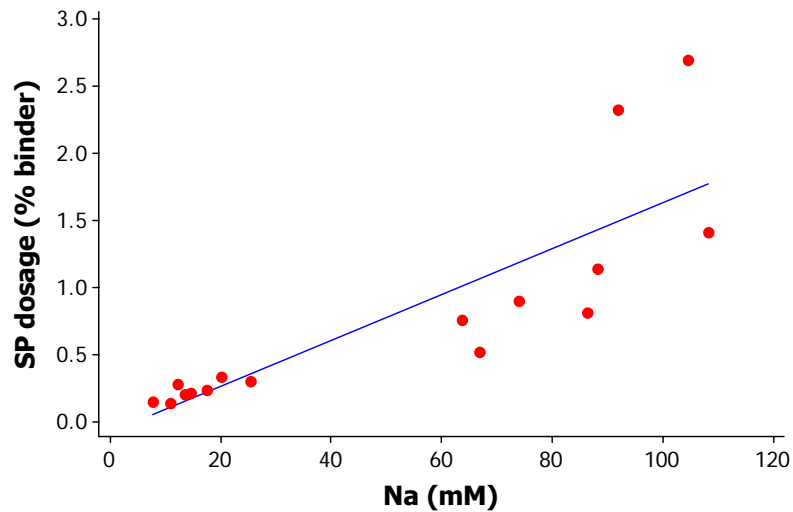


Figure 6-26. Relationship between superplasticizer demand and [Na]

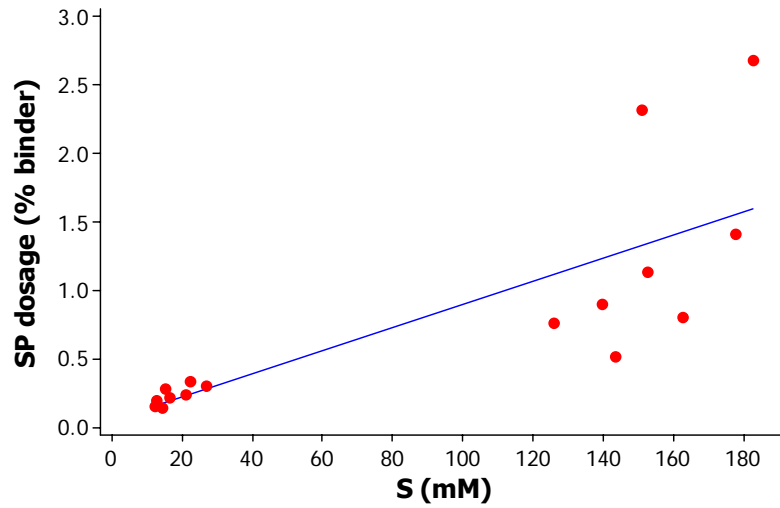


Figure 6-27. Relationship between superplasticizer demand and [S]

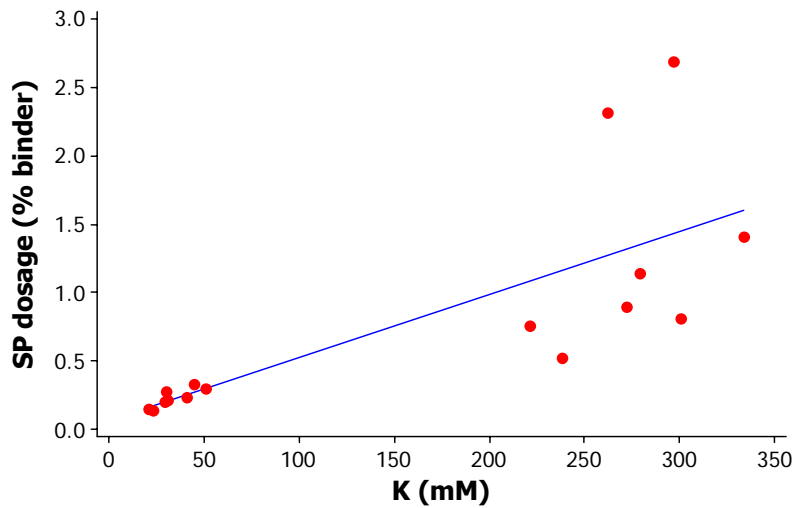


Figure 6-28. Relationship between superplasticizer demand and [K]

Increasing the concentration of soluble sulfates tends to decrease the fluidity due to a reduction in the adsorption of PC polymers on the cement particles and hydrated phases [116].

In addition, Kato et al. (as reported in [116]) indicated that in the presence of excess soluble sulfates the length of the polyethylene oxide (PEO) side chains contained in the admixture shrink, thereby decreasing the steric repulsion force. PC superplasticizers are preferentially adsorbed onto the aluminate phases (particularly C_3A) [15, 117], and, thus, an increase in C_3A content is likely to produce an increase in the superplasticizer demand. However, the formation of ettringite also involves the C_3A phases, and, thus, competitive adsorption between the sulfate ions and anionic PC molecules exists. Additional factors such as the particle size distribution, composition and structure of the tricalcium aluminate phase, type calcium sulfate, alkali sulfate composition, and hydration precipitates will also influence adsorption [15, 16, 96, 98, 118].

It is proposed that the vast differences in structural rebuilding behavior between the two cements used in this study are based on the following mechanisms:

- Anhydrous compounds of cement react with water. Rapid dissolution of alkali sulfates in combination with a release of Ca^{+2} and OH^- ions increases the attractive forces resulting in a more cohesive paste.
- Superplasticizer is introduced to the system. Superplasticizer preferentially adsorbs onto C_3A and ettringite resulting in higher superplasticizer demand in Cement 6 due to the higher C_3A content. Due to high concentration of sulfate ions in the pore solution, adsorption decreases causing the need for a higher superplasticizer dosage to achieve the target initial slump flow.
- Due to poorer and weaker adsorption, more “free” PC superplasticizer exists in the aqueous solution of pastes made with Cement 6. This results in a reservoir of superplasticizer molecules available to adsorb onto newly formed hydrates, which should result in better slump flow retention due to a more dispersed particle network.

6.4.1.2 Mineral Admixtures

Addition of silica fume was the second most significant factor influencing the rate of structural rebuilding. This is most likely due to the high fineness of the silica fume particles. As mentioned in Chapter 3, the strength of the force of attraction from the London dispersion forces is related to the size of the particle. Due to the high surface area of the silica fume particles, the London dispersion forces are likely to be more significant, and the silica fume particles will be more prone to aggregate. This would reduce the fluidity, and, as seen in Figure 6-29, superplasticizer demand increased when silica fume was added. In addition, since the cement substitution was conducted on a mass basis, the total solid volume content increased by approximately 2% when silica fume was used. (This is due to the differences in specific gravities among the cement, fly ash, and silica fume.) Although it is not possible to accurately determine how many more particles are in the paste, the increase in solid volume content may contribute to increasing the rate of structural rebuilding since the distance between flocs is reduced. However, it is likely that this effect is small since a faster rate of rebuilding was not seen in the fly ash mixtures. Rather the increase in structural rebuilding is probably more associated with the small size of the silica fume particles.

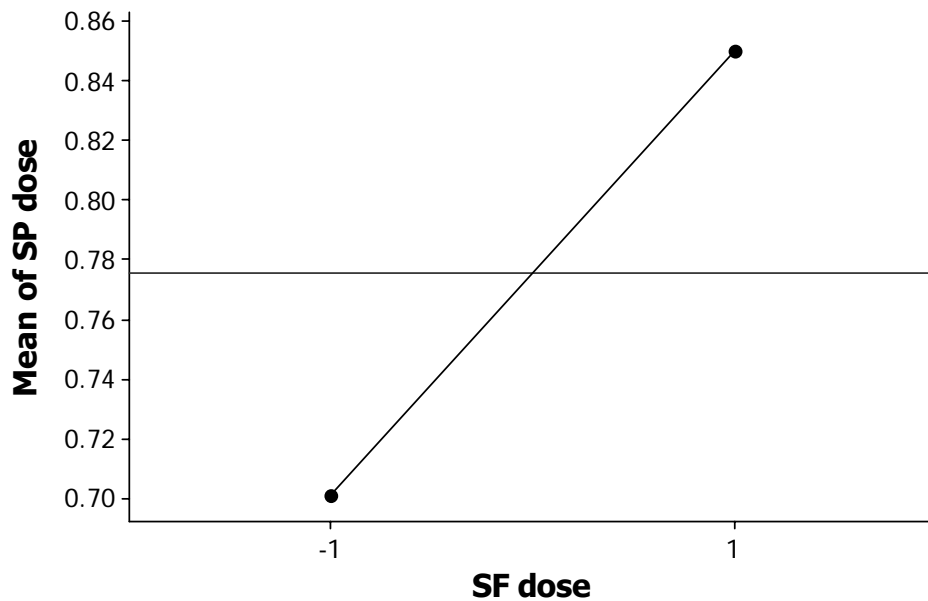


Figure 6-29. Main effect of silica fume on structural rebuilding

With regard to fly ash, a modest increase in the rate of rebuilding occurred when the fly ash dosage was increased in pastes made with Cement 5 (c.f. Figure 6-4). In mixtures made with Cement 6, an increase was seen in the rate of rebuilding. The particle size distributions of the cements were very similar, which should lead to similar packing densities and, thus, similar behaviors if the rate of stiffening is only due to particle packing. It was shown in Chapter 6.3.4 that increasing the fly ash content resulted in a reduction in sodium concentration, potassium concentration pH, and ionic strength when Cement 5 (high alkali / C_3A) was used. When Cement 6 (low alkali/ C_3A) was used, increasing the fly ash content decreased the pH and increased the ionic strength. Assuming the validity of the ICP test results, then the increase in the rate of rebuilding in the low alkali cements is related to an increase in the ionic strength of the pore solution.

6.4.1.3 Water-to-Binder Ratio and VMA

As expected, an increase in w/b ratio resulted in a decrease in the rate of structural rebuilding.

The solid volume concentration was reduced by approximately 13% when the w/b was changed from 0.3 to 0.4. Thus, there is more water to coat the particles and the interparticle forces of attraction are lessened.

Surprisingly, VMA was not a significant factor influencing the rate of rebuilding. Limited research has been conducted on VMA and structural rebuilding. Assaad and Khayat [119] evaluated the relationship between type and content of VMA on the structural rebuilding and formwork pressure of SCC. The cement composition, w/b ratio, and aggregate content were held constant. They observed that formwork pressure can be reduced when VMA is used at a low concentration. The authors also noted that using a cellulose-based VMA along with a PC-based superplasticizer increased the magnitude of the maximum pressure and decreased the rate of pressure drop when the VMA was used at low and medium dosage rates. This was attributed to an increase in fluidity retention from an increase in SP dosage. Furthermore, the authors noted that mixtures containing a cellulose-based VMA and a PC-based superplasticizer developed the highest degree of structural build-up, but at later ages the structural build-up decreased. Again, this was attributed to the increase in superplasticizer demand. Figure 6-30 shows a normal probability for the superplasticizer demand. The superplasticizer demand was most influenced by the cement type, w/b ratio, fly ash content, and interactions between the cement and fly ash. The presence of VMA resulted in a modest increase on superplasticizer demand (VMA is to right of zero) and, thus, does not fully account for the behavior seen in the VMA pastes. VMA will be discussed in more detail in Chapter 8.

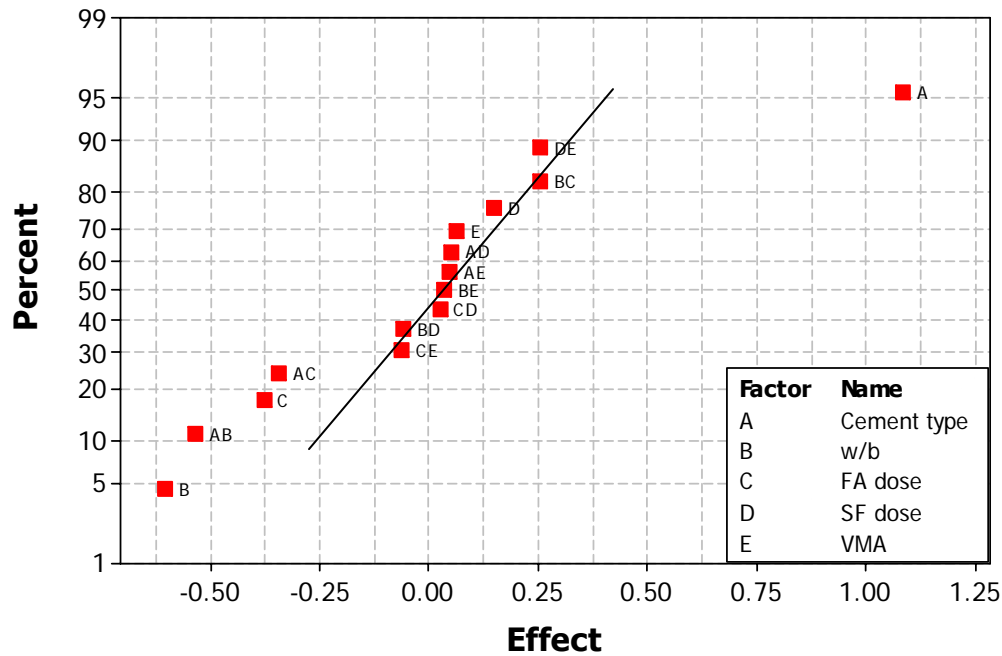


Figure 6-30. Factors influencing superplasticizer demand

6.4.2 Correlation between Structural Rebuilding and Initial Setting Time

The initial setting time signifies the physical state transition at which the consistency of fresh cement paste goes from an easily deformable paste to a non-moldable paste. Although initial set is based on an arbitrary measurement, the stiffening that occurs prior to initial set is controlled by chemical and physical processes from the hydration of cement compounds, and, hence, initial set is typically used for quality control to ensure that the cement does not abnormally set or to test for the response of a particular cement and admixture. As shown in Figure 6-7, the Vicat needle test is not sensitive enough to detect the gradual and progressive changes that are occurring within the microstructure. However, it is interesting to note that even though the time scale of the structural rebuilding and initial setting time tests were different (90 minutes compared to 300 - 600 minutes, respectively), an inverse relationship exists between the initial setting time and

rate of structural rebuilding (c.f. Figure 6-31). In general, mixtures exhibiting longer initial setting times had slower rates of rebuilding. As the superplasticizer demand increased, the setting time was delayed. This demonstrates that the cement-superplasticizer interactions also affect the physical processes controlling the development of the 3 dimensional network required for setting.

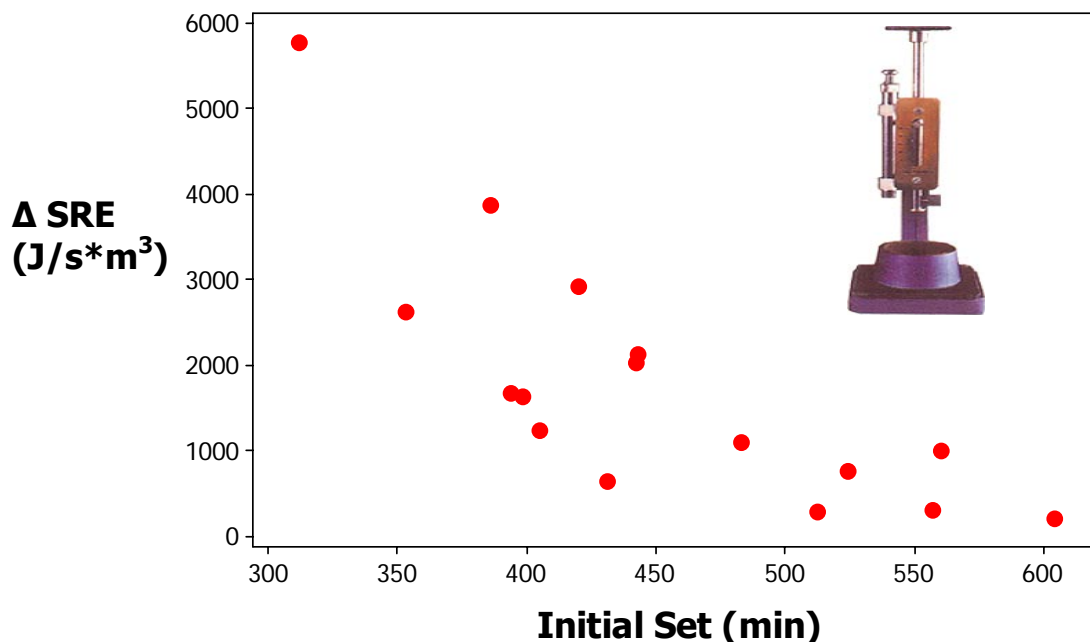


Figure 6-31. Relationship between rate of rebuilding and initial setting time

6.5 Chapter Summary

The objective of this chapter was to determine the relative importance of the different constituent ingredients on the structural rate of rebuilding. A comprehensive experimental program was carried out to determine the role of cement type, fly ash content, silica fume content, and w/b

ratio. Based on the foregoing data presented in this chapter, the following conclusions can be

made:

- Structural rebuilding is influenced by the mixture proportioning of the matrix phase. The rate of structural rebuilding was most strongly impacted by the cement chemistry, followed by silica fume incorporation, and changes in the w/b ratio. This is attributed to the significant impact that all three of these factors have on the interparticle interaction of the cement particles.
- The time-dependent structural build-up can not always be characterized using a linear function.
- Regardless of changes in the mixture proportioning, mixtures proportioned with the cement with the high alkali/C₃A content always had slower rates of rebuilding than the mixtures proportioned with the lower alkali/C₃A content. This highlights the importance of proper cement selection. The difference in the structural rebuilding is attributed to the significant differences in the pore solution chemistry, particularly the soluble alkalis and sulfate contents.
- A positive correlation was seen between the superplasticizer dosage and soluble alkalis and sulfates. This indicates that superplasticizer demand increased in order to counteract the reduction in the repulsive forces. Paste mixtures prepared with the lower alkali/C₃A cement necessitated lower superplasticizer demand than the higher alkali/C₃A cement. The results indicate that the structural rate of rebuilding is related to the interactions between the cement and superplasticizer.
- For a given fluidity condition and cement content, it was seen that the rate of structural rebuilding was more influenced by incorporation of silica fume versus changes in the w/b

ratio. Incorporation of silica fume increased the rate of rebuilding, and more pronounced effects were obtained when silica fume was added to mixtures with a lower w/b versus a higher w/b ratio. This is attributed to an increase in particle-particle interaction due to the reduction of interparticle spacing and enhanced flocculation from the silica fume particles.

- Changes in the fly ash content resulted in slight changes in the rate of rebuilding. The fly ash used in this study was coarser than Cement 5 and Cement 6. This is likely to decrease the packing density and interparticle spacing among the solid particles. No significant changes in the rate of rebuilding were seen with VMA addition.
- The time of initial set was found to be related to the overall rate of rebuilding. The rate of structural rebuilding decreased as the setting time increased. This indicates that the mechanisms governing the structural build-up and the mechanisms governing the development of the structural network required for initial set are closely linked.

**PART V: Relationship among Structural
Rebuilding, Formwork Pressure, and
Aggregation Mechanisms**

7 Chapter 7: Formwork Pressure of SCC and Influence of Structural Rebuilding

7.1 Introduction

As stated in Chapter 3, one of the key issues for designing formwork is determining the horizontal pressure exerted by the concrete during casting. This chapter presents a simple laboratory apparatus to simulate formwork pressure. The apparatus was used to monitor the formwork pressure behavior of three concrete mixtures. In Chapter 6, it was determined that the most significant factors influencing the rate of structural rebuilding were the type of cement and the incorporation of silica fume. The main intent of this Chapter is to determine to what extent the formwork pressure behavior of SCC can be changed through altering the mix composition. The results from this section of the work can aid in developing formwork pressure models that can better account for the influence of structural rebuilding.

7.1.1 Formwork Pressure Simulation Overview

7.1.1.1 Concept

In the field of soil mechanics, the triaxial shear test is a popular method for determining shear strength parameters. In this test, a specimen of soil encased in a rubber membrane is placed inside of a cylindrical chamber. The chamber is filled with a fluid, and then the specimen is subjected to a confining fluid pressure by applying an axial load to compress the fluid in the chamber [120]. Experience shows that the horizontal pressure exerted by the specimen is lower,

and decreases much faster, when the water is allowed to flow away from the pressurized volume (drained condition). In contrast, the values of the horizontal pressure are higher, and less prone to decrease, when an undrained condition is held during application of the axial load. The absence of water drainage and the immobility of the fluid phase (cement paste) are reasonable hypotheses for real casting of SCC into formworks. Hence, the same conditions have been maintained during the experiments reported in this paper, and special care was taken to avoid loss of fluid during the pressurization of the fresh concrete.

Simulation of the vertical loads generated by the concrete was performed using a cylinder-piston system (c.f. Figure 7-1) to pressurize cementitious materials. A laboratory mechanical testing system (MTS) was used to impose and control the axial load, and this force was transferred to the concrete via the piston. The simulation of the filling of a real formwork can be carried out by increasing the magnitude of the vertical load applied by the MTS to the piston. Different casting rates can be easily simulated by changing the rate at which the vertical load is applied. This device can be used to reproduce the real *in situ* casting conditions and, according to the structural needs, to better control the main variables involved in the selection of the best SCC mix and the placement procedure.

The height of any real column, H , can be simulated through the relationship:

$$N = A\gamma H$$

where A = cross sectional area of the piston

γ = specific weight of the material inside cylinder

N = vertical load applied by MTS

The applied vertical stress (hydrostatic stress) is calculated by dividing the vertical load applied by the cross-sectional area of the piston.

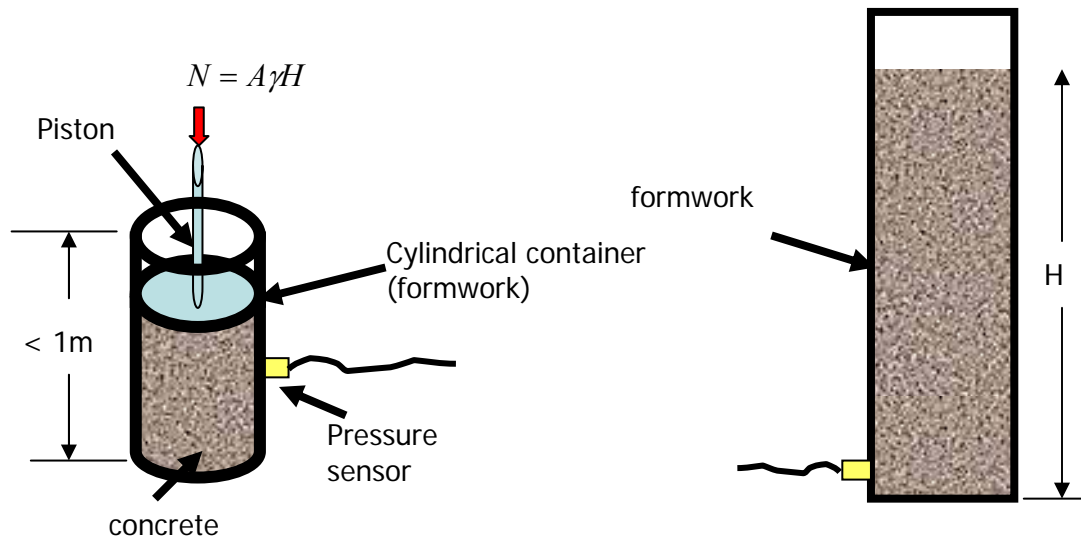


Figure 7-1. Schematic of formwork pressure system

The resulting lateral pressure depends on the vertical pressure applied. In the field of soil mechanics, there are three categories of lateral pressure, and each depends on the movement experienced by the vertical wall on which the pressure is acting. The *at-rest* lateral pressure occurs when the wall does not move laterally. The *active* lateral pressure develops when the wall moves away from the soil, whereas the *passive* lateral pressure occurs when the wall moves into the soil. In the field, formworks for concrete are designed to remain stationary and are constrained against lateral movement. In the case of the formwork simulation test, the walls of the cylinders do not move, and, thus, the lateral pressure developed is similar to the *at-rest* lateral pressure (for simplicity it will be referred to as lateral pressure). The horizontal stress acting in a

soil at a specific point can be determined by multiplying the vertical stress at that point by the ratio.

The ratio of the horizontal lateral stress (σ_h) to the vertical stress (σ_v) depends on lateral pressure coefficient, K:

$$K = \frac{\sigma_h}{\sigma_v}$$

The value of K can range from 0 to 1 and is related to shear strength of the soil. The shear strength of a soil is the internal resistance that the soil mass can offer to resist failure and sliding [120]. According to Mohr-Coulomb failure criteria, the shear strength of soil at failure is

$$\tau_f = c + \sigma_n \tan \varphi$$

where τ_f is the shear strength, c is the cohesion, σ_n is the normal stress acting on the failure plane, and φ is the internal angle of friction. Cohesion is the component of shear strength that is independent of interparticle friction, and it is related to the forces that hold together the particles within soil. Cohesion arises from electrostatic forces and adhesion from soil cementing processes. The angle of internal friction is the component of shear strength that is independent of cohesion, and it arises from the friction between grains [121]. Particle interlock also plays a role in shear strength and is believed to contribute to the component of strength that arises from the apparent cohesion [121]. In the formwork simulation test, K can be determined directly from the ratio between the measured pressure recorded by pressure transducers and the vertical pressure applied by mean of axial load. The K value will depend on the shear resistance of the concrete, and this shear strength is a function of the frictional resistance and interlocking

between the aggregate and cement powders [122], cohesion from the hydration process [5, 122], and cohesion due to thixotropic phenomena. If, and how much, K will change is essentially the major concern regarding the formwork pressure of SCC. Initially, during the casting of SCC the concrete behaves more fluid-like, and it may be assumed that K is equal to 1. However, in order for there to be a reduction in the maximum formwork pressure value, at some point during the casting, K must be less than 1 for the already placed concrete.

The role of friction at the interface between the formwork and concrete was recently investigated [123]. It was suggested that the lateral pressure decay is due to increasing shear stresses at the walls of the formwork. This increase in shear stress at the walls of the interface was shown to be correlated to the time-dependent yield stress due to structural build-up and formwork roughness. This suggests that formwork pressure will be higher when the frictional effects between the concrete and formwork are reduced (i.e. as in the case of a smooth column or formwork with oil) due to lower shear stresses at the wall. In other words, a rough wall would carry a larger part of the vertical load than a smooth wall, and, therefore, the lateral stress is higher than when a smoother wall is used (assuming all other parameters are equal). In the field, it can be assumed that the frictional effects between the concrete and form are minimized since form oil is used. The results from the aforementioned study also suggest that formwork pressure may be reduced when taller or wider forms are used due to increased shear stresses (assuming the material properties and casting rates are the same). Research by Assaad showed that the initial formwork pressure was slightly reduced when wider columns were used (96% hydrostatic pressure for the wider column compared with 99% hydrostatic pressure for the taller column); however, the same study showed that higher initial lateral pressures were seen in shorter columns

[5]. The influence of formwork geometry and interfacial friction between the form and concrete is an area that requires further study.

7.1.1.2 Formwork Pressure Apparatus and Measurement

A metal cylinder measuring 152 mm (6.0 inch) in the inner diameter, 300 mm (12.0 inch) in height, and 5 mm (0.20 inch) in wall thickness was used as the formwork (Figure 7-2). The cylinder consists of two symmetric halves held together by four screws and a circular steel plate as the base (which was bolted onto the cylinder). For ease of cleaning, it was decided to design the apparatus such that both sides of the cylinder and the bottom plate of the cylinder could be removed. The piston portion of the apparatus was made of steel, and a hole was drilled through the piston to serve as an air release valve. Duct tape was used to seal the gap where the cylinder's two ends were screwed together in order to prevent leakage from the system. Plumber paste and o-rings were used to seal the gaps between the inner surface of the cylinder and the perimeter of the piston and between the plate and the cylinder. Prior to each test, the inner surface of the form was lightly coated with form oil. Once the concrete was poured into the cylinder, the piston was lowered onto the surface of the fresh concrete. After ensuring that no external air was trapped in the system, the air release valves were closed and the pressurization began.

To record the horizontal pressures generated by the concrete during the pressurization, the cylinder was instrumented with two diaphragm pressure cells positioned at mid-height (Figure 7-2). Each pressure cell had a 400 kPa (58.0 psi) capacity and was calibrated prior to use. Measurement of total and pore water pressures were carried out: the former by a sensor directly touching the concrete surface (Figure 7-3), and the latter by means of a sensor in a

chamber hydraulically connected to the concrete volume inside the formwork (Figure 7-4).

An automatic acquisition system with scanning voltage of 4 V was used.

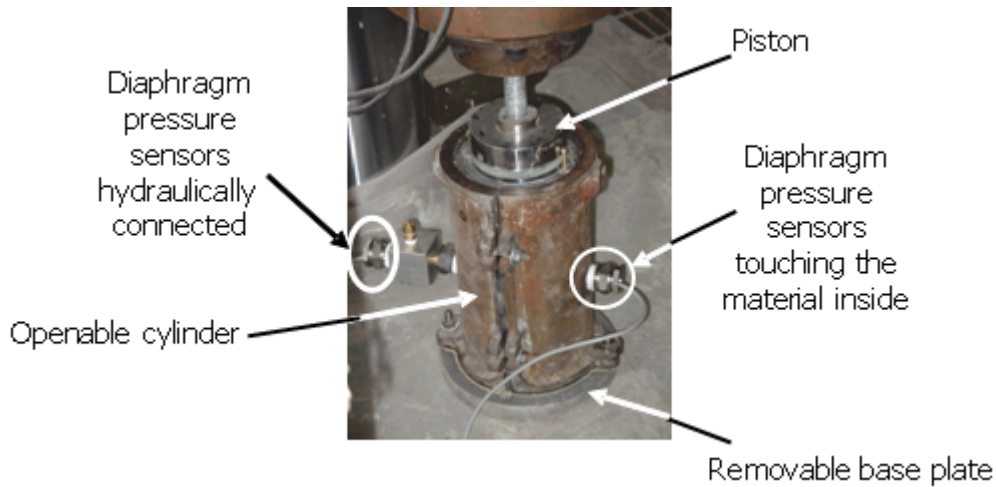


Figure 7-2. Instrumented Apparatus

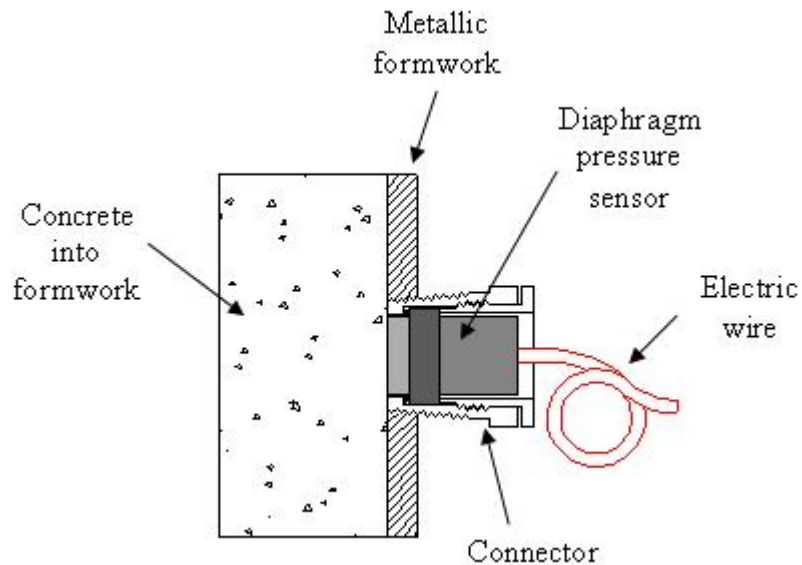


Figure 7-3. Measurement of the total pressure: the sensor touches the concrete

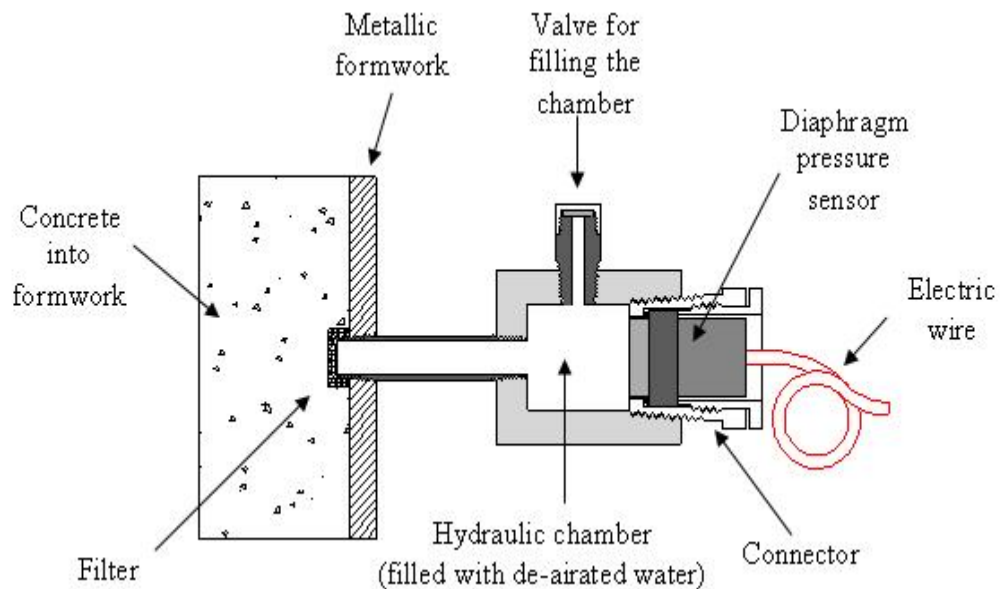


Figure 7-4. Measurement of the pore water pressure: the hydraulic connection

7.2 Experimental Program

7.2.1 Materials and Mixture proportions

Two cements, a class F fly ash, and silica fume were used. Cement replacement was conducted on a mass basis. The physical and chemical properties of the cements were given in Section Chapter 6.2. River sand and natural round coarse aggregate (nominal size 19 mm) were used, and the physical properties and particle size distributions of the sand and coarse aggregate are given in

Table 7-1. A liquid water-soluble cellulose-based VMA with a specific gravity of 1.002 and solids content of 70% was employed in all the mixtures. The superplasticizer was a carboxylated polymer in aqueous solution (specific gravity = 1.04).

Table 7-1. Particle size distribution and physical properties of coarse and fine aggregates

Screen Size (mm)	Coarse Aggregate % Passing	Screen Size (mm)	Fine Aggregate % Passing
19	100	9.5	99
12.5	74	4.75	88
9.5	46	2.36	68
5.66	23	1.18	46
4.75	19	0.6	30
2.36	10	0.3	16
1.18	10	0.15	4
		0.075	2
		0	0
Specific gravity	2.6	Specific gravity	2.7
Absorption (%)	2.1	Absorption (%)	2.3

The mix compositions of the concretes corresponding to a volume of 1000 m³ are presented in Table 7-2. The volume percentage of paste was kept constant at 36% for all the mixes. All mixtures were proportioned using a gravel-to-coarse aggregate ratio of 0.47 and w/b ratio of 0.35. The water content was adjusted to account for aggregate absorption. The superplasticizer dosage and VMA dosage were calculated based on the weight of the aqueous solution relative to the weight of the binder materials. A constant VMA dosage of 400 ml/100 kg of binder was used in all of the mixtures. The superplasticizer content was adjusted until an initial slump flow of 670 ± 12.5 mm (approximately 26.5 ± 0.5 inch). Paste mixtures corresponding to the matrix phase of the concrete were also tested.

Table 7-2. Composition of SCC mixtures

Mix	Cement Type	Cement (kg)	Fly ash (kg)	Silica Fume (kg)	Water (kg)	Gravel (kg)	Sand (kg)	VMA (kg)	SP (kg)
Cem6	Cement 6	417	104	0	182	792	893	2.09	1.93
Cem6SF	Cement 6	369	103	41	180	792	893	2.06	3.57
Cem5	Cement 5	417	104	0	182	792	893	2.09	6.2

7.2.2 Sample Fabrication and Testing Program

7.2.2.1 Concrete Fabrication

The concrete mixtures were prepared using a drum mixer with a rotation speed of 25 – 27 rpm. The mixes were prepared by mixing the coarse and fine aggregates for 1 minute, after which one-third of the water was added, and mixing continued for an additional minute. Then the cement, mineral admixtures, superplasticizer, and additional one-third of the water were added. After 3 minutes of mixing, VMA and the remaining portion of water were added and mixing progressed for another 3 minutes. The total preparation time for the concrete sample was 10 minutes. The concrete temperature was recorded after mixing followed by the (inverted) slump test (ASTM C1611 [10]) and visual stability test. The visual stability test is a subjective test that is used to determine the dynamic stability of slump flow patty [124]. In addition to the T_{50} time, the time at which the concrete reached the final slump flow diameter, T_f , was recorded. Recent studies indicate that time to final spread is better correlated to the viscosity than T_{50} time [125]. After measuring the slump flow, the concrete was remixed for 30 seconds prior to conducting the formwork pressure tests.

7.2.2.2 Concrete Formwork Pressure Simulation Procedure

The elapsed time between the end of mixing and the beginning of the formwork pressure simulation test was approximately 10 – 15 minutes. The casting height selected for the simulations was 10 m (33 ft), and the casting rate was fixed at a rate of 7 m/hr (23 ft/hr). For an average concrete density of 2400 kg/m³ (150 lb/ft³) and the dimensions of the formwork pressure simulation apparatus, this requires the final load of approximately 4.2 kN (944 lbf) to be reached in about 1.5 hours. The load was maintained until the time of pressure cancellation. During the simulations, the values of the horizontal pressure exerted by the tested material were recorded each minute during the period in which the load was being increased, and then once every five minutes during the period in which the axial load was held constant. In all tests, a displacement ranging from 3 -5 mm was noticed during the time in which the axial load was kept constant by the MTS. This can be due to rearrangement and consolidation of the particles. In addition, it may also provide some insight about volumetric changes due to autogenous shrinkage; however, that is outside of the scope of this dissertation.

7.2.2.3 Paste Fabrication

Paste mixtures corresponding to the tested concrete mixtures given in Table 7-2 were prepared for structural rebuilding rheological testing (c.f. Chapter 4). A Hobart mixer (ASTM C305[126]) was used to mix all the paste samples. The mixing sequence consisted of homogenizing the dry powder materials for one minute using at speed 1 (136 rpm), and then over a 30 second interval a portion of the water and all of the superplasticizer were added. The paste was mixed for 2 minutes (at 136 rpm) after which the mixing was paused so that the sides of the bowl could be cleaned. The VMA and the remaining water were added, and mixing was restarted for an

additional 2 minutes (at 136 rpm). The mixing was paused a second time to clean the bowl, and finally the paste was mixed for an additional 2 minutes at the higher speed (281 rpm). The total preparation time for the paste mixtures was 9 minutes. After mixing, the paste slump flow tests and rheological tests were conducted.

7.3 Results

7.3.1 Fresh State Properties

Table 7-3 summarizes the fresh state properties for the concrete and the corresponding paste matrix. The viscosity was calculated by fitting the Bingham model to the data shown in Figure 7-9. Cem5 was very cohesive and exhibited excellent segregation resistance, which can be attributed to the high viscosity of this mix (as seen in the T_{final} reading of the concrete and the viscosity measurement of the paste). The paste flow diameter of Cem6 was slightly lower than the other two mixtures indicating that the paste matrix of this concrete would have a higher yield stress. Slight bleeding and more air bubbles were observed in Cem6, and, in general, this mixture was more sensitive to variations in the superplasticizer dosage than the other mixtures. The T_{50} time indicates that the relative viscosity of the Cem6SF concrete is likely to have a higher viscosity than the Cem5 concrete. However, due to the very short duration of the T_{50} test and it being very sensitive to the user, it is difficult to compare results conducted by different users due to the high variability of this test. Although the results show that the relative viscosity of Cem6 is probably lower than Cem6SF, it can not be said with absolute certainty that the relative viscosity of Cem5 differs from that of Cem6 or Cem6SF. The T_{final} time indicates that relative viscosity of the Cem6SF concrete is likely to have a higher viscosity than Cem5 concrete. Furthermore, the T_{final} time is able to distinguish between the flow behavior of the

Cem5 concrete and Cem6 concrete, and the results indicates that the relative viscosity of

Cem5 concrete is higher than Cem6 concrete. The T_{final} time does not show an appreciable difference between the viscosity of the Cem5 concrete and Cem6SF concrete, which is probably due to the influence of the yield stress. However, it is seen that the slump flow diameter of Cem5 concrete is slightly larger than the slump flow diameter of Cem6SF concrete. Taking this into account, the results seem to suggest that the Cem5 concrete has a slightly lower viscosity than Cem6SF concrete.

Table 7-3. Fresh state properties for concrete and paste

Mix	Concrete						Paste	
	Slump flow (mm)	T_{50} (s)	T_{final} (s)	VSI	Temp. ($^{\circ}\text{C}$)	Density (kg/m^3)	Slump flow (mm)	Viscosity (Pa/s)
Cem6	667	4.8	37	1.5	22.9	2440	421	0.050
Cem6SF	667	10.0	67	0	22.9	2424	405	0.066
Cem5	673	7.0	66	0	24.1	2446	391	0.098

7.3.2 Lateral Pressure and Pore Water Pressure Evolution

Initially, both pressure sensors followed similar paths, and the total lateral pressure (P_t) and the pore pressure (P_p) both reached similar peak values. However, after the peak, the two readings deviated and the total lateral pressure remained constant, whereas the pore water pressure continued to decrease. The constant lateral pressure reading is due to the fact that the concrete is able to retain its shape, and the deformation of the transducers diaphragms is no longer reversible against the hardening material. Improvements could be obtain by applying a thicker layer of grease between the concrete and the total lateral pressure transducer, but this only resulted in slightly delaying the time at which the total lateral pressure deviated from the pore pressure

This was similar to results reported in another study [127] and brings into question the use of monitoring lateral pressure with diaphragm pressure transducers that are in direct contact with a stiffening material. An alternative procedure to monitor the evolution of the total pressure using a displacement-type control procedure was developed by Andriamanantsilavo et al. [128]. The total pressure exerted by cement paste against a mobile surface was measured against that required to push against another fluid on the other side of the same surface and keep it in equilibrium. By using this rigorous technique, it was demonstrated that the total and pore water pressure decreased identically to zero during setting of cement pastes showing the equivalence between total and pore water pressure when proper testing protocol is used. Since the pore water pressure is easier to directly measure than the total lateral pressure for a setting material, the pore water pressure was used to monitor the pressure evolution in this study.

In soil mechanics, the total stress at a particular point in a saturated soil mass (with no seepage of water) is divided into two parts: pore water stress and the effective stress [120]. The pore water stress is the stress carried by water in the continuous void spaces, and it acts in equal intensity in all directions. The effective stress is the stress carried by the soil solids at their points of contact [120]. This is represented mathematically by the following formula

$$\sigma = \sigma' + u$$

where σ is the total stress at the elevation of the considered point, σ' is the effective stress, and u is the pore stress. The above equation is based on the assumption that the fraction of cross-sectional area occupied by the solid-to-solid contacts is negligible. As mentioned in the preceding paragraph, it was found that in concrete the total stress was equal to the pore pressure.

Thus, it appears that prior to pressure cancellation the concrete acts as a granular structure in a dispersed state.

7.3.3 Formwork Pressure of SCC

The variation of lateral pressure with respect to time and concrete head is shown in Figure 7-5. The SCC mixture proportioned with Cement 5 exerted the lowest lateral pressure: 70% of the hydrostatic value, compared to 80 % for mixture Cem6SF and 88 % for mix Cem6. Figure 7-6 shows evolution of pressure from the point at which casting has ended (denoted at $t = 0$ hr) to the time of pressure cancellation (note: the lateral pressure at each time was normalized with respect to the maximum lateral pressure exerted by the concrete). Although mixture Cem5 developed the lowest maximum pressure, it displayed the slowest pressure drop. The fastest pressure decay was seen in mixture Cem6, while both mixtures made with Cement 6 cancelled out at the same time.

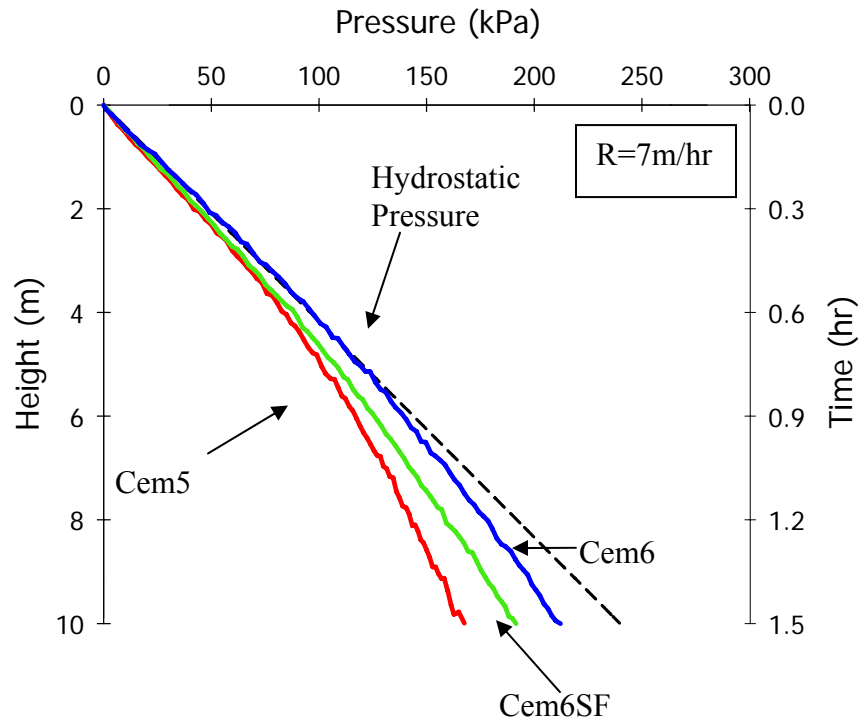


Figure 7-5. Evolution of pressure reading with time and height

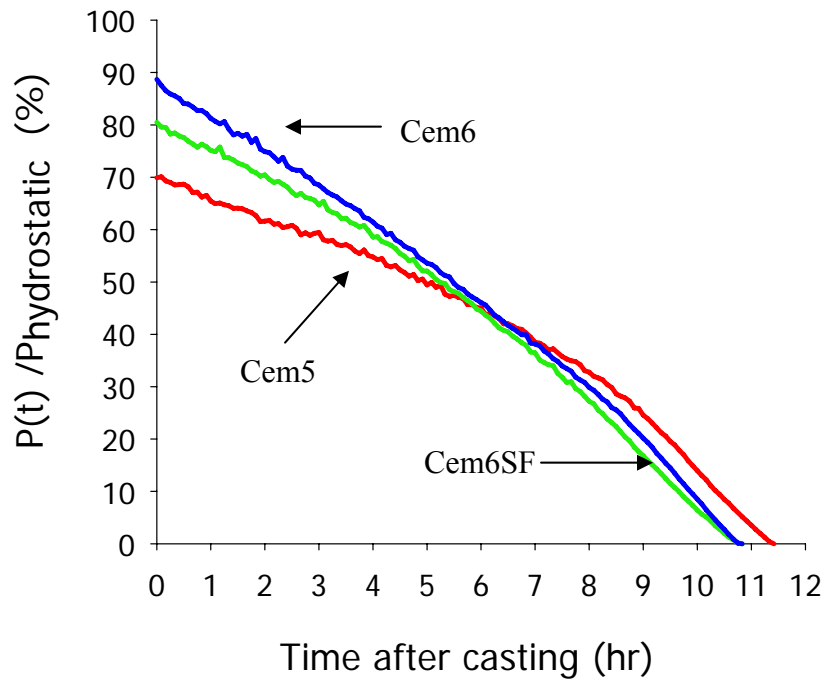


Figure 7-6. Evolution of pressure decay, where $P(t)$ is the formwork pressure of the concrete at a specific time. $P_{\text{hydrostatic}}$ is constant and corresponds to the total vertical pressure applied at the end of casting (approximately 240 kPa).

7.3.4 Structural Rebuilding

Attempts were made to characterize the mortar phase by sieving portions of the fresh concrete through a 0.15mm sieve, but difficulties in extracting a representative sample lead to highly variable results. Thus, the rheological protocol described in Section 4.4 was used to characterize the structural rebuilding within the representative paste matrix. The paste samples were prepared using the same mix proportions as the concrete, except that no aggregates were used and no additional water was added to account for aggregate absorption since no aggregates were used. The structural rebuilding curves are plotted in Figure 7-7.

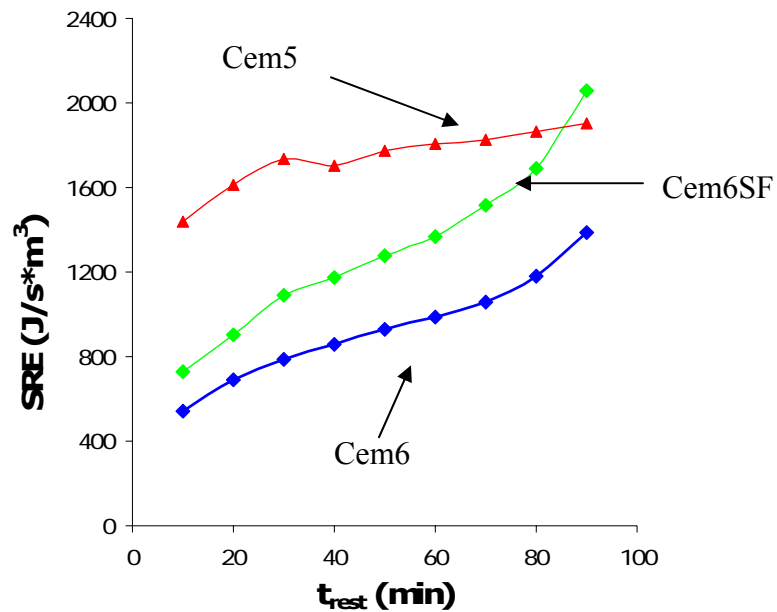


Figure 7-7. Structural rebuilding curves for representative paste matrix

The results were in agreement with those obtained in Chapter 6. The paste mix proportioned with Cement 5 had the lowest ΔSRE value (approximately half that of the mix made with Cement 6), and the cement paste proportioned with Cement 6 with silica fume had the fastest rate of rebuilding. The reasons for this behavior were discussed in Chapter 6.

7.4 Discussion

Surprisingly, the paste matrix that had the slowest rate of rebuilding yielded a concrete that had the lowest maximum pressure. This is attributed to the different rheological properties of the mixtures. Although Cem5 had the lowest ΔSRE value, the magnitude of its SRE values (c.f. Figure 7-7) was much higher than those of Cem6 and Cem6SF. This was especially true during

the earlier time periods. The magnitude of the SRE value indicates the total amount of structural build-up that occurred in the mix, while the slope of the line SRE vs. t_{rest} curve and the Δ SRE value provide a measure of the rate of change of structural rebuilding. Hence, although Cem5 did not have a fast rate of structural rebuilding, the total amount of structural build-up is higher than the other two mixtures (except for at 90 min at which Cem6SF has a higher structural build-up than Cem5).

From Figure 7-7, it is seen that during the first 80 minutes, the SRE value for Cem5 is higher than that of the other two mixtures, and, at 90 minutes, the magnitude of SRE is higher in Cem6SF than Cem5 (2060 J/s*m³ compared to 1903 J/s*m³). (Cem6 had the lowest SRE at 90 minutes (1390 J/s*m³.) 90 minutes corresponds to the end of the casting (c.f. Figure 7-5), and, at time, the vertical stress corresponds to the full weight of the simulated concrete column. The high SRE value of Cem5 indicates that a high amount of cohesion was developed in this mixture. Thus, the yield stress and/or apparent viscosity of Cem5 are likely to be substantially greater than the other two mixtures. Since SRE is calculated based on the area between the upcurve of each hysteresis loop and the equilibrium line (see Section 4.3.3), the position of the equilibrium line is important. Let us suppose we have a mixture called Mix X. If Mix X has the highest SRE value and the position of its equilibrium line is higher than, or similar to, the other mixtures that it is being compared against, then the microstructure of Mix X (at the particular resting time in question) is more aggregated, which would result in an increase in at least one of the rheological properties (yield stress or viscosity). Similarly, if Mix X has the lowest SRE value and the position of the equilibrium line is less than, or similar to, the other mixtures, then there are fewer (or weaker) particle associations within the microstructure of Mix X, which should lead to a lower rheological response. However, an interesting situation can arise if the SRE value is high

and the equilibrium line is lower than the other mixtures. Consider Figure 7-8a: it is obvious that Mix A has a higher SRE value than Mix B. However, if the rheological properties were extrapolated from fitting the Bingham model to the up curve, then the viscosity and yield stress of a given mixture depends on the shape of the up curve. As shown in Figure 7-8b, the yield strength of Mix B is greater than the yield strength of Mix A even though $SRE_B < SRE_A$ (Mix B and Mix A has the same viscosity). In Figure 7-8c, $SRE_C < SRE_A$ and the yield stress of Mix C is less than Mix A, but the viscosity of Mix C (characterized by the slope of the up curve) is greater than the viscosity of Mix A. Hence, the SRE value should be used in conjunction with the equilibrium line value in order to fully describe the behavior of a mixture. The equilibrium lines are plotted in Figure 7-9. The equilibrium line of Cem5 is higher than the other two mixtures, and this supports the hypothesis that the high SRE value in Cem5 is indeed due to a more aggregated microstructure.

Using the rheological studies results, a better understanding of the resulting formwork behavior is obtained. Initially the concrete behaves as a fluid and produces a hydrostatic pressure that acts laterally on the vertical forms; all the vertical stresses are transferred to horizontal stresses (i.e. $K = 1$). However, if cast slowly enough or if at rest, then the concrete has time to build up an internal structure, and all the vertical stresses are not entirely transferred into lateral stresses (i.e. $K < 1$). All of the tested concretes exerted lateral pressures that were less than hydrostatic, ranging anywhere from 70 to 90 percent of the hydrostatic value. Formwork pressure can be reduced from aggregate interlocking [97], but since all the mixtures were designed to have the same granular phase, it is assumed the differences in formwork pressure behavior are due to the changes in the matrix phase.

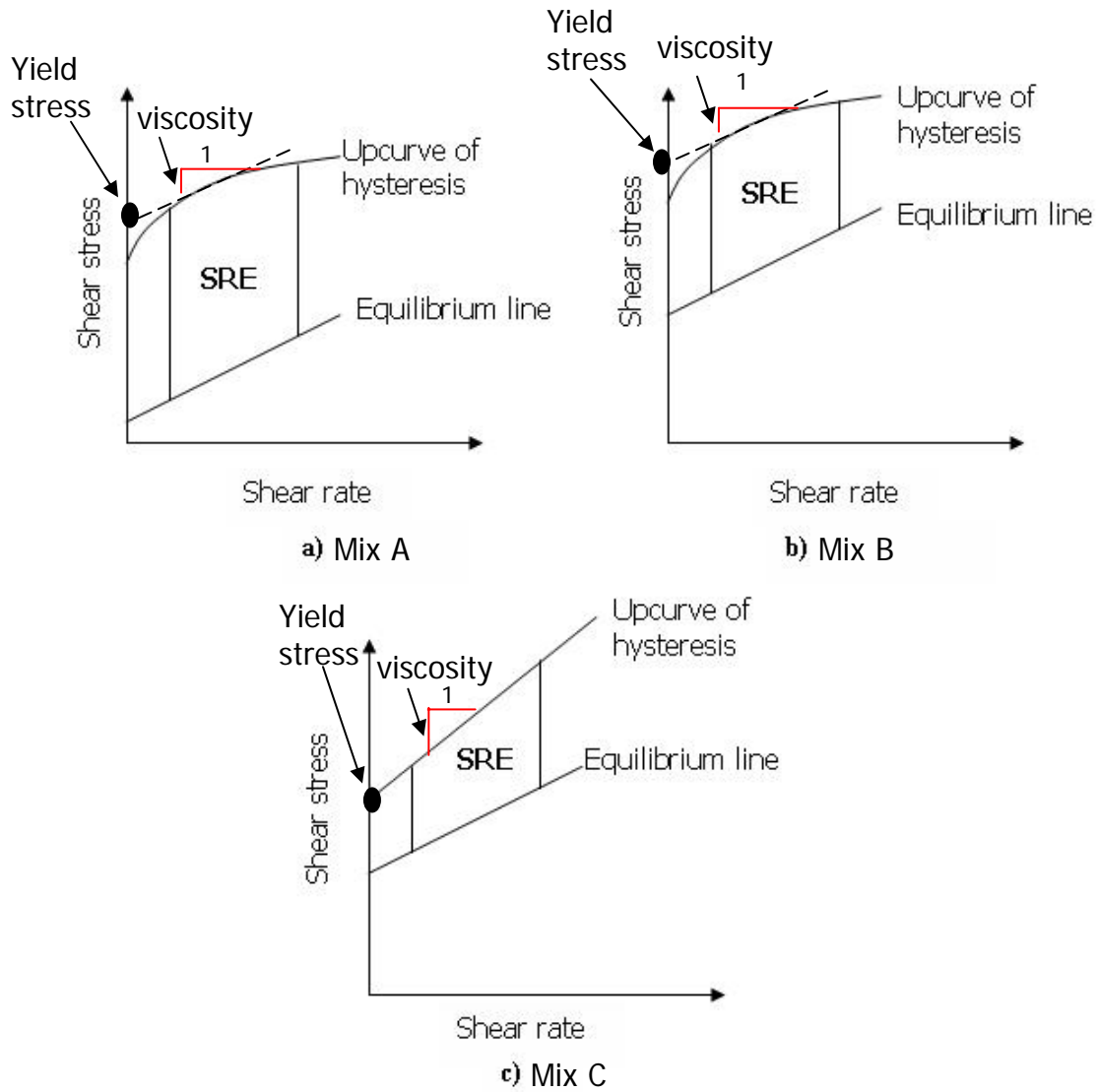


Figure 7-8. Depiction of different structural conditions based on equilibrium line location

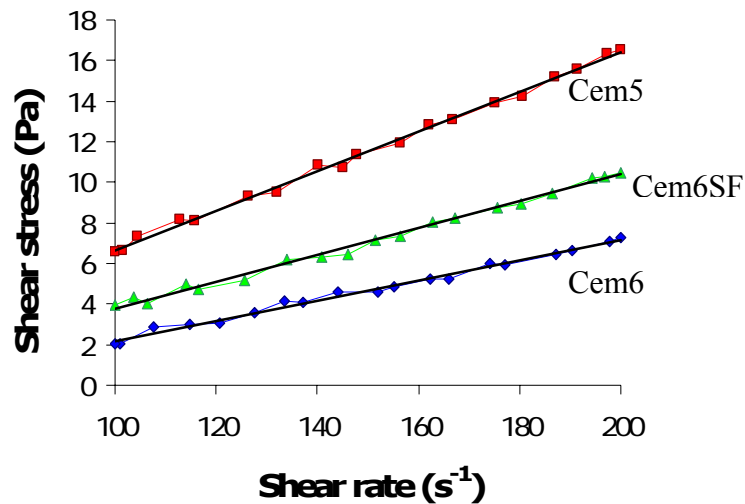


Figure 7-9. Equilibrium lines for representative paste mixtures

During the 90 minute casting process Cem5 had the lowest rate of rebuilding, but as shown in Figure 7-7, and Figure 7-9 (imagine extrapolating to zero shear rate), the shear strength (represented in the yield stress value) of this mix was significantly higher than the other two mixtures. This difference in mechanical behavior is largely attributed to differences in the pore solution chemistry. Paste mixtures proportioned with Cement 5 showed extremely high concentrations of alkali and sulfate, which effectively compressed the electrical double layer and reduced the electrostatic repulsive forces among the particles so that more interparticle links developed. The higher shear strength of Cem5 enables less of the vertical stresses to be transferred horizontally. Cem5 illustrates the importance of both the instantaneous amount of structural rebuilding (i.e. the magnitude of structural rebuilding at a given time) and the overall rate of structural rebuilding (determined from the difference between SRE values at 90 minutes and 10 minutes). It was already shown that the high value for the instantaneous amount of structural rebuilding in Cem5 is related to its high degree of structure and that when left at rest

this structure quickly rebuilds to its preferred state of structure (i.e. fast instantaneous rebuilding). However, throughout the duration of the test, little additional rebuilding occurred past this preferred state of structure, and, thus, the overall rate of structural rebuilding is low. As mentioned earlier, the lateral pressure will depend on how much of the vertical stresses are transferred through the material. Throughout the casting process a lower amount of the vertical stresses were transferred due to the high state of structure in Cem5, which resulted in a significant reduction in the formwork pressure.

With regard to Cem6SF, the reason why this mixture has a lower maximum formwork pressure than Cem6 is related to both the higher degree of structure (c.f. Figure 7-9) and the faster rate of structural rebuilding (both instantaneous and overall). During the first hour of casting, Cem6 exerts hydrostatic pressure ($K \sim 1$). After this point, enough shear strength is developed in this mixture so that the pressure begins to deviate from the hydrostatic value. This increase in shear strength could be attributed to an increase in inter-particle forces due to the small surface area of the silica fume particles [129] and improved packing density [14, 129, 130]. The faster rate of structural rebuilding may also be due to the silica fume serving as a nucleation site during the first few hours of hydration [131]. This can result in an enhanced precipitation of hydration products on the surface of the silica fume particles, which may serve as bridges to other particles [16]. Thus, the particle-particle contact is increased and promotes the development of a continuous contact network. In order to fully characterize the initial formwork pressure, the structural rebuilding parameter needs to include both the magnitude of structure (i.e. value of static yield stress) and the rate of rebuilding. One way to account for the degree of structure and the rate at which the structure rebuilds is to calculate the area under the SRE versus t_{rest} curve (c.f. Figure 7-10).

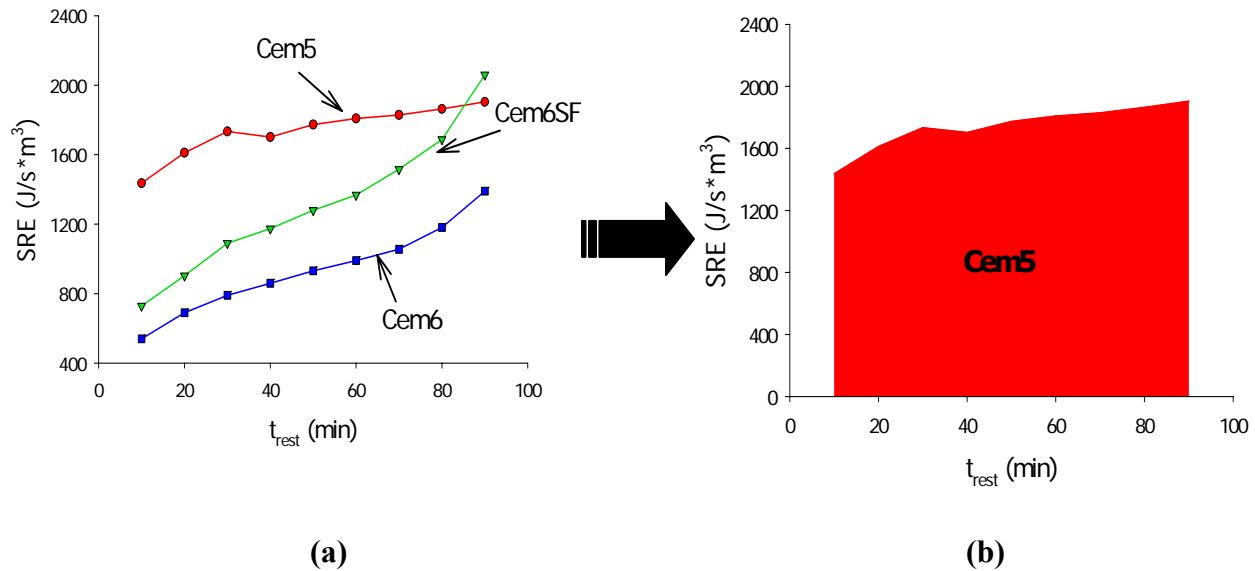


Figure 7-10. Transformation of SRE to take into account initial degree of structure and rate of rebuilding (a) SRE curves (b) Transformation of Cem5 SRE curve to area

This area has the units of J*min/s*m³ (or alternatively, Pa*min/s) and represents the total power (per unit volume and time) transferred into the system to break down the structure. The power takes into account both the rate and the magnitude of the initial structure. Higher power values should correlate to a lower formwork pressure. As shown in Figure 7-11, the behavior of Cem5 is accounted for when this approach is used.

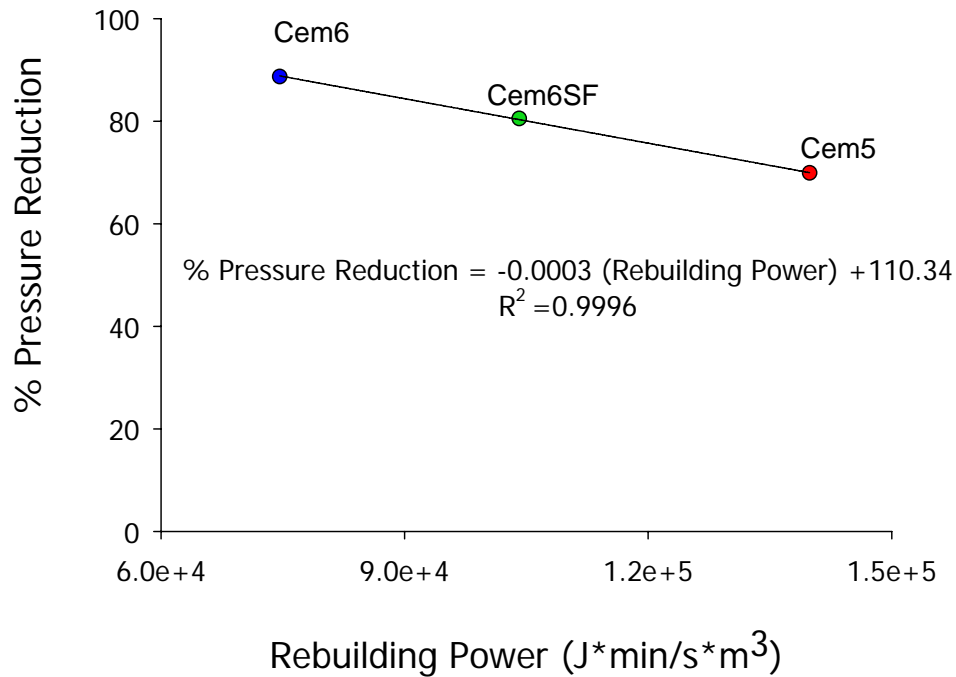


Figure 7-11. Rebuilding Power versus Pressure Reduction Percentage

Research on formwork pressure behavior showed that the rate of pressure decay is inversely proportional to the maximum initial pressure [5, 127, 132]. This is not to say that mixtures with higher maximum formwork pressures always cancel out an earlier time, but rather the relative rate at which the pressure decays tends to be faster when the initial maximum pressure is high. Indeed, this same behavior was seen in this study (c.f. Table 7-4). Cem5 had the slowest cancellation rate. This may be expected since this mixture had the lowest rate of rebuilding. The time at which the pressure cancels corresponds to the point at which lateral pressure against the formwork is equilibrated by the internal strength of the concrete, and the cancellation time provides an indication of when the forms can be removed [133]. Studies have shown that longer setting times are associated with longer cancellation times [5, 128]. The

kinetics of pressure decay was decelerated in Cem5. This is likely to be related to the higher superplasticizer demand, which prevents the particles from flocculating and delays the onset of setting. With regard to Cem6 and Cem6SF, it is seen that although Cem6 had a lower initial degree of structure it had a slightly faster rate of cancellation. Since the time of cancellation occurs several hours after the end of mixing, it is closely linked to the rate of hydration. Since the Cem6 mixture had a higher cement content, its rate of hydration is faster, and, thus, the pressure decay is faster. In the case of longer casting times, it is possible that Cem6 and Cem6SF may actually develop a lower initial formwork pressure than Cem5 since the effects of hydration will be more significant when the casting period is extended.

Table 7-4. Maximum initial pressure and rate of pressure cancellation

Mix	Hydrostatic Pressure (%)	Max Pressure (kPa)	Cancellation* (hr)	Rate of Cancellation (kPa/hr)
Cem6	89	212	10.95	19.4
Cem6SF	81	194	10.95	17.7
Cem5	70	168	11.38	14.7

*after the end of casting

An aspect that has not been discussed is whether the structural rebuilding in the mixtures is reversible (i.e. thixotropic) or irreversible. Irreversible structural rebuilding in cementitious materials occurs from aging effects due to a decrease in the superplasticizer efficiency and hydration. Figure 7-12 shows the structural breakdown curves for the three mixtures. These structural breakdown curves were conducted in order to establish the equilibrium condition (see

Section 4.3.3) at the beginning of the tests (right after loading the sample into the rheometer).

In these curves, the equilibrium shear stress (τ_{1eq}) is the shear stress value at 480 seconds. Similarly, τ_{2eq} defines the equilibrium shear stress for structural breakdown tests conducted at the end of the 90 minute rheological tests. The difference between τ_{2eq} and τ_{1eq} provides insight into the amount of irreversible and reversible bonds within the microstructure. τ_{2eq} corresponds to the shear stress value as it approaches an equilibrium value. If steady-state is achieved, and then it is independent of shear history [51]. Smaller differences were seen between τ_{2eq} and τ_{1eq} for Cem5 and Cem6SF (c.f. Table 7-5), indicating that these microstructures are characterized by a large amount of reversible (thixotropic) interparticle links (c.f. Figure 7-13 and Figure 7-15). The opposite is seen in Cem6 (c.f. Figure 7-14), and the structural rebuilding due to irreversible bonds from hydration and/or coagulated flocs plays a larger role in this mix. The τ_{2eq} stress of Cem5 is greater than the τ_{2eq} stress of Cem6, which indicates a higher degree of agglomeration.

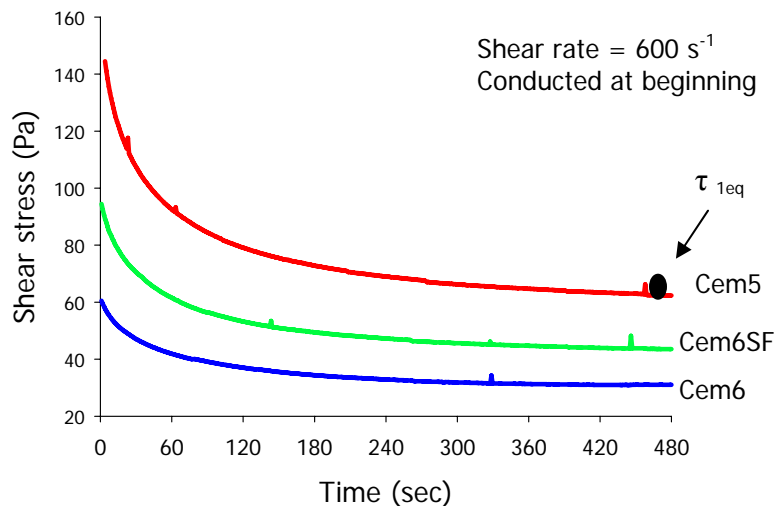


Figure 7-12. Initial structural breakdown curves

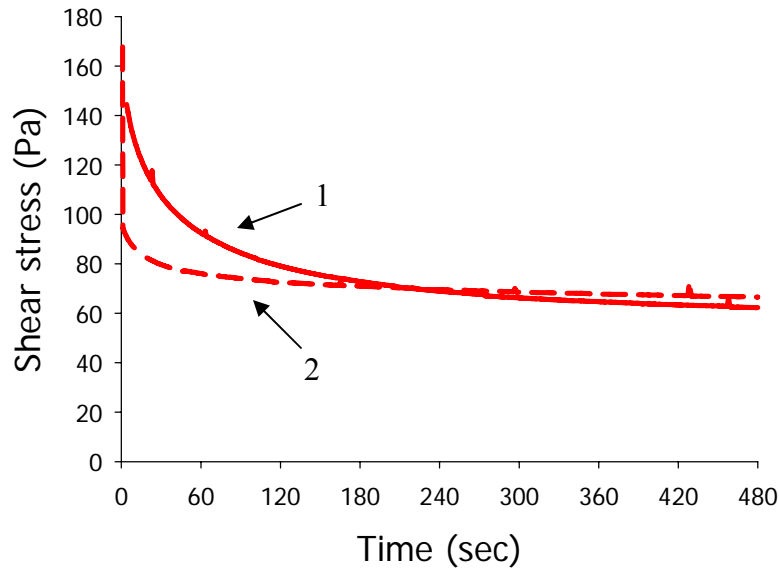


Figure 7-13. Contribution from irreversible structural buildup in Cem5 (1 represents structural breakdown curve conducted at beginning of test, 2 represents structural breakdown curve conducted after 90 minutes)

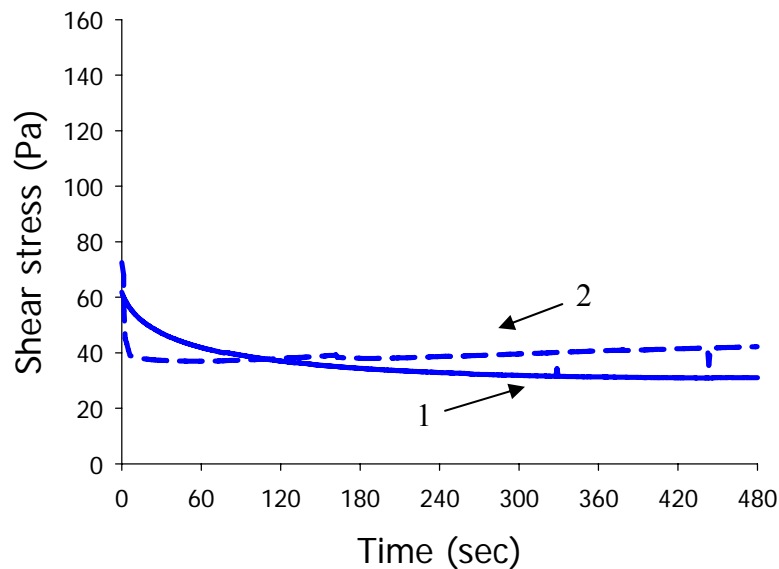


Figure 7-14. Contribution from irreversible structural build-up in Cem6 (1 represents structural breakdown curve conducted at beginning of test, 2 represents structural breakdown curve conducted after 90 minutes)

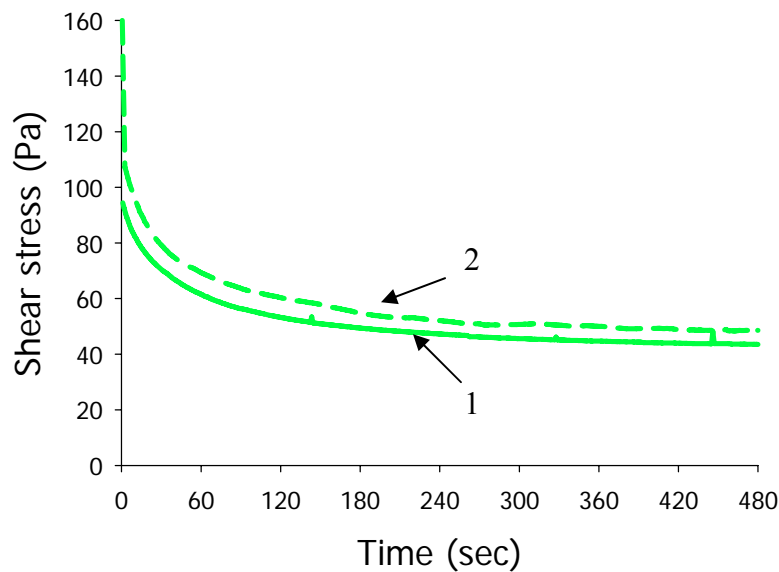


Figure 7-15. Contribution from irreversible structural buildup in Cem6SF (1 represents structural breakdown curve conducted at beginning of test, 2 represents structural breakdown curve conducted after 90 minutes)

Table 7-5. Equilibrium shear stress measured at beginning and end of rheological test. The % increase represents contribution from irreversible build-up, where a 0% increase means that all structural build-up was due to reversible (thixotropic) build-up.

Mix	τ_{eq1} (Pa)	τ_{eq2} (Pa)	$\Delta \tau_{eq}$ (Pa)	% increase
Cem5	62.3	66.5	4.2	6.7
Cem6SF	43.5	48.6	5.1	11.7
Cem6	31.0	42.2	11.2	36.1

7.5 Chapter Summary

The use of a simple laboratory device to investigate the parameters influencing the formwork pressure development of SCC was developed. The device allows for the study of the formwork pressure of SCC without having to cast large scale columns and walls. With this device, an experimental program to investigate the influence of structural rebuilding on formwork pressure behavior was conducted. Based on the results of the study the following conclusions can be made:

- The rheological properties of the paste matrix and its evolution with time can be used as indicators of the formwork pressure behavior of the concrete.
- Replacing portions of cement with silica fume increases both the degree of structure and the rate of structural rebuilding. This results in a decrease in the maximum formwork pressure.
- Formwork pressure is related to both the initial degree of internal structure (i.e. magnitude of yield stress) and the rate at which the structure rebuilds with time (Δ SRE). Both of these parameters should be accounted for in order to fully characterize the initial formwork pressure exerted by an SCC mixture.
- A two parameter model that predicts the formwork pressure behavior solely using the rate of rebuilding and the time of cancelation may not be sufficient to describe the formwork pressure behaviors. The rate of rebuilding during the casting period is not the sole predictor of the time at which pressure will decay. Mixtures with slower rates of rebuilding can cancel out similar times than mixtures with faster rates of rebuilding depending on the rate of hydration. A two-component model may be

required to describe formwork pressure of SCC: one part describes the formwork pressure until the end of casting and the second part describes the formwork pressure decay.

- Formwork pressure decay is influenced by superplasticizer dosage. For a given w/b ratio, higher superplasticizer demand reduces the rate of pressure decay.

8 CHAPTER 8: Flocculation of Cement Pastes

8.1 Introduction

The objective of this phase of the work was to evaluate the flocculation mechanisms in cement pastes and to determine how the flocculation mechanisms are influenced by mixture proportioning. With the growing use of SCC and the development of other unique rheological cementitious suspensions, it is important to have a fundamental understanding of parameters that can affect the flow behavior in order to better predict and control fresh state properties. The results of the previous chapter showed that the structural rebuilding within the cement paste matrix significantly influences the macroscopic behavior of concrete in its fresh state, in particular the formwork pressure behavior. Aside from formwork pressure concerns, there are limited studies that have focused on the flocculation mechanisms of high fluidity cement pastes. The rheology of cement-based materials is related to the flocculation/deflocculation process between particles, and current methods to examine the flocculation process involve sample dilution or sampling or can not be measured in-process. To overcome these limitations, a new experimental method was used to measure *in situ* the particle size evolution in real time. To the author's knowledge, this is the first time that *in situ* particle size measurements have been measured in concentrated cement pastes while the paste is being subjected to shear-induced stresses. Much research has focused on the microstructure and, more recently, on the nanostructure of hardened concrete. However, there is still a lack of information about the structure of cementitious materials in the fresh state, which is especially important during processing of concrete (from mixing to pumping to formwork pressure to shrinkage). The results

from this stage of the project contribute to the state of knowledge regarding fresh state microstructure and provide some answers to the following questions: What is the microstructure of fresh cement paste? How do the different material ingredients influence the rate of flocculation and floc strength? How does the microstructure of fresh cement paste respond to shear-induced stresses?

8.2 Focused Beam Reflectance Measurement (FBRM)

The FBRM method is a particle size analysis measurement technique that provides information about particle chord length distribution in real-time. While there are various techniques for determining particle size, few can be applied *in situ* or used for materials with high solid volume concentrations[134]. The significant advantage of the FBRM method is that no dilution or sampling of the suspension is required. The FBRM technique has been widely used in crystallization studies and flocculation studies of dilute suspensions. Information obtained from FBRM tests is often correlated to a process parameter. The chord length, defined as the intersection of the scan plane with the projected-area of the particle in the focal plane, is a characteristic measurement of particle geometry. FBRM instruments operate by scanning a highly focused laser beam across particles in a suspension and measuring the time duration of back scattered light from individual particles [135]. The laser beam is focused to a small beam spot and scans at a fixed velocity. When the focused beam intersects the edge of a particle, the particle will backscatter the laser light. Light will continue to backscatter until the beam reaches the opposite edge of the particle (c.f. Figure 8-1).

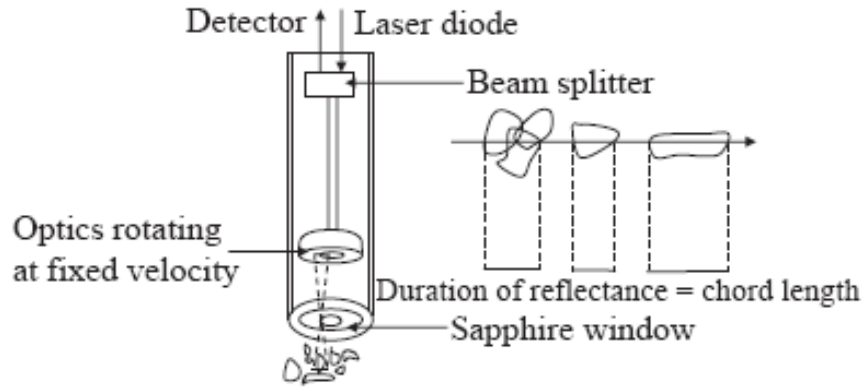


Figure 8-1. FBRM probe operation method [136]

A discriminating electronic circuit was used to isolate the time period of the backscattered light, and this time period is multiplied by the scan speed to obtain a distance [137]. It is this distance that is used as a measure of the particle size and is referred to as the chord length in FBRM terminology. The mean chord length, \bar{C} , is defined as follows,

$$\bar{C} = \frac{\sum_{i=1}^k \left[\left(\frac{n_i}{\sum_{i=1}^k n_i} \right) M_i \right]}{\sum_{i=1}^k \left(\frac{n_i}{\sum_{i=1}^k n_i} \right)} = \frac{\sum_{i=1}^k n_i M_i}{\sum_{i=1}^k n_i M_i^0}$$

where n_i is the counts in an individual measurement channel, M_i is the midpoint of an individual channel, and k is the upper channel number [137]. Every second thousands of chord length data measurements are collected. The FBRM laser probe used in this research can obtain particle size measurements in the range of 1 –1000 μm , and these measurements can be represented with different distribution statistics (ex. median, mean, number of counted chords between any size

interval, cubic mean, square mean, etc.). Generally, the distinction between chord length data and actual particle size is neglected because the considered process parameter has been found to be related to some aspect of the chord length data [134]. An advantage of FBRM is that it can be used in opaque materials, such as cement paste, since the chord length is determined by the time (pulse width) rather than the intensity (pulse amplitude). A problem associated with backscatter light measurements is difficulty in detecting smooth spherical particles or particles suspended in liquids with refractive indices close to that of the particle [136, 138]. A schematic of the FBRM probe and the actual experimental setup are shown in Figure 8-2 and Figure 8-3.

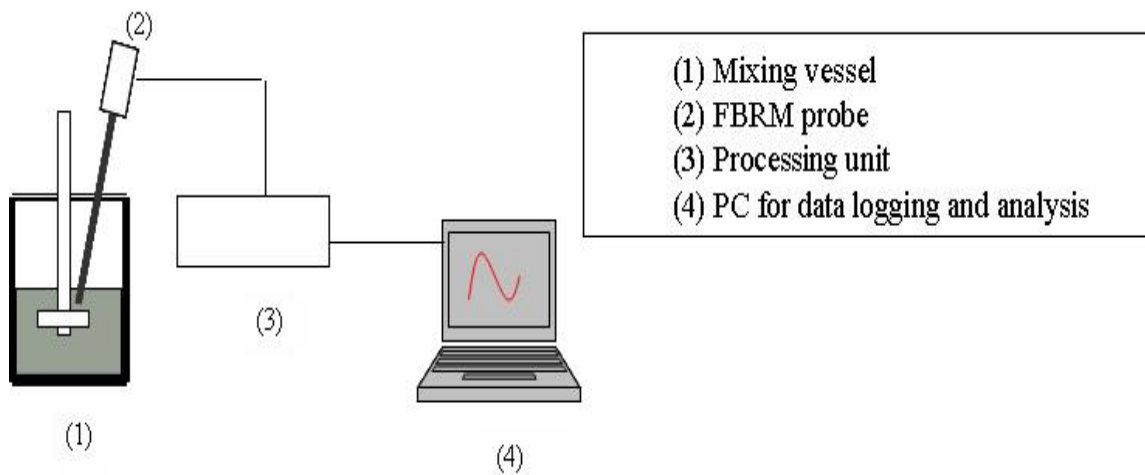


Figure 8-2. Schematic of FBRM system

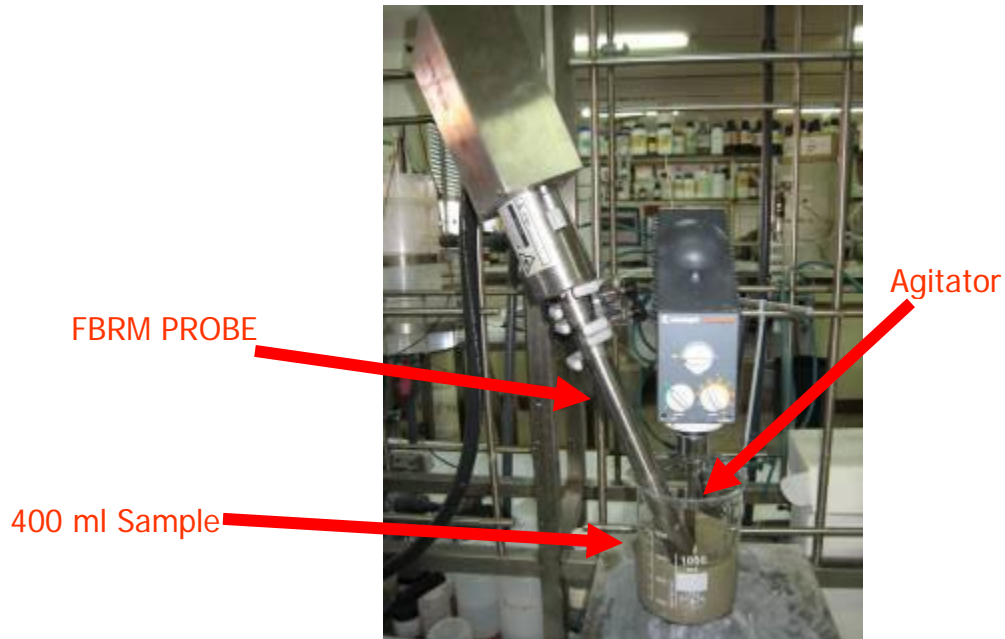


Figure 8-3. FBRM experimental set-up

Further insight about the evolution of flocs was obtained by modeling the process of aggregation and breakage. This process depends on the rate of different processes, namely, the aggregation rate of particles forming flocs and the rate of floc breakage from shear induced forces. Based on the Smoluchowshi classical theory of flocculation [139], a model based on second-order kinetics for aggregation and first-order kinetics for disaggregation was used to study the particle concentration evolution. This model is represented by the equation

$$\frac{dn}{dt} = -k_1 n^2 + k_2 n$$

where n is the number of particles, t is the time (in seconds), k_1 is the aggregation kinetic constant, and k_2 is the disaggregation kinetic constant. The model assumes that flocculation occurs from the collision of two particles (i.e. no multi-particle collision). The integrated solution of this differential equation has been used to fit the experimental evolution of flocs and is shown below:

$$n = \frac{\frac{k_2}{k_1}}{1 - \left[\left(\frac{n_0 - \frac{k_2}{k_1}}{n_0} \right) e^{-k_2 t} \right]}$$

where n_0 is the initial number of particles.

8.3 Experimental Program

8.3.1 Mix Proportions and Materials

Two cements and a class F fly ash were used. Cement replacement was conducted on a mass basis. The physical and chemical properties of the cements and fly ash were given in Section 6.2. The superplasticizer was a carboxylated polymer in aqueous solution (specific gravity = 1.04). A liquid water-soluble cellulose-based VMA with a specific gravity of 1.002 and solids content of 70% was employed in some of the mixtures. The manufacturer's recommended dosage for the VMA ranged from 130 to 910 mL/100 kg of cementitious materials. Based on these recommendations, two dosages were used in this study: a medium dosage of 400 mL/100 kg of cementitious materials and a high dosage 800 mL/100 of cementitious materials. Two clays were also used. The first clay was metakaolinite in pulverized powder form.

Metakaolinite is composed of pozzolonic amorphous alumino silicates formed by controlled calcination of kaolinite. The chemical formula for kaolinite is $\text{Al}_2\text{Si}_2\text{O}_5(\text{OH})_4$ and the general formula for metakaolinite is $\text{Al}_2\text{Si}_2\text{O}_7$. The second clay was sepiolite in slurry form. Sepiolite is a hydrated magnesium silicate serpentine mineral commonly found in Spain. The chemical formula for sepiolite is $\text{Mg}_4\text{Si}_6\text{O}_{15}\cdot 6(\text{H}_2\text{O})$. Both clays were used as a cement replacement, at a dosage of 1.5% solids content. The water content for the sepiolite mix was adjusted to take into account the water in the sepiolite slurry. The mix compositions are shown in Table 8-1:

Table 8-1. Mix compositions for FBRM studies

Mix	Cement Type	w/b	FA (% binder)	SF (% binder)	SP dosage (% binder)	Cellulose VMA dosage	Meta-kaolinite Clay	Sepiolite Clay	ϕ (solids volume fraction)
Mix 9 ^(*)	Cement 6	0.3	20	8	0.33	-----	-----	-----	0.540
Mix 10 ^(*)	Cement 5	0.3	20	8	2.69	medium	-----	-----	0.540
P1	Cement 5	0.4	-----	-----	-----	-----	-----	-----	0.442
P1-SP	Cement 5	0.4	-----	-----	0.8	-----	-----	-----	0.442
P1-VMA H	Cement 5	0.4	-----	-----	-----	high	-----	-----	0.442
P1-SP-VMA H	Cement 5	0.4	-----	-----	0.8	high	-----	-----	0.442
P1-SP-VMA M	Cement 5	0.4	-----	-----	0.8	medium	-----	-----	0.442
P2	Cement 6	0.4	-----	-----	0.14	-----	-----	-----	0.442
P2-VMA	Cement 6	0.4	-----	-----	0.14	medium	-----	-----	0.442
P2-Meta	Cement 6	0.4	-----	-----	0.14	-----	yes	-----	0.443
P2-Sepio	Cement 6	0.4	-----	-----	0.14	-----	-----	yes	0.443
FA0	Cement 6	0.3	-----	-----	0.33	-----	-----	-----	0.514
FA20	Cement 6	0.3	20	-----	0.33	-----	-----	-----	0.532

* Corresponds to the same mix discussed in Chapter 6

8.3.2 Sample Preparation

8.3.2.1 For effect of shear history, VMA, and clays

Each suspension was prepared by manually mixing all the powder materials for 60 seconds. Then, the superplasticizer and 90% of the deionized (DI) water were added while stirring, and the sample was manually mixed for 60 seconds. Following the hand-mixing, the cement paste suspension was mixed with a hand-held blender at low speed for 50 seconds, and then the remaining water was added over a 10 second interval. When used, the viscosity modifying agent or sepiolite slurry was also added during the 10 second interval. Next, the paste suspension was mixed for 50 seconds at speed 1, after which the sides of the container were cleaned with a spatula over a 30 second interval. Within a 40 second time interval the sample was transferred to the FBRM. Hence, the FBRM experiment was started 4 minutes after the addition of water.

8.3.2.2 For studies of mix 9, mix 10, and fly ash

The suspensions were prepared by manually mixing all the powder materials for 60 seconds. The superplasticizer and 90% of the DI water were added while stirring, and the sample was manually mixed for 90 seconds. Following the hand-mixing, the cement paste suspension was mixed with a hand-held blender at low speed for 50 seconds, and the remaining water and viscosity modifying agent (when used) were added over a 10 second interval. Then, paste suspension was mixed for 90 seconds at speed one, and over a thirty second interval the sides of the beaker were cleaned with a spatula. Within a 60 second time interval the sample was transferred to the FBRM. Hence, the FBRM experiment was started 5 minutes after the addition of water.

8.4 Results and Discussion

8.4.1 Effect of shear history

Figure 8-4 and Figure 8-5 show the mean chord length evolution and total number of particle counts recorded for Mix9 subjected to different shearing conditions over a 30 minute period:

- Stirred at 40 rpm for 30 min
- Stirred at 400 rpm for 30 min
- Stirred at 40 rpm for 10 min, 400 rpm for 10 min
- Stirred at 400 rpm for 10 min, 40 rpm for 10 min, 400 rpm for 10 min
- Stirred at 400 rpm for 10 min, 175 rpm for 10 min, 400 rpm for 10 min

A new sample was prepared for each test. It can be seen that both the particle size and the total number of particles were influenced by the shearing conditions. Floccs can form from three possible mechanisms: perikinetic aggregation, orthokinetic aggregation, and differential sedimentation. Perikinetic aggregation refers to aggregation caused by Brownian (i.e. random) motion whereas orthokinetic aggregation refers to aggregation caused by shear induced (i.e. deterministic forces) forces [140]. Differential sedimentation occurs from particle collisions due to differences in the settling velocities [141].

The Peclet number is a dimensionless quantity that characterizes the ratio of shear to diffusion [142]:

$$P_e = \frac{\pi \eta a^3 N}{10 k_B T}$$

where η is the shear-rate dependent suspension viscosity, a is the size of the particle, N is the stirrer intensity in rpm, k_B is Boltzmann's constant, and T is the absolute temperature in Kelvin

[143]. Aggregation is perikinetic when the Peclet number is lower than 1 and orthokinetic

when the Peclet number is higher than 1. It is generally accepted that under shearing conditions orthokinetic aggregation will dominant over differential sedimentation [134]. Hydrodynamic effects increase with particle size and become more dominant for particles larger than a few micrometers [144].

Table 8-2 shows the particle diameter that makes $P_e = 1$ under different shearing conditions and viscosities. The viscosity values were selected to represent those of typical SCC pastes [145, 146]. It is seen that perikinetic flocculation is the dominating mechanism for sizes less $\sim 0.1 \mu\text{m}$. The minimum measurement range of the FBRM device was 1 micrometer; therefore, aggregation of all of the measurable particles is orthokinetic. Increasing the stirring intensity decreases the size limit of particles subjected to perikinetic aggregation, and in cementitious suspensions this will increase the number of particles undergoing orthokinetic aggregation. Increased mixing promotes particle-particle collisions (this can produces floc growth); however, if the agitation is too vigorous, the turbulent shear forces that are produced will break up the floc into smaller particles [147].

Table 8-2. Particle Size for when Peclet number equals 1

η (Pa/s)	N (rpm)	a (μm)
0.05	40	0.19
0.05	400	0.09
0.10	40	0.15
0.10	400	0.07
0.20	40	0.12
0.20	400	0.05
0.30	40	0.10
0.30	400	0.05

8.4.1.1 Flocc Growth

As shown in Figure 8-4, when $N = 40$ rpm, the mean chord size gradually increased over time, which shows that the kinetics of aggregation dominates at this stirring intensity. During the thirty minute test, the average chord length increased by 5.5%, from $11.00 \mu\text{m}$ to $11.60 \mu\text{m}$. The number of counts decreased when the chord size increased (c.f. Figure 8-5). This reflects the phenomena of floc formation (i.e. individual particles/flocs coming together to form a larger floc). As shown in Figure 8-6, the majority of the measured chords ($\sim 80\%$) are within less than 22 microns, and, thus, the average size of the suspension will tend towards the smaller values. In addition to looking at the average chord size and total number of counts, the overall change of different floc sizes provides further insight into the aggregation and breakage kinetics.

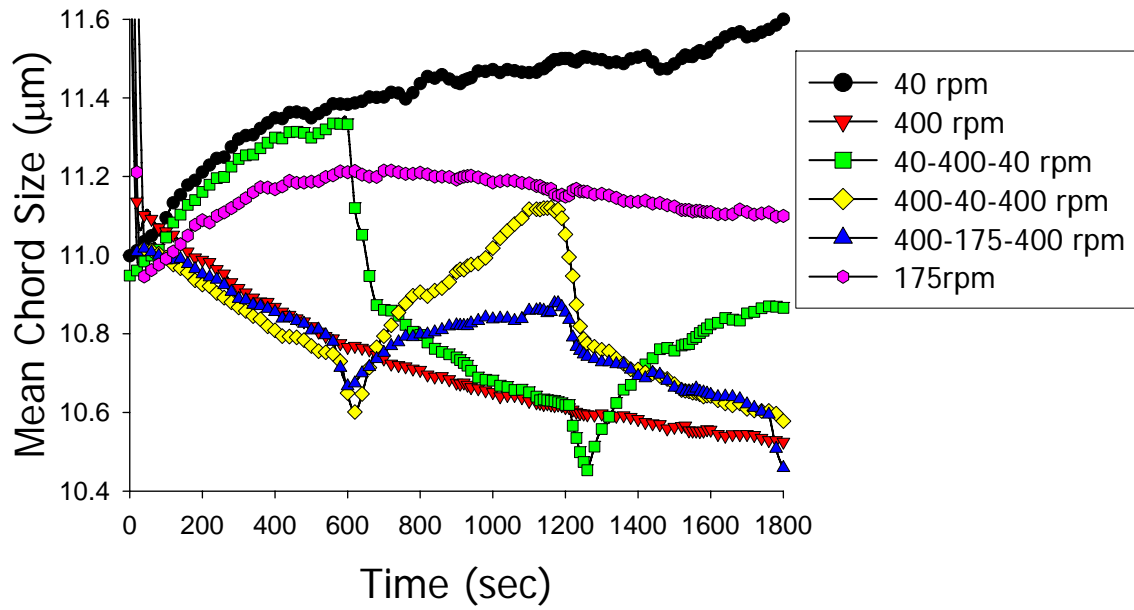


Figure 8-4. Mean chord size evolution when subjected to different shearing conditions

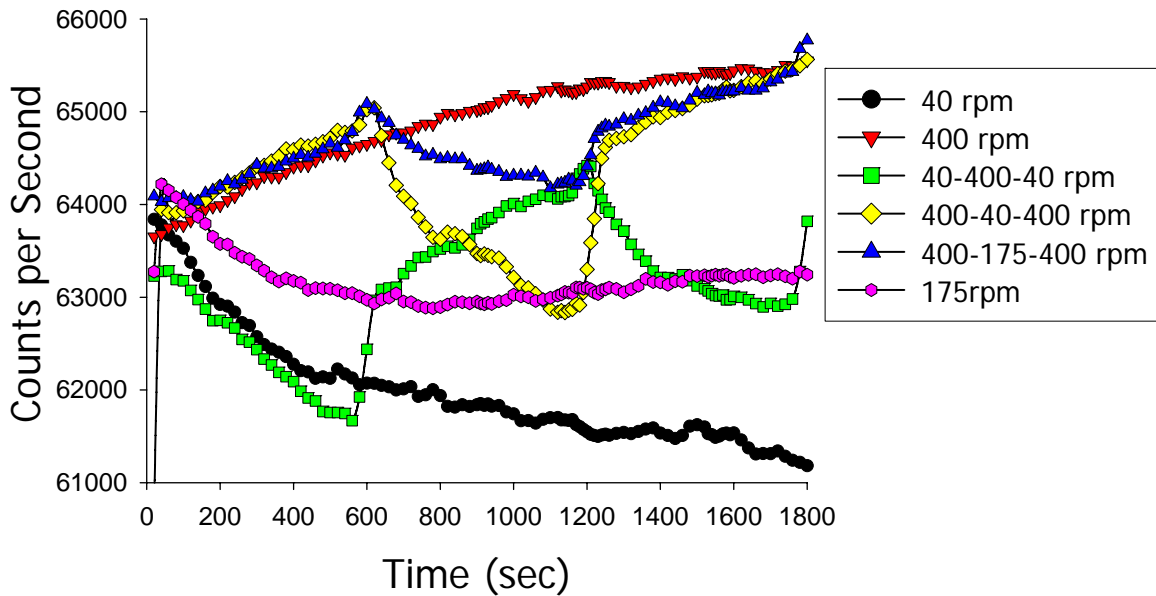


Figure 8-5. Counts per second when subjected to different shearing conditions

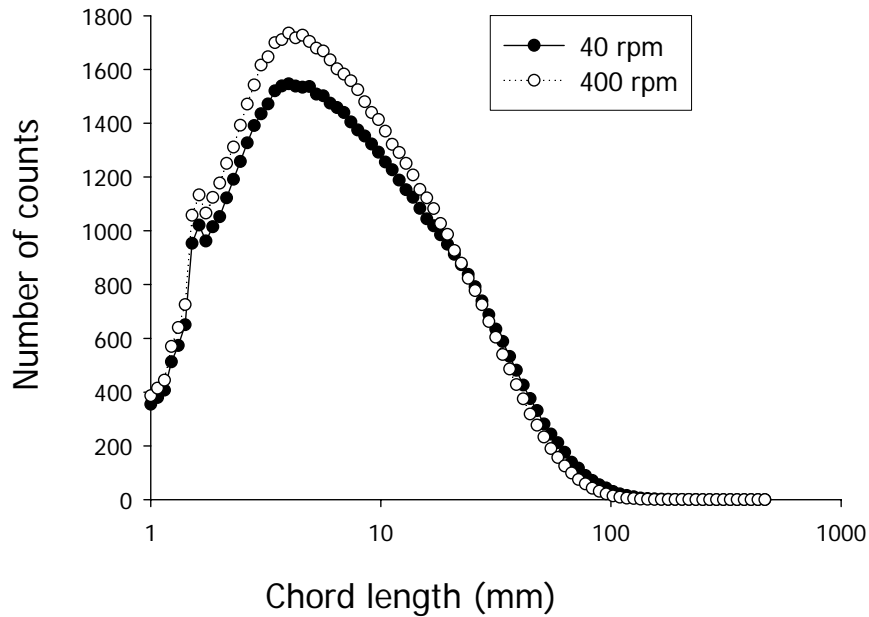


Figure 8-6. Chord size distribution curves for cement suspension at $t= 1800s$ for samples subjected to constant shearing of 40 rpm and constant shearing of 400 rpm

Figure 8-7 shows the percent difference of the various chord lengths from the beginning and end of the thirty minute test. Negative values indicate a net loss of flocs within a particular chord length range, whereas positive values indicates that the number of flocs within a given size range increased. When the stirring intensity was held constant at 40 rpm, the number of flocs in the smaller ranges decreased (6.5% reduction for flocs within 1-5 microns and 4.9% reduction for flocs within 5-22 microns), while the number of larger flocs increased (3.9% increase for flocs within 22-67 microns, 19.9% increase for flocs 67-95 microns, and 44.6% increase for flocs within 95-250 microns).

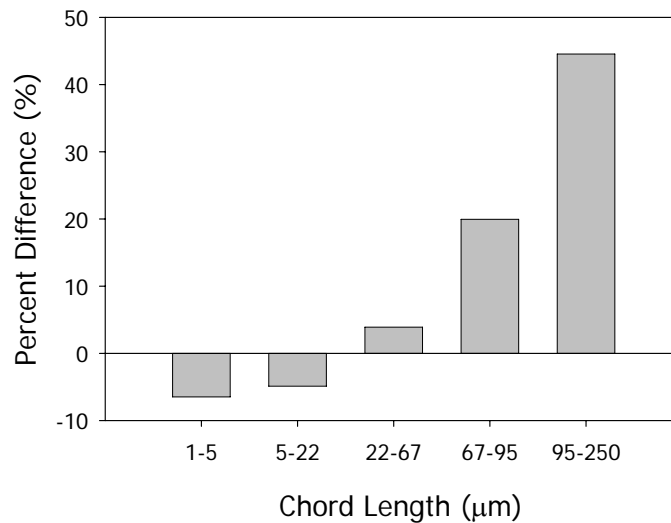


Figure 8-7. Percent difference when N= 40 rpm during 30 minute interval

This further demonstrates the dynamic process of rebuilding. Within a particular chord range X , there will be a gain of flocs due to smaller particles outside of that size range aggregating to form a new floc that is within range X . However, there will also be a loss of flocs in chord range X , due to flocs that are within size range X now aggregating to form a new floc that is larger than size X (c.f. Figure 8-8). The reduction in the number of flocs within the 1 – 22 micron range was very small (less than 10%), which indicates that there was a rapid formation of new flocs from particles that were previously not measurable. The newly formed flocs counterbalanced the decrease in the particle counts lost due to the particles within that size range forming larger flocs. Regardless of the cement paste suspension tested, under a stirring intensity of 40 rpm, the maximum floc size measured was 250 micron. Based on this observation, it is hypothesized that even when the aggregation processes dominate, there is a limiting size criterion for particle aggregation during the early stages of the induction period under shear. The concept of a maximum stable aggregate size under orthokinetic aggregation has been seen in

polystyrene suspensions [148], scheelite particles in a sodium oleate solution [149] and suspensions of quartz [150], and other systems [151-153]. It is likely that flocs that are larger than 250 microns are weakly bonded and are unstable. Thus, when subjected to even a low stirring intensity of 40 rpm their interparticle links are ruptured and the floc is broken up into smaller sizes.

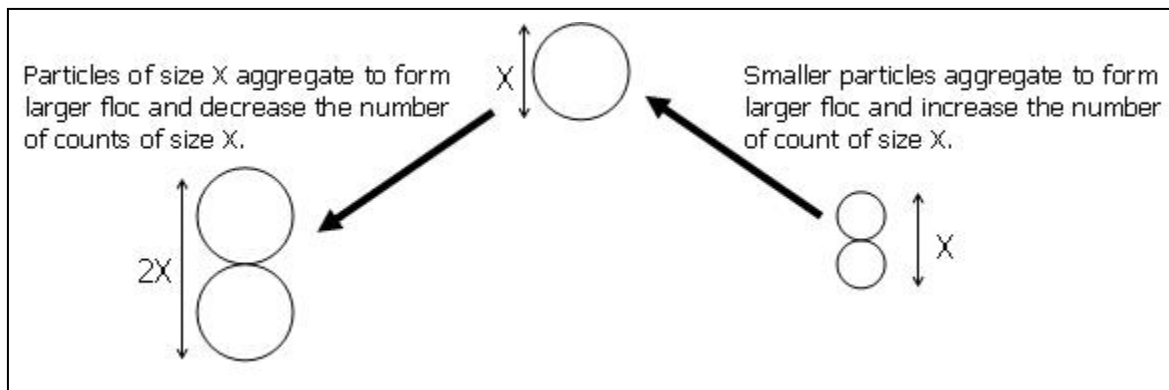


Figure 8-8. Loss and gain of particle size X

8.4.1.2 Floc Breakage

When $N = 400$ rpm, the mean chord length decreased from $11.93 \mu\text{m}$ to $10.53 \mu\text{m}$, representing a 11.7% decrease (c.f. Figure 8-4). A reduction in particle size typically indicates disaggregation is occurring, and, as seen in Figure 8-5, the number of counts increased during the same period. Thus, when $N = 400$ rpm, the disaggregation process dominates, and larger flocs are broken into smaller flocs due to the high stirring intensity decreasing the aggregation efficiency [154]. Figure 8-9 shows the percent difference of the various chord lengths from the beginning and end of the thirty minute test. A substantial decrease is seen in the larger chord length ranges and almost all of the flocs within the 95 – 250 micron range are destroyed at the end of the 30

minutes. The high rate of breakage further supports the premise that the floc strength is inversely proportional to floc size. As the floc size increases, it is more prone to breakage since the probability of finding flaws (i.e. weaker bonds) also increases. Yang et al. examined the sediment from differential sedimentation experiments and reported that the sediments from cement-water systems were voluminous and soft, whereas sediments from cement-alcohol systems were closely packed, dense, and hard [80]. Thus, as the floc size increases, it is likely that the microstructure of the floc is more porous (i.e. less compact), and this would also increase the susceptibility to breakage.

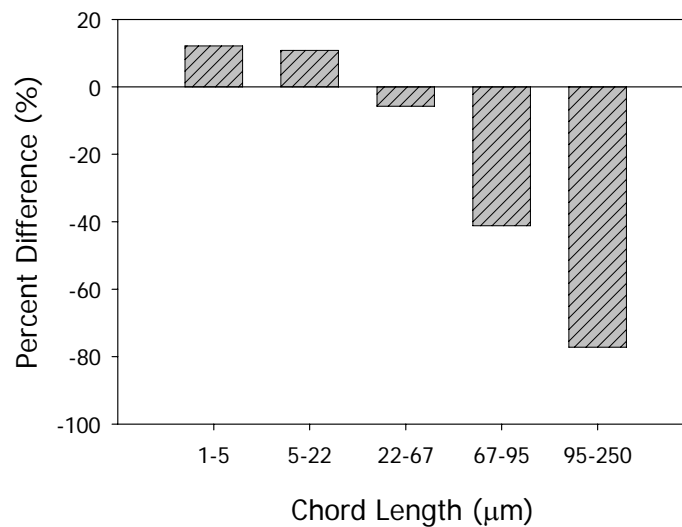


Figure 8-9. Percent difference when N= 400 rpm during 30 minute interval

8.4.1.3 Influence of shear regime on aggregation and breakage

Consider the cement sample that is subjected to the 40-400-40 rpm stirring intensity cycle (c.f. Figure 8-4). It is seen that the when the stirring intensity is initially increased from 40 rpm to 400 rpm, an immediate decrease in the average chord size was seen. This is accompanied by an

increase in the number of counts (c.f. Figure 8-5) and is a result of the disaggregation process

dominating over the aggregation kinetics. By the end of the 400 rpm stirring period, the average chord size is the same as that of the suspension that was sheared continuously at 400 rpm. However, the total number of counts was slightly less than that of the specimen that was sheared at 400 rpm continuously, indicating that the arrangement of the microstructures of the two systems are different. This can be due to re-conformation of the adsorbed polymers and rearrangement of the flocs during the flocculation and breakage process. Negro et al. [154] studied the flocculation of 5% w cement suspension cycled from low to high stirring intensities and also found different tendencies in the aggregate size based on the shearing condition. A significant increase in the mean chord sized evolution was seen when the stirring intensity was cycled from 300-800-300 rpm versus being held constant at 300 rpm. This was believed to be due to the smaller flocs sizes during the 800 rpm cycle increasing the hydration reaction.

The effect of shearing when the suspension is cycled from a high to low shearing condition is demonstrated in samples subjected to the 400-40-400 cycle and the 400-175-400 cycle. At $N = 400$ rpm, the shearing forces generated by the high stirring intensity are strong enough that breakage mechanisms dominate, and all samples sheared at 400 rpm achieved a similar microstructural condition. When the stirring intensity was reduced, the sample began to rebuild, and average particle size gradually increased. However, the microstructures of the samples subjected to 400 rpm stirring intensity prior to rebuilding were different than the microstructures of the samples that were continuously rebuilding. Samples subjected to shear prior to rebuilding had smaller average chord sizes and a greater number of particle counts than the sample that was allowed to continuously rebuild. Figure 8-10 shows the measured number of

counts at the $t = 1200$ s for the sample stirred at 40 rpm continuously and the sample subjected to 10 minutes of stirring at 400 rpm prior to being stirred at 40 rpm.

Table 8-3 shows the percent difference in the number of measured counts measured at 1200 seconds between the 40 rpm sample and 400-40-400 rpm sample and the 175 rpm sample and 400-175-400 rpm sample. Negative values indicate that less flocs were measured (in a specific size range) in the sample subjected to a cycled shearing regime versus the sample that was sheared at a constant stirring intensity. In both cases, the microstructure of the sample subjected to the high stirring intensity prior to rebuilding contained a smaller quantity of flocs larger than 65 micrometers and had a larger quantity of flocs smaller than 22 microns. This highly suggests that the aggregation mechanisms are directly influenced by different shear forces. When the stirring intensity is reduced the sample is able to rebuild, but the kinetics of re-aggregation are relatively slow and more time is needed in the sample that was subjected to a high stirring intensity to rebuild back to its initial state. This is further demonstrated in the reflocculation potential of the sample subjected to the 40-400-40 rpm stirring intensity cycle. After being subjected to a stirring intensity of 400 rpm, the stirring intensity was reduced back to 40 rpm. As seen in Figure 8-4, the mean chord size measured at the end of the second rebuilding cycle ($t = 1800$ s, mean chord size = 11.743 μm) is smaller than the mean chord size measured after the first rebuilding cycle ($t = 600$ s, mean chord size = 10.742 μm). Yoon and Deng [155] quantitatively studied reflocculation behavior of clay suspensions under turbulent and defined the following reflocculation index:

$$\text{Reflocculation index (\%)} = \left(\frac{X_{i+1} - Y_i}{X_1 - Y_1} \right) \quad (i=1,2,3\dots)$$

where i is the cycle number of the shear condition change, X is the floc size at low shear condition, and Y is the floc size at high shear condition. Using this equation, the reflocculation index at $t = 1800$ seconds for the 40-400-40 sample is 0.51. As stated earlier, the aggregation kinetics are influenced by the shearing condition, and, thus, the differences in the mean chord size is also attributed to different microstructural states prior to the beginning of each rebuilding period. The sample is less agglomerated at $t = 1200$ s than at $t = 0$ s, and it takes a longer time for the interparticle bonds that were ruptured during the 400 rpm stirring period to be recovered.

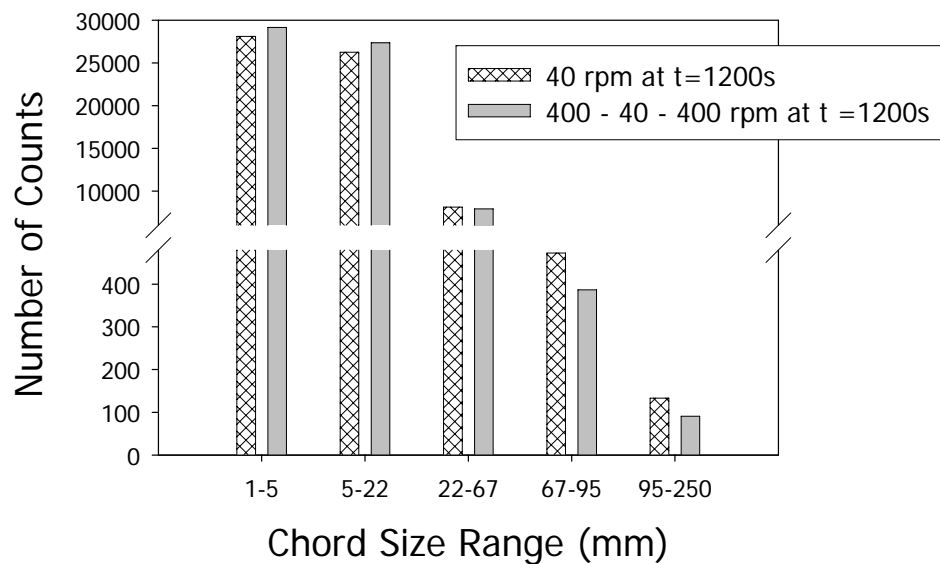


Figure 8-10. Number of counts at end of 20 min period for sample allowed to continuously rebuild and sample subjected to shear prior to rebuilding

Table 8-3. Influence of shearing regime on rebuilding of flocs

Chord size range at t = 1200 seconds (μm)	1 - 5	5 - 22	22 - 67	67 - 95	95 - 250
Percent difference between 40 rpm sample, and 400-40-400 rpm sample	3.7 %	4.2 %	-2.6 %	-18.2 %	-31.7 %
Percent difference between 175 rpm sample, and 400- 175-400 rpm sample	3.2 %	2.3 %	-1.5 %	-12.7 %	-22.7 %

Examining the microstructures the samples at t = 1800s for the samples subjected to the 400-40-400 rpm and 400-175-400 rpm cycles provides information about the reversibility of the bonds formed during the rebuilding periods. From Figure 8-4, it is evident that the flocs formed during the rebuilding period were easily broken up when the stirring intensity increased. In both cases, when the stirring intensity was increased to 400 rpm, the mean chord size and total number of counts approached values that were similar to the sample that was continuously stirred at 400 rpm. Table 8-4 shows the percent difference in number with respect to the number of measured counts measured in the 400 rpm sample at 1800 seconds and the number of measured counts in the 400-40-400 rpm sample and 400-175-400 rpm sample at the end of the second breakdown period at t=1800s. Negative values indicate that less flocs were measured (in a specific size range) in the sample subjected to a cycled shearing regime versus the sample that was sheared at a constant stirring intensity. A larger number of large flocs were seen in the mixture subjected to the 40 rpm rebuilding period indicating that the bond strength of the larger flocs increased during the rebuilding period and that they are less susceptible to breakage. However, it is likely that with continued shearing at 400 rpm, the majority of those flocs would also be broken down. Overall, little difference is seen in the chord size distributions of the

samples; this indicates that the system tends towards an equilibrium microstructural condition when the breakage kinetics dominates.

Table 8-4. Influence of shearing regime on breakdown of flocs

Chord size range at t = 1800 seconds (μm)	1 - 5	5 - 22	22 - 67	67 - 95	95 - 250
Percent difference between 400 rpm sample, and 400-40-400 rpm sample	0.2 %	-0.6 %	1.1 %	5.2 %	11.3 %
Percent difference between 400 rpm sample, and 400- 175-400 rpm sample	0.4 %	0.6 %	-0.8 %	-3.0 %	-1.7 %

8.4.2 Effect of VMA and clays

Figure 8-11 shows the influence of VMA on chord count evolution for Cement 5. No substantial changes in the number of chords counted were seen when VMA was added (compare P1 and P1-VMA_H), which is an indication that additional flocculation of the cement particles did not occur from the incorporation of this VMA. Rather, the superplasticizer was shown to be the dominating factor affecting the chord length measurements, and a significant increase in the number of counts occurred when superplasticizer was used (compare P1 with P1-SP). This shows that the superplasticizer molecules directly interact with the cement particles such that the cement particles are deflocculated, which increases the number of particles in the system. When both VMA and superplasticizer are used in a paste, the flocculation behavior is more similar to that of a paste with just superplasticizer. This behavior is seen regardless of the VMA dosage (compare P1-SP, P1- SP-VMA_M, and P1-SP-VMA_H). It can be concluded that the VMA did

not interact with the cement particles, or if it did, it did not have any influence on the flocculation properties. Thus, it is likely that the increase in cohesiveness when this particular VMA is used is garnered from the polymers binding to the water phase.

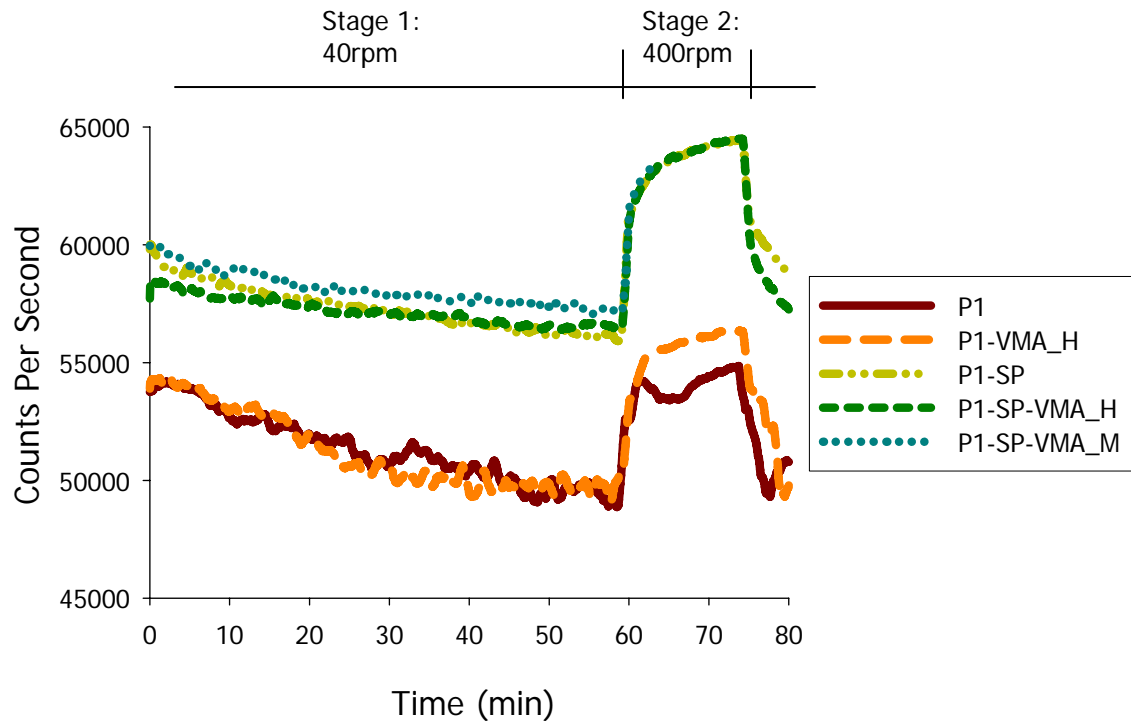


Figure 8-11. Influence of VMA on number of particles for Cement 5

Figure 8-12 shows the influence of VMA and clays on mixtures made with Cement 6. Similar to the behavior seen above, no significant changes were seen in the aggregation and breakage behavior due to the incorporation of VMA. These results further support the hypothesis that this particular VMA increased the stability of the paste matrix by adhering to the aqueous phase. The results indicate that VMAs that increase the stability of the paste matrix by

adhering to the mixing water will not significantly change the rate of structural rebuilding even though it may alter the viscosity of the paste.

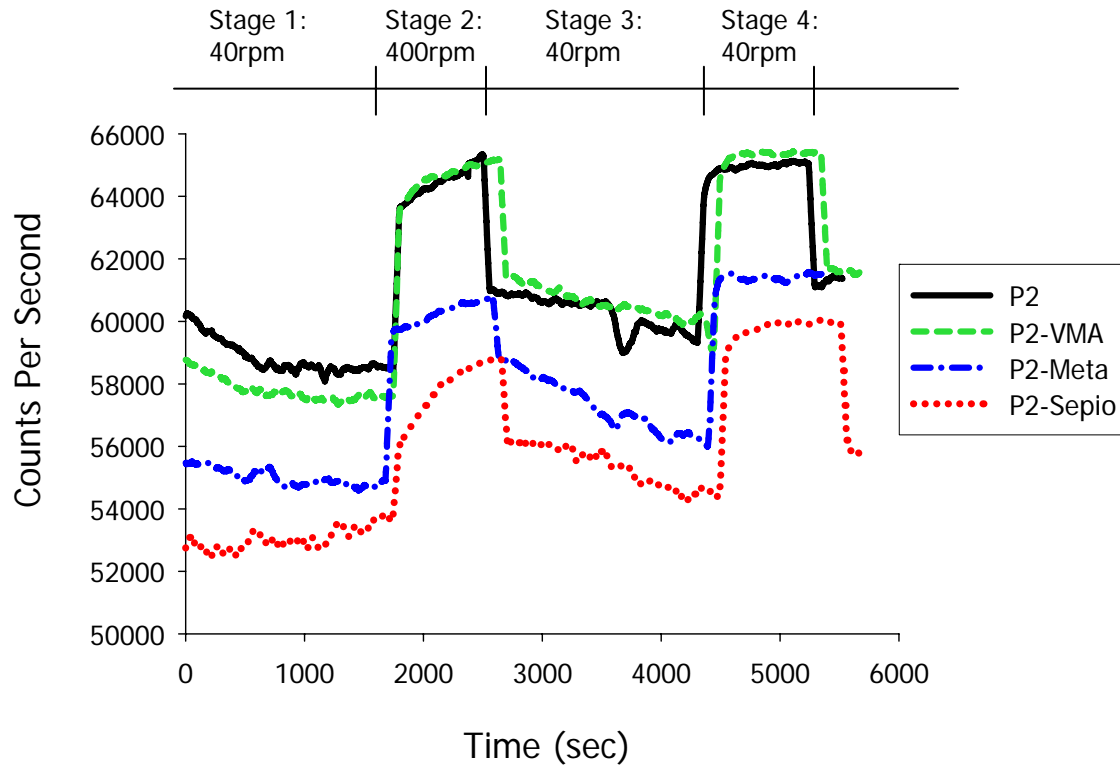


Figure 8-12. Influence of clays and VMA on particle counts evolution

However, significant changes in the aggregation and breakage kinetics were seen when clays were used. Regardless of the clay used in this study, an increase in the average chord size and degree of flocculation occurred when they were employed. The increase in mean chord size indicates that even at a small dosage of 1.5 %, the clays interact with the cement particles such that particle-particle agglomeration is enhanced. This may be due to the clay particles serving as micro/nano-fillers and increasing the connectivity of the network structure. In addition, due to

surface charge effects, clay minerals can form card-house type structures, and it is likely that

the formation of these card-house type structures will also contribute towards increasing the rate of aggregation. The aggregation mechanisms are influenced by the morphology of the particles. From previous investigations [156, 157], it was determined that the particles of the metakaolinite clay are characterized by spiny needles on their surface, and it has been demonstrated that small additions of this clay can improve the consolidation and shape stability of concrete mixtures during the slipform paving process [156]. In addition, Kuder and Shah [158] showed that kaolinite clays improve the cohesiveness of cement-based extruded materials at very low dosages [158]. Sepiolite particles are also composed of needle-shaped structures [159-162], and their ability to yield high viscosity values at low solid concentrations provides a major advantage over other clays [161]. Thus, the increased mean chord size may also be attributed to the needle-like particles of the clays increasing the flocculation efficiency by interweaving and bridging the solid particles. Modification of natural clays using netting and bridging reagents has been shown to increase the flocculation efficiency of solids particles in a freshwater system [160]. It is also likely that interactions between the chemical admixtures and the clay particles can also influence the flocculation and breakage kinetics. Studies have shown that clay surface modifications can affect the adsorption behavior of the polymers [142], flocculation mechanisms and rheological behavior of suspensions [142, 155], and hardened state microstructures [163].

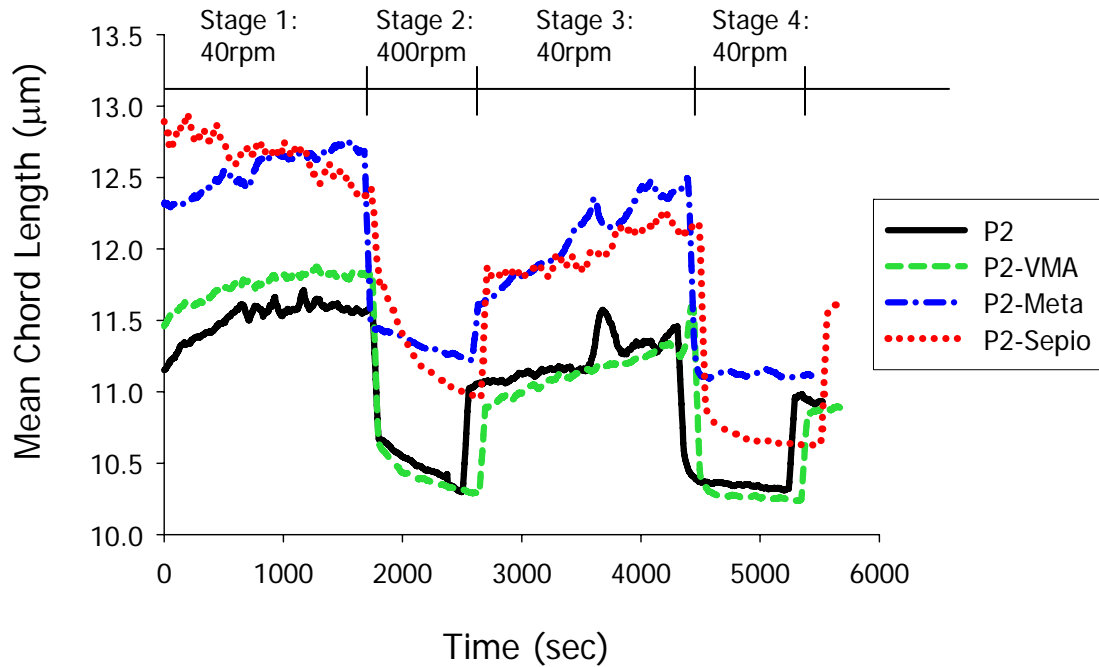


Figure 8-13. Influence of clays and VMA on average chord length

During Stage 2, a reduction in the number of measured chords was seen in all mixtures except P2-Sepio. The number of counts slightly increased in the P2-Sepio mix, which is indicative of floc breakage of the larger, weaker flocs. Information about the reflocculation ability is obtained from the behavior of the mixtures during Stage 4 (i.e. the recovery of the system after being subjected to the 400 rpm stirring intensity). When the stirring intensity was reduced, two responses were seen: an immediate reduction in the number of counts and then a more gradual reduction in the number of counts. Mixtures with clays showed the smallest immediate decrease in the number of counts, indicating that the flocs in these mixtures were more stable. With respect to the gradual evolution of measured counts, the P2-Meta exhibited the fastest decay, whereas the rate of reflocculation was relatively the same for the other mixtures. The faster rate of flocculation in P2-Meta is attributed to a larger quantity, and faster

rate of formation, of flocs within the 22 – 95 micron range, which may be due to enhanced particle aggregation from the needles bridging to form flocs.

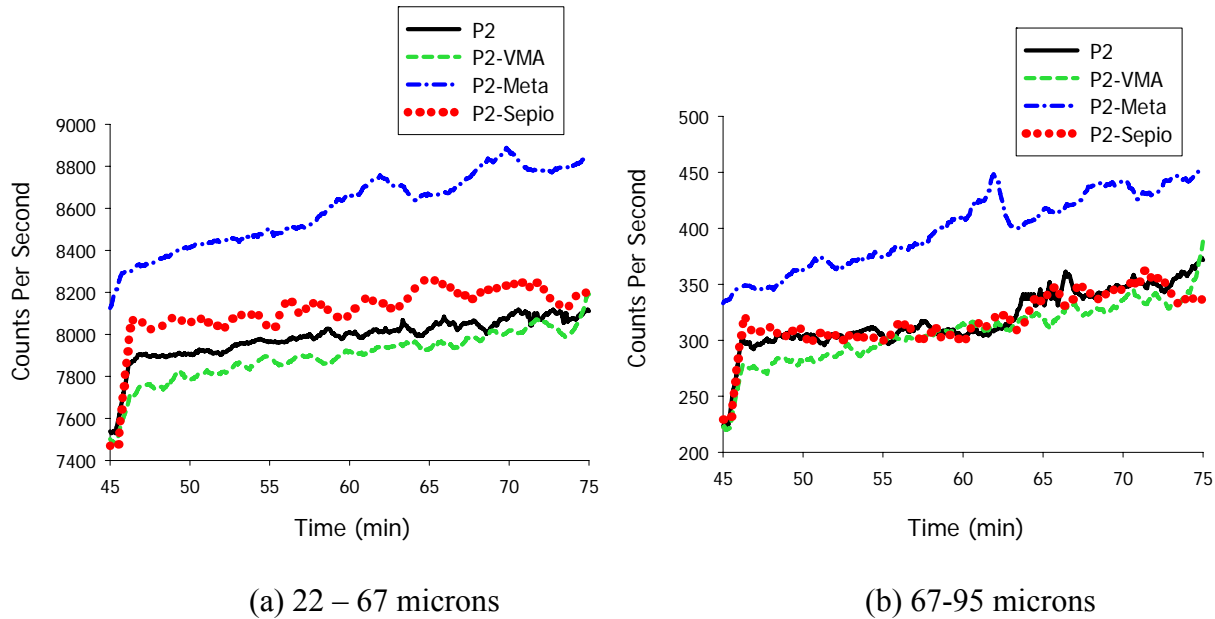


Figure 8-14. Influence of clays and VMA on floc size evolution during Stage 3

The response of the mixtures during Stages 2 and 4 provides information about the shear resistance of the flocs. As evidenced by the increase in the measured counts, when the stirring intensity is increased, the interparticle links ruptured. The reduction in the mean chord size provides further confirmation that the breakage process is the dominating mechanism during this stage. Table 8-5 contains the breakage kinetic constants for Stages 2 and 4. During Stage 2, lower k_2 values were seen in mixtures containing clays, and the flocs produced in P2 and P2-VMA were the least stable and most sensitive to shear-induced changes. The relatively low k_2 value of P2-Sepio is an indication that this mix is characterized by highly stable flocs with strong interparticle bonds that are able to oppose floc breakage and floc erosion. Analysis of the chord length distribution curve (not displayed here) showed that this was related to the stability of flocs

within the 1-5 micron range. In the case of mix P2-Meta, the relatively low k_2 value is attributed to highly stable large flocs, and as illustrated in Figure 8-15, a high percentage of the flocs greater than 65 micron were retained during the high stirring intensity phases. Thus, there exists a distinctive aggregation mechanism for the metakaolinite clay that encourages the development of large flocs that have a high resistance to shear forces.

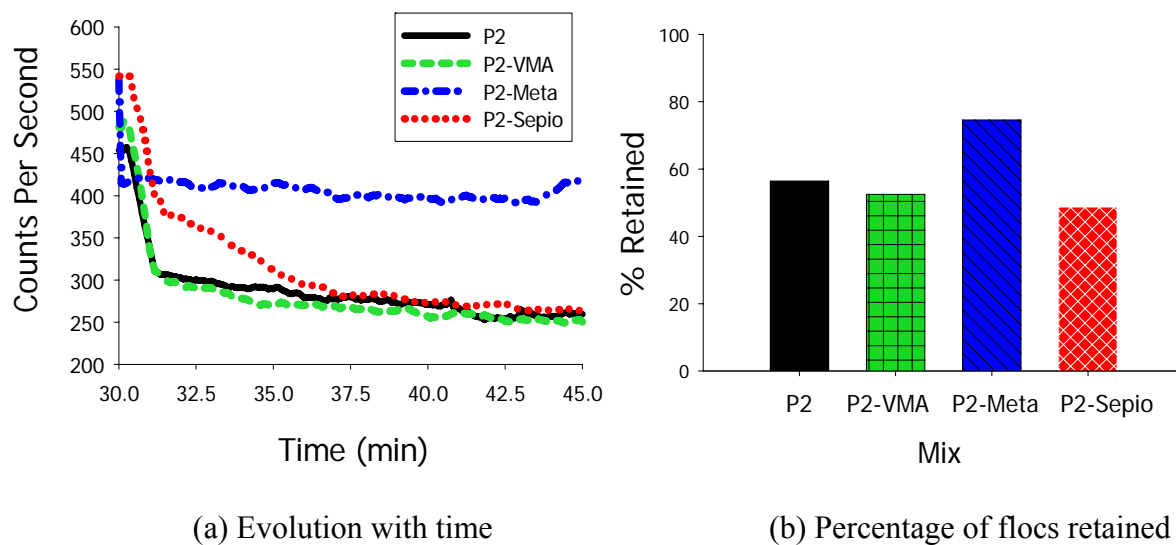


Figure 8-15. Floc sizes greater than 65 microns during Stage 3

Comparing P2 and P2-VMA showed that VMA did not significantly change the breakage kinetics of the paste mixture. This suggests that the VMA polymers induce a softer structural network and that a stiffer structural network is formed when the viscosity is increased using mechanisms that encourage particle-particle aggregation rather than mechanisms that thicken the aqueous phase. From the second deflocculation period, it is seen that all of the mixtures had good reversibility, and, in fact, higher k_2 values (and total number of particle counts) were measured during this stage. The results show that the deflocculation kinetic is highly influenced

by the shear history of the sample. This is directly related to the different microstructural arrangement of pastes (as seen in the differences in chord length and number of counts) in Stages 1 and 3. The high increase in P2-Sepio shows that its floc structure is more sensitive to shear history than the other mixtures.

Table 8-5. Breakage kinetics constant (N=400 rpm)

Mix	Stage 2 (t = 30 – 45 min)	Stage 4 (t = 75 – 90 min)
	k_2 (s ⁻¹)	k_2 (s ⁻¹)
P2	0.0205	0.0298
P2-VMA	0.0212	0.0356
P2-META	0.0149	0.0345
P2-SEPIO	0.0045	0.0277

8.4.3 Relationship between Structural Rebuilding and Flocculation

Figure 8-16 shows the chord length evolution for mixtures 9 and 10. Both mixtures had the same initial slump flow (330 mm), yet significant differences in the aggregation kinetics were displayed. During the first stage, little change is seen in the chord size evolution of mixture 9 indicating that the mixing energy during the sample preparation stage was high enough to disperse the cement particles. The higher mean chord size value of mixture 10 (in addition to the lower number of particle counts) shows that the microstructure of mixture 10 is more agglomerated than mixture 9. However, the flocs formed in the mix are unstable, and, because the flocs are held loosely together, they can be easily broken down and rearranged under shear. This results in a decrease in average chord size of mix 10 during Stage 1.

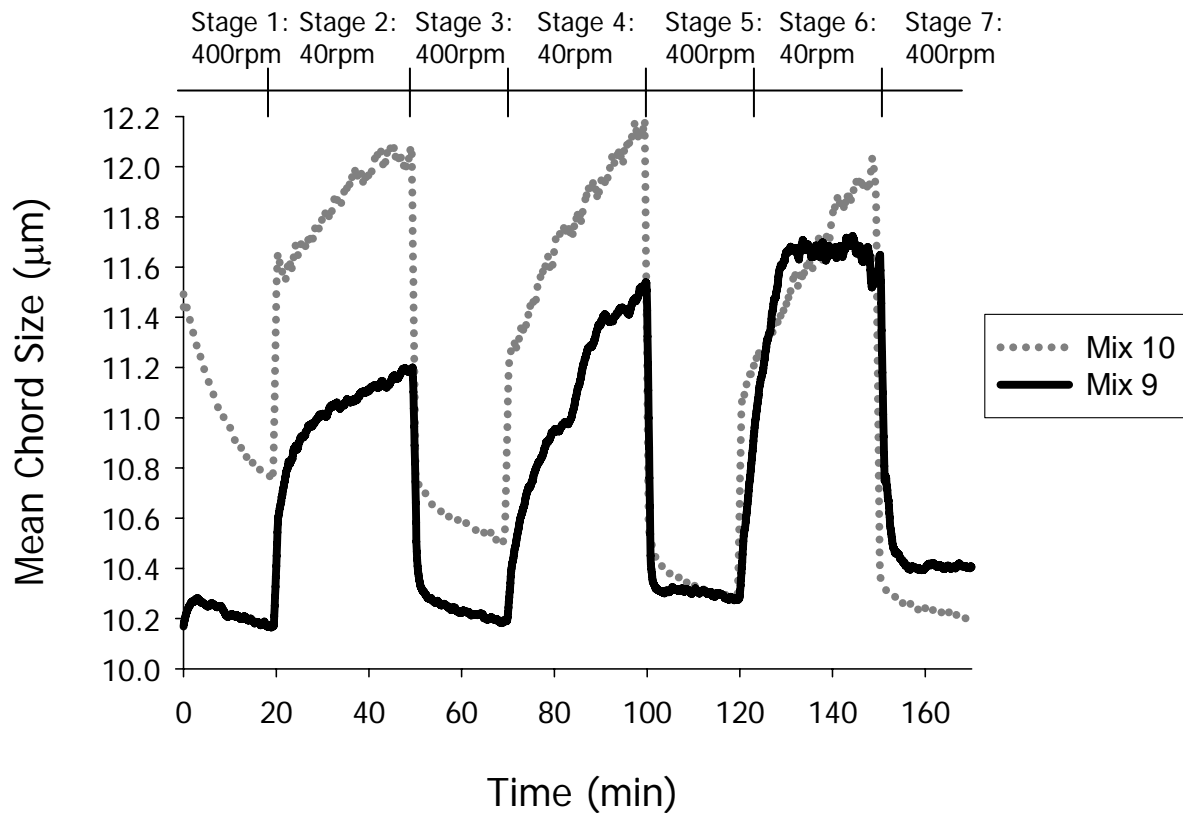


Figure 8-16. Chord Size Evolution for Mix 9 and Mix 10

As shown in Section 6.3, the initial SRE of mix 10 is greater than the initial SRE value of mix 9 (2678 J/m³*s versus 1339 J/m³*s). It was hypothesized that the magnitude of the SRE value can be used as an indication of the degree of structure. The results show that there is a correlation between the magnitude of the SRE value and the degree of structure and that the high SRE value of mixture 10 was indeed attributed to a more flocculated microstructure. Changes in the mean chord size were seen when the stirring intensity was cycled from 40 rpm to 400 rpm. Initially, all mixtures displayed a high degree of reversibility. Mixture 10 had the highest amount of reversibility, and the mean chord size at Stage 7 was actually smaller than the mean chord size at Stage 1. This indicates that there are weak bonds among the flocs formed in mix 10

and that these bonds are easily broken under the influence of shear. The reflocculation potential of mixture 10 is also very high, and it can be seen that the mean chord size at the end of Stages 4 and 6 are similar to the mean chord size at the end of Stage 2. This is very interesting because it shows that the microstructure of mix 10 tends to quickly restructure itself back to its initial state (and nothing greater) after a disturbance. This agrees with the results seen in Figure 6-2 in which mixture 10 had a high SRE with a relatively slow rate of rebuilding. The reversibility potential of mixture 9 decreases with time, and it is seen that the mean chord size at the end of Stage 7 was greater than the mean chord size at the end of Stage 5. This may be due to an increase in the bond strength within the flocs, which makes them less susceptible to breakage from the hydrodynamic forces, or to the formation of irreversible flocs due to the hydration process. The reflocculation potential increases over time and mean chord size at the end of Stage 4 was greater than the mean chord size at Stage 2 in mix 9.

As shown in the previous chapter, in terms of formwork pressure, a mixture that exhibits a behavior like mixture 10 would be preferable for reducing the maximum value of the formwork pressure when the casting process is relatively short because the highly flocculated state of mixture 10 provides more rigidity and the mixture quickly rebuilds back to this highly flocculated state after a disturbance. This would be useful in cases in which the concrete is refluidized during and after casting. However, mixture 9 would be preferable if the goal is to have the quickest cancellation time due to its fast rate of structural build-up.

8.4.4 Formwork pressure and flocculation

Figure 8-17 shows the chord counts evolution for mixes FA0 and FA20. Both mixtures were subjected to the following experimental protocol: stirring at 400 rpm for 20 minutes followed by

decreasing the stirring intensity to 40 rpm for 30 minutes to promote flocculation. Based on previous studies, it was found that the superplasticizer dosage used in this stage would give an average paste slump flow diameter of 330 mm for both mixes. Figure 8-18 shows the percent difference in the number of counts at the beginning versus the end of the 40 rpm flocculation. The percent difference can also provide insight into the rate of flocculation in which a higher magnitude represents a faster rate of flocculation. It can be seen that FA0 had a 103% increase in the number of flocs within the 95 – 250 micron range, whereas FA20 experienced a 31% increase. Similarly, for flocs within the 65 – 95 micron range, the percent increase was 52% and 25% for the FA0 and FA20, respectively. The reduction in the number of flocs within the 1 – 22 micron range was very small (less than 10% for both mixes), which hints at a rapid formation of new flocs from particles that were previously not measurable. The newly formed flocs counterbalance the decrease in the particle counts lost due to the particles within that size range forming larger flocs.

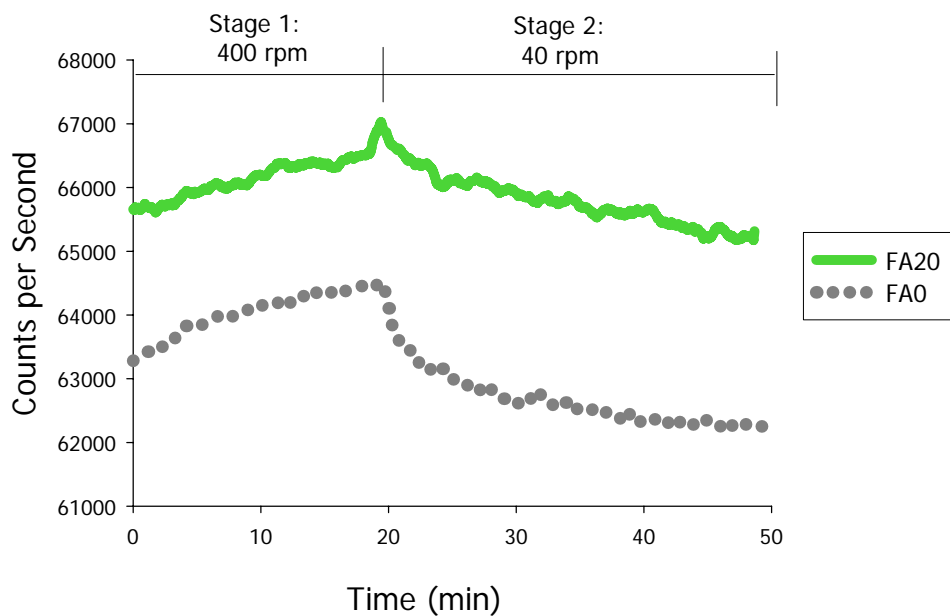


Figure 8-17. Influence of fly ash on number of counts

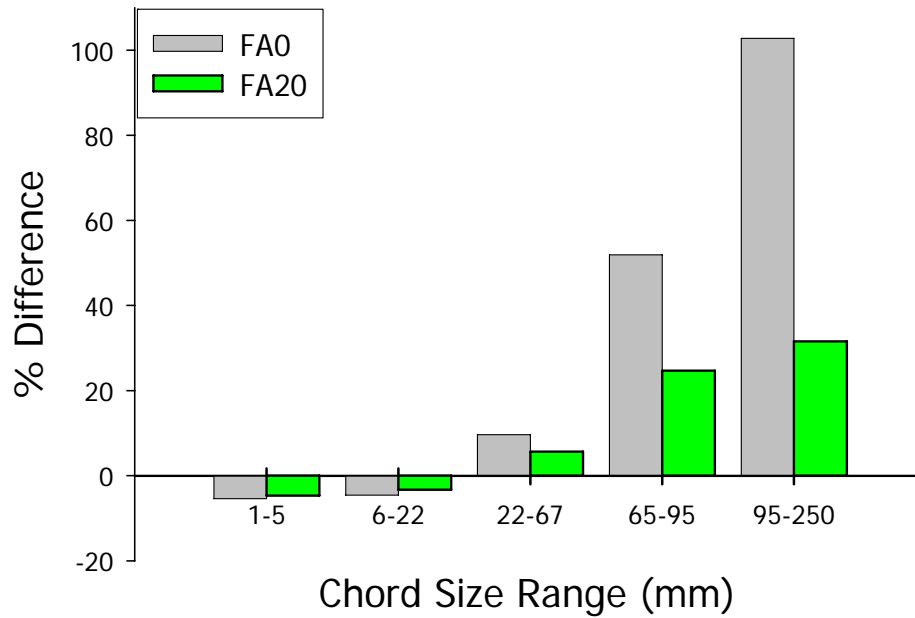


Figure 8-18. Percent difference from beginning to end of flocculation period

Values for the aggregation constant (k_1) and disaggregation constant (k_2) are given in Table 8-6. The k_1 value for FA0 is approximately two and a half times larger than the k_1 value for the FA20. Higher k_1 values indicate a faster aggregation process. Thus, the addition of fly ash is shown to decrease the rate of particle aggregation.

Table 8-6. Aggregation and disaggregation kinetics

Mix	k_1 (1/count *s)	k_2 (1/s)
FA0	3.21E-08	0.0020
FA20	1.39E-08	0.0009

The rheological method showed that fly ash typically yields a reduction in the rate of structural rebuilding, and the flocculation studies showed that a larger number of smaller flocs develop when fly ash is used. Insight into the stability of the flocs can be obtained from the disaggregation constant. Higher k_2 values indicate a lower floc strength (i.e. it is easier to separate the particles within the flocs). The value of k_2 decreases when fly ash is added. Hence, the floc strength increases when fly ash is added and causes the formation of smaller, more compact flocs (c.f. Figure 8-19). Since the flocs are more compact, more energy is required to break the interparticle bonds.

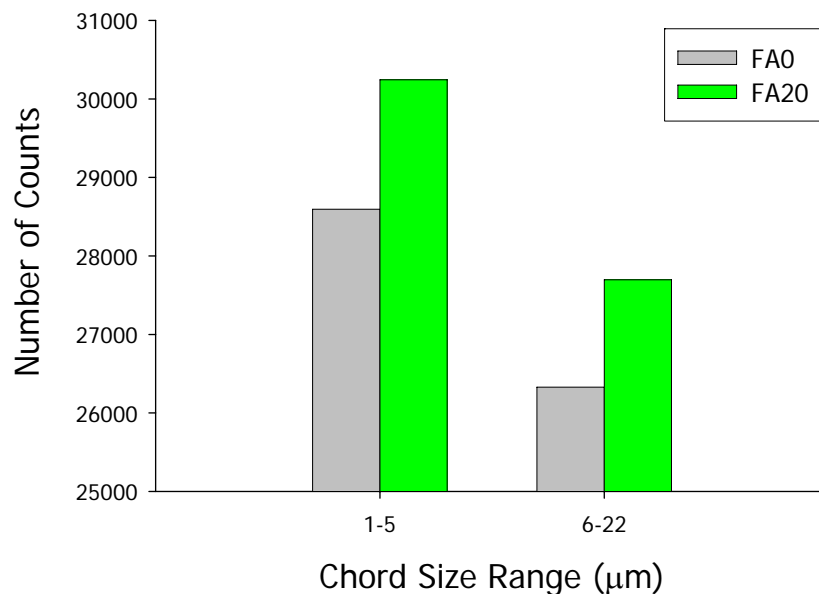


Figure 8-19. Comparison of small particles/flocs for mixtures made with and without fly ash

The effect of fly ash on the formwork pressure in cement paste was measured using the NU formwork pressure simulation device [132]. The simulated casting height was 14 m (46 ft) cast at a rate of 7m/hr (23 ft/hr). These tests were conducted using a different cement and higher fly ash dosage (30%). Figure 8-20 shows the structural rebuilding curves for these mixtures. As seen in the figure, the mixture made without fly ash had a faster rate of rebuilding and a higher magnitude of the SRE value than the mix made with fly ash. The pastes in the flocculation and structural rebuilding studies all showed a decrease in rate of rebuilding when fly ash is used. Although the flocculation data can not be directly compared to the formwork pressure data, the information from the flocculation study provides some insight into the behavior that was seen when fly ash reduces the rate of rebuilding.

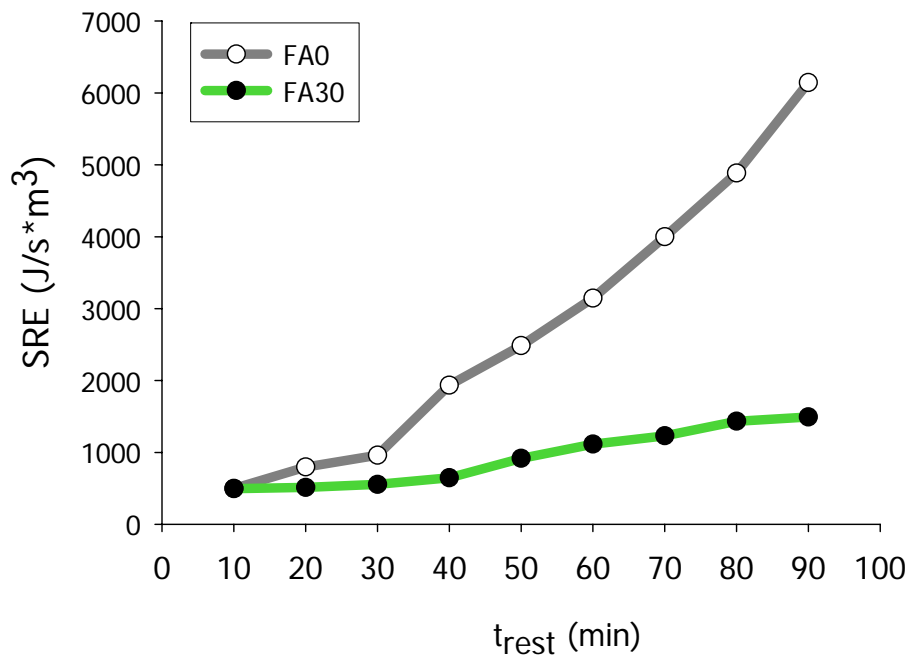


Figure 8-20. Structural rebuilding curve showing slower rate of rebuilding when fly ash is used

It is shown in Figure 8-21 that the addition of fly ash increased the peak pressure and reduced the initial rate of formwork pressure decay (note, hydrostatic pressure is 350 kPa.). This behavior can be explained from the slower rate of aggregation, which decreased the rate of structural buildup. During the casting process, the formwork pressure is reduced when fly ash is added because it takes more time for the flocs to form. The aggregation constant, k_1 , will play a large role in both the initial formwork pressure and formwork pressure evolution. High k_1 values will decrease the initial formwork pressure and increase the rate at which the formwork pressure decays due to the fact that the structural network develops at a faster rate. However, k_2 values will primarily influence the initial formwork pressure. A low k_2 value may reduce the initial formwork pressure since the structural network will be more effective in resisting shear-induced forces from the casting process (ex. pouring and pumping). The key lies in achieving the optimal balance between k_1 and k_2 . This behavior is seen in Figure 8-21 in which the initial formwork pressure between the two mixes were much closer due to the contribution from k_2 .

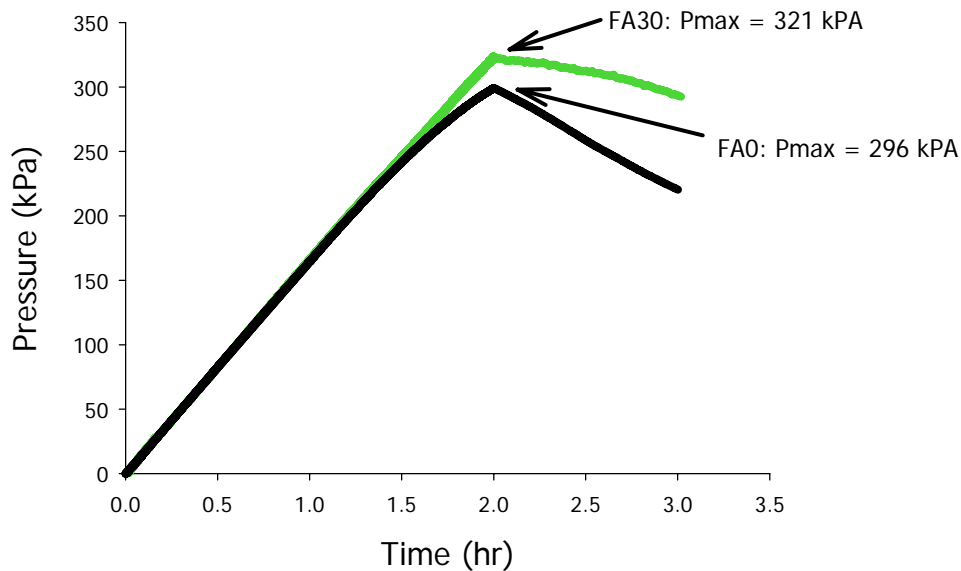


Figure 8-21. Influence of fly ash on formwork pressure

In the fly ash mix, although the flocs form at a slower rate once they form, they are harder to break apart. Once the pouring has ceased, the k_1 value is more significant since breakage of flocs from shear induced forces is no longer a factor. The formwork pressure decays faster for the mix made without fly ash, which has a k_1 value that is approximately 2.5 times greater than the mix made with fly ash. The particle size distribution of the fly ash varies widely. The fly ash used in this study was coarser than the cement. An improvement in the formwork pressure and rate of aggregation may occur when the fly ash particles are finer than the cement (i.e. using ultra fine fly ash).

8.5 Chapter Summary

The objective of this phase of the work was to gain further insight into the flocculation mechanisms within cement pastes. In order to accomplish this objective, a novel experimental technique was used to obtain *in situ* information about the evolution and size distribution of flocs within the fresh state microstructure. A kinetic model was successfully used to study the aggregation and breakage kinetics. Based on the research presented, the following conclusions can be drawn:

- The FBRM technique can be used to evaluate the particle size evolution of concentrated cement paste suspensions, and this technique can be used to identify the flocculation mechanisms of chemical admixtures.
- Aggregation and breakage kinetics of the paste matrix are highly influenced by shear history. The kinetics of re-aggregation is relatively slow, and the time scale for recovery is longer than the time needed to break down the structure.
- Aggregation and breakage kinetics in cement pastes can be modified based on the material ingredients. Floc stability is strongly influenced by the presence of chemical admixtures. The addition of SP drastically reduces the mean floc size, which creates a more dispersed suspension due to a reduction in particle aggregation.
- The largest chord size measured was 250 microns. Thus, there appears to be a limiting size to which the flocs will grow. Floc strength generally decreases as the floc size increases.
- VMAs that increase the stability of the paste matrix by adhering to the mixing water will not significantly change the rate of aggregation.

- Clays increased the average chord size of the paste suspensions, and resulted in a more flocculated network. The aggregation kinetics depends of the type of clay used. Both clays produced highly stable flocs with strong interparticle bonds that were able to oppose floc breakage and floc erosion. In the metakaolinite mix, this was characterized by the larger flocs (greater than 65 micron), whereas in the sepiolite mix, the increased floc strength was related to the smaller flocs (less than 5 microns).
- Replacement of cement with fly ash results in a less flocculated structure. Fly ash reduces the rate of rebuilding, but the floc strength increases when fly ash is added. This results in the formation of smaller, more compact flocs. This indicates that the rheological properties of cement pastes depend not only on the particle concentration and size distribution, but also on the floc strength.
- A relationship exists between the flocculation kinetics, structural rebuilding, and formwork pressure. The maximum lateral pressure is related to both the floc strength and the rate of aggregation, and the lateral pressure decay is related to the rate of aggregation.

PART VI: Conclusion

9 CHAPTER 9: Conclusions and Future Work

9.1 Introduction

The main objective of the research presented in this dissertation was to identify the effect of mixture composition on the rate of structural rebuilding and relate it to the lateral pressure generated by SCC. After the overview section, the manuscript was divided into five sections. Section I provided essential background information to understand the context of the work. Section II presented an experimental method based on rheological measurements that was developed to measure the structural rebuilding in cement pastes. Section III investigated the impact of mixture proportioning on influencing the structural rebuilding. Section IV investigated the relationship among formwork pressure, flocculation mechanisms, and structural rebuilding. This final section summarizes the major conclusions of this body of work and provides ideas about prospective work.

9.2 Conclusions

An experimental program was conducted to gain insight into the role of mixture proportioning on the structural rebuilding of high fluidity cement pastes and its implications for formwork pressure development. Emphasis was given to understanding the role of aggregation and breakage mechanisms on structural rebuilding. Based on the results of this research and well-established principles from the literature, the following key findings and conclusions were made:

- ***The rheological properties of the paste matrix and its evolution with time can be used to predict the formwork pressure behavior of the concrete.*** The use of rheological techniques provides a macroscopic way to measure structural rebuilding and to evaluate indirectly the microstructure of the fresh cement paste.

- ***Formwork pressure and structural rebuilding of SCC is highly influenced by the mixture proportioning of the paste matrix phase.*** With proper design of the matrix phase it may be possible to achieve significant reductions in lateral pressure development. A 30% reduction in formwork pressure (with respect to the hydrostatic pressure) was produced solely by altering the paste matrix to one in which the network interactions among particles were high.

- ***The lateral pressure exerted by SCC is related to both the rate at which structural rebuilding occurs and the total amount of structural build-up*** (i.e. overall degree of structure). The higher the total amount of structural build-up, the higher the shear strength properties of the concrete. This provides for further resistance to the applied vertical stresses and less initial pressure will develop. The formwork pressure cancellation is related to the rate of rebuilding and hydration effects. Mixtures with slower rates of rebuilding can cancel out similar times (or even earlier) than mixtures with faster rates of rebuilding depending on the rate of hydration.

- ***Shear rheology can be effectively used to monitor and evaluate the rate of structural rebuilding in fresh cement pastes.*** A rheological protocol was developed to measure the

structural rebuilding within cement pastes. The developed protocol provides an effective method to determine the rate at which a sample rebuilds. The effects of yield stress and viscosity are coupled together, and the protocol provides a measure of the total rate of rebuilding that occurred in a sample. The major advantage of this protocol is that it requires only one sample providing a relatively quick way to determine the rate of structural rebuilding.

- ***For a given consistency level, higher rates of structural rebuilding were seen when the consistency level was achieved by using a lower w/b ratio and slightly higher superplasticizer dosage versus using a higher w/b ratio and slightly lower superplasticizer dosage.*** Reducing the w/b ratio reduces the thickness of the water layer around the particle, which decreases the inter-particle distance. Furthermore, the solid volume content is increased when the w/b ratio is reduced, which results in more interparticle contacts. In addition, the ion concentration of the pore solution increases when the w/b ratio decreases, which also favors structural build-up due to a reduction in the repulsive interaction forces.

- Ideal thixotropic fluids show a decrease in viscosity when subjected to a constant shear stress, or shear rate, and then a gradual recovery when the stress or shear rate is removed. Furthermore, the recovery and breakdown is reversible. ***During the early stages of the induction period (~ first 30 minutes), it is possible for plasticized cement paste to show ideal thixotropic behavior.*** The length of time at which a cement paste will display ideal thixotropic behavior is largely dependent on the cement-superplasticizer interaction. ***At***

later stages in the induction period, cement pastes do not display ideal thixotropic

behavior due to irreversible structural build-up, which results in the formation of a stronger network structure.

- The Vicat needle test is not sensitive enough to monitor changes in the microstructure prior to initial set. However, a relationship was seen between the rate of structural rebuilding during the first 90 minutes and the time of initial set. ***Mixtures with faster rates of rebuilding showed lower initial setting times.*** This suggests that the mechanisms influencing structural rebuilding are closely linked to the mechanisms influencing stiffening.

- The FBRM technique can be used to evaluate the flocculation behavior of concentrated cement paste suspensions. This technique provides a direct way to evaluate structural rebuilding in fresh cement pastes on a microstructural level. ***A relationship exists among the FBRM particle size measurements, structural rebuilding determined with rheological tests, and the lateral formwork pressure. The maximum lateral pressure appears to be related to both the floc strength and the rate of aggregation while the lateral pressure decay is mainly a function of the rate of aggregation.*** The degree of structure (i.e. magnitude of SRE) is related to the total number of flocs and the size of the flocs. Generally, floc strength decreases as the particle size increases. However, mixtures with a high degree of structural build-up are characterized by lower particle counts and larger chord sizes. This suggests that connectivity of the structural network

increases when the particle size increases and that the rheological properties are also influenced by the floc strength.

- ***The FBRM technique can be used to identify the impact of chemical admixtures on the flocculation/dispersion of cement particles.*** When superplasticizers were added to the paste suspension, an immediate increase in the number of measured counts occurred. This indicates that superplasticizers increase the workability of concrete through mechanisms that cause a de-structuring of the interparticle network such that the cement particles are dispersed.

- ***Aggregation and breakage kinetics of the paste matrix are highly influenced by the shear history.*** The kinetics of re-aggregation is relatively slow, and time scale for recovery is longer than the time needed to break down the structure.

- When subjected to shear-induced stresses, the microstructure of cement paste suspension will tend towards an equilibrium condition or steady state condition in which the rate of breakage and rate of aggregation are equal. The response of the system will depend on the initial state of the structure. ***When the aggregation kinetics dominates, network interactions among particles develop and the average floc size increases. When the breakage kinetics dominates, the network interaction among particles is broken accompanied by a decrease in the average floc size. This gives rise to the macroscopic flow behavior of shear thinning and thixotropy.***

- The results suggest that there is *a limiting size to floc growth*. The maximum floc size correlates with the stirring intensity and is indirectly proportional to the stirring intensity. Thus, the maximum floc size decreases as the shear intensity increases.

The above statements are holistic conclusions and findings that can be applied to other studies regarding cement paste suspensions. However, no statements were made concerning the influence of materials on proportioning on the structural rebuilding and formwork pressure of SCC. As with any other concrete, the nature of the constituent ingredients of SCC depends on the availability of local materials, and the properties of the raw materials will vary from SCC to SCC. As such, different behaviors can occur based on the specific properties of raw materials. With these considerations in mind, the following conclusions can be drawn regarding the role of mixture proportioning:

- Structural rebuilding is highly influenced by the cement composition. The results show that the rate of structural rebuilding is related to the C_3A , soluble alkali content, and sulfate ion concentration. Faster rates of structural rebuilding were seen in pastes with lower contents of C_3A , soluble alkali, and sulfate ion. The superplasticizer demand was also found to be highly dependent of the cement composition, and higher superplasticizer demand was found in the cement with the higher C_3A and alkali content. The results indicate that cement-superplasticizer interaction plays a large role on the structural rebuilding behavior, and it was suggested that this is largely due to the amount of superplasticizer adsorption.

- The rate of structural rebuilding is significantly affected by the incorporation of silica fume. Faster rates of rebuilding were seen in paste mixtures in which silica fume was used as a cement replacement. This appears to be more related to physical effects and less to changes in the pore solution chemistry. The increase in the structural rate of rebuilding is attributed to stronger attractive forces due to the high surface area of the silica fume particles. The flocs made in pastes that contain silica fume are likely to be less susceptible to breakage and have faster rates of aggregation due to the increased colloidal forces of attraction. This increases the shear resistance of the concrete and reduces the initial formwork pressure.

- On a microscopic level, the rate of particle aggregation is reduced and the microstructure is less agglomerated when fly ash is used as a cement replacement. Thus, it takes longer for the structural network to build-up, and slower rates of structural rebuilding were seen when fly ash was added. These effects resulted in a higher formwork pressure value for the mixture containing fly ash. No significant difference was seen in the rate of rebuilding when the fly ash dosage was increased, which suggests that the initial formwork pressure behavior would be similar for the two fly ash contents. Thus, in terms of cost and environmental considerations, a higher dosage of fly ash could be used without detrimentally affecting the initial formwork pressure.

- The results indicate that VMAs that increase the stability of the paste matrix by adhering to the mixing water will not significantly change the rate of aggregation. This suggests

that the rate of structural rebuilding is largely controlled by factors influencing particle-particle interaction of the structural network.

- Minor additions of clays can significantly impact the structural network development of cement paste suspensions and result in a more flocculated structure. The flocs produced by the clays were highly stable flocs with strong interparticle bonds that were able to oppose floc breakage and floc erosion.

9.3 Future Work

The results from this study have many implications for the cement and concrete industry. The results showed that there is a linkage between structural rebuilding, flocculation, and formwork pressure. It was found that formwork pressure is highly influenced by the total degree of structure and that changing the composition of the paste matrix phase can significantly impact the formwork pressure behavior. Furthermore, for the first time the response of the microstructure to applied shear forces was directly explored. The results showed that the response of the system will depend on the aggregation and breakage kinetics. While much information has been gleaned, much research still remains to be conducted. The following section highlights a few areas where future research would be very valuable:

- **Focus Beam Reflectance Measurement Studies:** The results show that the FBRM technique could be used to assess the particle formation and breakage process in cement paste suspensions. However, there is still much work to be done concerning the

relationship between the chord length measurement and the actual floc size measurements in cement pastes. The relationship between the actual particle size and the chord length measurement can be analyzed by conducting FBRM studies on cement-alcohol suspensions and fly-ash-water suspensions. The particle size distribution via the FBRM study would be compared against the particle size distribution obtained from other methods in order to determine if the FBRM method is under-predicting or over-predicting the particle size.

- **Structure-Property Relationships:** As we move to concrete with more complicated rheological properties, it is important to know how the microstructure of the fresh paste matrix is influenced by processing. An experimental set-up in which the FBRM is used with a rheometer would provide a direct way to develop structure-property relationships. Research in this area could also focus on detailed characterization of the microstructure using FBRM and microscopy techniques to investigate the role of mixing energy, temperature, and delayed superplasticizer addition on the structure. Furthermore, studies can also be conducted to examine the relationship between the fresh state microstructure, fresh state properties, and hardened state properties. For example, the porosity of the hardened paste is directly related to the connectivity of the solid phases. The addition of micro/nano-particles to cementitious mixtures may result in a more homogeneous and finer pore structure, and, based on the results of this study, these fine particles will also influence the flocculation mechanism and microstructure of the fresh state. This leads to the following question: What role does the fresh state microstructure and rheology play in the development of the hardened state pore structure? And if there is a correlation

between the fresh state structure and pore structure, the next question is: Can the pore structure be micro-structurally engineered by controlling the fresh state microstructure?

- **Paste/Mortar Rheological Models:** Cement pastes and concrete are typically modeled as Bingham or Herschel-Buckley fluids. Due to structural rebuilding, the rheological properties are time-dependent and rheological models for cementitious suspensions are needed that can account for this behavior. Phenomenological equations have been developed to express the viscosity of a suspension as a function of the viscosity of the suspending medium, the solid volume fraction, and the characteristics of the particles. Work in this area could focus on developing a phenomenological equation that is applicable to concentrated cement pastes and mortars.

- **Relationship between Paste, Mortar, and Concrete Rheology:** This is perhaps the most fundamental issue concerning concrete rheology and has been the subject of much research. The relationship among paste, mortar, and concrete rheology is complex, but the ability to link these two would be beneficial because this would allow for the prediction of concrete rheology solely from the characterization of the paste or mortar phase. Research in this area should not only focus on the impact of aggregate volume and size, but also on the role of aggregate morphology and packing density on the rheological properties of the paste matrix. Computational modeling of flow behavior is an area that needs to be further explored.

- **Cement-Superplasticizer Interaction:** It was shown that a correlation existed between the rate of rebuilding and the chemistry of the aqueous phase. The rate of rebuilding is influenced significantly by type of cement selected, and the results showed that this is attributed largely to interactions between the cement and superplasticizer. Further, parameters should be investigated to evaluate how cement-superplasticizer interactions influence the rate of rebuilding and flocculation mechanisms. This includes studying the influence of superplasticizer adsorption, cement fineness, C₃A reactivity and dissolution, alkali sulfates, and the role of cement phases and hydrates.

- **Influence of hydration on structural rebuilding and initial set:** Structural rebuilding in cementitious materials is influenced by reversible and irreversible structural rebuilding due to hydration effects. Determining the contribution of irreversible structural rebuilding can provide insight into the mechanisms controlling initial set. Isothermal calorimetric studies could be used to evaluate the amount of irreversible structural rebuilding from hydration.

- **Role of clays on flocculation:** Significant increases in the floc size were seen with minor additions of clays. Research in this area could focus on determining the effects of clay morphology and clay particle surface modification on flocculation and rheology.

- **Formwork Pressure:** Comprehensive studies on the influence binder composition on the formwork pressure of concrete when subjected to various casting rates are needed. Other possible areas of further research include modifying the formwork pressure

simulation device to make it field-friendly, developing comprehensive formwork pressure models that account for structural rebuilding, and investigating the influence of frictional effects on maximum lateral formwork pressure.

- **Structural Rebuilding and Fresh State Properties:** The mechanisms that give rise to structural rebuilding, time-dependent viscosity, and time-dependent yield stress are interconnected. However, it is possible that the time-dependent viscosity may be more important for a particular fresh state property, and vice-versa. Research in this area could focus on evaluating how different properties (i.e. formwork pressure, segregation) are influenced by the time-dependent yield stress and time-dependent viscosity.

10 CHAPTER 10: REFERENCES

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APPENDIX A: Zeta Potential Testing

A.1. Introduction

Concrete is a concentrated suspension, and, as shown in the previous section, formwork pressure is influenced by the degree of aggregation within the paste matrix. Zeta potential provides a measure of the electrical potential of a particle and indicates to what degree particles within a liquid will tend to aggregate. In pure cement paste, the zeta potential typically ranges from -10 to 20 mV [82]. Chemical admixtures, such as water-reducing agents and viscosity-modifying agents (VMA), are commonly used to alter the particle dispersion and the aggregation mechanisms. In ordinary cement pastes, zeta potential studies have been used to investigate particle coagulation [85], flocculation mechanisms [86], and the effect of SPs on rheology [87]. In most of these studies very high dilution factors (ex. $w/c = 500$) were used, which altered the pore solution and, this will affect the zeta potential. Hence, in order to accurately measure the zeta potential it is important that one tests the particle in its original pore solution.

The objective of this stage of the research was to evaluate the capability of using zeta potential measurements as a means to evaluate the rate at which structural rebuilding occurs within a cement paste matrix.

A.2. Experimental Protocol

A Zetasizer Nano manufactured by Malvern Instruments was used in this study. This device uses electrophoresis to determine the zeta potential and is capable of measuring the zeta potential of particles ranging from 5 nanometers to 10 microns. One of the features of the device is that particle movement is tracked using phase angle light scattering, which is sensitive to very small

shifts in movement. Slow field and fast field reversal of the electric field are also used to determine the zeta potential. Slow field reversal reduces polarization of the electrodes and it provides a measure of the zeta potential distribution. Fast field reduces effects from electroosmosis so that particle movement is primarily due to electrophoresis.

Figure A- 1 shows the phase plot for a “standard” zeta potential fluid manufactured by Malvern Instruments. As shown in the figure, during the fast field reversal period the graph should show oscillations as the sample responds to the quick phase shifts.

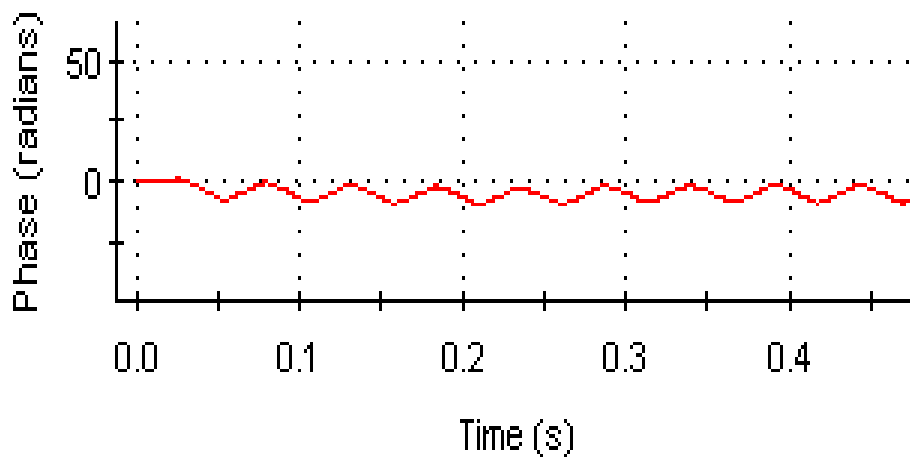


Figure A- 1. Phase plot for “standard” suspension during fast field reversal

Proper sample preparation is important in order to ensure reliable and accurate measurements. If the sample concentration is too low, not enough light will be scattered to make a measurement; however, if the concentration is too high, then multi-particle scattering may occur, which can prevent the laser from reaching the detector. In the Zetasizer Nano device, the laser power is automatically attenuated so that the count rate from the sample, especially high scattering

samples, is within acceptable limits. An attenuation index of 11 denotes no attenuation (full laser power), while 0 denotes full attenuation (total laser block) [164]. Poor quality data with an attenuation index of 11 can indicate that the particle concentration is too low and the sample concentration should be increased. Similarly, poor quality data with a low attenuation value of 0 can signify that particle concentration is too high and the laser light can not be transmitted. Hence, the particle concentration should be decreased.

During this stage of the program, careful attention was given to develop a sample preparation protocol that would yield results that are representative of the *in situ* conditions. After mixing the paste samples with a Hobart mixer, the samples were prepared for zeta potential testing. Initially, five different sample preparation sequences were evaluated:

- 1) Mix--Centrifuge--Filter--Zeta Potential Testing
- 2) Mix--Centrifuge--Filter--Suspend--Zeta Potential Testing
- 3) Mix--Centrifuge--Filter--Suspend--Rest--Zeta Potential Testing
- 4) Mix--Centrifuge--Filter--Rest--Zeta Potential Testing
- 5) Mix--Centrifuge--Rest--Zeta Potential Testing
- 2) Mix--Centrifuge--Filter--Suspend--Stir--Zeta Potential Testing

Mix refers to the 5 minute process in which the cement paste suspension was prepared with a Hobart Mixer. Centrifuge refers to the 8 minute process in which the samples were centrifuged. Filter refers to the process in which the supernatant from the centrifuging process was filtered through a syringe filter. Suspend refers to the process in which a portion of the original cement sample was added to the supernatant extracted from the centrifuge step. Stir refers to the process

in which the supernatant sample (with or without suspended particles) was agitated with a stirring rod. Zeta potential testing refers to the point at which the supernatant sample was transferred to the zeta potential device.

A.3. Results

During this stage of the study, all testing was performed on a cement sample with a w/c ratio of 0.5 and DI water. Figure A- 2 shows the results obtained using the first sample preparation protocol in which the supernatant was filtered through a 0.2 micron syringe filter prior to testing. The average zeta potential was 6.416 mV, and the first standard deviation around the mean was 164.8 mV. The attenuation value was 10 and this indicated that the sample concentration was too low.

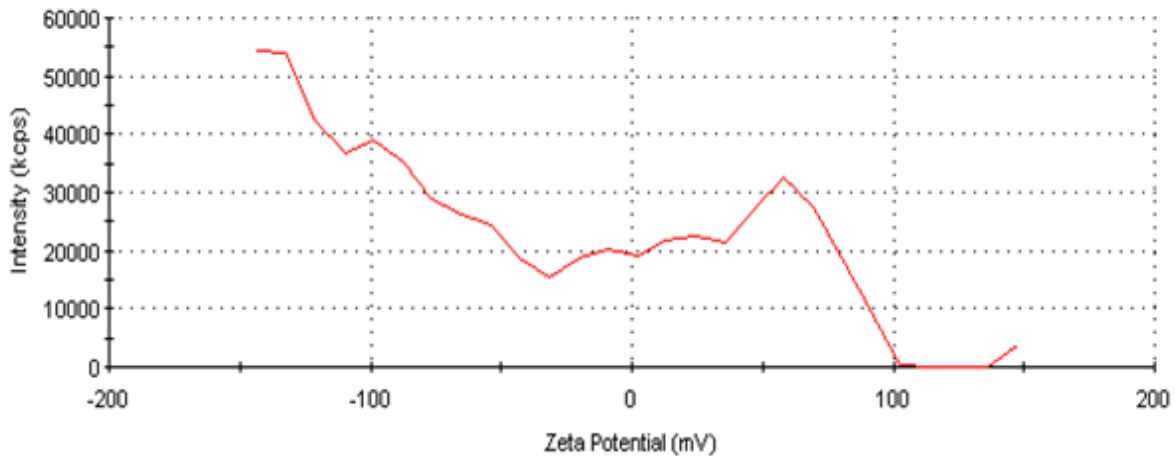


Figure A- 2. Zeta Potential Distribution for sample prepared using sample preparation protocol 1

The phase plot exhibited no oscillations and is shown in Figure A- 3. The fact that the zeta potential distribution has a wide distribution, a high attenuation value, and a smooth phase plot were all indicators that the data quality was not good and that the sample preparation technique needed to be modified.

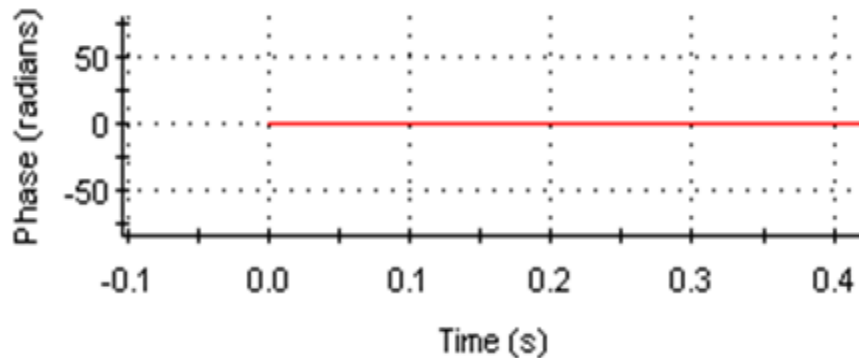
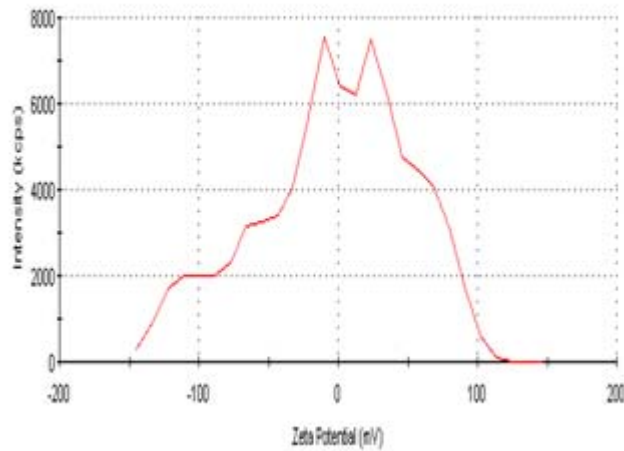


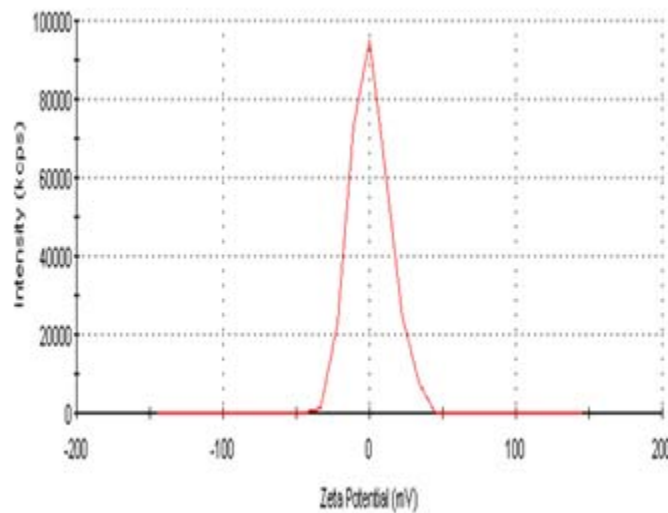
Figure A- 3. Phase plot for sample prepared using sample preparation protocol 1

The zeta potential distribution for a sample tested using sample preparation technique 2 is shown in Figure A- 4. The sample was filtered through a 1.2 micron syringe filter and 1 drop of cement paste was added to the supernatant prior to zeta potential testing in order to increase particle concentration. Five measurements were taken on this sample. Figure A- 4a shows the results of the first measurement and Figure A- 4b shows the results from the last measurement, which was taken eleven minutes after the first measurement. A broader distribution was seen in the first measurement with two distinct peaks obtained at -19.53 mV and 34.91 mV. The average zeta potential, zeta potential deviation, and attenuation value were 1.745 mV, 123.4 mV, and 9, respectively. In Figure A- 4b, the average zeta potential, zeta potential deviation, and

attenuation value were 0.6317 mV, 13.32 mV, and 9, respectively. Only one peak was seen in the measurement and the distribution was much smaller.



(a) initial



(b) final

Figure A- 4. Zeta potential distribution for sample prepared using sample preparation protocol 2

It must be noted that the zeta potential is actually the average value based on the results of the zeta potential distribution of all the particles in the suspension. It was very common to obtain zeta potential with very broad distributions (as shown in Figure A- 2). Since cement is a multiphase particle, this distribution varied considerably from measurement to measurement, and it was very difficult to obtain the same results even if the testing conditions were carefully repeated. For a suspension composed of homogenous particles, the distribution is narrow with one well-defined peak. It was observed that the zeta potential distribution for the cement paste sample tended to a distribution with a sharper peak as the number of measurements on a sample increased. However, numerous measurements (at least 10 measurements of 25 runs) were needed in order to improve data quality. While some results were clearly outliers, it was often difficult to decide which values were due to true differences in material behavior and not just noise or experimental error. The zeta potential measurement is based on the movement of the particles to the applied electric field, and movement due to sedimentation (i.e. gravitational forces) can cause erroneous values since this movement is not accounted for in the measurement principles of the zeta potential device. Peaks from particle sedimentation would cause the values for the zeta potential to change with time since the location of the particles in the measurement cell is continuously changing due to the settling of the particles. However, if the multiple peaks were due to the heterogeneous nature of the cement particles, then these peaks should still appear in the zeta potential graphs at later times since heterogeneity is not temporal. In Figure A- 4b, a single, distinct peak occurs suggesting that the multiple peaks seen in the earlier measurements were due to the particle movement from sedimentation.

The influence of particle concentration is shown in Figure A- 5. The sample was prepared using sample preparation technique 3 in which a 1.2 micron syringe filter was used and 6 drops of cement paste were resuspended in the supernatant. The sample was allowed to rest for 10 minutes before the top portion of the suspension was transferred to the zeta potential device for testing. Increasing particle concentration did not increase data quality. Broader distributions and multiple peaks were obtained when the particle distribution was increased. This is due to the greater number of particles settling in the system. Even though the sample was allowed to rest before testing, while transferring the sample to the testing cell, it is likely that large particles were also transferred and settled during the measurement.

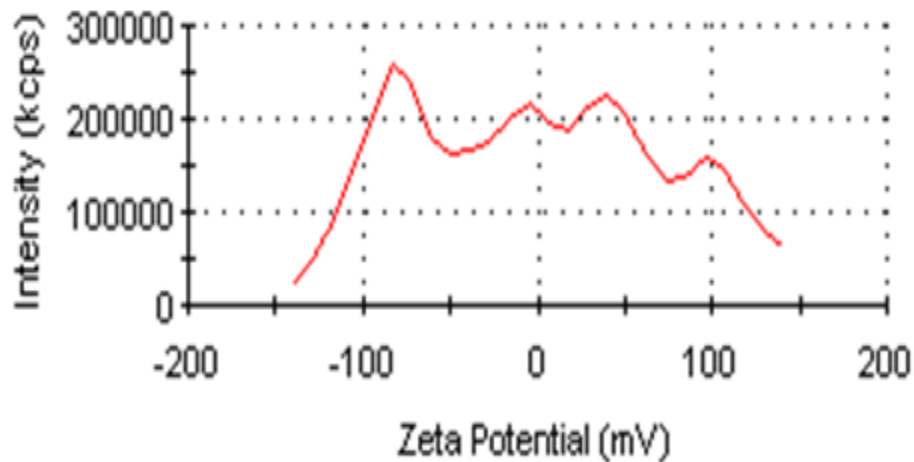


Figure A- 5. Zeta potential distribution for sample prepared using sample preparation protocol 3

Figure A- 6 shows the temporal evolution of the zeta potential distribution of the sample shown in Figure A- 5. As seen in Figure A- 6, the number of peaks decreased with measurement time. In the first measurement (labeled 1) four peaks were seen in the measurement. In the second measurement (labeled 2), which was taken two minutes after the first measurement, 3 peaks were seen. However, in the third measurement (labeled 3), 1 major peak and 2 smaller humps were seen in the graph indicating that with more time the measurement would be more stable and better quality data would be obtainable.

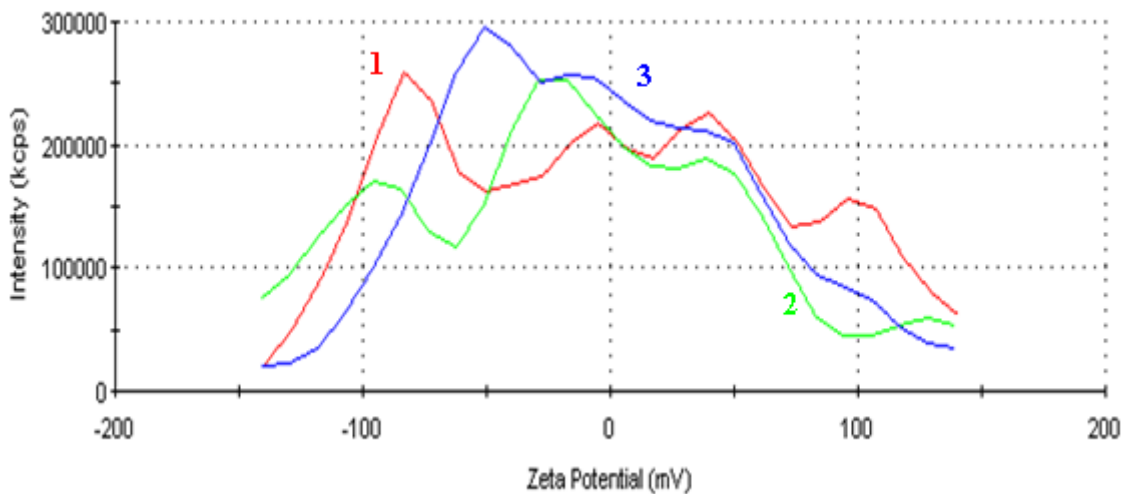


Figure A- 6. Temporal evolution of zeta potential distribution for sample prepared using sample preparation protocol 3 (numbers corresponds to different measurement times, with 1 = first measurement, 3 = third measurement)

Based on the above results, a protocol that involved centrifuging the cement paste and filtering out the large particles was developed. The protocol consisted of centrifuging the samples for five minutes at 10,000 rpm in a Sorvall RC 5B centrifuge. Approximately 20 mL of pore solution was extracted from 240 mL of centrifuged paste, and this pore solution was filtered through a 1.2 μm filter in order to remove the large particles that may settle during the zeta potential testing. The filter size of 1.2 μm was large enough so that the solids concentration of the filtered solution was high enough for the ZetaSizer Nano instrument, but large amounts of particle settling did not occur. Other filtering approaches were tried but produced poor results. A 0.2 μm filtered solution had too low of a particle count rate for the machine. A 0.8 μm filter yielded variable results that were sometimes acceptable but several times the count rates were too low for the ZetaSizer Nano instrument. Another approach was to use unfiltered pore solution. This yielded results with high deviations, but, after about 8 measurements of 50 runs, the data became fairly consistent. Re-suspending particles from the initial cement paste after filtering was investigated but was found to disrupt zeta potential measurements as large particles settled in the sample cell during testing. Care was taken to cover filtered and unfiltered pore solution to reduce effects associated with carbonation.

After developing the protocol, the zeta potential of four mixtures was measured at different times.

Table A- 1 contains the mix design of the mixtures evaluated. The w/c ratio was held constant at 0.35, and, in mixtures in which superplasticizer was used, the superplasticizer dosage corresponded to the dosage necessary to achieve an initial slump flow of approximately 330 mm. The superplasticizer was a carboxylated polymer in aqueous solution (specific gravity = 1.04) and the dosages shown in Table A- 1 were calculated based on the mass of the aqueous solution

with respect to the mass of the cement. The physical and chemical properties of the cements were given in Table 6-1.

Table A- 1. Mixture proportions for zeta potential tests

Mixture Name	Cement	SP dosage (% cement)
LowCem	Cement 6 (low alkali/C ₃ A cement)	0
HighCem	Cement 5 (high alkali/C ₃ A cement)	0
LowCemSP	Cement 6 (low alkali/C ₃ A cement)	0.14
HighCemSP	Cement 5 (high alkali/C ₃ A cement)	0.80

The zeta potential was measured at approximately 40-50, 120-150, and 200-250 minutes (taken with respect to the time centrifuging commenced). The range of times reflects differences in the amount of time it takes to transfer the cement to centrifuge tubes, which depends on the consistency of the paste.

Figure A- 7 shows the results for mixtures made without any superplasticizer. The zeta potential of HighCem is negative, whereas LowCem has a positive zeta potential. Ignoring the sign of the zeta potential, the magnitude of both mixtures increases over time. Initially the zeta potential of HighCem was -3.8 mV compared to an initial zeta potential of +1.8 mV for LowCem. A high zeta potential ($\zeta > 25$ mV) indicates that a solution or dispersion will resist aggregation. The low zeta potential value in both mixtures indicates that the suspensions are coagulated (strongly aggregated). Initial set occurred in HighCem before the third measurement was conducted, and, thus, only two readings were recorded.

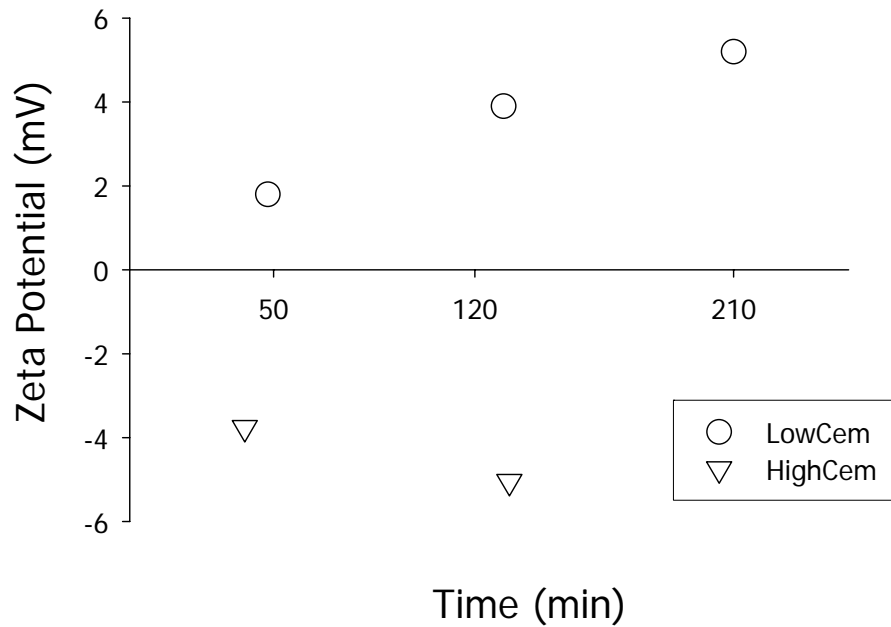


Figure A- 7. Influence of cement type on zeta potential

Figure A- 8 and Figure A- 9 show the effect of superplasticizer on the zeta potential. Regardless of the mix, the addition of superplasticizer increased the initial magnitude of the zeta potential. A sign reversal was seen in the mixture made with the high alkali/ C_3A cement (HighCem) when superplasticizer was added. This indicates that the superplasticizer is dispersing the cement particles through electrostatic and electrosteric repulsion mechanisms.

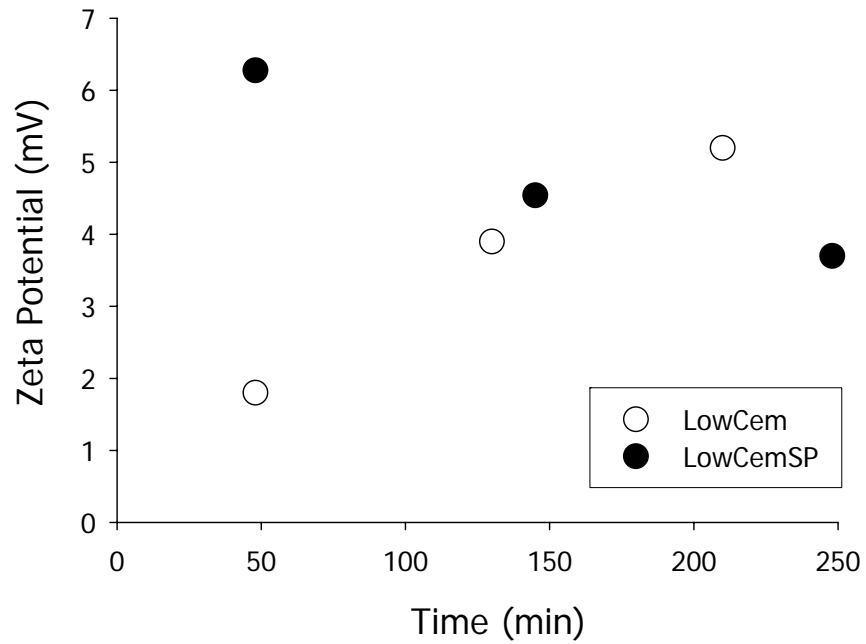


Figure A- 8. Influence of superplasticizer on low alkali/C₃A cement

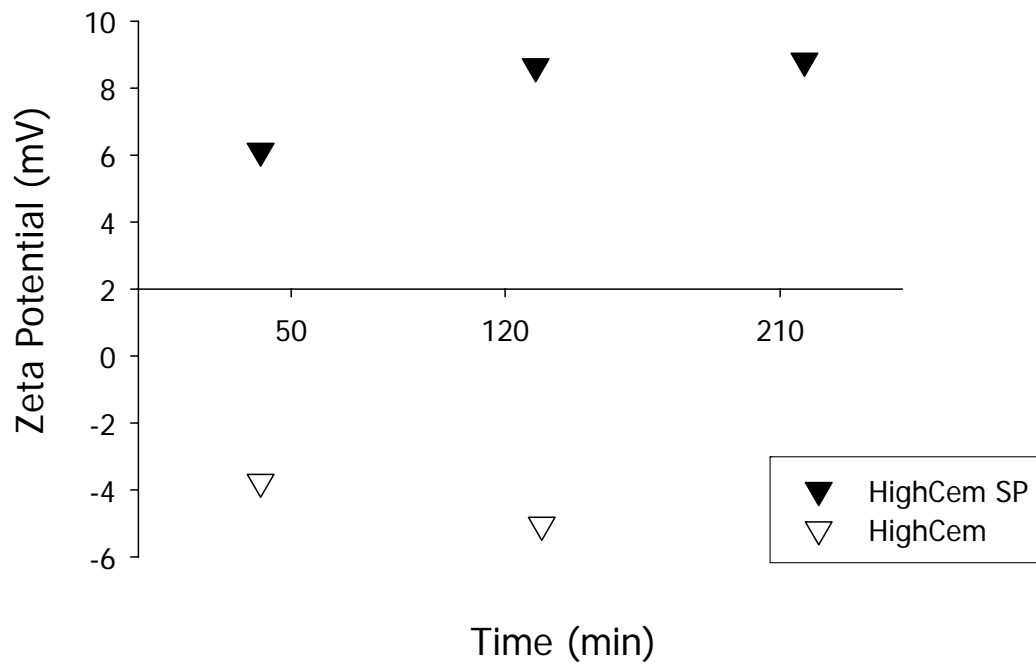


Figure A- 9. Influence of superplasticizer on high alkali/C₃A cement

In general there does not seem to be significant changes in the zeta potential values measured over time (exceeding a magnitude of 5 mV with respect to the preceding measurement). Slight changes may exist, but with the difficulty of interpreting data, these changes are hard to distinguish from the inherent variation of the cement paste system if they do not have a strong and obvious effect on the average zeta potential value.

VITA

Raissa Patricia Ferron was born on April 10, 1980 in Brooklyn, New York, under her maiden name Raissa Patricia Douglas. She is the daughter of Celestine Harold Douglas and Patresia Bernadette Douglas. Raissa defended her Ph.D. work on November 13, 2008. She has accepted a faculty position at the University of Texas at Austin, and will start there in January 2009 as an Assistant Professor in the Department of Civil, Architectural and Environmental Engineering.

M.S. in Civil Engineering; Northwestern University, Evanston, IL, Granted June 2004, Thesis: Development of Self-Consolidating Concrete with Type F Fly Ash: With a Verification of the Minimum Paste Volume Method, *Advisor: Professor Surendra P. Shah*

B.S. in Civil Engineering; Howard University, Washington, D.C., Granted May 2002; *Summa Cum Laude*

Research Experience

Research Assistant, 09/2002 – Present, Northwestern University

Studied the fresh state and hardened state properties of self-consolidating concrete (SCC).

- ❖ Developed rheological protocol to quantify the structural rebuilding in the paste matrix.
- ❖ Developed a field-friendly viscometer for high fluidity cement pastes and mortars.
- ❖ Devised experimental laboratory device to simulate formwork pressure and developed prediction models for SCC formwork pressure.
- ❖ Evaluated flexural strength, compressive strength, fracture properties, and shrinkage of steel reinforced and polypropylene reinforced SCC specimens.
- ❖ Evaluated the influence of fly ash on chloride permeability resistance.

Visiting Researcher, 06/2007 – 08/2007, *Universidad Complutense de Madrid, Madrid, Spain*

- ❖ Investigated and modeled the flocculation mechanisms of cementitious materials.

Teaching and Research Mentoring Experience

❖ **Properties of Concrete**

- *Undergraduate Student Research Advisor (Spring 2007):* Developed unique research projects for students to conduct as part of their term project. Directly supervised the research of four undergraduate students. Mentored students about career plans and graduate school
- *Teaching Assistant (Spring 2005):* Developed and graded exams. Organized and updated class lecture notes and syllabus. Scheduled guest lecturers. Managed course website.

- *Laboratory Instructor (Spring 2005)*: Prepared and organized laboratory handouts. Gave lectures. Demonstrated and assisted students in laboratory techniques regarding rheological testing, concrete mechanical testing, concrete durability testing, and mechanical testing. Graded laboratory reports.
- ❖ **Reinforced Concrete Design**
 - *Teaching Assistant (Spring 2006)*: Developed and graded homework and exams. Organized and updated class lecture notes and syllabus. Scheduled guest lecturers. Managed course website.
- ❖ **Mechanics of Materials**
 - *Grader (Winter 2007)*: Graded homework assignments.
- ❖ **Modern Materials and Society**
 - *Science/Engineering Research and Teaching Synthesis Advisor (Spring 2007)*: Created a research project for four non-science/engineering undergraduate. Directly supervised and advised students.
- ❖ **Accelerated Inorganic Chemistry**
 - *Guest Lecturer (Fall 2005-2007)*: Gave a lecture on the use of chemistry in concrete technology.
- ❖ **Center for Advanced Cement-Based Materials**
 - *Undergraduate Student Research Coordinator (Summer 2005)*: Trained and supervised an undergraduate student conducting research.

Honors and Awards

- ❖ National Science Foundation US/Africa Materials Research Institute Travel Fellowship, 2007
- ❖ Featured in Winter 2007/Spring 2008 issue of Diversity/Careers in Engineering & Information Technology Magazine
- ❖ Georgia Institute of Technology Focus Fellow, 2007
- ❖ Southern Regional Education Board Doctoral Scholar, 2007
- ❖ ASTM International Katharine and Bryant Mather Scholarship, 2006
- ❖ McCormick School of Engineering I &G Dissertation Year Fellowship, 2006
- ❖ Illinois Ready Mixed Concrete Association Harvey H. Hagge Concrete Scholarship, 2006
- ❖ Invited Participant of National Science Foundation ADVANCE Workshop on Transforming the Professoriate: Preparing Women for Careers in Science and Engineering, Blacksburg, Virginia, June 20-22, 2006
- ❖ Invited Participant of NSF-funded Workshop for the Advancement and Retention of Underrepresented and Minority Engineering Educators, Arlington, Virginia, March 5-8, 2006
- ❖ First Place Award ACBM Poster Competition, 2005
- ❖ Alliance for Graduate Education and the Professoriate (AGEP) Scholar, 2005
- ❖ American Concrete Institute Student Fellow, 2002
- ❖ Portland Cement Association Fellow, 2002
- ❖ Illinois Minority Graduate Incentive Fellow, 2002
- ❖ National Science Foundation Fellowship—Honorable Mention, 2002
- ❖ Northwestern University P. Murphy Fellowship, 2002
- ❖ CDM Certificate of Commendation for Technical and Service Excellence, 2002
- ❖ UNCF/Camp Dresser & McKee Fellow, 2001
- ❖ Tau Beta Pi Nagel Scholar, 2001
- ❖ All American Scholar, 2001
- ❖ Who's Who in American Colleges and Universities, 2001
- ❖ Professionals Dedicated to Excellence Coin Presented by Japan Engineer District Commander, 2001
- ❖ National Society of Collegiate Scholars, 2000
- ❖ Howard University's Legacy Scholarship, 1998
- ❖ Essence and Pepsi-Cola Youth Leadership Award, 1998

Research and Travel Grants

- 2007 Northwestern University Graduate School/ Alliance for Graduate Education and the Professoriate**
 Proposal Title: Flocculation Mechanisms in Cement Pastes
 Awarded \$3000 travel grant to conduct research overseas
- 2006 National Science Foundation (NSF)**
 Division of Civil and Mechanical Systems (CMS); Infrastructure Materials and Structural Mechanics Program
 Proposal Title: Thixotropy and Formwork Pressure of SCC; Grant No: 0625606
- 2005 Federal Highway Association (FHWA)**
 Proposal Title: Highways 2008: Innovative Infrastructure Materials and Quality Control Techniques
 Award No: DTF H61-05-C-00001
- 2004 Ready-Mix Concrete Research Foundation; American Concrete Institute Strategic Development Council**
 Proposal Title: SCC Formwork Pressure

Leadership Positions

- ❖ **Treasurer**, CEE Graduate Student Board, 09/2006 – Present, Northwestern University
- ❖ **Civil Engineering Delegate**, Engineering Graduate Leadership Council, 09/2006 –08/2007, Northwestern University
- ❖ **President**, Black Graduate Student Association, 05/2004 – 05/2005, Northwestern University
 Chairperson of the 9th Annual Graduate and Professional Student Research Conference held in April 2005 which had over 100 attendees from all over the country. To date, the conference was the best attended one in BGSA history.
- ❖ **Secretary**, CEE Graduate Student Board, 05/2003 – 05/2004, Northwestern University
- ❖ **President-Elect**, Black Graduate Student Association, 05/2003 – 05/2004, Northwestern University
- ❖ **President**, American Society of Civil Engineers, 09/2000 – 06/2002, Howard University
- ❖ **Vice-President**, Tau Beta Pi National Engineering Honor Society, 09/2000 – 06/200, Howard University

Professional Development

- ❖ **Preparing Future Faculty Program Participant**, 10/2004 – 06/2005, Northwestern University
 Attended monthly workshops focused on preparing doctoral students for the professoriate
- ❖ **ACBM/Portland Cement Association Undergraduate Faculty Enhancement Workshop**, 07/2004, Skokie, IL
 Participant in a four day program geared towards effectively teaching concrete materials.

University Service and Outreach Activities

- ❖ **Northwestern University Graduate Student Recruiter**, 03/2003 – Present, Northwestern University
- ❖ **Student Life and Multi-cultural Affairs Graduate Student Mentor**, 08/2003 – Present, Northwestern University
 Planned activities that focused on the recruitment, retention and graduation of underrepresented students in the fields of science, technology and engineering.

- ❖ **Graduate Leadership Council (GLC)**, 05/2003 – 05/2005, Northwestern University
Served as a member of a council composed of ten graduate students and representatives from The Graduate School, and The Office of the Vice President for Student Affairs. Presented issues to the Board of Trustees concerning graduate student needs. Met with internal and external reviewers for the program review of Northwestern University.
- ❖ **Seminar Coordinator**, 2004, Northwestern University
Organized First Joint Industrial Engineering Departmental Seminar Series and Distinguished Minority Engineering Lecture Series.
- ❖ **Career Day for Girls Lab Tour Host**, 02/2003 and 02/2004, Northwestern University
- ❖ **Summer Research Opportunity Program (SROP) Mentor**, Summer 2003 and 2004, Northwestern University
- ❖ **Communications Task Force Committee Member**, 06/2003, Northwestern University
Served on a committee composed of faculty, staff and graduate students which focused on improving communication across and within constituencies of the College of Engineering

Industry Experience

- ❖ **Civil Engineer Intern**, Camp, Dresser, & McKee (CDM); Summer 2002, New York, NY
Performed quality control verification on air analysis tests and soil tests to ensure that design details were accurate. Wrote property summary reports to inform residents on the amount of contamination that was found on their property.
- ❖ **Engineering Technician**, U.S. Army Corps of Engineers (AMIE Program); Summer 2001, Camp Zama, Japan
Accompanied design engineer for site inspections of facilities prior to designing drawing plans. Conducted Design Reviews and Bid ability, Construction ability, Operation ability, and Environmental reviews. Accompanied resident engineer on safety inspections and safety surveys during construction phase. Accompanied resident engineer for site visits to inspect concrete cylinders and pile driving.
- ❖ **Assistant Superintendent**, York Hunter Construction Services (INROADS, NYC), Summer 2000, New York, NY
Helped organize and update manpower reports, daily logs, requests for information, and construction schedule. Created a Safe Work Plan for Office of Self-Insurance Plans (OSIP) which outlined the major activities occurring on site and the safety guidelines to follow to take minimize risk of any accidents.
- ❖ **Engineering Intern**, Brooklyn Union Gas Company, Summer 1998 and 1999, Brooklyn, NY
Edited maps using Integrating Mapping System (IMS) and AUTOCAD.

Professional Memberships

- ❖ American Ceramic Society
- ❖ Materials Advantage
- ❖ American Society for Testing and Materials
- ❖ American Concrete Institute
- ❖ Tau Beta Pi
- ❖ American Society of Civil Engineers
- ❖ National Society of Black Engineers

Professional Service

- ❖ Organizing Secretary, 3rd North American Conference on the Design and Use of Self-Consolidating Concrete
- ❖ Reviewer for technical journals: ACI Materials Journal, Materials and Structures

Professional Licenses

- ❖ Engineer-In-Training (EIT), certified in the District of Columbia, EIT 900753

Publications (* indicates presenter at conference)

Journals

- ❖ Kwon, S., **Ferron, R.**, Akkaya, Y., and S.P. Shah. "Cracking of fiber-reinforced self-consolidating concrete due to restrained shrinkage," Korea Concrete Institute Journal of Construction Material and Structures (submitted)
- ❖ Gregori, A., **Ferron, R.**, Sun, Z. and S.P. Shah. "Experimental simulation of SCC formwork pressure," American Concrete Institute Materials Journal, v105(1), January/February 2008
- ❖ **Ferron, R.**, Gregori, A., Sun, Z. and S.P. Shah. "Rheological method to evaluate the thixotropy of cement pastes for SCC," American Concrete Institute Materials Journal, v104(3), May/June 2007
- ❖ Sun, Z. Gregori, A., **Ferron, R.**, and S.P. Shah. "Developing a falling-ball viscometer for high flowable cement-based materials," American Concrete Institute Materials Journal, v104(2), March/April, 2007

Refereed conference proceedings

- ❖ Kwon, S., **Ferron, R.**, Akkaya, Y., and S.P. Shah. "Restrained shrinkage cracking of fiber-reinforced self-consolidating concrete," 6th RILEM International Conference on Pavement Cracking, June 9-11, 2008, Chicago, IL (submitted)
- ❖ **Ferron, R.**,* Gregori, A., Sun, Z. and S.P. Shah. "The effect of superplasticizers on the thixotropy of cementitious material," *Proceedings of the Eighth CANMET/ACI International Conference on Superplasticizers and Other Chemical Admixtures in Concrete*, Supplementary Vol., pp 175-185, October 29-November 1, 2006, Sorrento, Italy
- ❖ **Douglas, R.**, Gregori, A., Sun, Z. and S.P. Shah. "Investigations of the properties of SCC: a method for measuring thixotropy and viscosity," *Proceedings of Knud Hojgaard Conference on Advanced Cement-Based Materials: Research and Teaching*, pp 19 -30, June 12-15, 2005, Lyngby, Denmark
- ❖ **Douglas, R.P.***, Bui, V.K., Akkaya, Y., and S.P. Shah. "Properties of self-consolidating concrete containing class f fly ash: with a verification of the minimum paste volume method," *ACI Committees 236 and 237 Special Publications: Workability of SCC: Roles of Its Constituents and Measurement Techniques*, CD-ROM, April 17 - 21, 2005, New York, NY

Non-refereed conference proceedings

- ❖ Shah, S.P., **Ferron, R.**, Ferrara, L., Tregger, N., and S. Kwon. "Research on SCC: some emerging themes," *Proceedings of the 5th International RILEM Symposium on Self-Compacting Concrete*, v1, pp 3 – 14, September 2007, Ghent, Belgium
- ❖ Kwon, S., **Ferron, R.**, and S.P. Shah. "Development of a prediction model for SCC formwork pressure," *Proceedings of the National Ready Mix Concrete Association 2007 Concrete Technology Forum*, CD-ROM, May 2007, Dallas, TX
- ❖ **Douglas, R.***, Gregori, A., Sun, Z., Shah, S.P. and D. Bonen. "The effect of ingredients and shear history on the thixotropic rate of rebuilding of SCC," *Proceedings of the 2nd North American Conference on the Design and Use of Self-Consolidating Concrete and the 4th International RILEM Symposium on Self-Compacting Concrete*, v1, pp 591 - 596, October 30 - November 3, Chicago, IL
- ❖ Gregori, A., Sun, Z., **Douglas, R.**, Shah, S.P. and D. Bonen. "The evaluation of viscosity by using a novel viscometer for SCC," *Proceedings of the 2nd North American Conference on the Design and*

Use of Self-Consolidating Concrete and the 4th International RILEM Symposium on Self-Compacting Concrete, October 30 - November 3, 2005, v1, pp 775 – 782, Chicago, IL

- ❖ **Douglas, R.P.***, Bui, V.K., Akkaya, Y., and S.P. Shah. “Properties of self-consolidating concrete containing class f fly ash,” *Proceedings of the RILEM International Symposium on Advances in Concrete through Science and Engineering*, Evanston, IL, CD-ROM, March 21-24 2004,.

Other Publications (Magazine Articles, Technical Reports)

- ❖ Contributing Author of White Paper on Self-Consolidating Concrete prepared for Portland Cement Association, February 2007, (<http://www.cement.org/bookstore/profile.asp?itemid=RP440>)
- ❖ Featured in article “Coming to terms with SCC: a short list of high-tech terms lets you talk the talk with the experts,” *The Concrete Producer*, October 2005

Select Oral and Poster Presentations (9 out of 18)

- ❖ **Ferron, R.**, Fuente, E., Negro, C., and S.P. Shah. “Aggregation mechanisms in pastes for self-consolidating concrete,” The 4th International Conference of the African Materials Research Society, December 13, 2007, Dar es Salaam, Tanzania
- ❖ **Ferron, R.** and S.P. Shah. “Flocculation mechanisms of SCC cement pastes,” American Concrete Institute Fall Convention, October 15, 2007, Fajardo, Puerto Rico
- ❖ **Ferron, R.** and S.P. Shah. “Structural rebuilding of cementitious materials using rheological and particle size measurements,” 109th Annual Meeting of the American Ceramics Society, September 17, 2007, Detroit, MI
- ❖ **Ferron, R.** “Thixotropy, workability, and formwork pressure of SCC,” Universidad Complutense de Madrid, July 26, 2007, Madrid, Spain
- ❖ **Ferron, R.** “Concrete: it’s more than cement, water, and gravel,” BGSA 11th Graduate and Professional Student Research Conference, April 28, 2007, Evanston, IL
- ❖ **Ferron R.** and S.P. Shah, “Structural stiffening and flocculation in cementitious materials,” ACBM Semi-Annual Meeting, Fall 2006, Evanston, IL
- ❖ **Douglas, R.**, Gregori, A., Sun, Z. and S.P. Shah. “The effect of mixture ingredients on the thixotropic rate of rebuilding of self-consolidating concrete (SCC),” ACI Student Fellowship Showcase, April 2005, New York, NY
- ❖ **Douglas, R.P.**, Bui, V.K., and S.P. Shah. “Application of minimum paste volume method in designing self-consolidating concrete with class f fly ash,” 7th Annual Joint IMGIP, ICEOP, and King-Chavez-Parks Future Conference, November 13–16 2003, Lansing, MI,
- ❖ **Douglas, R.P.**, Bui, V.K., and S.P. Shah. “Development of self-consolidating concrete containing type f fly ash,” Portland Cement Association Fall Meeting, April 29 2003, Chicago, IL,

Invited Presentations

- ❖ “Rheology, Flocculation, and Formwork Pressure of Self Consolidating Concrete,” Civil and Environmental Engineering Departmental Seminar, Princeton University, November 29, 2007